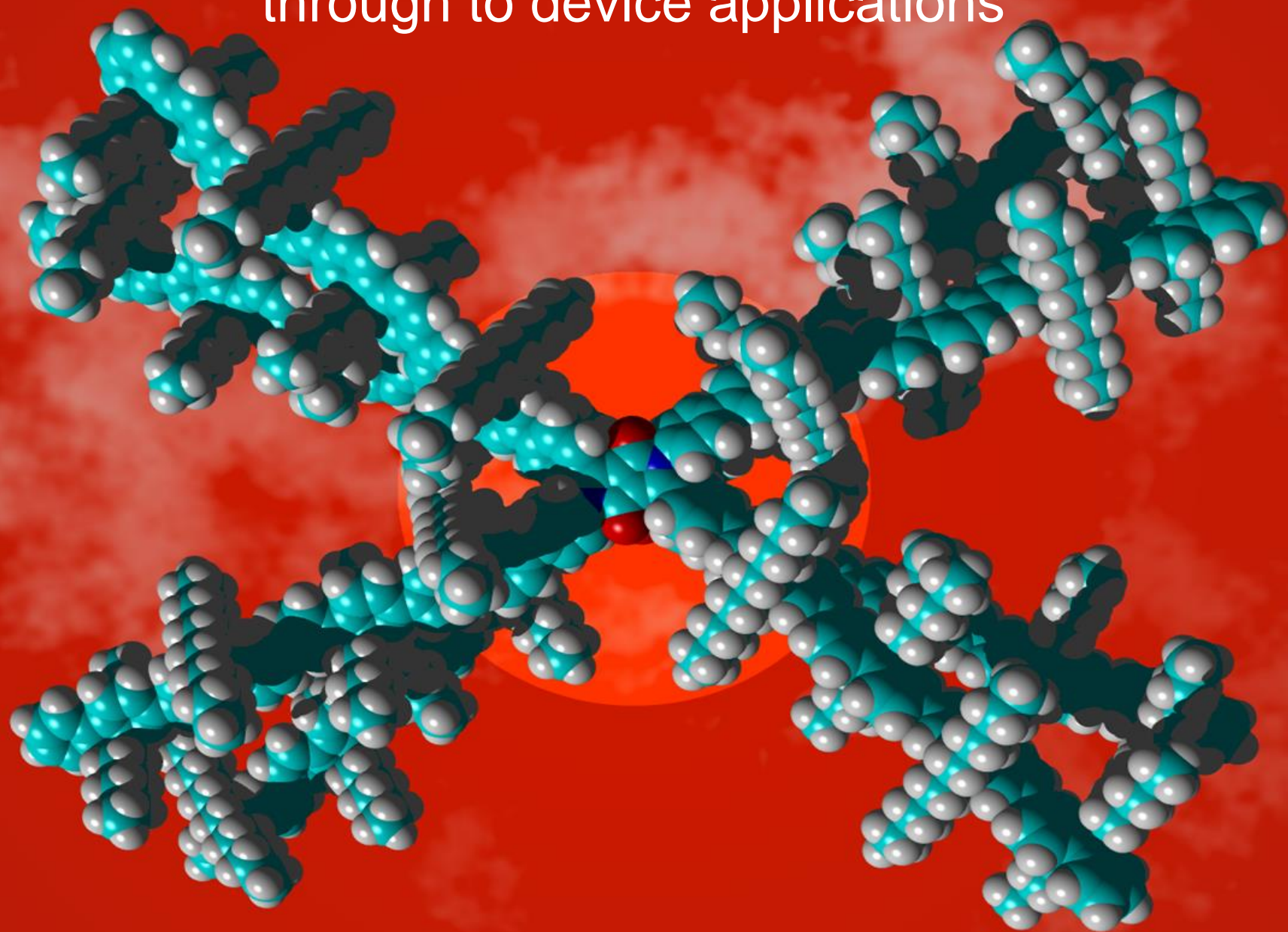


Complex conjugated architectures – from synthesis through to device applications



Acknowledgements

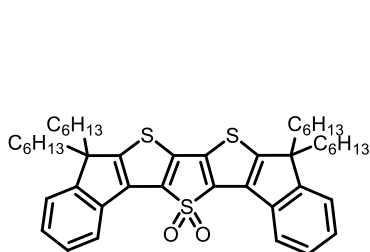
Alexander Kanibolotsky
Clara Orofino
Rory Berridge
Dave Crouch
Filipe Vilela
Hao Pang
John Forgie
Iain Wright
Greg McEntee
Diego Cortizo Lacalle
Saad Elmasly
Zuzka Vobecka
Neil Findlay

Simon Coles
Mike Hursthouse
Ross Harrington
Bill Clegg

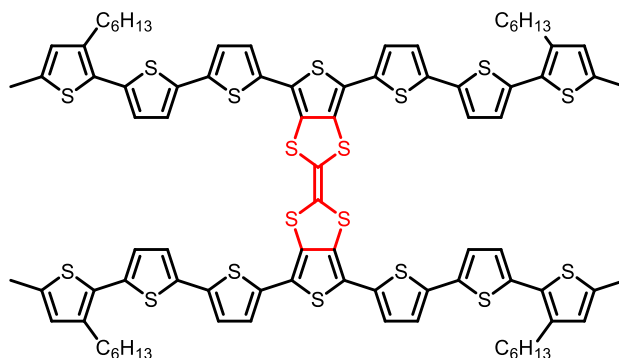
Martin Dawson
Ifor Samuel
Graham Turnbull
Donal Bradley
Colin Belton
Paul Stavrinou
Igor Perepichka



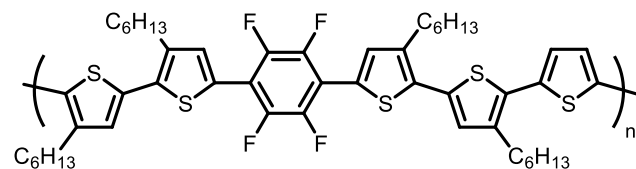
- Synthesis of small molecules, macromolecules and polymers: as organic semiconductors
- Electrochemical characterisation
- Device fabrication (solar cells, OLEDs, sensors, OFETs, batteries/capacitors, hybrid MEMS devices)
- Fundamental interest in non-covalent interactions and how they affect conformation



small molecules



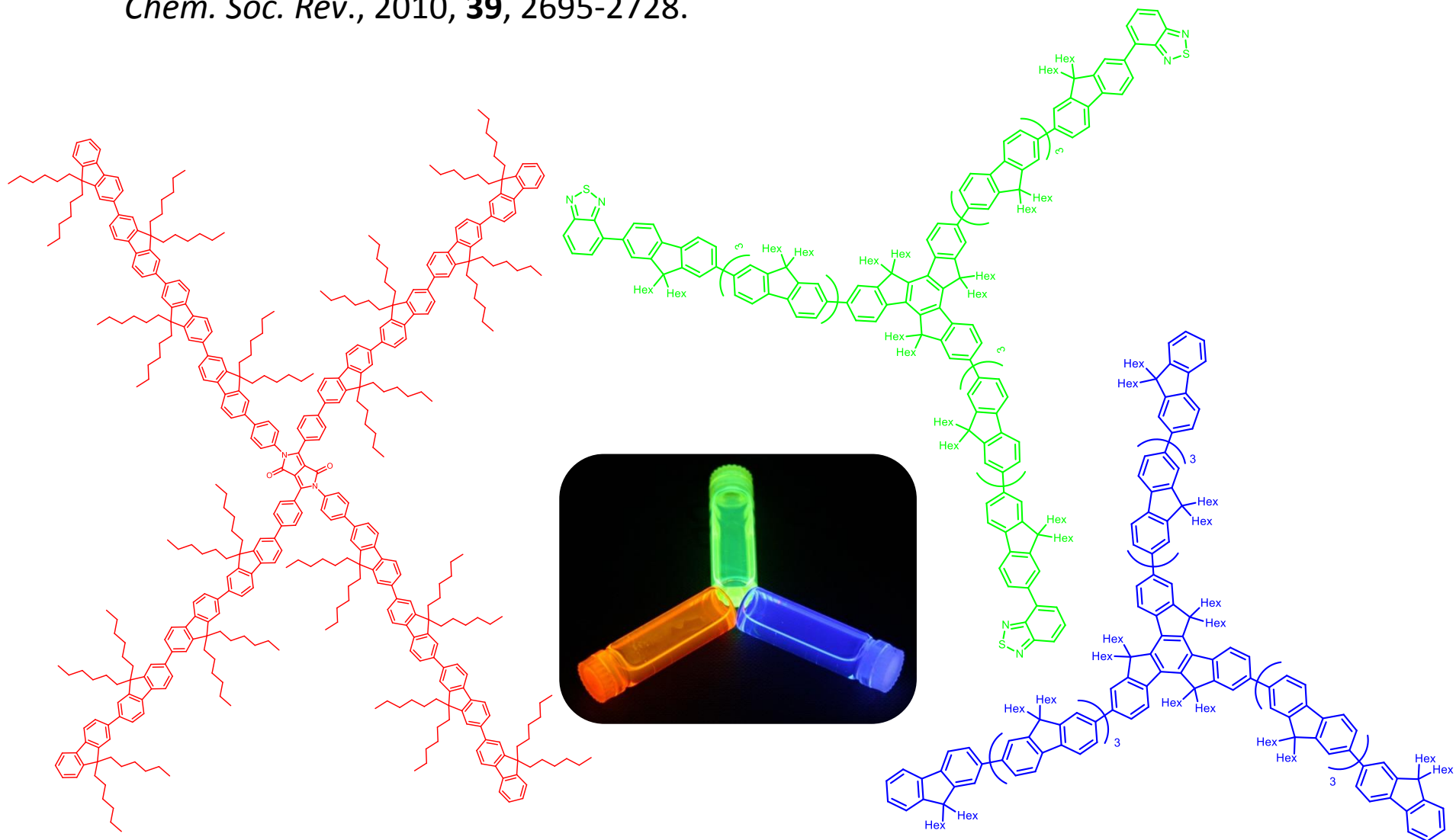
macromolecules



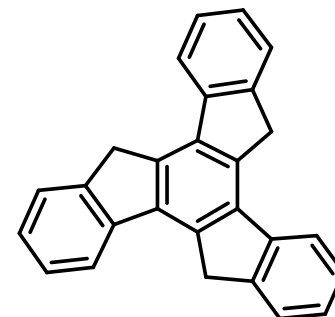
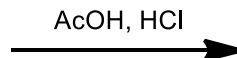
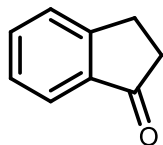
polymers

Star-shaped conjugated macromolecules

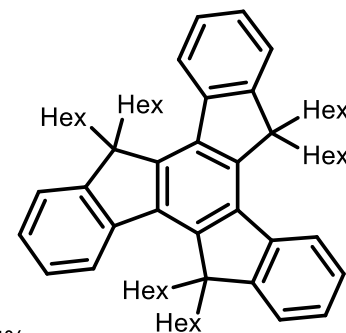
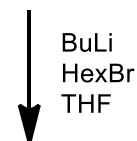
See: *Star-shaped π -Conjugated Oligomers and Their Applications in Organic Electronics and Photonics*, A. L. Kanibolotsky, I. F. Perepichka and P. J. Skabara, *Chem. Soc. Rev.*, 2010, **39**, 2695-2728.



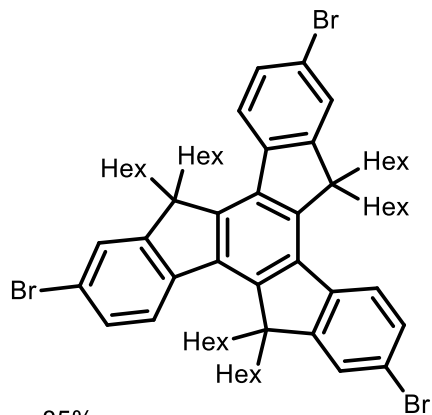
Star-shaped Oligofluorenes



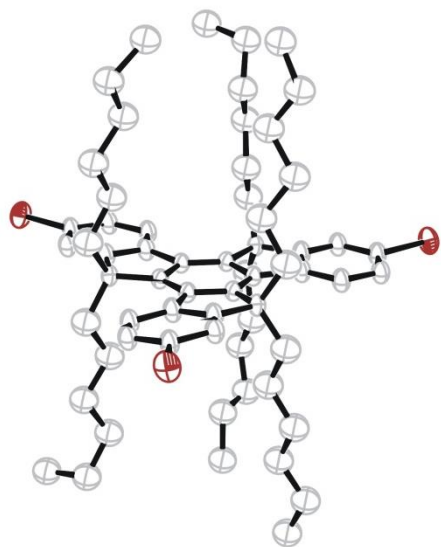
90-95%

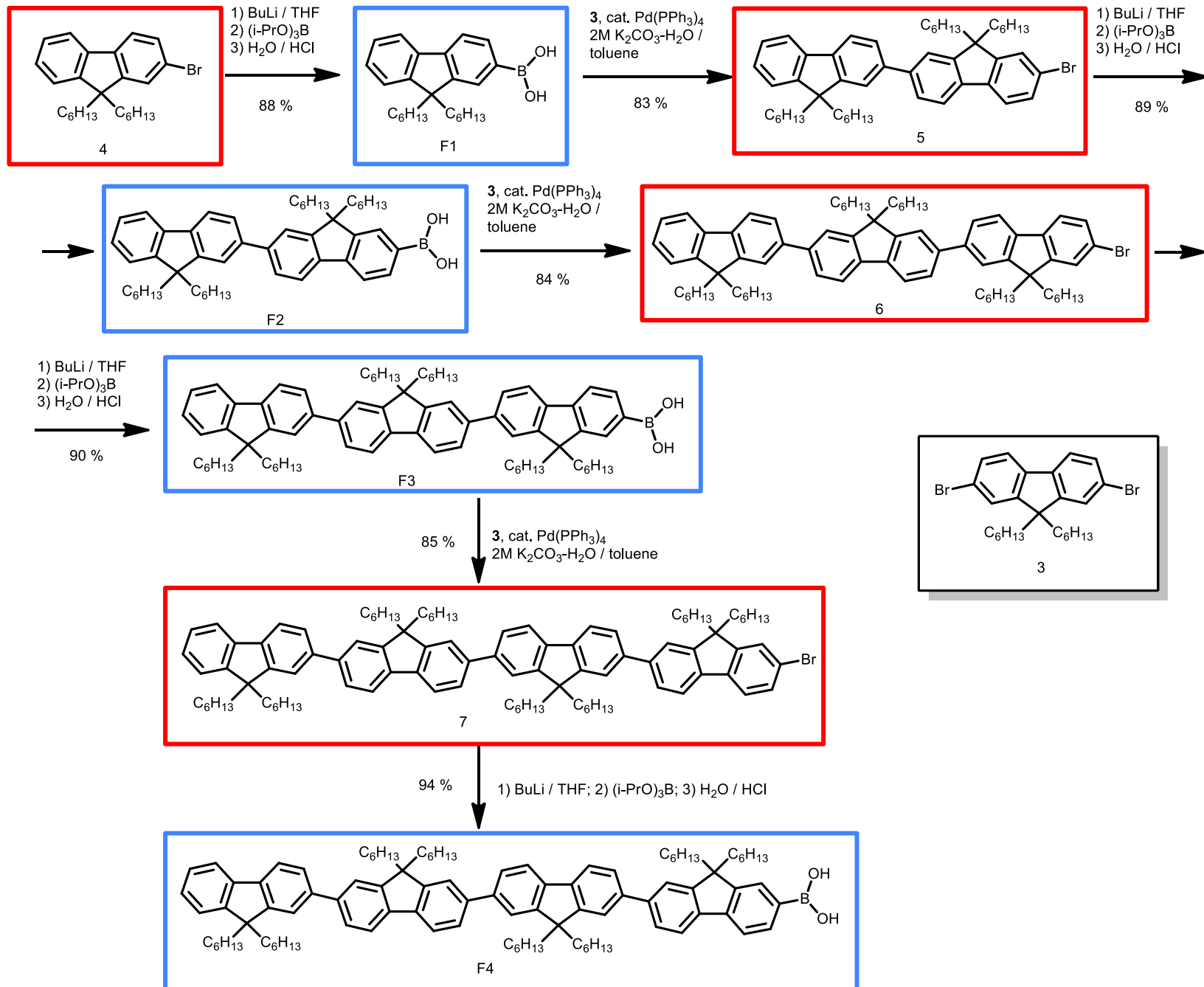


85%

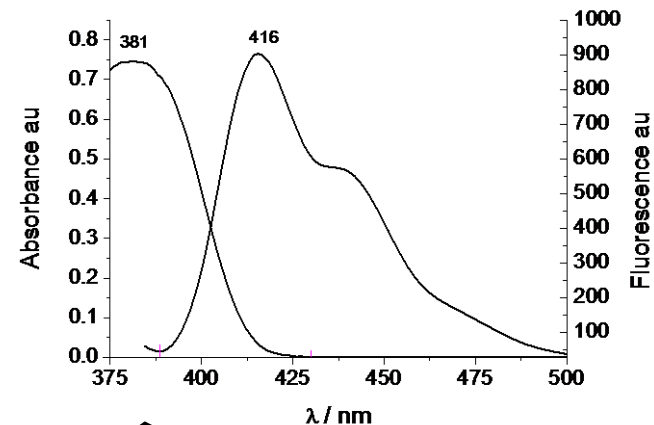


85%



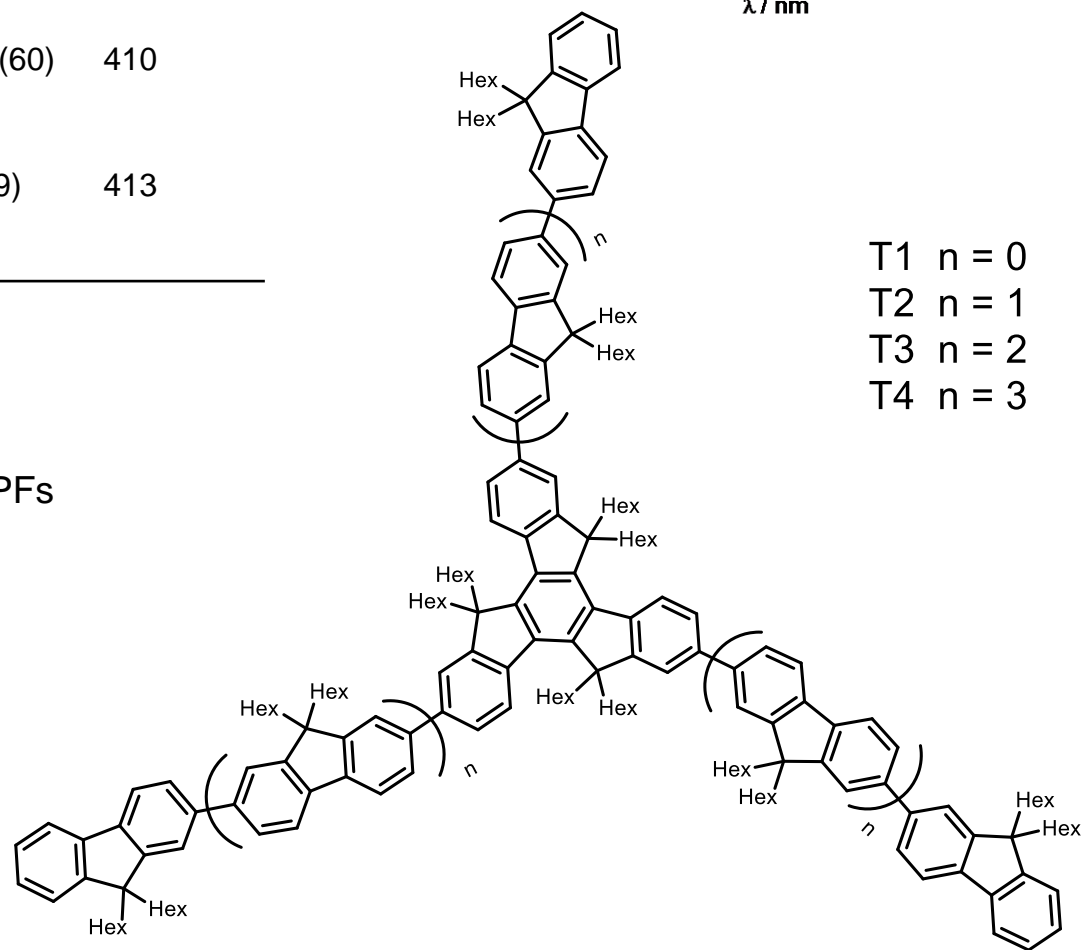


cmpd	M_w , g mol ⁻¹	λ_{abs} , nm film	λ_{PL} , nm (Φ_{PL} , %) film	TGA, °C [5% loss]
T1	1844.95	343	380sh, 398, 419.5 (43)	401
T2	2842.52	359	404, 425.5, 449 (51)	408
T3	3840.08	369	417sh, 436, 462sh (60)	410
T4	4837.65	372	422, 442, 467sh (59)	413

