

Young Chemist's Panel - Review Meeting 2012

26thth November 2012

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- The term "nucleophilic catalysis" is arguably most associated with pyridines.
- More specifically, 4-dimethylaminopyridine (DMAP) is the catalyst most associated with "nucleophilic catalysis."
- Pyridine demonstrated to promote acetylation of alcohols and phenols by Einhorn and Hollandt (Munich, 1898).
- In 1967, Litvinenko and Kirichenko (USSR) systematically study the benzoylation of anilines catalysed by structurally diverse pyridines. The presence of DMAP leads to a rate enhancement of x150,000 against the analogous use of Et₃N, and x6000 compared to pyridine.
- In 1969, Steglich and Höfle (Munich) independently demonstrate the catalytic activity of DMAP and soon after, 4-pyrrolidinopyridine.



- The chemistry that grew from the original acylation synthetic methodology work concerned pyridine acting as a *nucleophilic* additive.
- However, the nucleophilic catalyst functions by forming an acylating species, which has enhanced *electrophilicity*.
- In more general terms, DMAP acts as a "Lewis basic catalyst", reacting with electrophiles. (Denmark discusses in some detail: *Angew. Chem. Int. Ed.* 2008, 47, 1560.)

• More generally, Jansen has discussed 9 possible Lewis acid/base combinations based on the nature of the electron pair donator and acceptor:

Donor		Acceptor	
	n*	σ*	π^*
n	n→n*	n→σ*	n→π*
σ	σ→n*	$Q \rightarrow Q *$	$\sigma \rightarrow \pi *$
π	$\pi \rightarrow n^*$	$\pi \rightarrow \sigma^*$	$\pi \rightarrow \pi *$



Strategies for Stereoselective Synthesis via Acyl Transfer

• Contextualise common stereocontrol strategies used in asymmetric nucleophilic catalysis by alcohol acylation.

$$C = \frac{ee}{ee + ee'} 100 \qquad S = \frac{k_{fast}}{k_{slow}} = \frac{In[(1 - C)(1 - ee)]}{In[(1 - C)(1 + ee)]}$$



Strategies for Stereoselective Synthesis via Acyl Transfer

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- Consider the DMAP-catalysed reaction of an acid anhydride and an alcohol.
- Equilibrium formation of a highly electrophilic *N*-acylpyridinium salt.
- Donation of electron density from substituent is crucial (no catalysis with pyridine C-2 substituents).
- Reaction of nucleophile at *N*-acyl pyridinium, carboxylate acts as base.



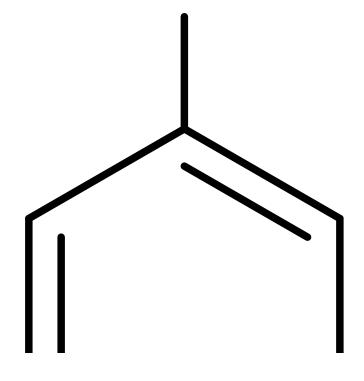
Chiral Aminopyridines – Structure/Activity



- The powerful catalysts DMAP and PPY are highly symmetrical with two planes of symmetry.
- Reactive site is pyridine nitrogen (N-1)
- No catalytic activity if substituted at C-2
- How does one effectively desymmetrize a distant lone pair?



Chiral Aminopyridines – Vedejs Reagent





Chiral Aminopyridines – Fuji Catalyst



Chiral Aminopyridines – Yamada Catalyst



Chiral Aminopyridines – Yamada Catalyst



Chiral Aminopyridines – Connon Catalyst



Chiral Aminopyridines – Jeong's Kemp Triacid









Chiral Aminopyridines – Fu Planar Chiral Catalyst Resolution Examples



Chiral Aminopyridines – Fu Planar Chiral Catalyst Resolution Examples



Chiral Aminopyridines – Fu Planar Chiral Catalyst Resolution Examples



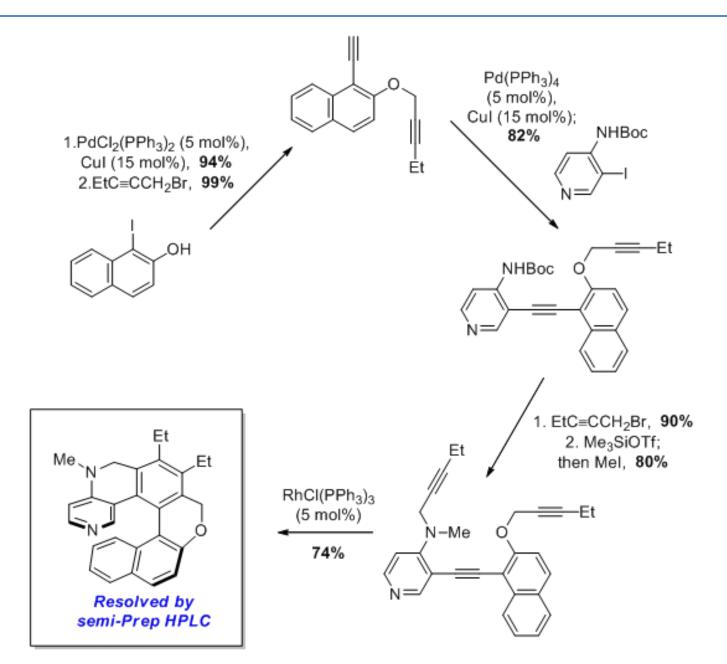
Chiral Aminopyridines – Fu Planar Chiral Catalyst Fu DKR



Chiral Aminopyridines – Carbery Helical Catalyst



Chiral Aminopyridines – Carbery Helical Catalyst





Chiral Aminopyridines – Carbery Helical Catalyst



Catalyst is Scalable, Selective, Resolution-Free











Chiral Imidazoles – Histidinyl Systems



Chiral Amidines – Birman Catalysts

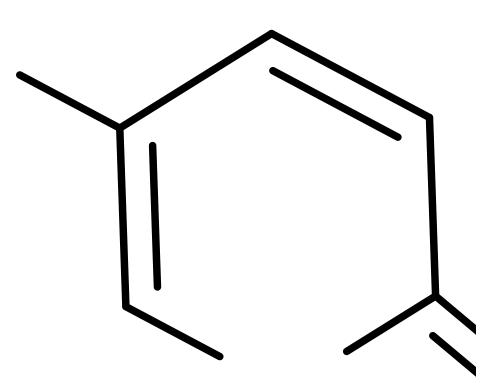


Chiral Amidines – Birman Catalysts



Chiral Amidines – Birman Catalysts







Chiral Amidines – Deng/Fossey Amidine



Chiral Vicinal Diamines – Oriyama Catalysts





Chiral Phosphines – Vedejs Catalysts



Chiral N-heterocyclic Carbenes



Alcohols

Thorough body of work, but, scope for 3° and saturated substrates (alkyl/alkyl 2°).

Amines

Remains a huge challenge. Competitive nucleophilicity. Other elements (P?)

Catalysts

Commercial availability. Multi-functionality/co-operativity. Modelling and understanding.

Reactions

C-C bond formations. Almost all are derived from early DMAP chemistry...creativity.

Electrophiles

E-X..... E = sulfonyl, phosphoryl, silyl etc