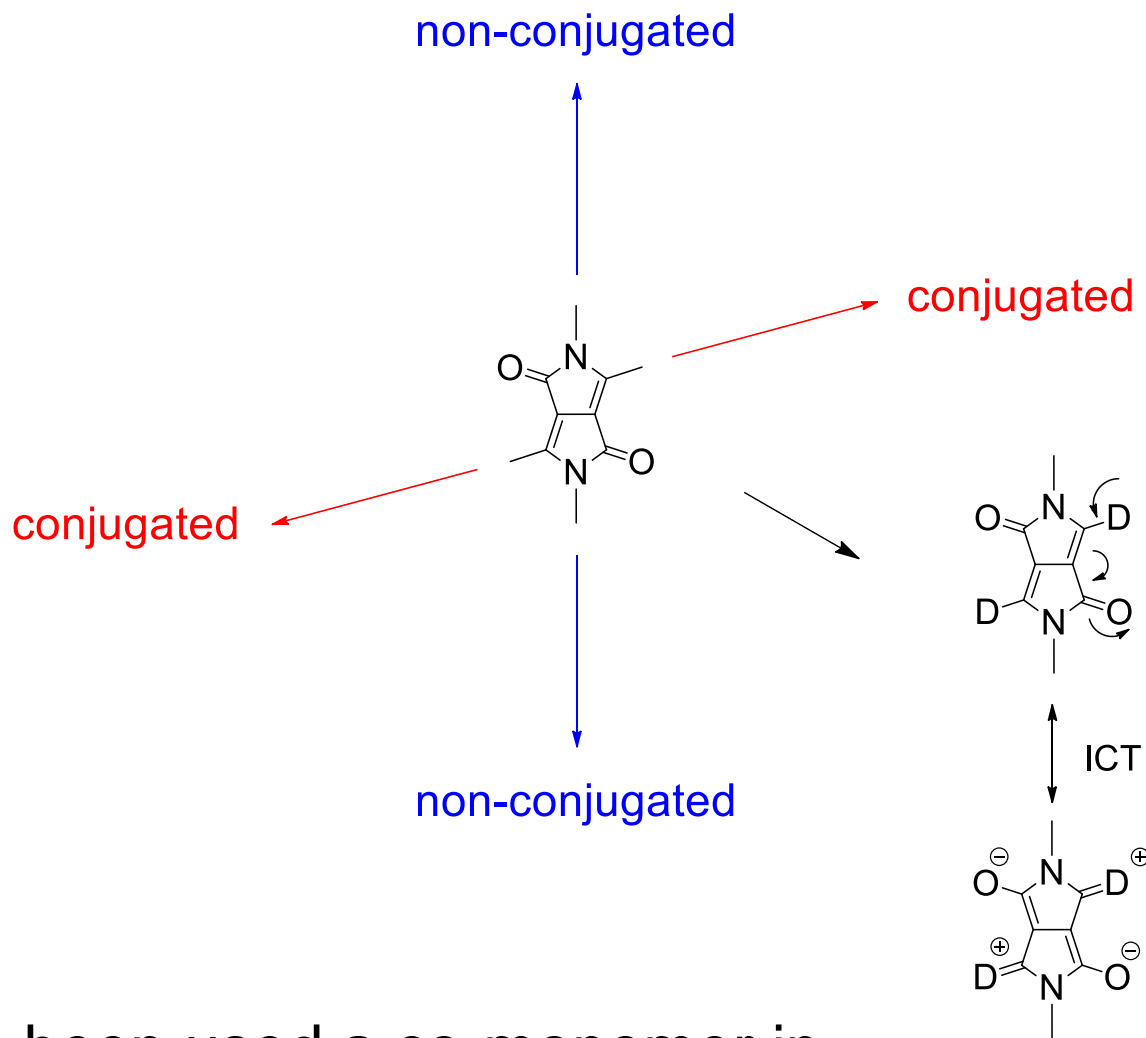


- Sharp PL characteristics
- Very accurate HOMO/LUMO levels
- PL efficiencies comparable to analogous PFs
- High degree of purity
- Good thermal stability
- Excellent solubility
- Improved stability over PFs
- **Synthetic reproducibility**

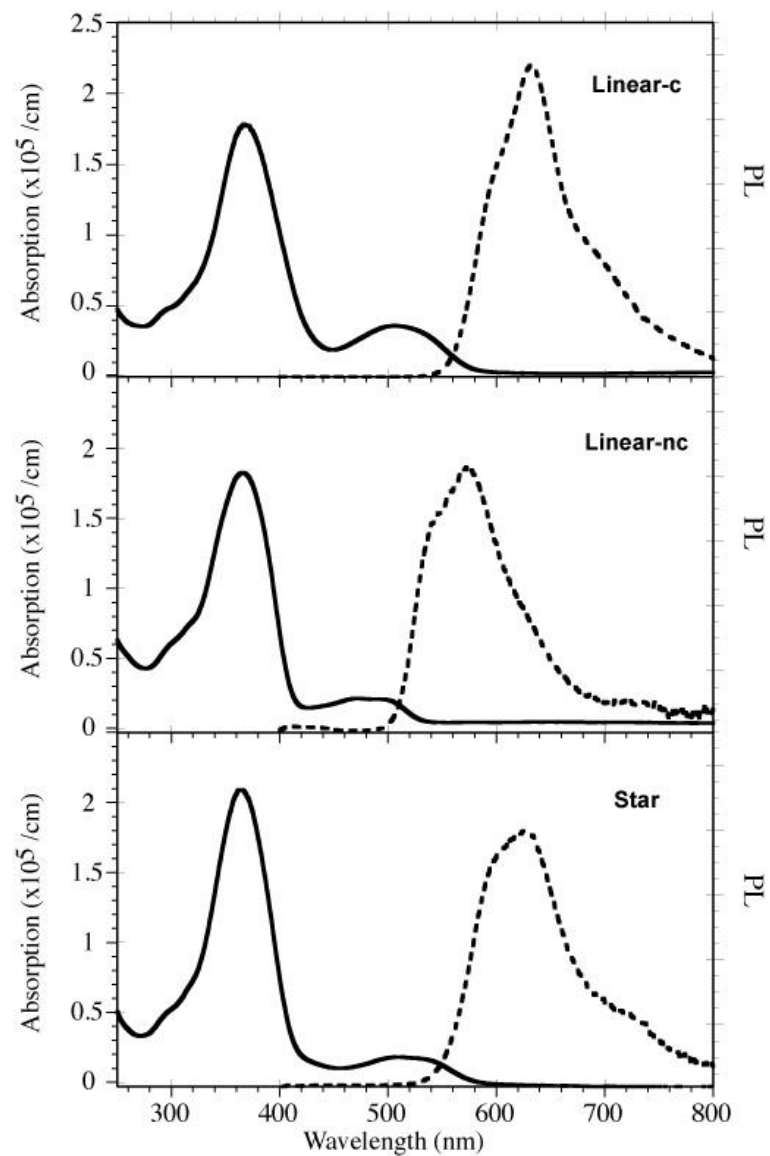
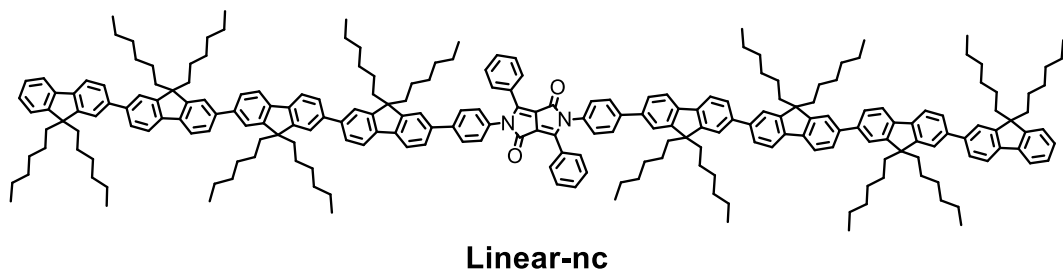
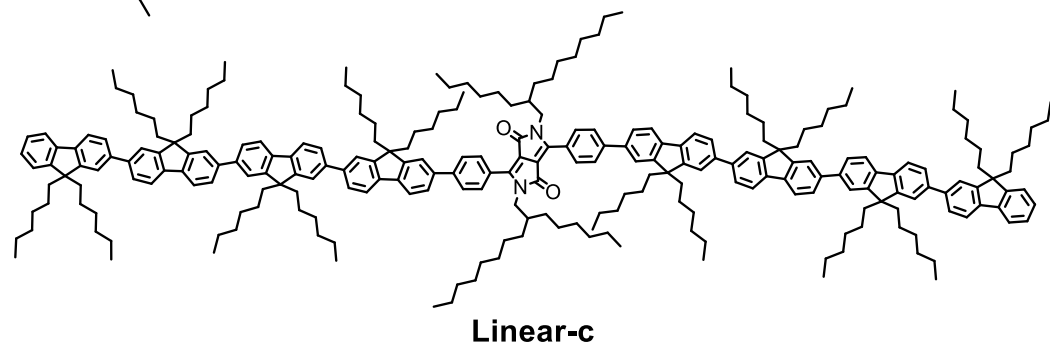
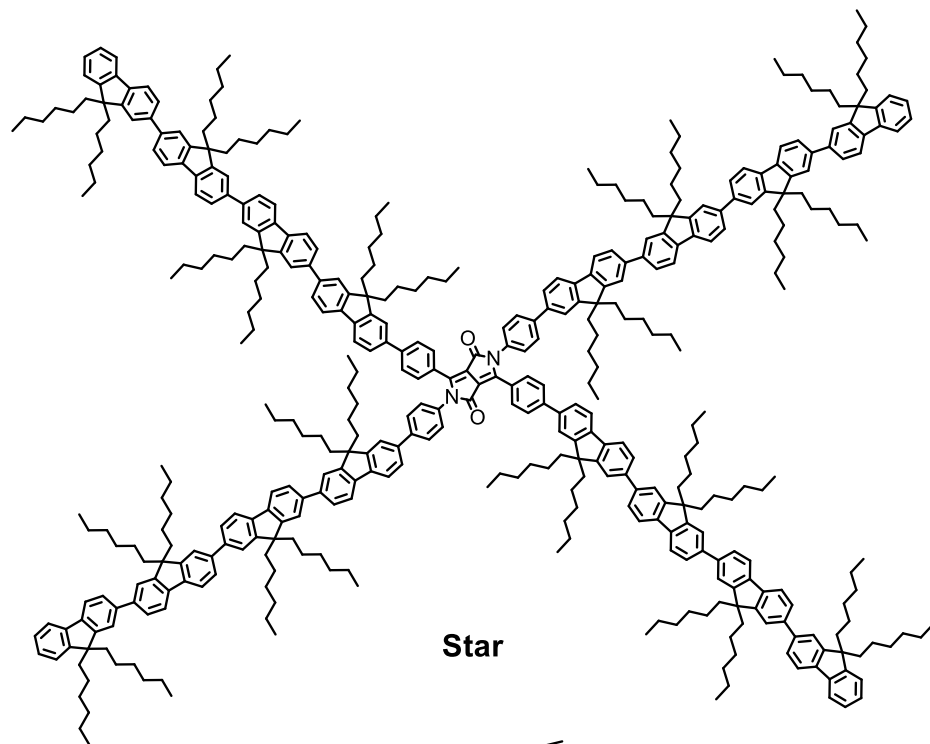
J. Am. Chem. Soc., 2004, **126**, 13695



Alternative cores: 1,4-diketo-pyrrolo[3,4-c]pyrrole (DPP)

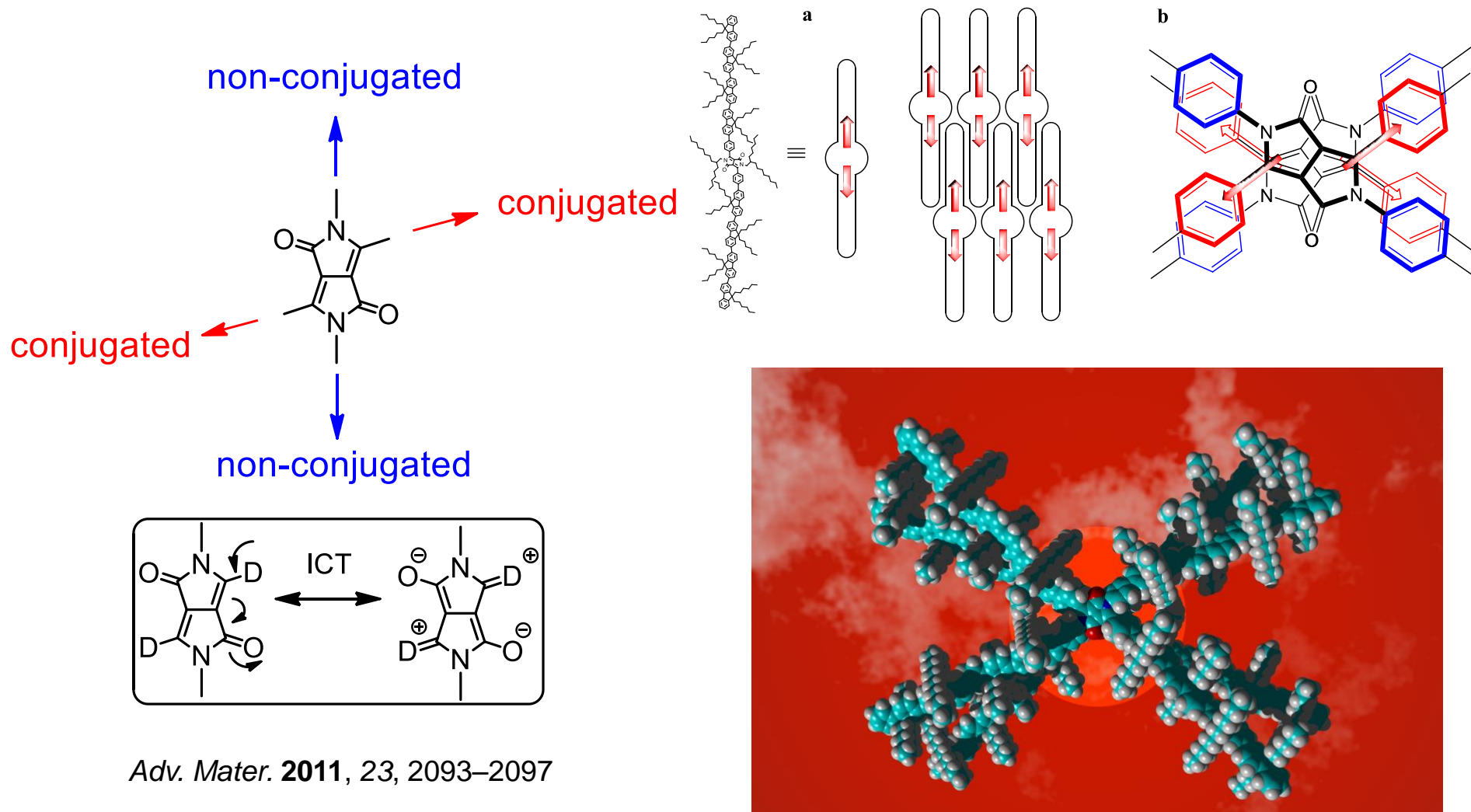


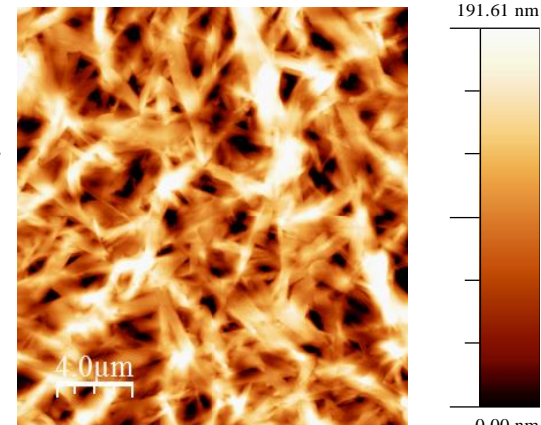
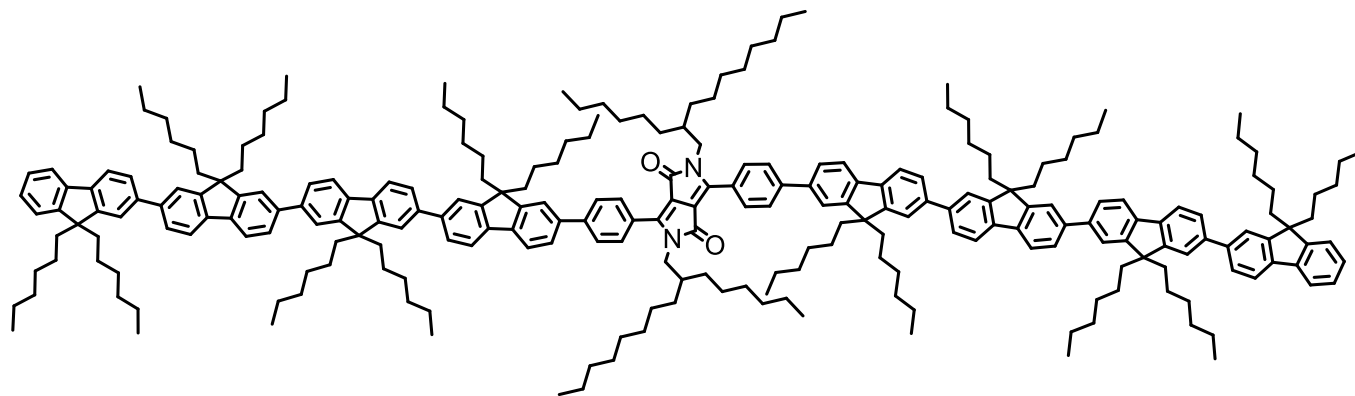
DPP has been used as a co-monomer in materials for OPV, OFET and EL (red emitter)



Charge transfer and aggregation

PLQY for neat films ranges 2-20% due to aggregation through dipole interactions.

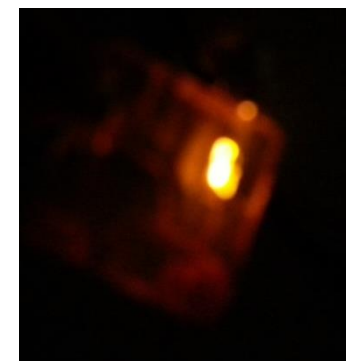
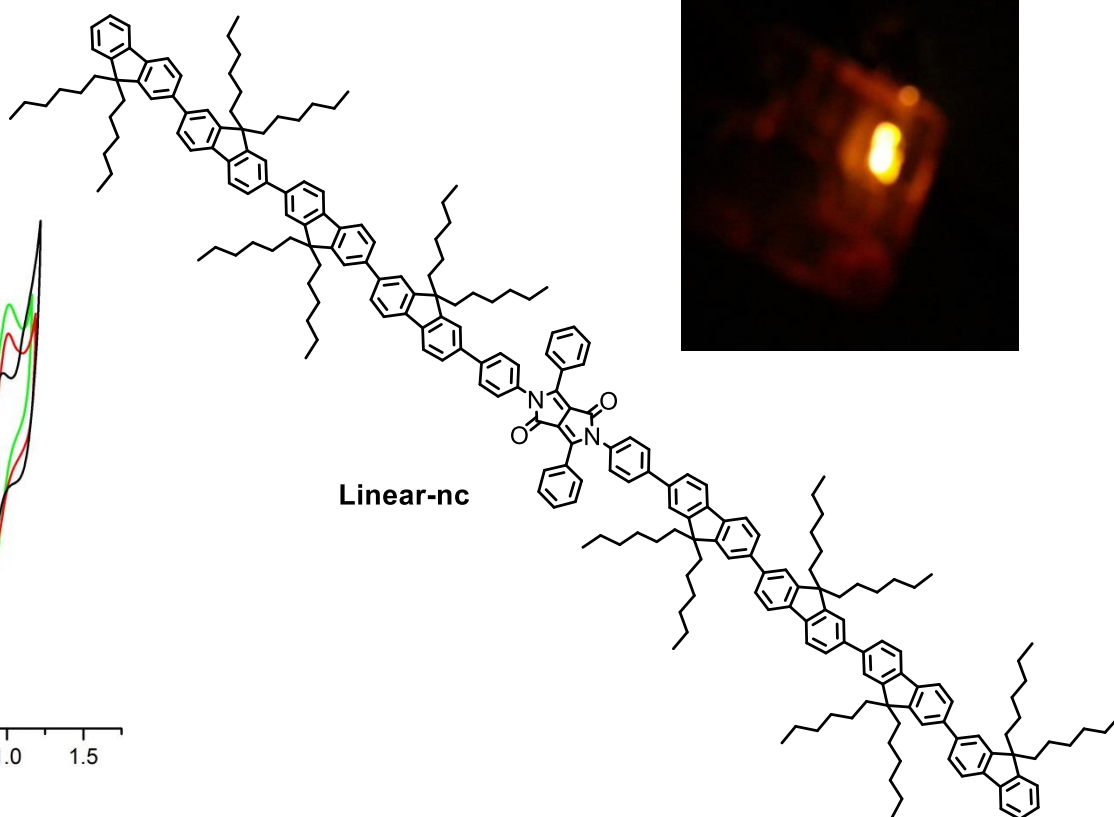
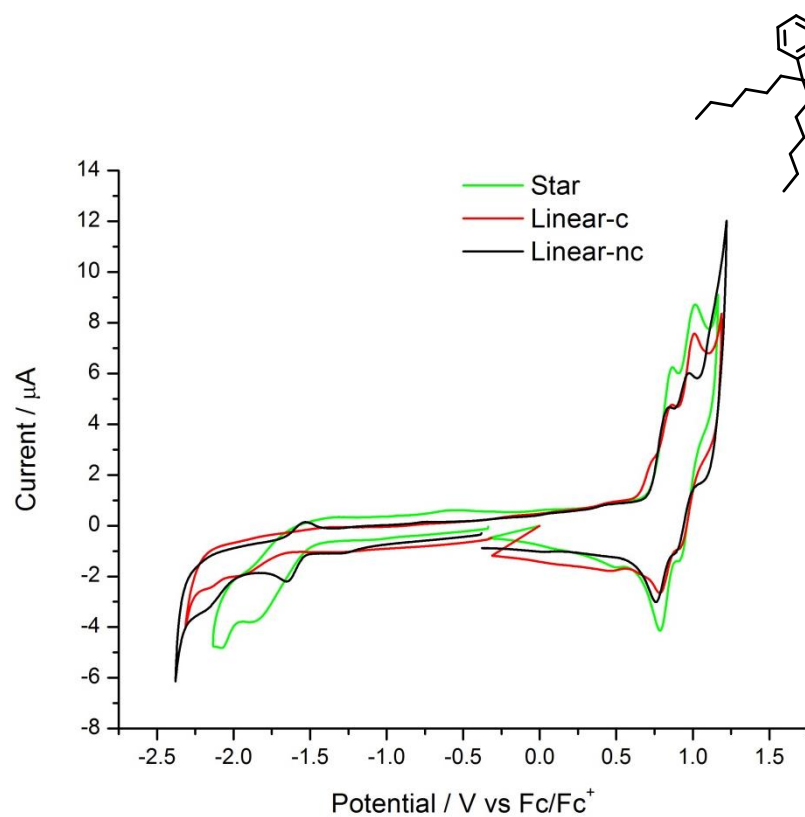




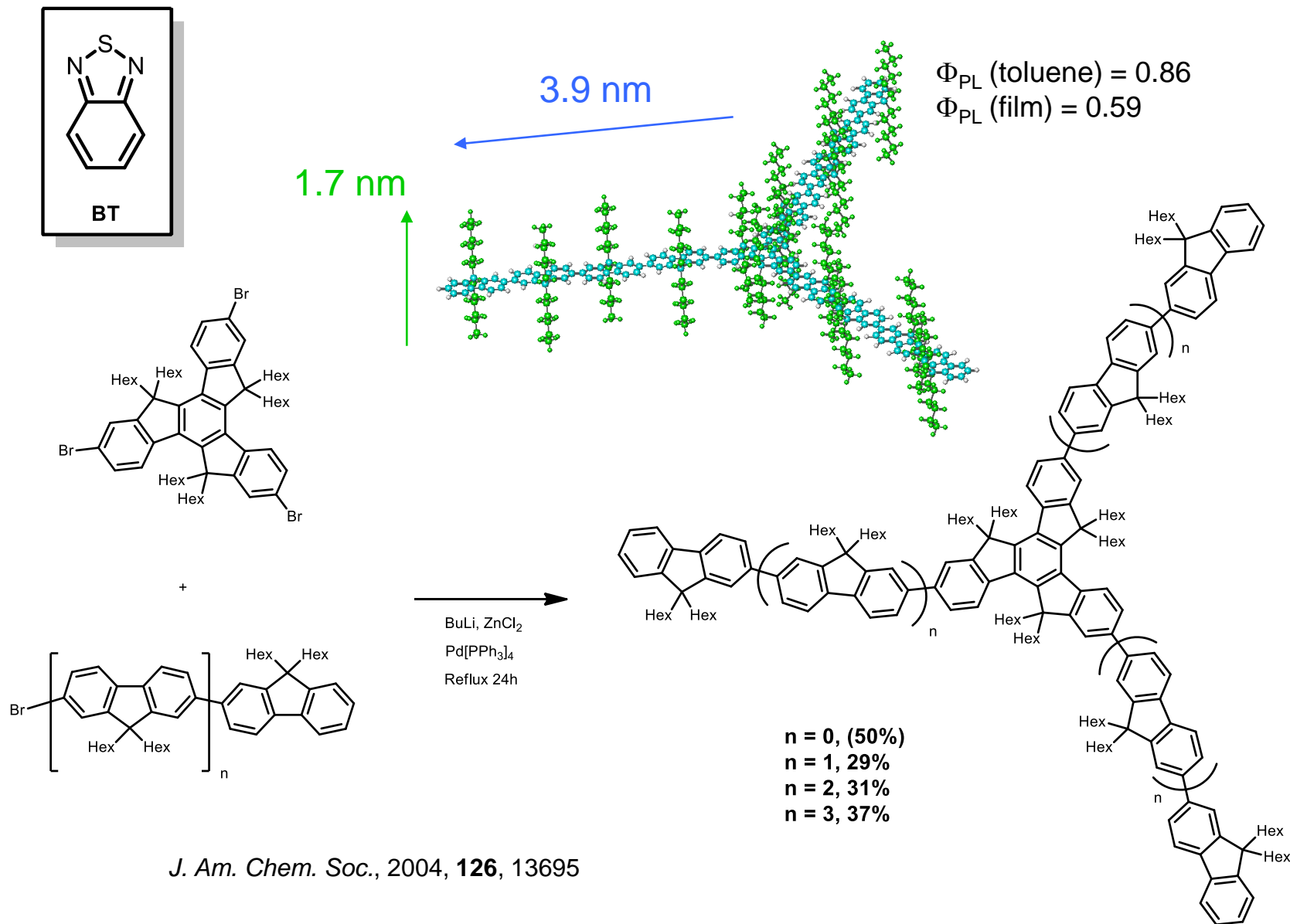
Isr. J. Chem., 2014, in press

30mg of **Linear-c** in 1ml of chloroform, PFBT 20 seconds, annealed at 140 °C for 30 min

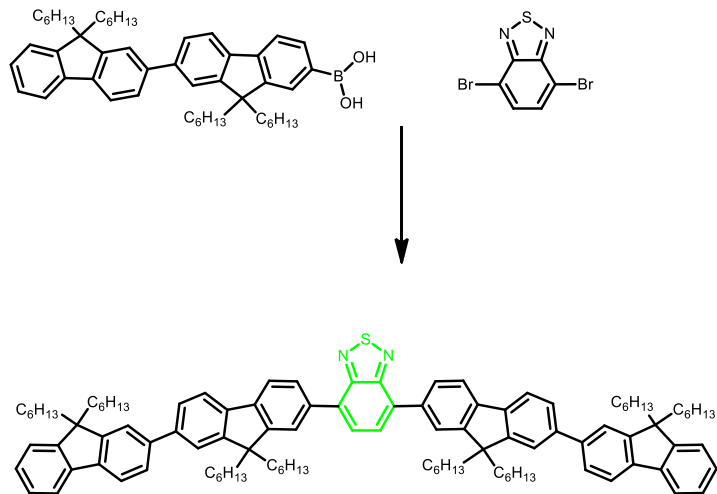
Hole mobility $10^{-4} \text{ cm}^2/\text{Vs}$



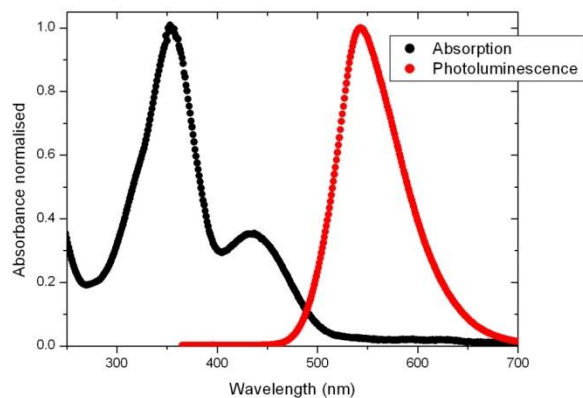
From blue to green: incorporating BT into oligofluorene-truxenes



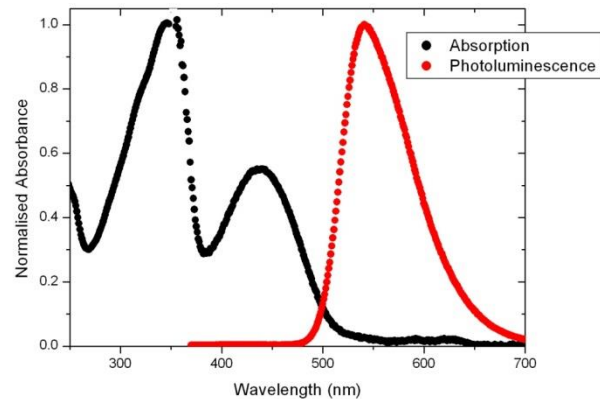
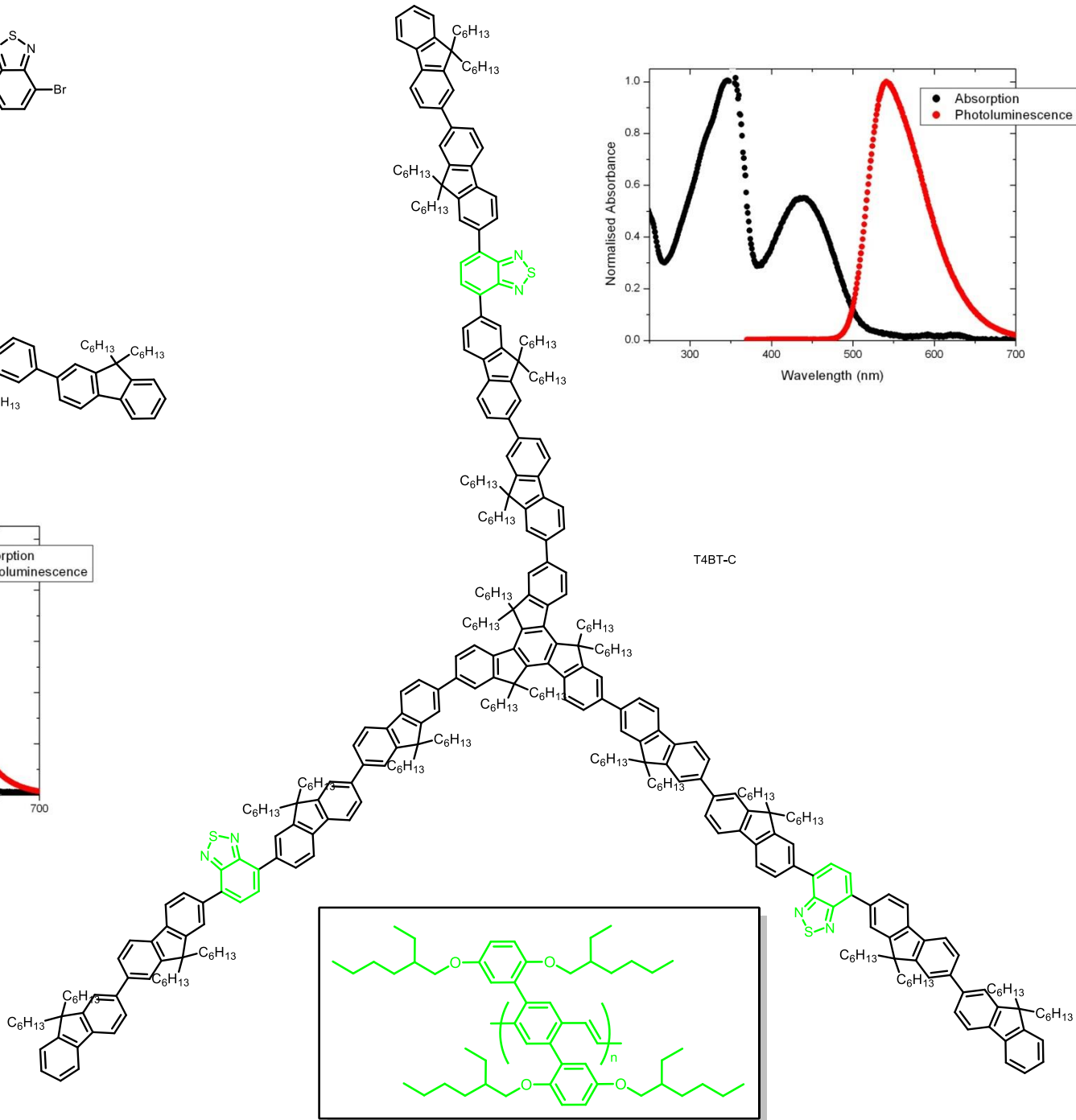
J. Am. Chem. Soc., 2004, **126**, 13695



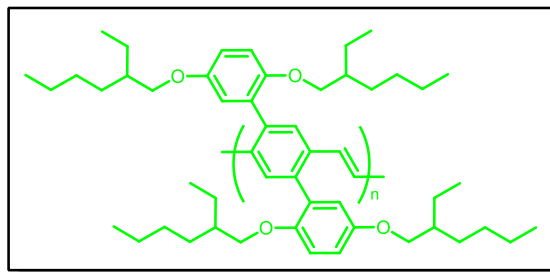
97.8%

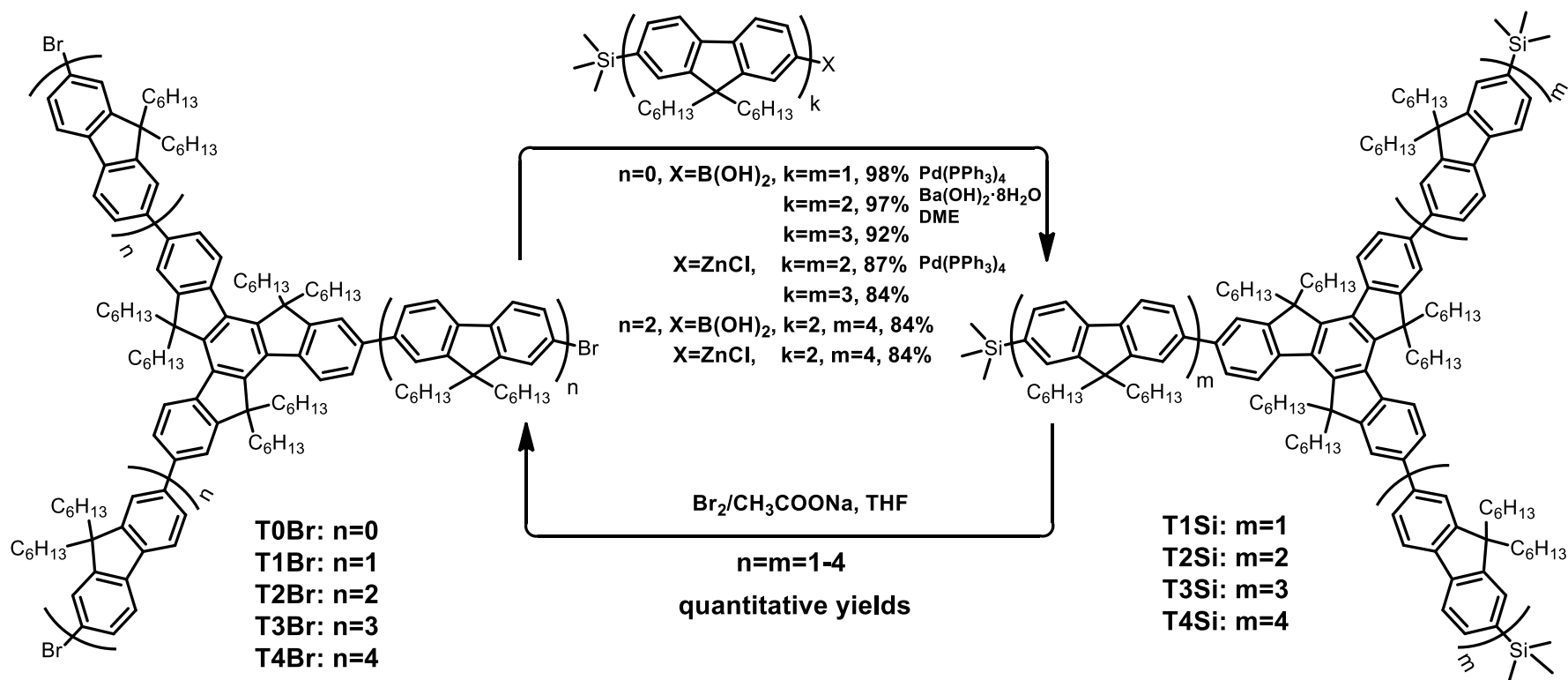
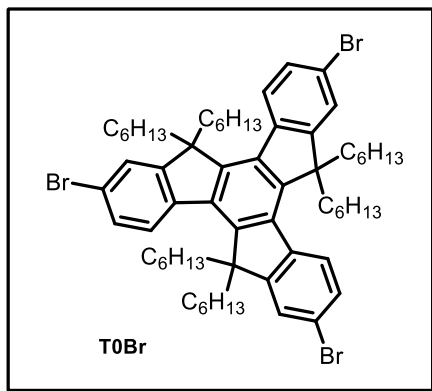


Solid state PLQY for both compounds is ca. 88%
Fluorescence lifetime is ca. 4 ns for the star

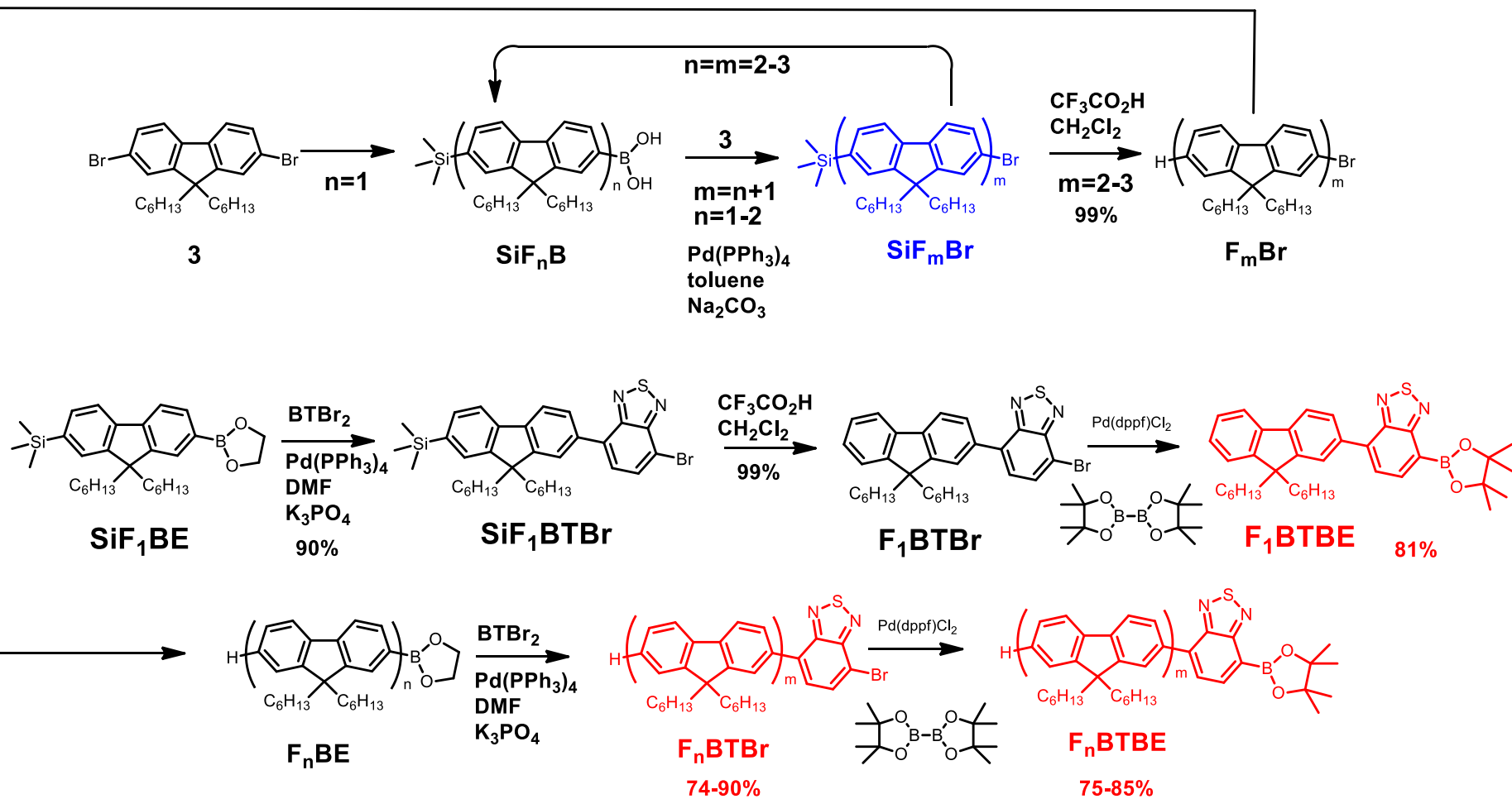


T4BT-C



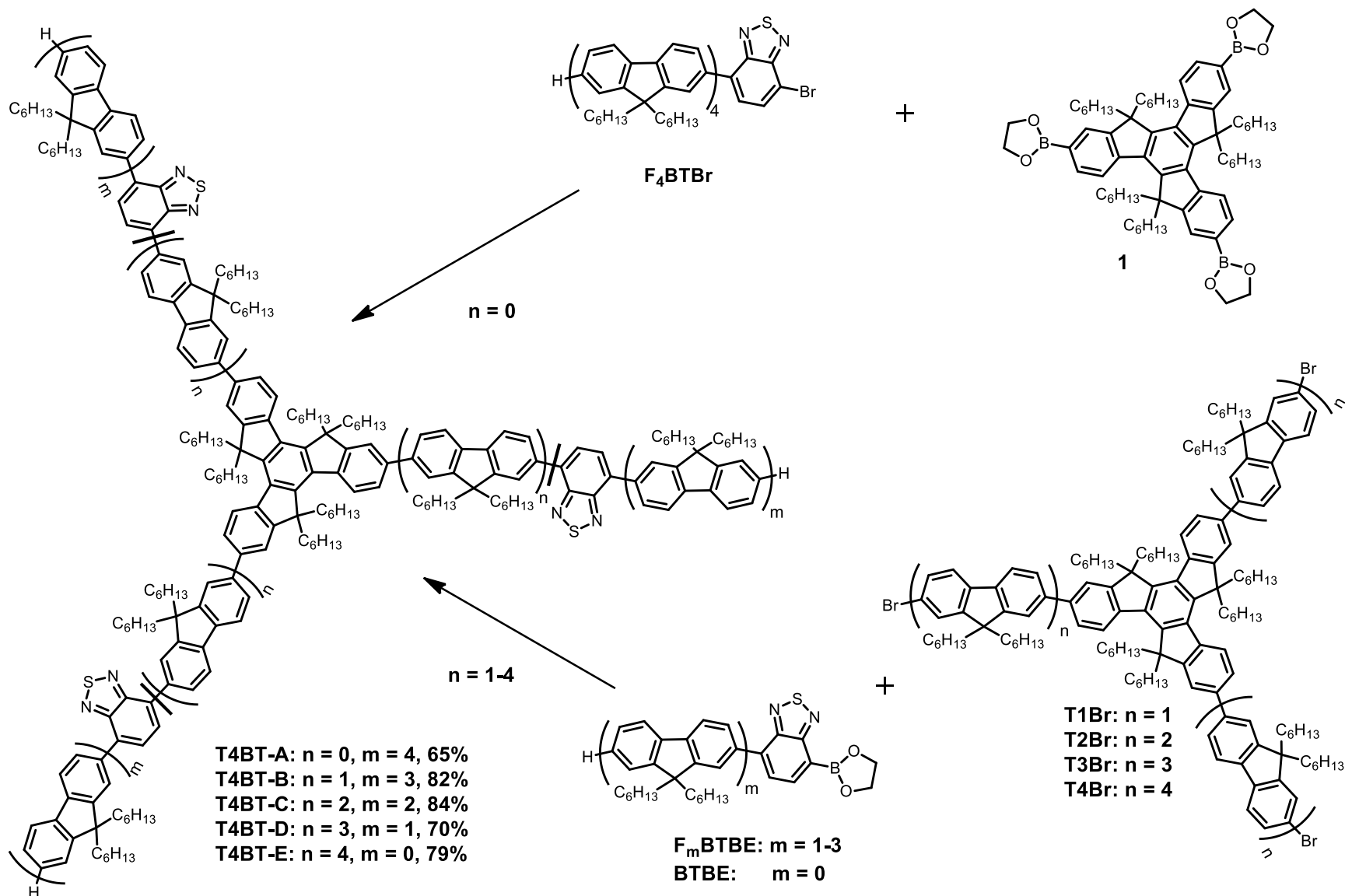


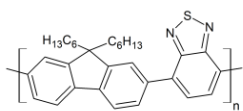
Synthesis of T1Si-T4Si by modified Suzuki coupling and bromination in mild conditions



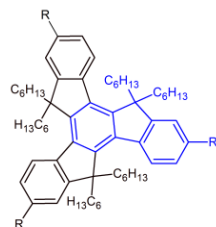
Synthesis of the oligofluorene-BT arm precursors F1BTBE - F3BTBE

New greens – BT series

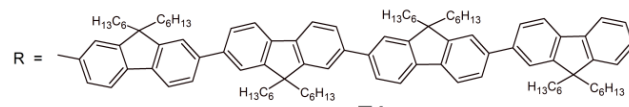




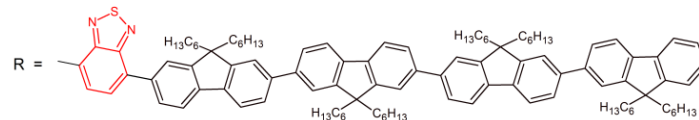
F8BT



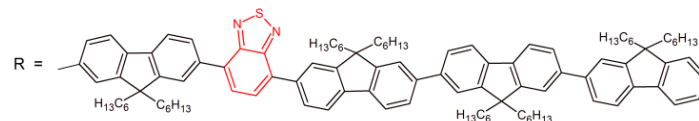
**Truxene
Core**



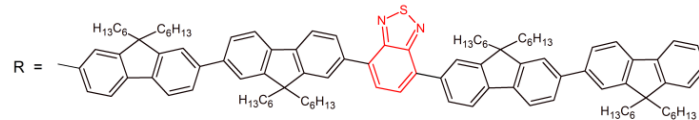
T4



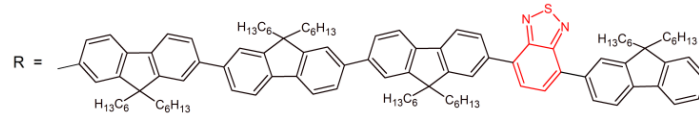
T4BT-A



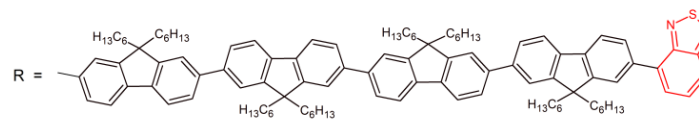
T4BT-B



T4BT-C

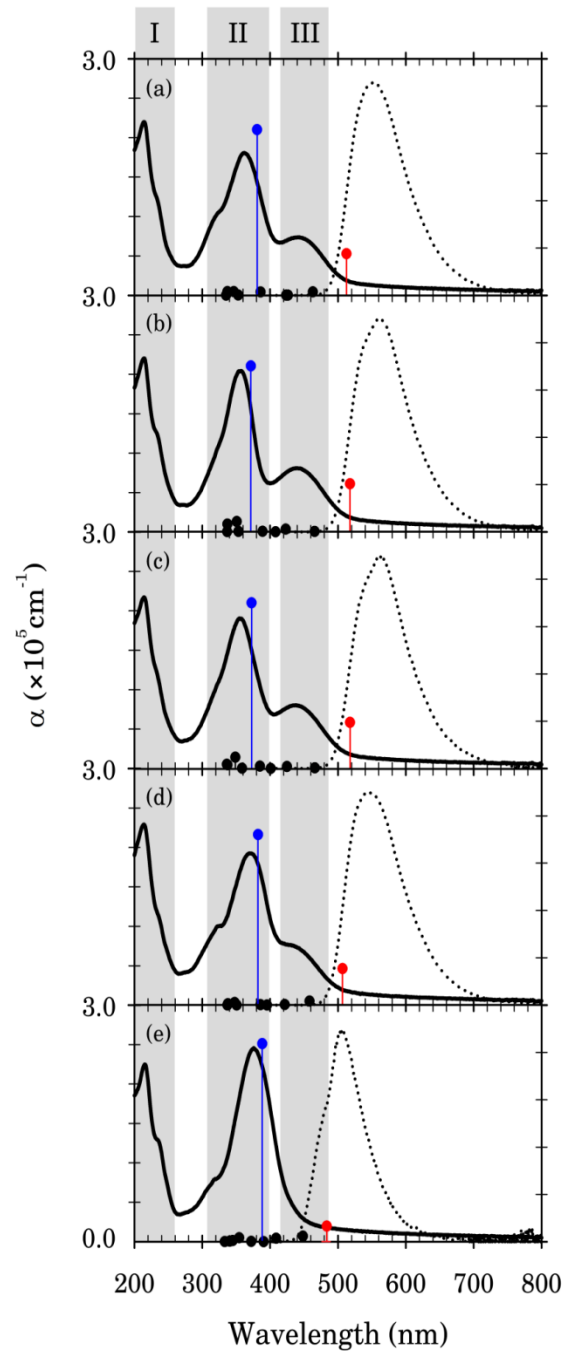
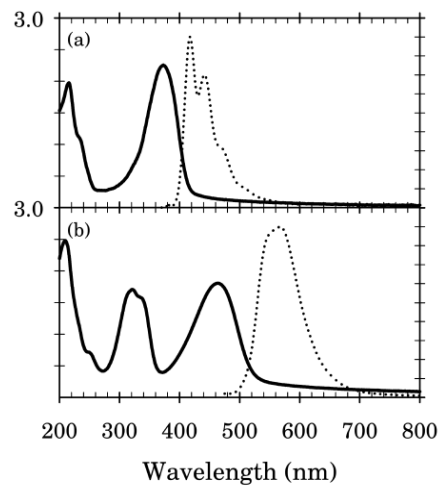


T4BT-D



T4BT-E

Normalised Photoluminescence

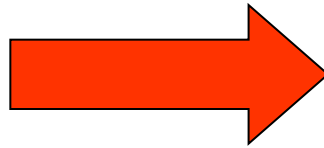


Normalised Photoluminescence

Shrinking Polymer Laser systems



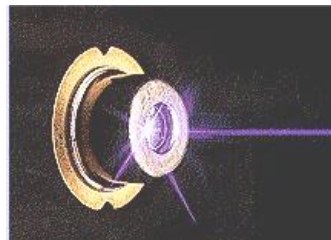
1995 Regenerative amplifier
(Tessler)



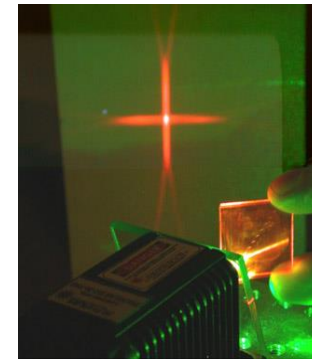
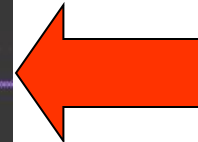
~2000 Q-switched
Nd:YAG



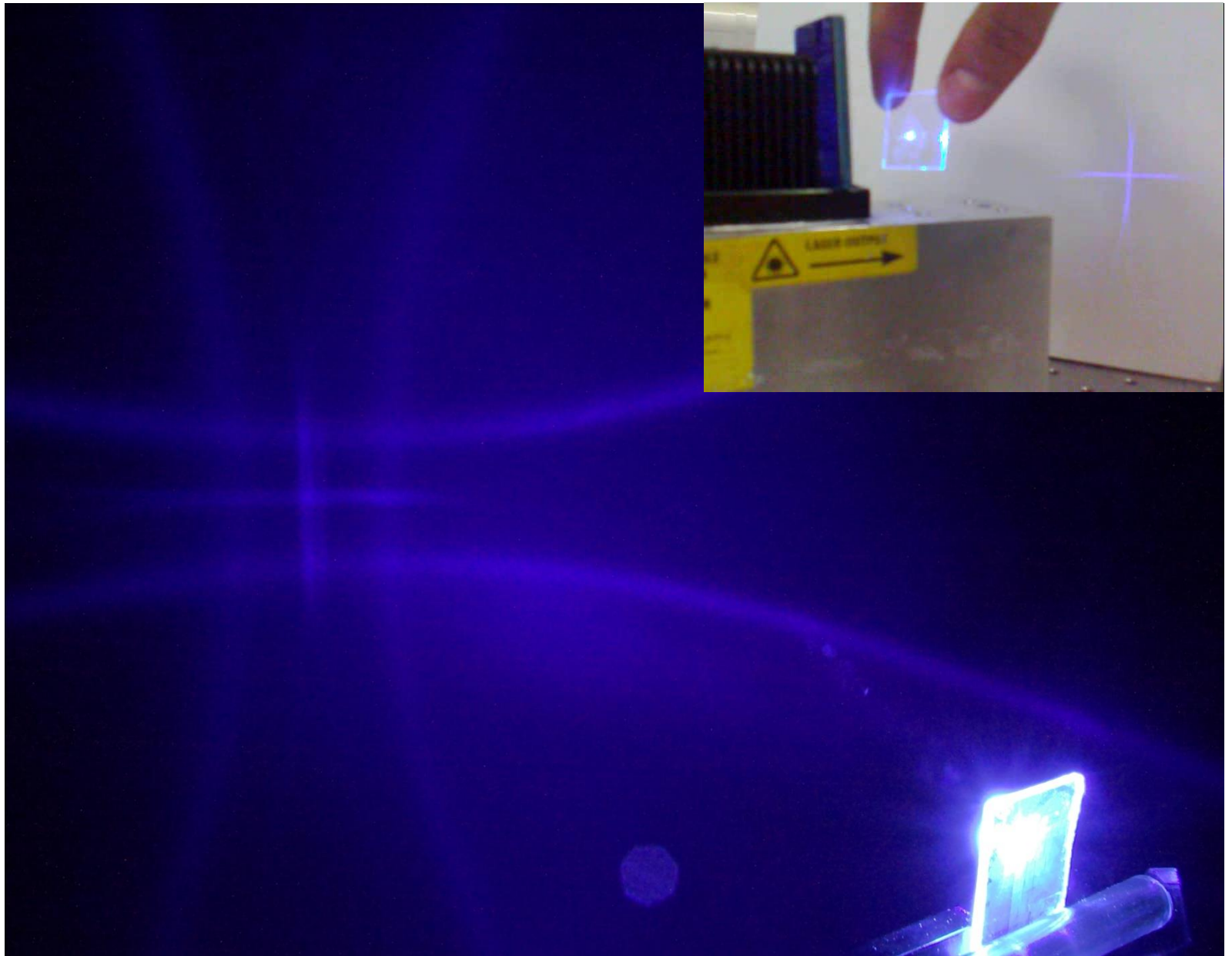
2008 LED
pumped



2006 Diode
pumped



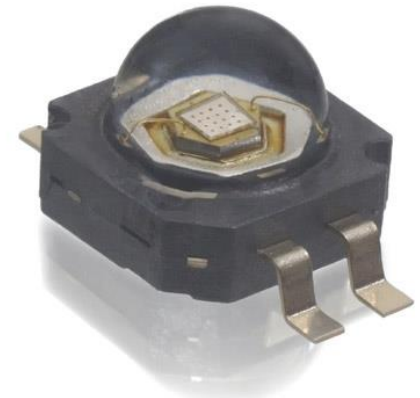
2003 Microchip laser



LED pumped polymer lasers

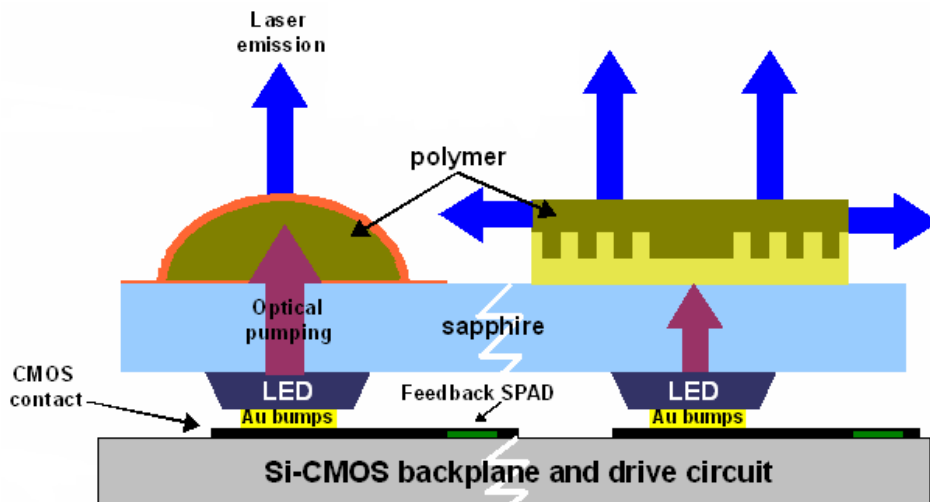
Advantages

- ▶ Exploits high mobility of III-V semiconductors
- ▶ Separates charge injection from gain medium
- ▶ Much lower cost than laser pumped system
- ▶ Very compact, electrically controlled



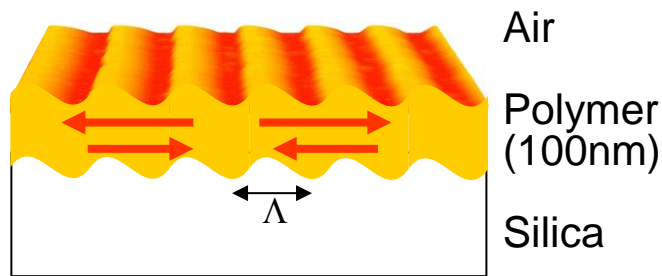
www.lumileds.com

All the benefits from direct electrical pumping

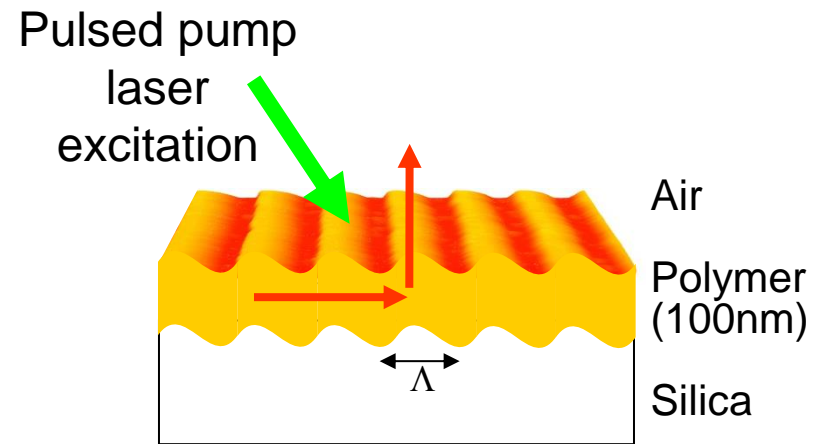


Surface-emitting distributed feedback lasers

$$m\lambda = 2n_{\text{eff}}\Lambda$$

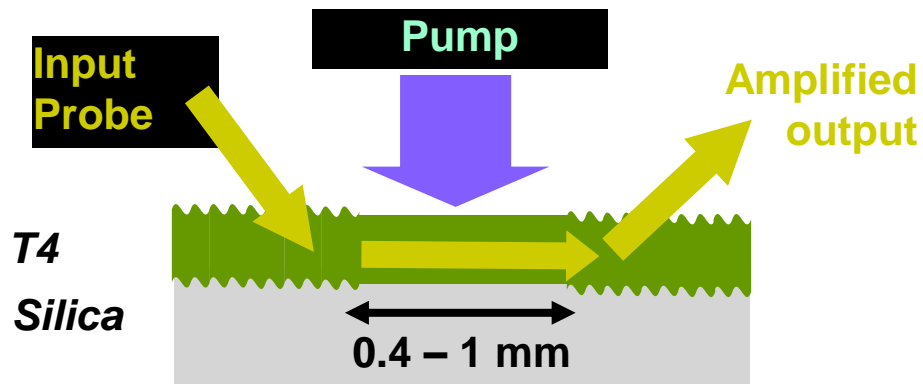


1st order scattering provides distributed feedback



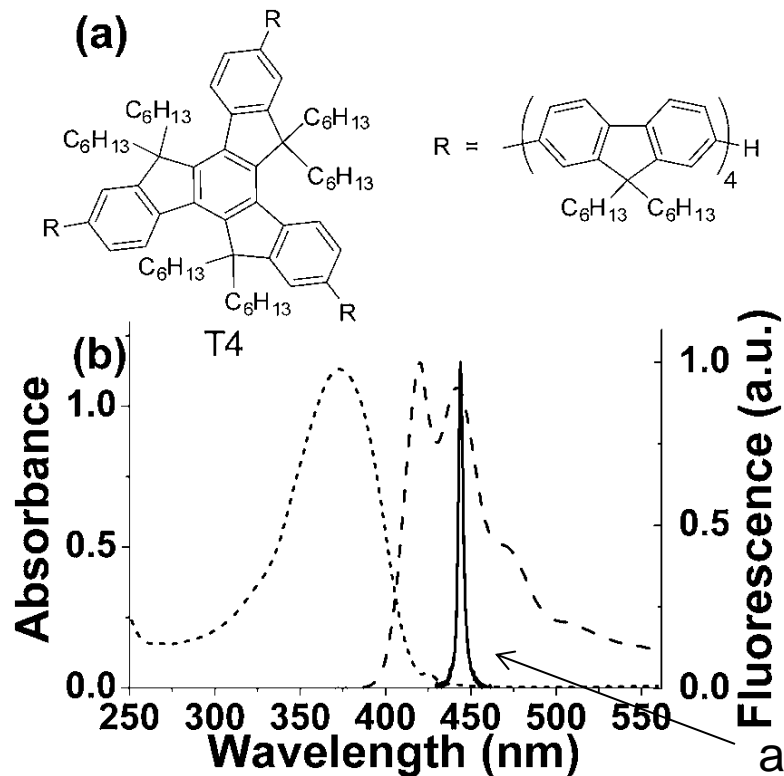
2nd order scattering provides surface output coupling

Truxene based lasers



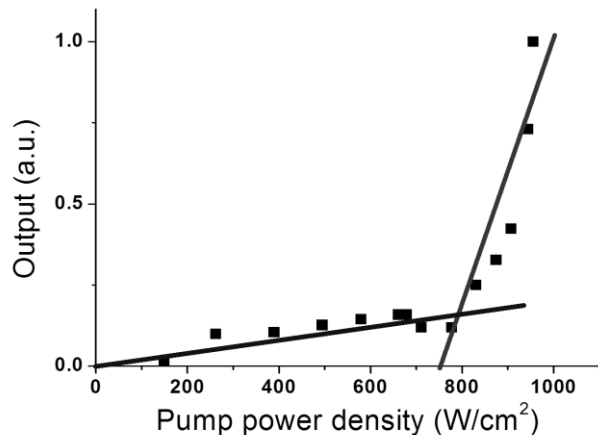
- ▶ T4 film (100-400 nm) forms optical waveguide
- ▶ Pump λ absorbed in ~ 100 nm
- ▶ Total internal reflection confines light in polymer film
- ▶ Signal amplified by 1000 times in 1 mm propagation through film
- ▶ T4 exhibits low-threshold (270 W cm^{-2}) and very low optical losses (2.3 cm^{-1}), one of the lowest report for an OSC

Appl. Phys. Lett., 2009, **94**, 243304.

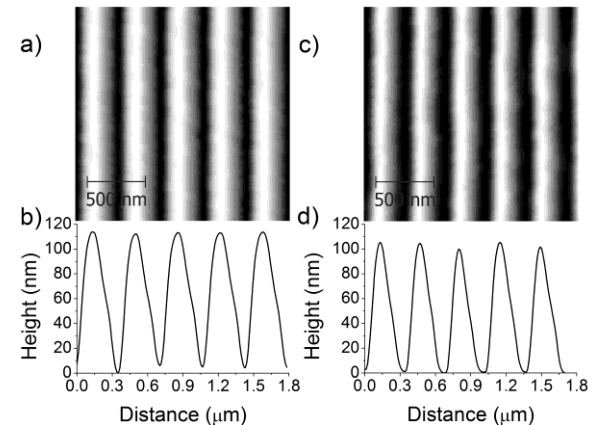
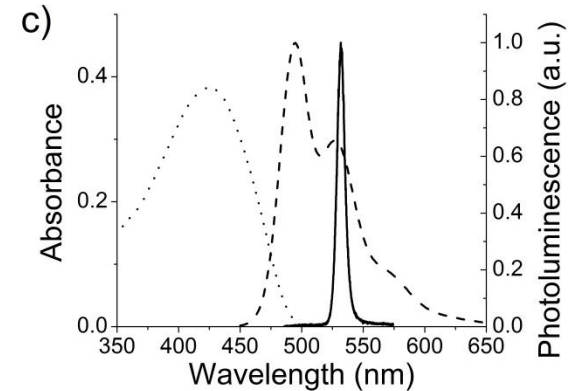
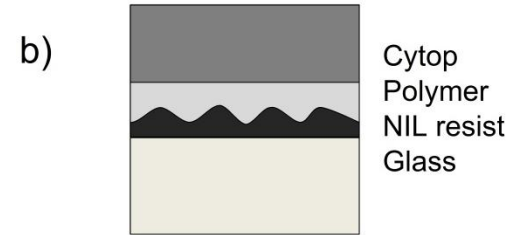
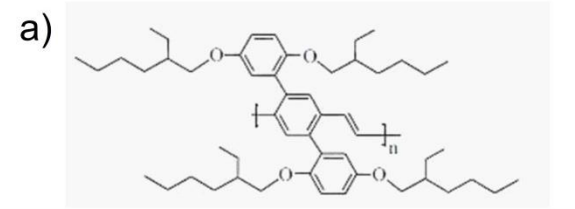


Nanoimprint lithography (NIL)

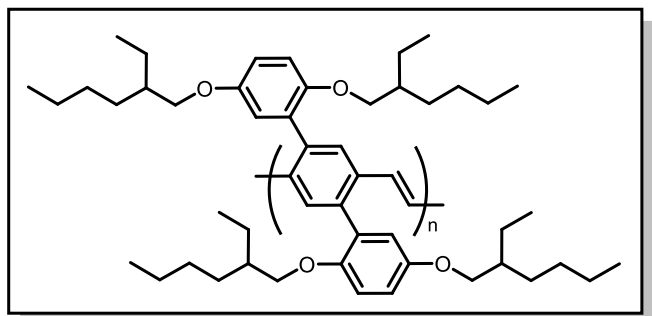
- Replicates a single high-quality master structure using a stamp-to-imprint procedure in a suitable photoresist under some combination of pressure and heat or UV illumination.
- Can be used as a roll-to-roll process.
- Low-cost.
- High volume.
- Commercial LED (Philips Luxeon Rebel royal blue) emitting at 448 nm; max power 1 kW/cm²
- Previous lasing thresholds from NIL: 2 kW/cm²



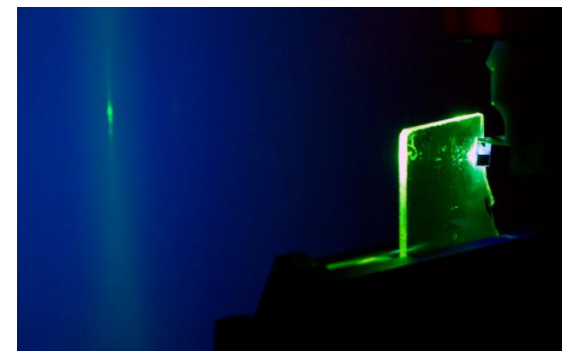
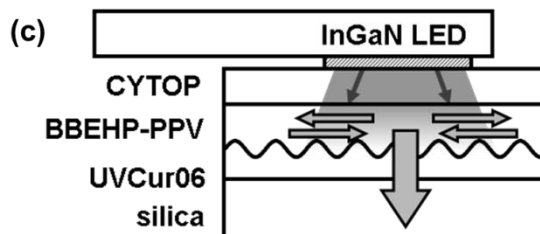
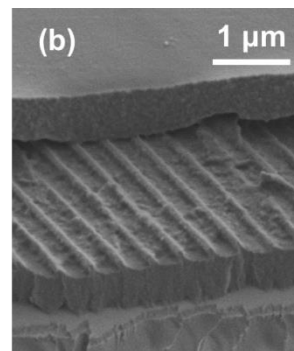
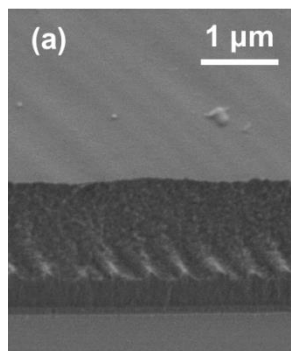
Threshold 770 W/cm²



Case study on polymers: BBEHP-PPV



First synthesised and reported by:
A. Rose, Z. G. Zhu, C. F. Madigan, T. M. Swager, V. Bulovic, *Nature* **2005**, 434, 876.



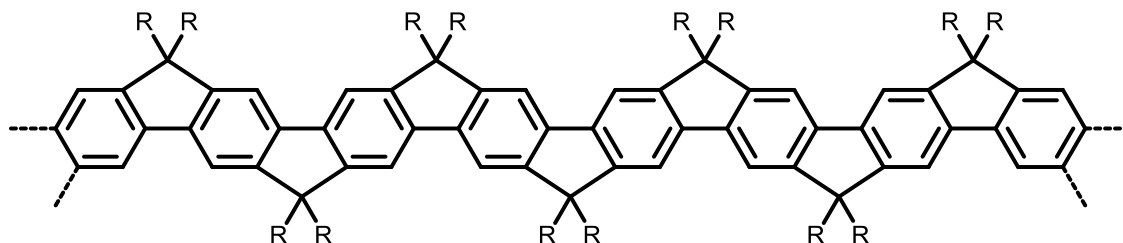
Plastic laser pumped by a cheap commercial LED!!!

Sample number	Isolation Method (from synthesis batch)	Appearance	M_w	M_n	PDI	ASE [Wcm ⁻²] threshold
1	Soxhlet, acetone (batch c)	Powder	28 430	28 270	1.01	488
2	Soxhlet, CH ₂ Cl ₂ (batch d)	Powder	202 800	93 220	2.18	216
3	Soxhlet, CH ₂ Cl ₂ (batch d)	Powder	341 300	68 860	4.96	167
4	Reprecipitation (batch b)	Powder	409 700	65 840	6.22	266
5	Soxhlet, CH ₂ Cl ₂ (batch c)	Powder	463 300	77 570	6.0	270
6	Soxhlet, CH ₂ Cl ₂ (batch b)	Powder	510 400	88 470	5.77	277
7	Reprecipitation (batch a)	Powder	582 800	54 790	10.6	302
8	Soxhlet, CH ₂ Cl ₂ (batch d)	Fibre	551 800	115 800	4.77	638

Materials for organic solar cells and field effect transistors

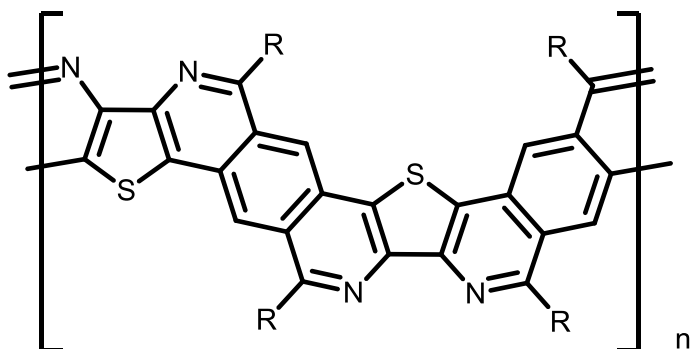
Towards planar structures

How important are non-covalent interactions vs covalent links?

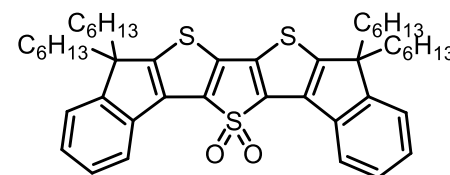
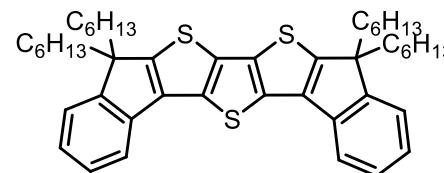
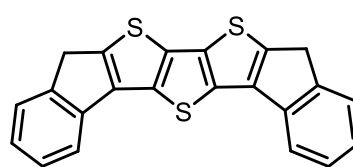


Scherf, Müllen

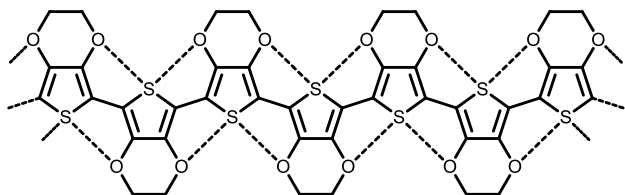
Ladders polymers – planar, but at the expense of bulky substituents to allow solubility



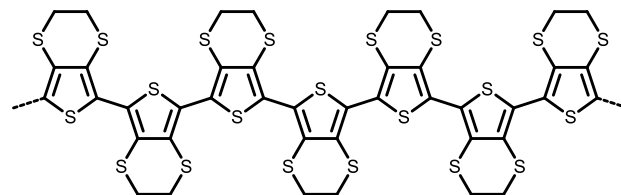
J. M. Tour, *J. Org. Chem.*, 2007, **72**, 7477.



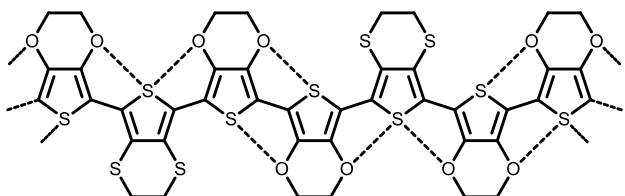
J. Mater. Chem., 2010, 20, 1112



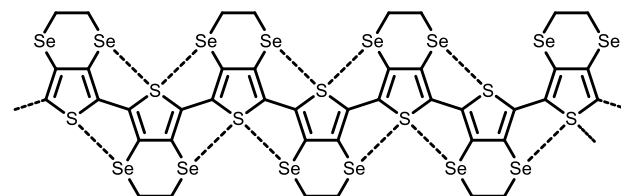
PEDOT
Eg = 1.6 eV



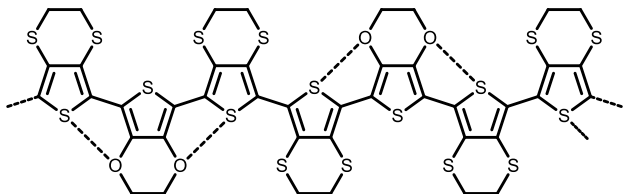
PEDTT
Eg = 2.2 eV



POSO
Eg = 1.6 eV



PEDST
Eg = 1.8 eV



PSOS
Eg = 2.1 eV

Chem Mater., 2007, 19, 301

J. Mater. Chem., 2005, 15, 4783

Contact Radii :	C	H	F	S	N	O	Se	B
(Angstrom)	1.70	1.20	1.47	1.80	1.55	1.52	1.90	1.63

S---O interactions are 2.9-3.2 Å as opposed to 3.32 Å for S + O

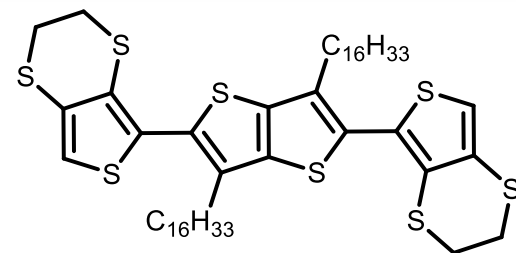
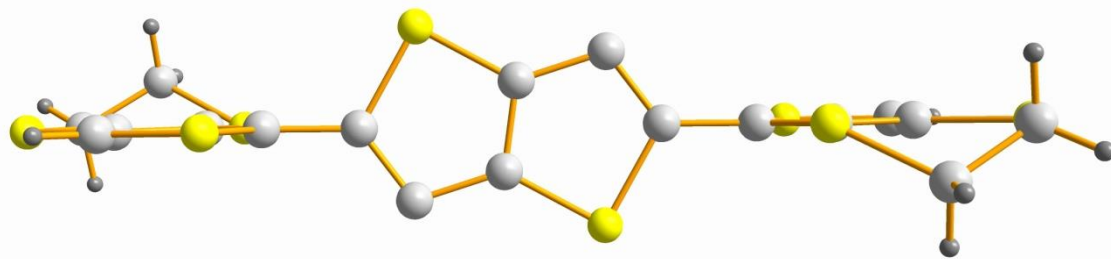
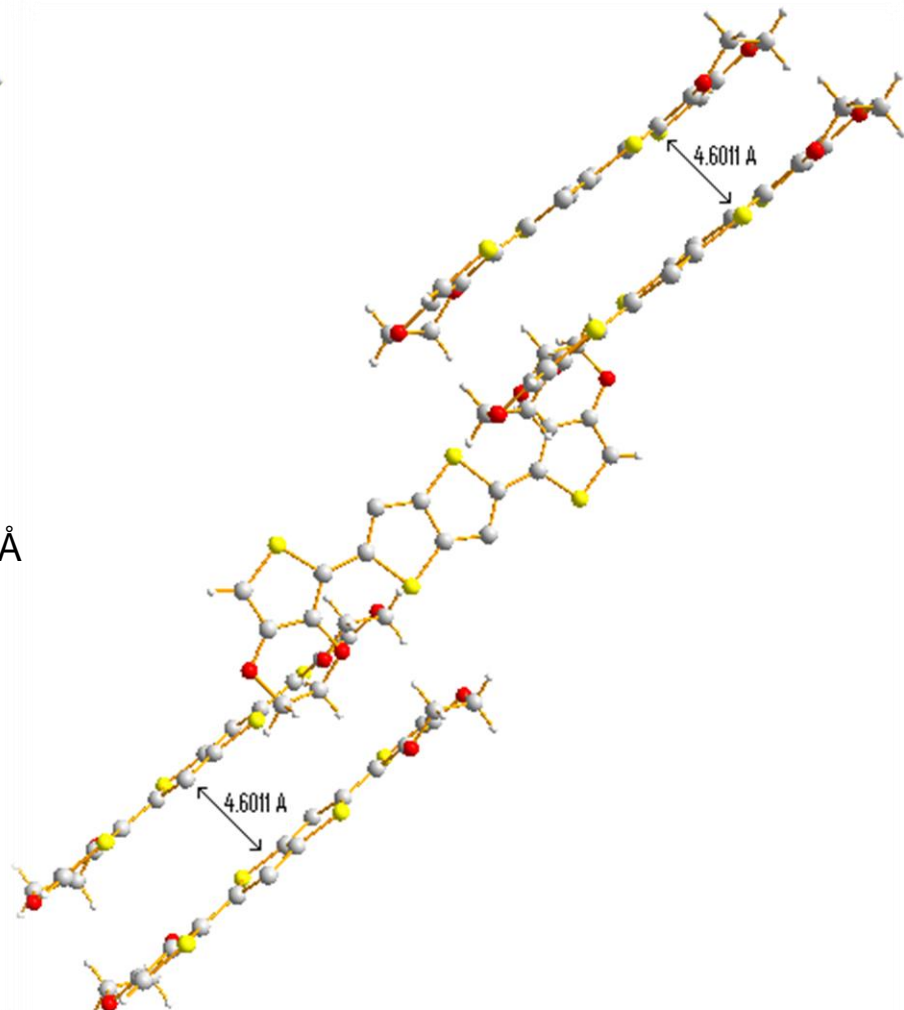
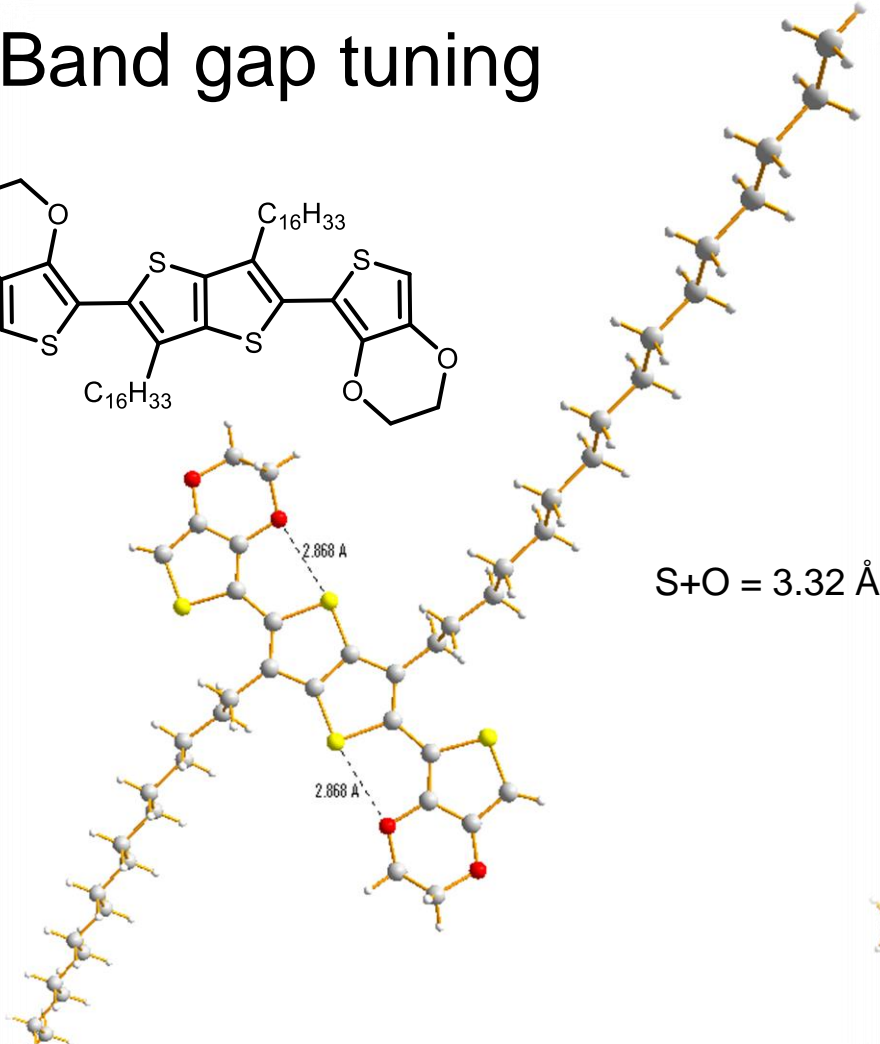
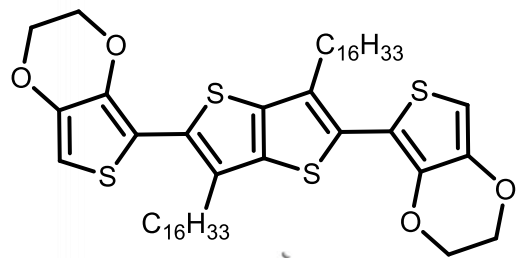
Evidence: crystal structures, abs data, molecular modelling,
systematic studies by structural variation

How significant are non-covalent interactions in conjugated polymers?

Pathways to exploitation:

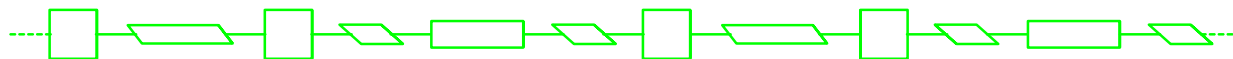
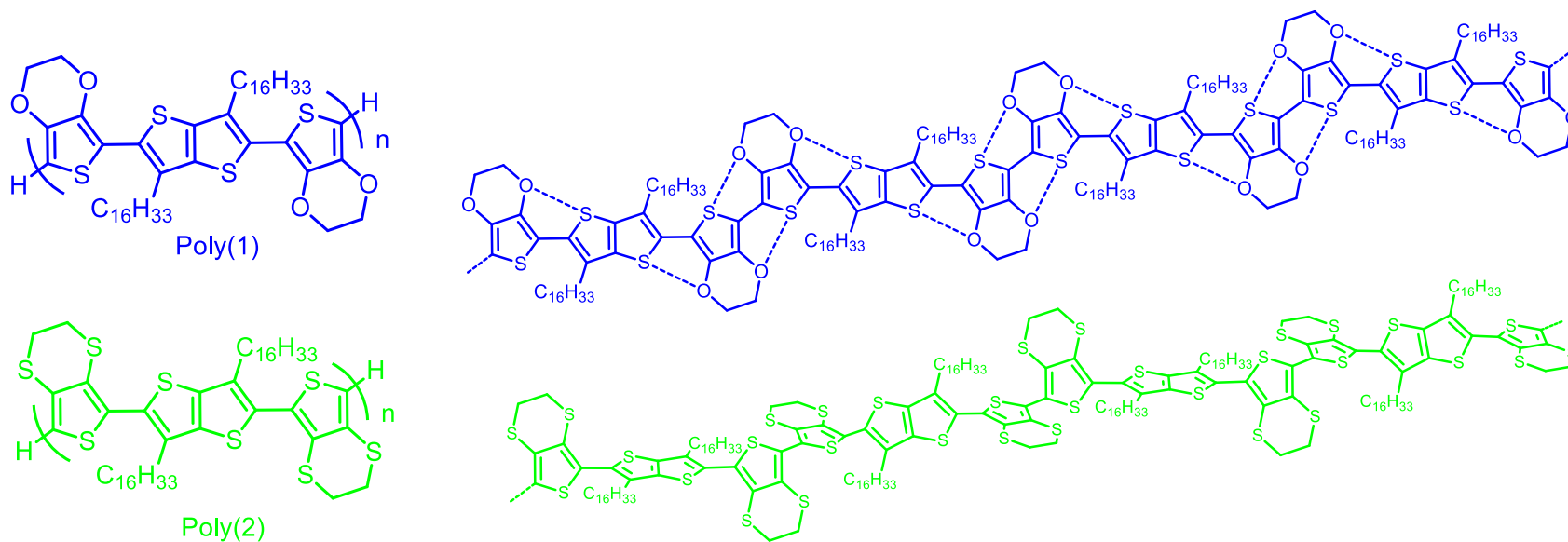
1. Band gap tuning
2. Enhanced stability of intermediate charged states
3. Processability
4. Efficient orbital overlap in the bulk in 2 or 3 dimensions

1. Band gap tuning

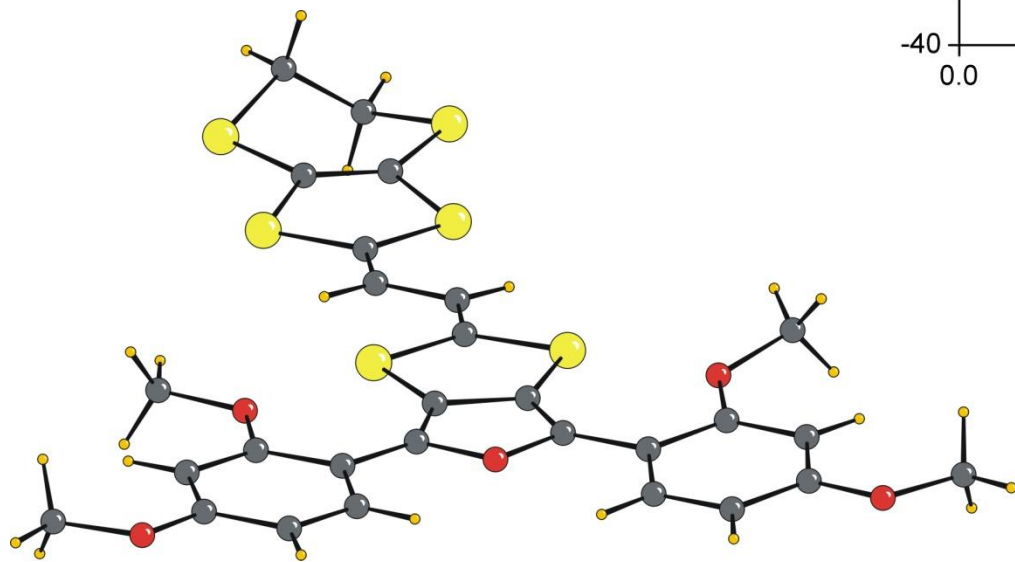
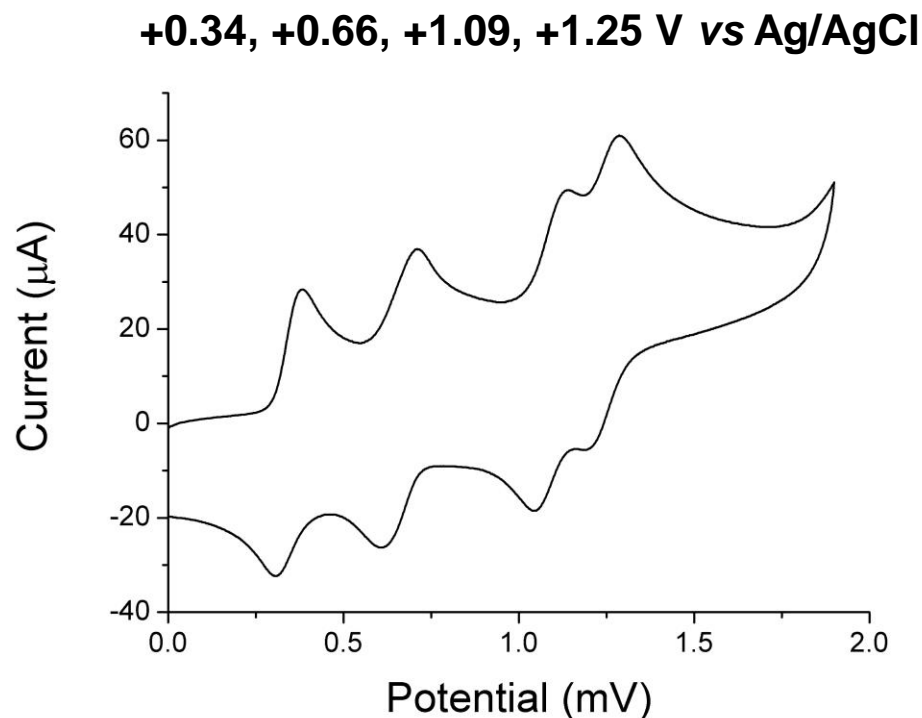
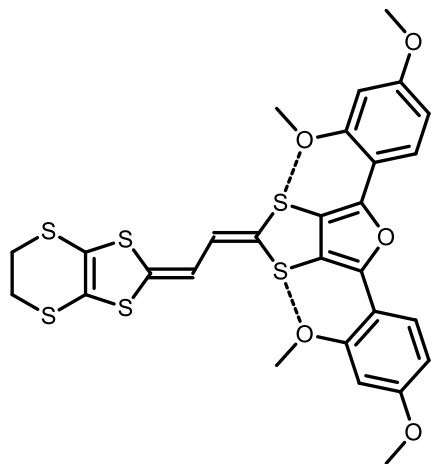


	E_{1ox} / V	E_{2ox} / V	E_{1red} / V	HOMO / eV	LUMO / eV	Electrochemical ^a HOMO-LUMO gap / eV	λ_{max} / nm
1	+0.46	+0.95	-1.98	-5.18	-2.97	2.21	347
2	+0.71/0.51	+1.00	-1.97	-5.42	-2.97	2.45	310

	E_{1ox} / V	E_{2ox} / V	E_{1red} / V	HOMO / eV	LUMO / eV	^a E_g eV
Poly(1)	-0.33	+0.31	-2.02	-4.24	-2.71	1.53
Poly(2)	+0.64/0.52	+0.74	-2.05	-5.39	-2.9	2.49



2. Enhanced stability of intermediate charged states

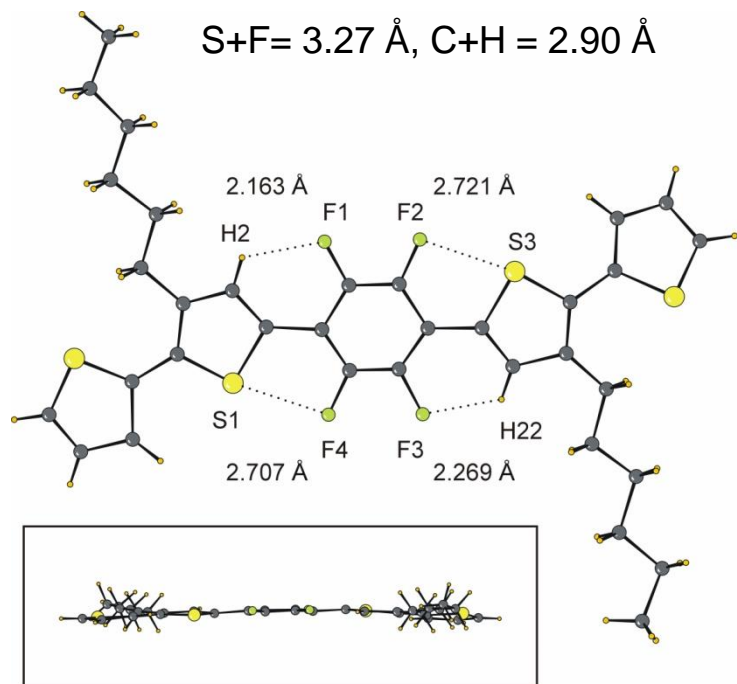


Tetrahedron Lett., 2004, **45**, 2535

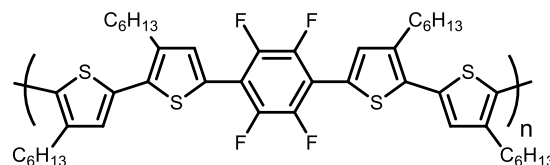
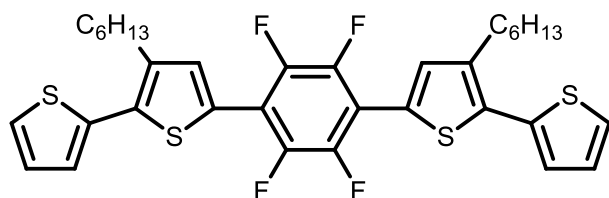
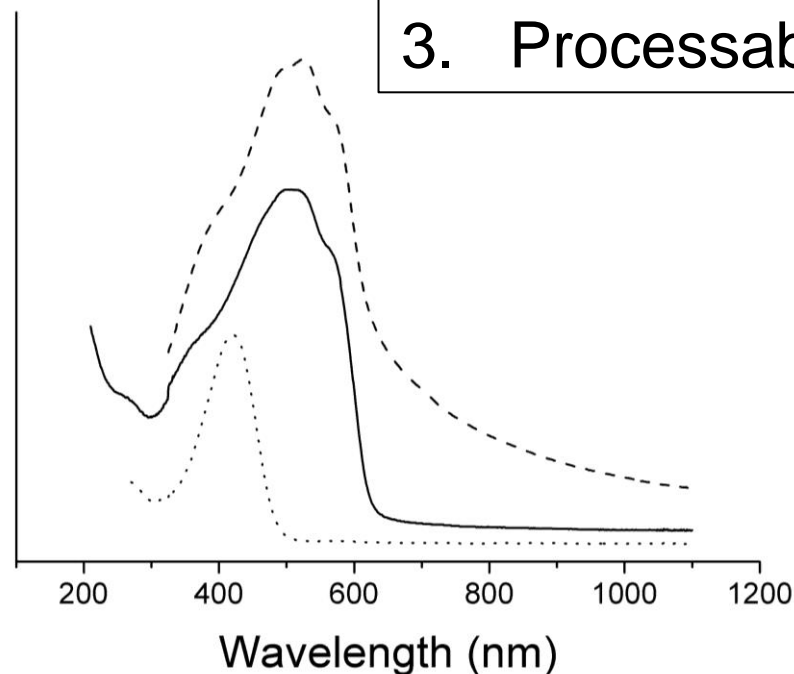
See also:

Chem. Soc. Rev., 2005, **34**, 69.

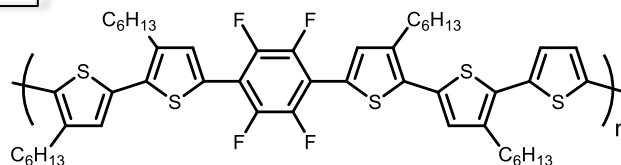
3. Processability



Absorbance (au)



Dashed line
Electropolymerised
Solid state



Solid line – solid state
Dotted line – solution state

$M_n = 10,500$ (70 aryl units)

Excellent solubility

λ_{\max} (solution) = 420 nm

λ_{\max} (film) = 566 nm

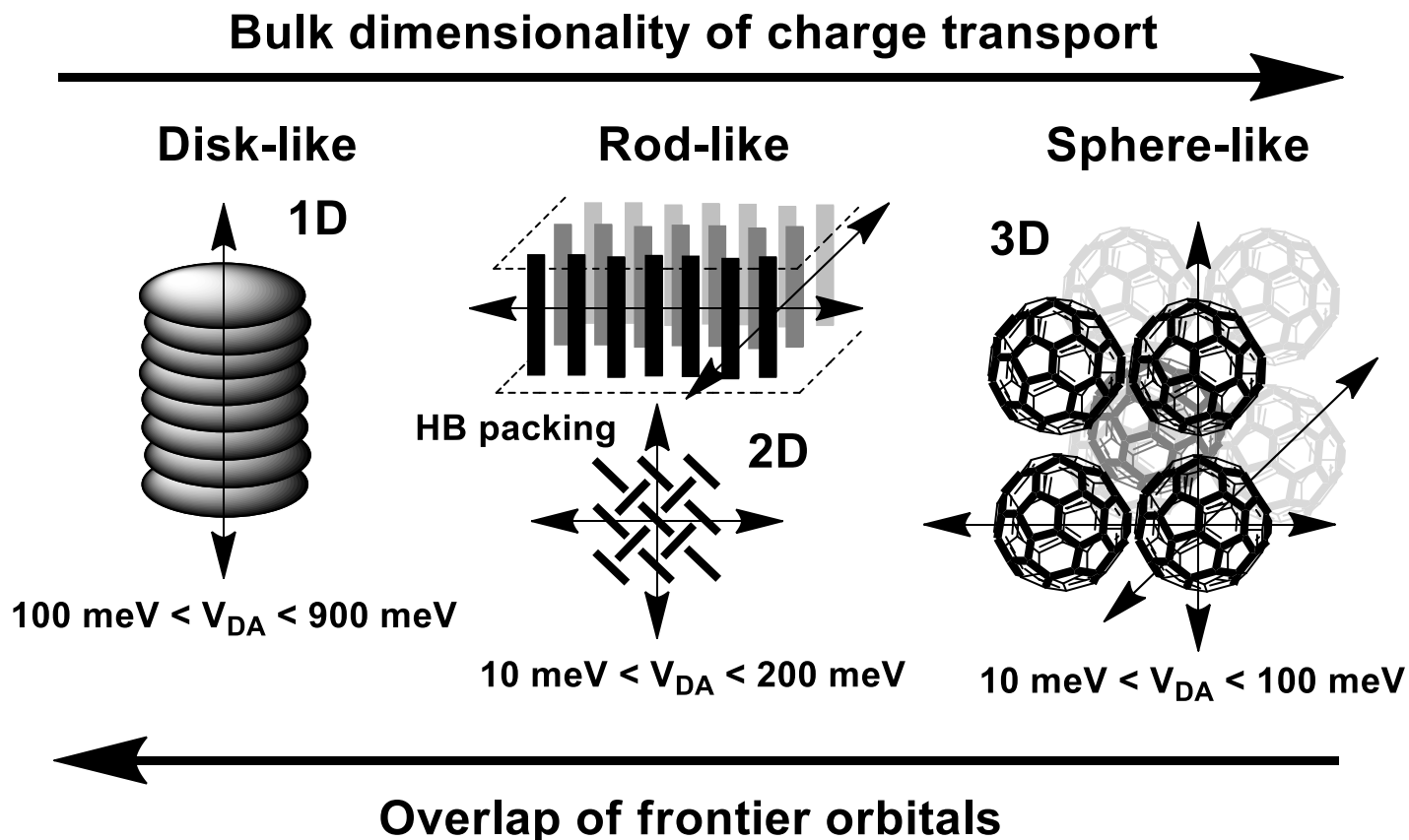
Mobility = $1 \times 10^{-3} \text{ cm}^2/\text{Vs}$

On/off ratio $10^4 - 10^5$ Turn on voltage 2 V

Chem. Commun., 2005, 1465.

Chem. Mater., 2005, 17, 6567.

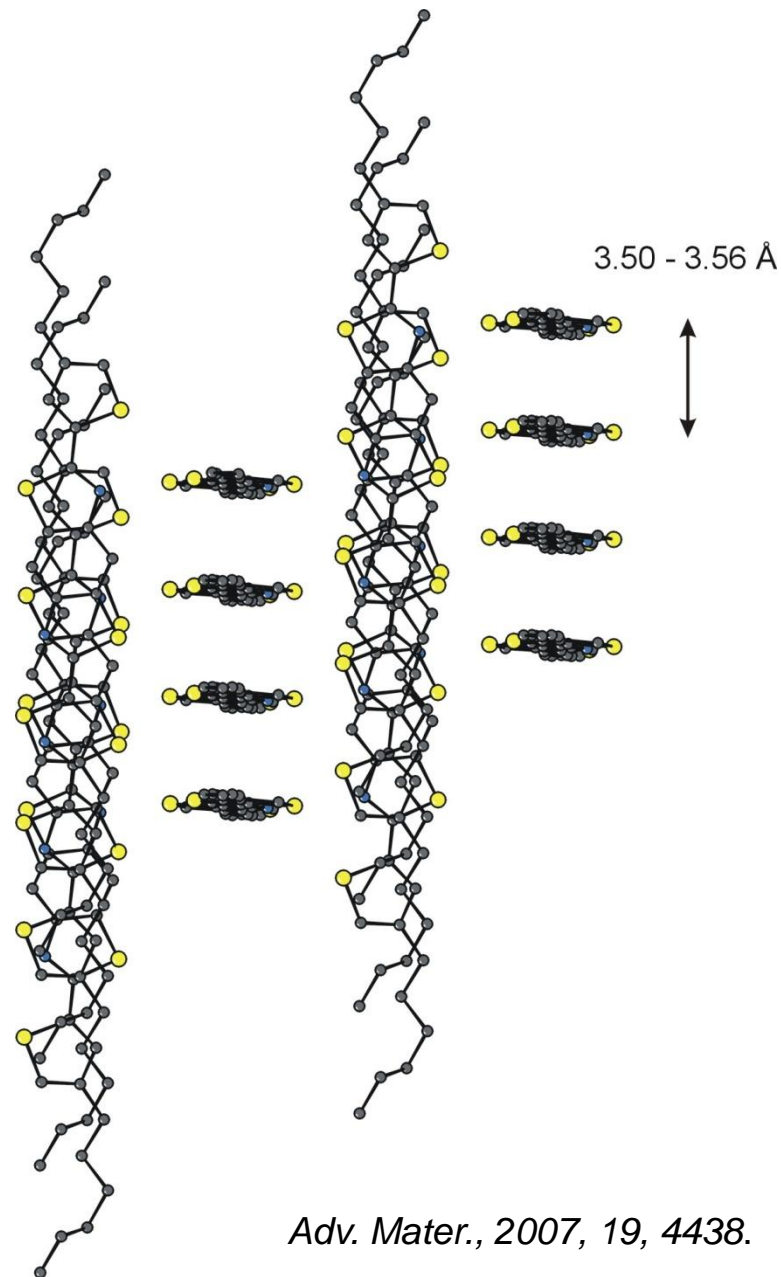
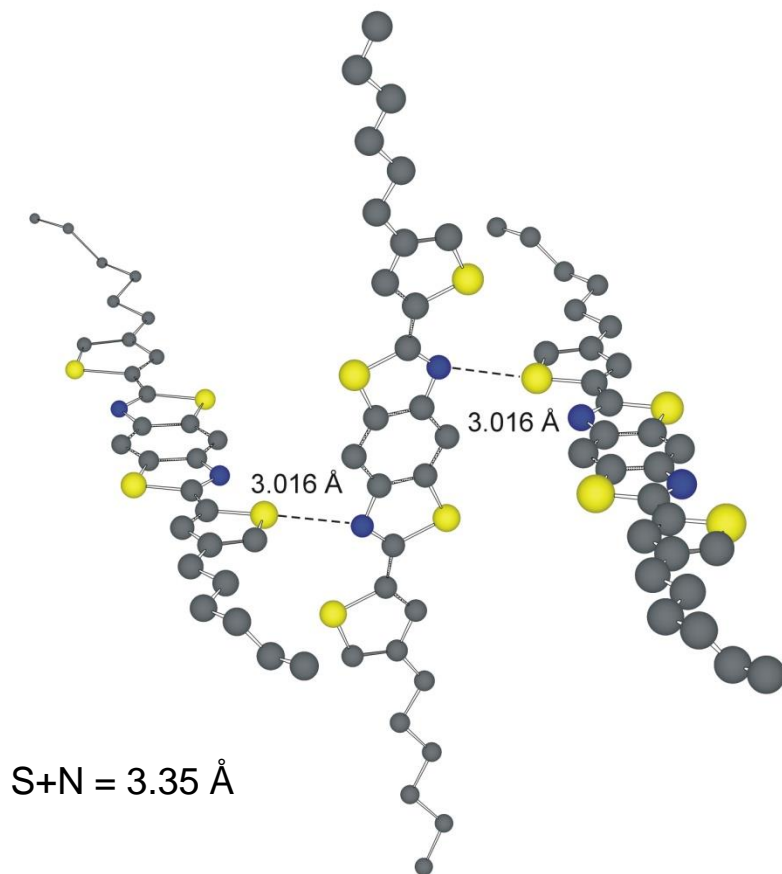
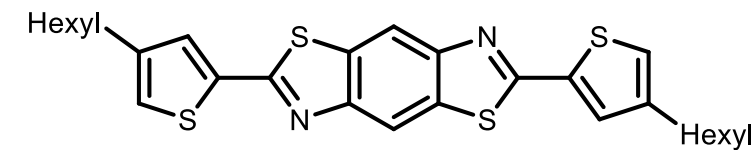
Charge transport in single molecule systems



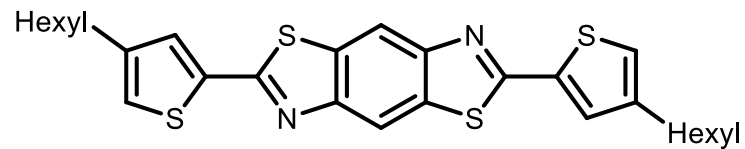
Schematic drawing of 1D, 2D, and 3D charge carrier materials with indicative values of transfer integrals. Values of transfer integrals, V_{DA} , should be considered as orders of magnitudes.

See: *Close Encounters of the 3D Kind – Exploiting High Dimensionality in Molecular Semiconductors*, Skabara, Arlin, Geerts, *Adv. Mater.*, 2013, **25**, 1948-1954.

4. Efficient orbital overlap in the bulk in 2 or 3 dimensions

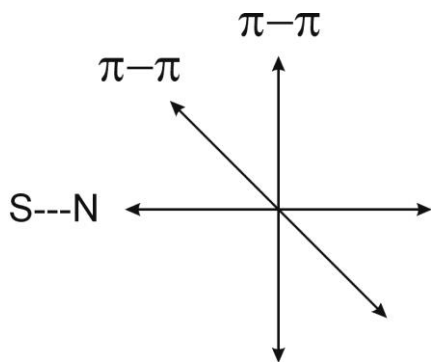
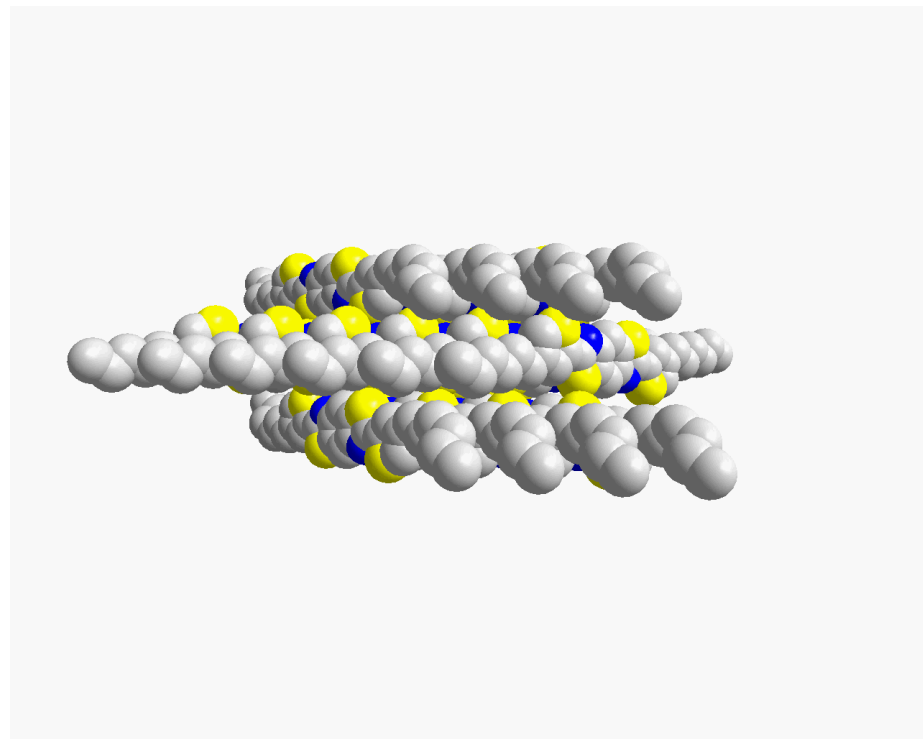
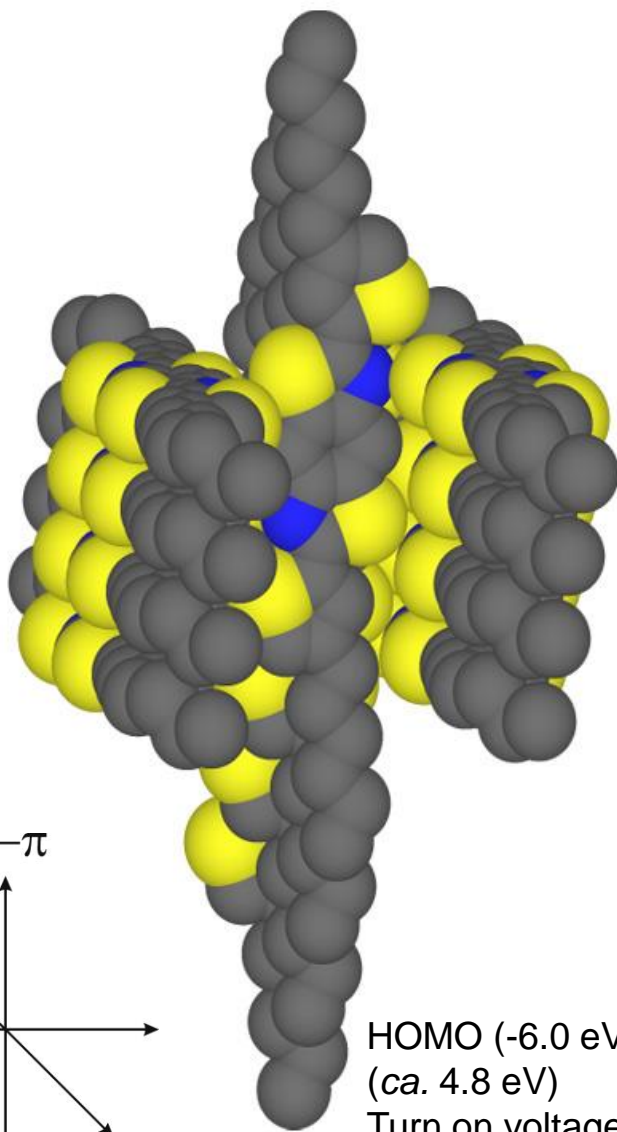


Adv. Mater., 2007, 19, 4438.



HOMO = -6.0 eV

LUMO = -2.7 eV



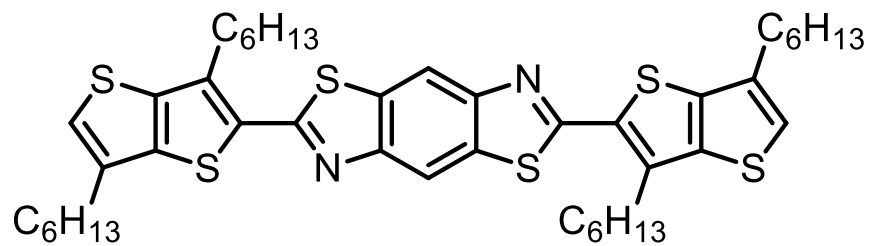
HOMO (-6.0 eV) appears to be far away from the Fermi level of Au (ca. 4.8 eV)

Turn on voltage -20 V

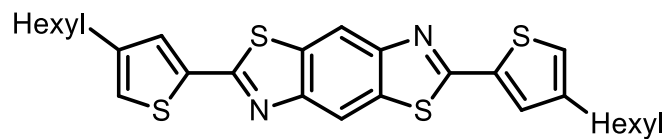
On/off ratio 10^5

Hole mobilities are approximately $0.01 \text{ cm}^2/\text{Vs}$ (bottom gate, bottom contact)

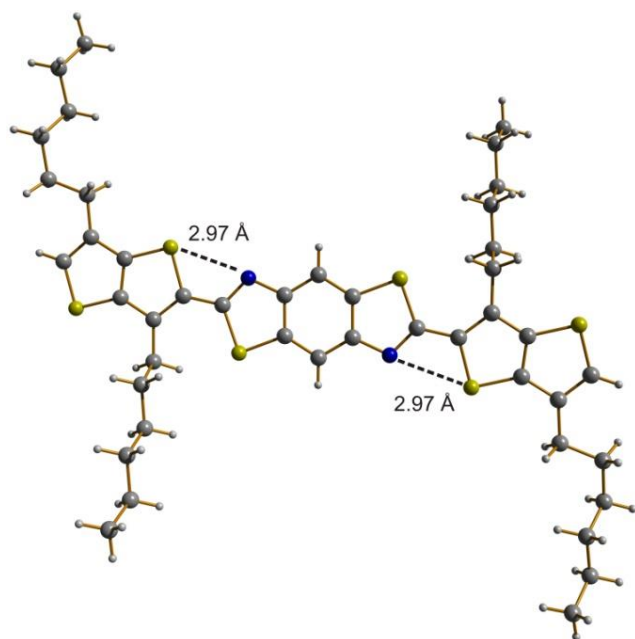
Adv. Mater., 2007, 19, 4438.



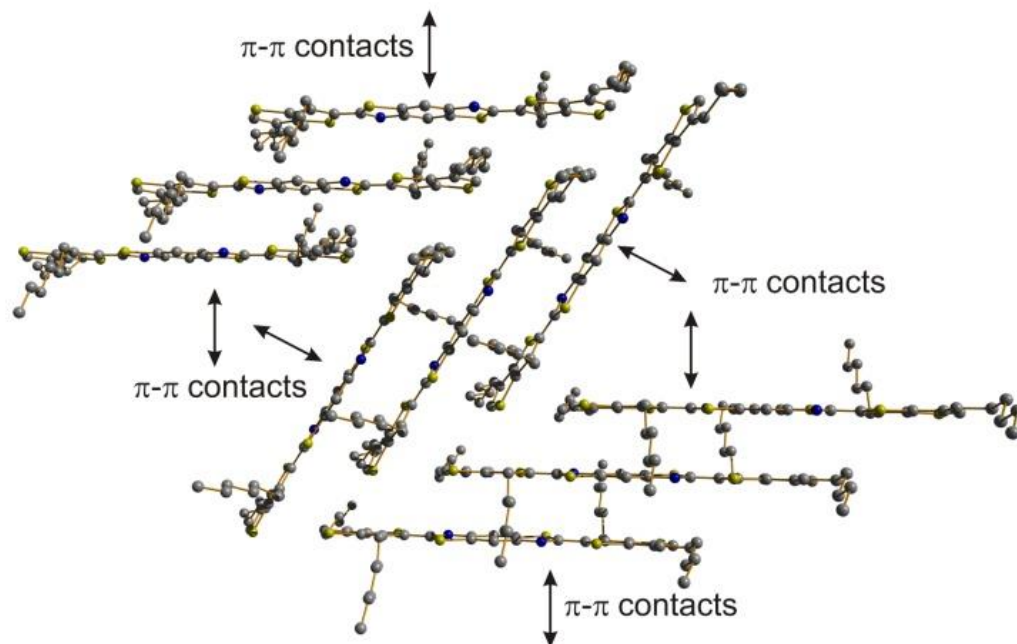
HOMO = -5.7 eV
LUMO = -2.9 eV



HOMO = -6.0 eV
LUMO = -2.7 eV



S+N = 3.35 Å



π - π = 3.52 Å



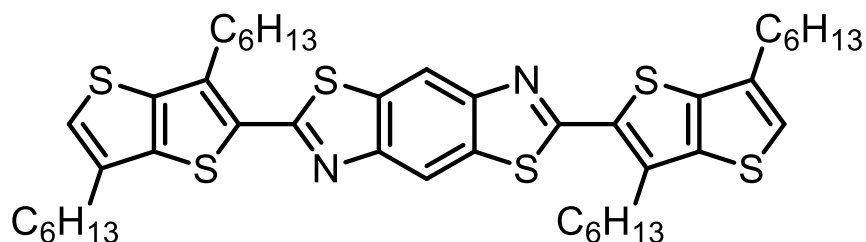
RSC Publishing

PAPER
Gang Li, Michael et al.
Self-assembly and charge transport
properties of a benzothiadiazole end-
capped with oligothiophene
units

International Year of
CHEMISTRY
2011



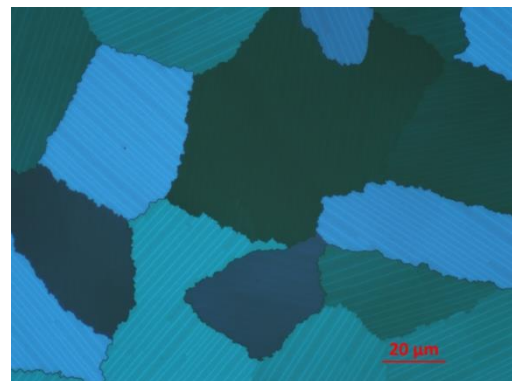
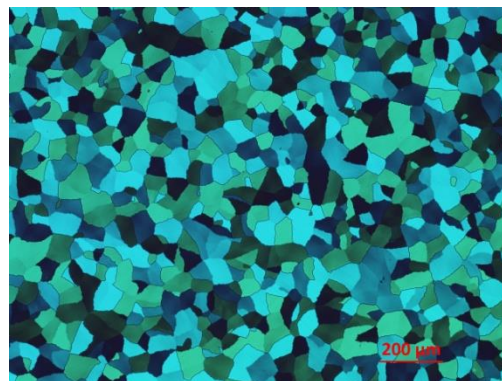
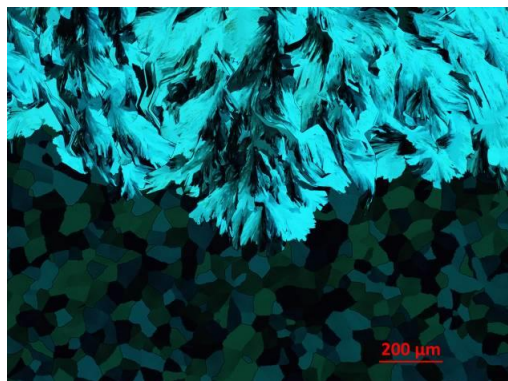
0950-0804(201102)21:7;1-D



J. Mater. Chem., 2011, 21, 2091



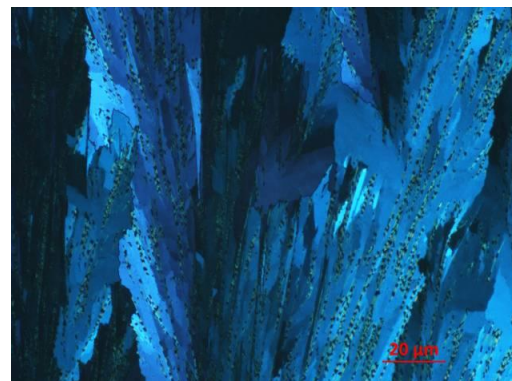
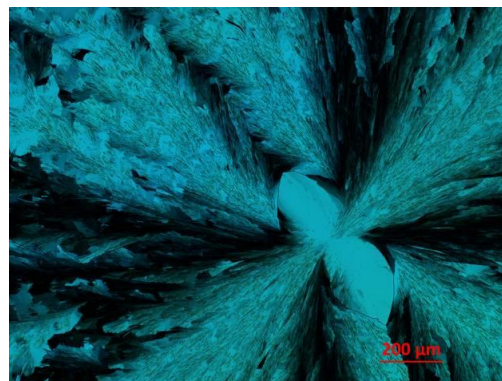
Before annealing
(50× magnification)



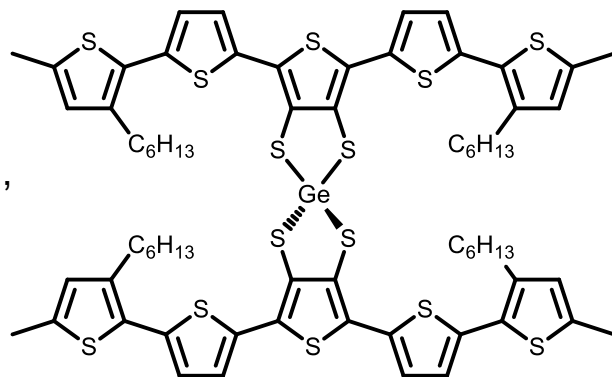
After annealing for 16 hours at 80°C Phase 1 (crystal B) above, Phase 2 (crystal E) below

After annealing for a further 2 hours at 100°C
only Phase 2 persists.

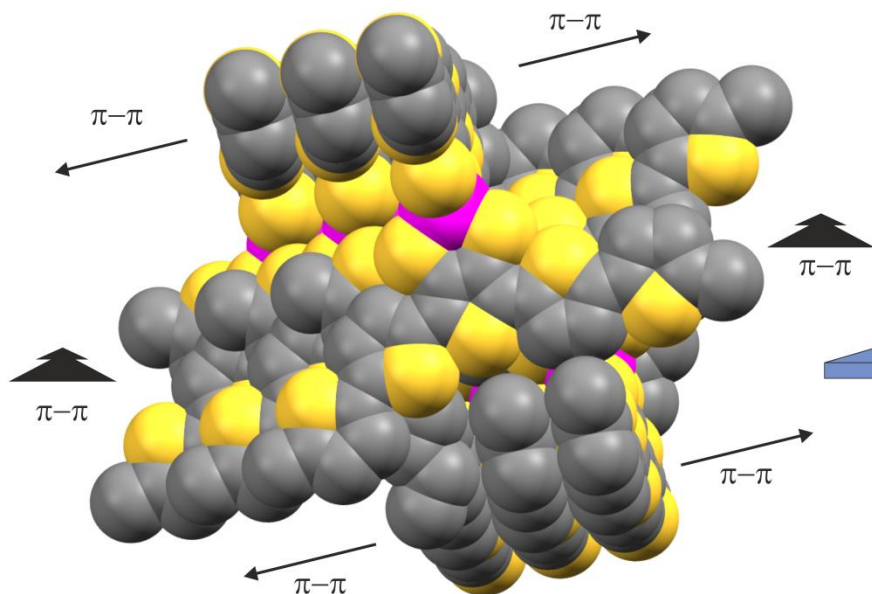
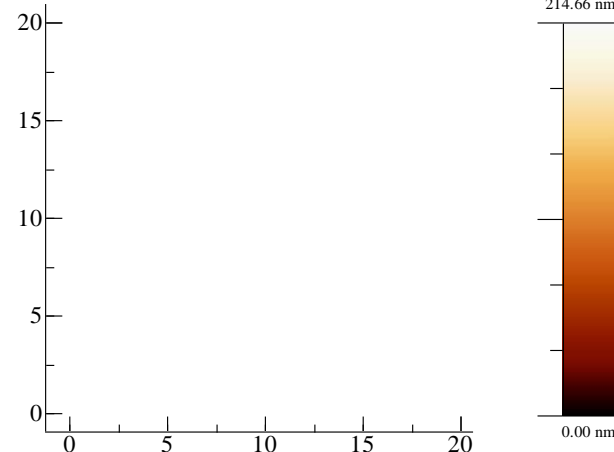
Hole mobilities of $3 \times 10^{-3} \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$
after annealing at 70°C for 15 hours
Bottom contact, top gate device
On/off ratio of $10^4 - 10^5$
(slight drop in Phase 2 OFETs)



Angew. Chemie. Int. Ed.,
2012, **51**, 4562

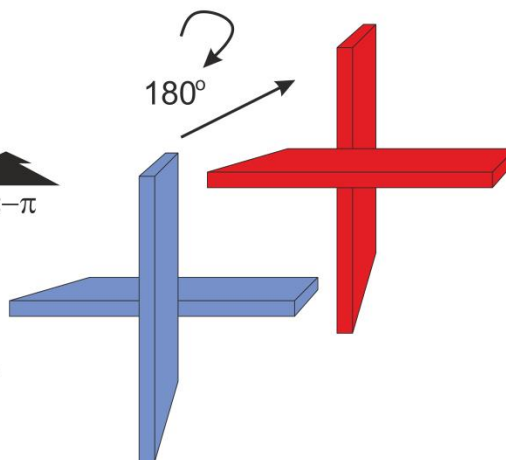


Y[μm]

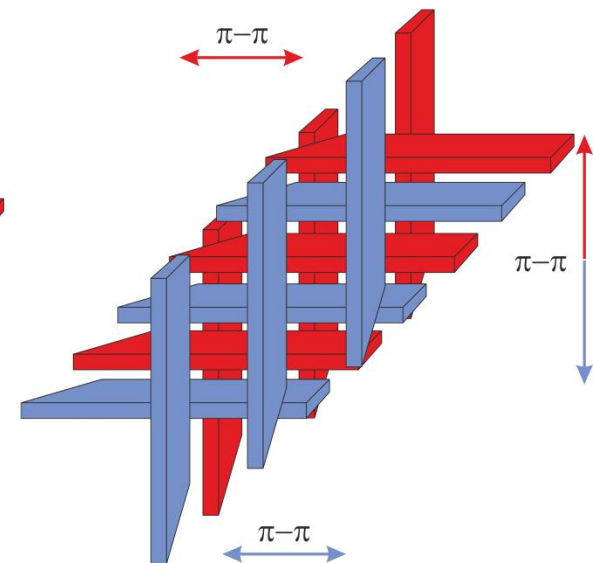


(a)

$$\pi-\pi = 3.3 - 3.6 \text{ \AA}$$

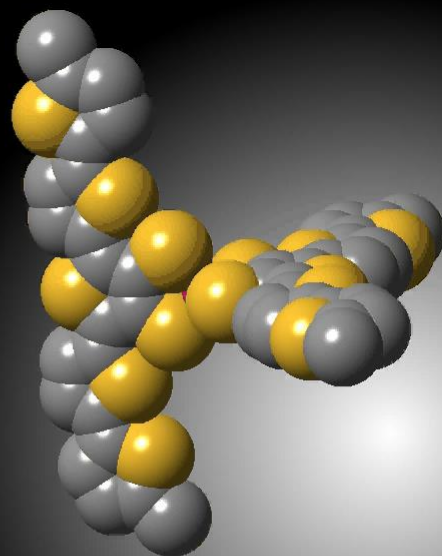


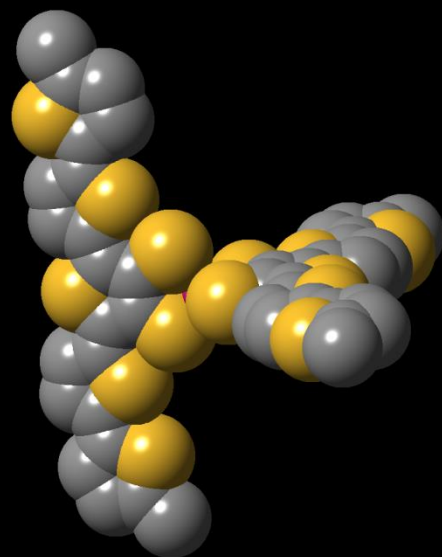
(b)

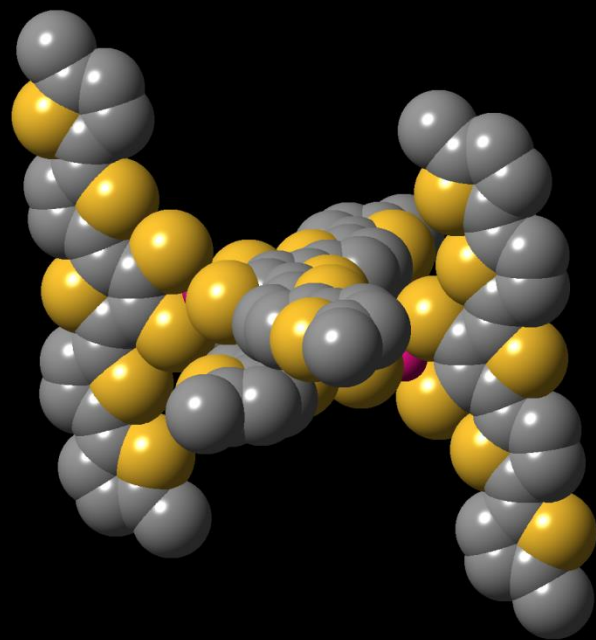


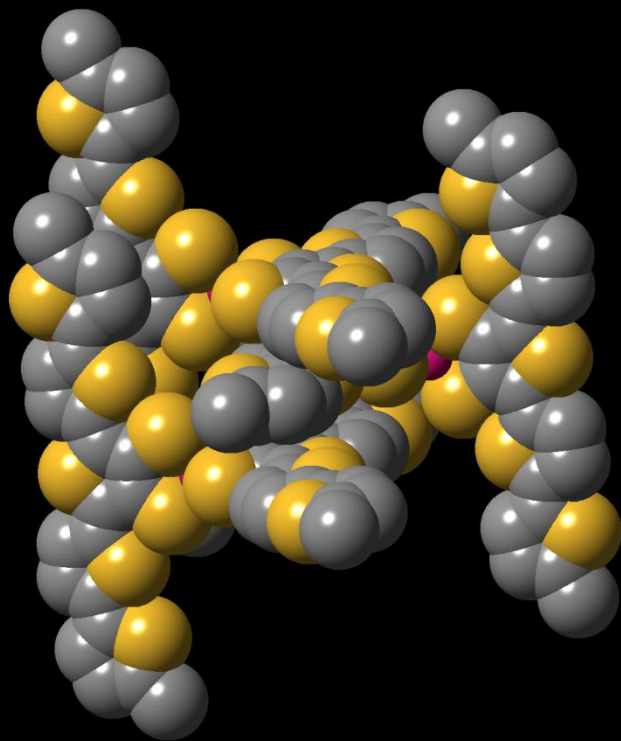
(c)

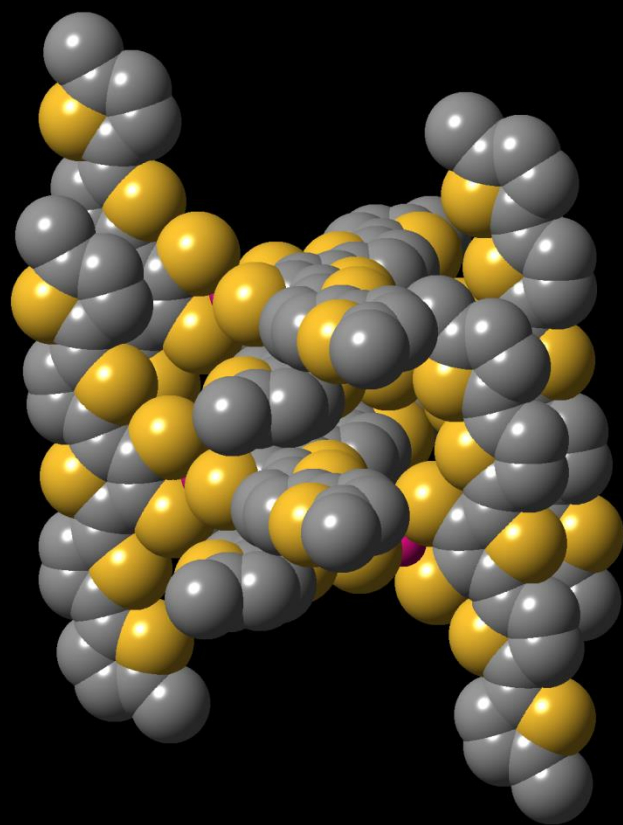
Self-assembly of cruciform 5T derivative: (a) packing structure derived from X-ray crystallography, showing π -stacking into and along the plane of the page (alkyl chains omitted; Ge atom coloured purple); (b) two orientations of the cruciform; (c) how these orientations interdigitate between horizontal chains to give a tightly packed structure with infinite π -stacks in two orthogonal directions.

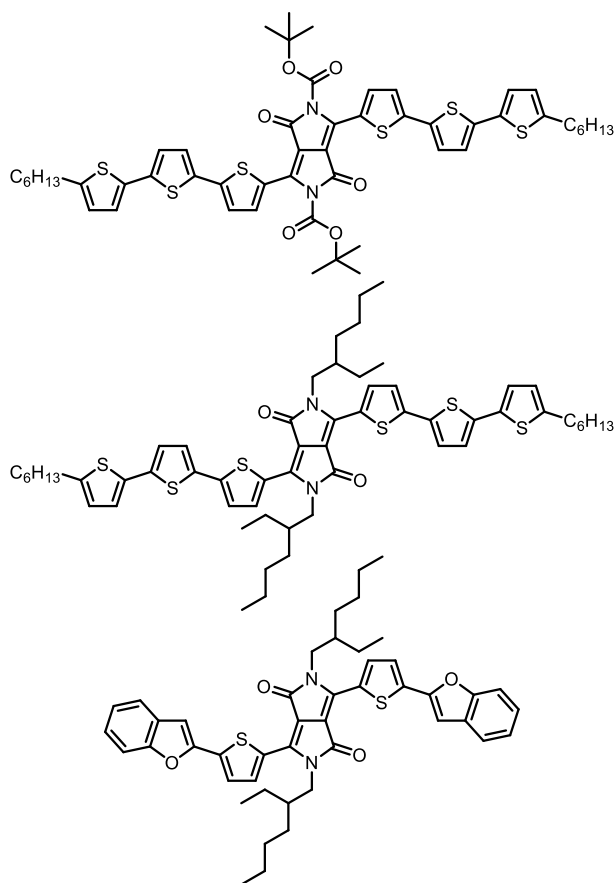












Nguyen et al., J. Phys. Chem. C, 2008, 112, 11545
2.3%

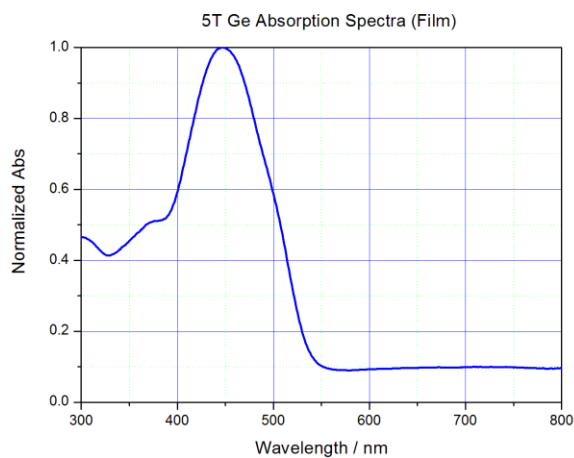
HOMO-LUMO gap 1.5 eV

Appl. Phys. Lett., 2009, 94, 103301
3.0%

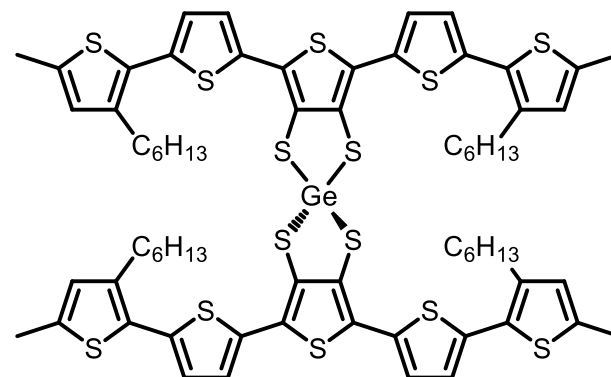
HOMO-LUMO gap 1.5 eV

Adv. Funct. Mater., 2009, 19, 3063
4.5%
Adv. Energy Mater., 2011, 1, 610
5.2%

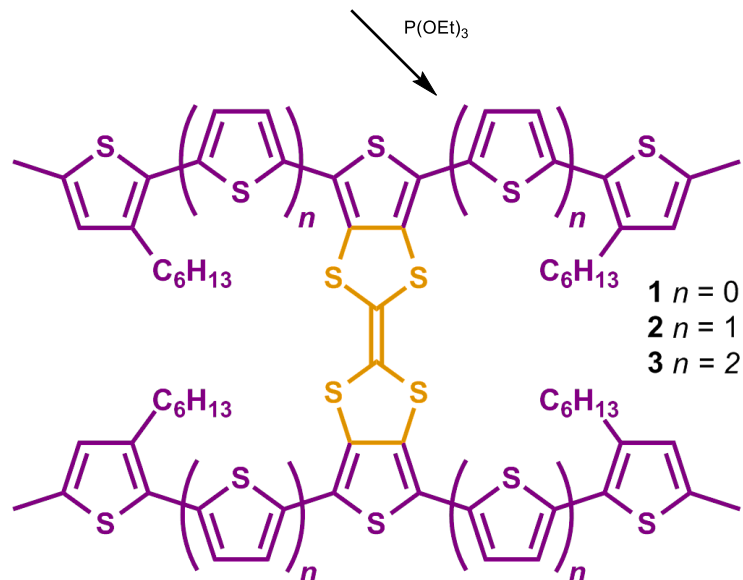
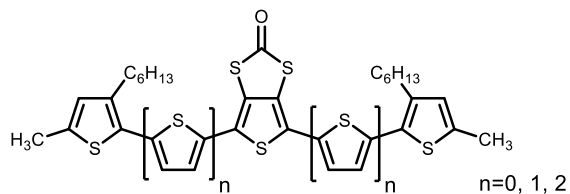
HOMO-LUMO gap 1.8 eV



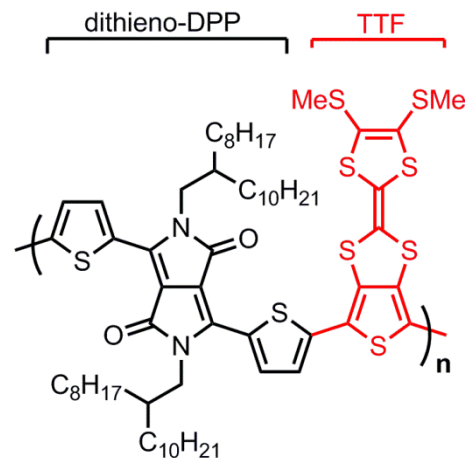
$J_{SC} = 9.57 \text{ mA cm}^{-2}$
 $V_{OC} = 0.63 \text{ V}$
 $FF = 0.37$
 $PCE = 2.26\%$
 $E_g = 2.3 \text{ eV}$



nT-TTF

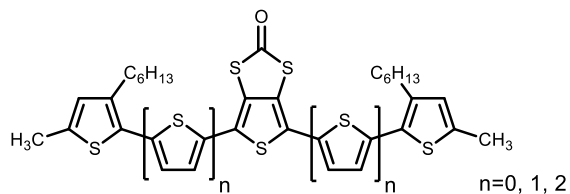


J. Mater. Chem. C, 2014, **2**, 2674

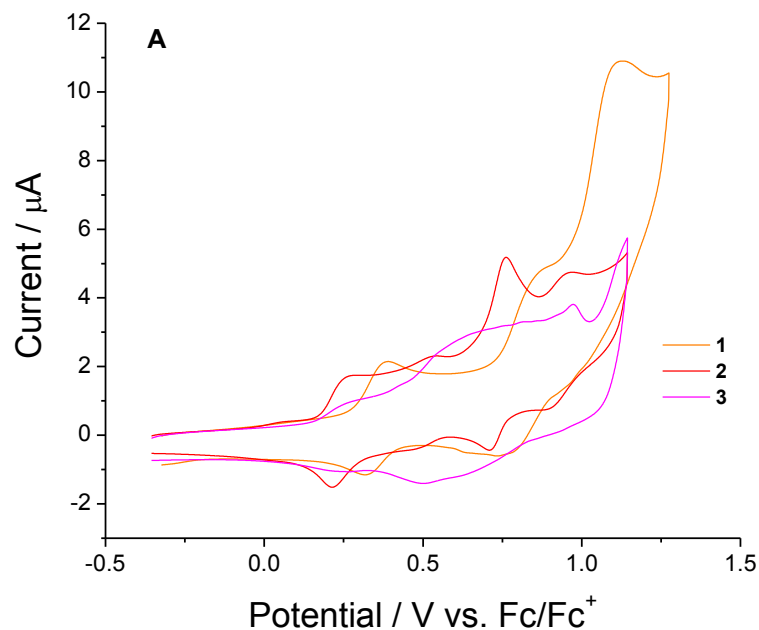
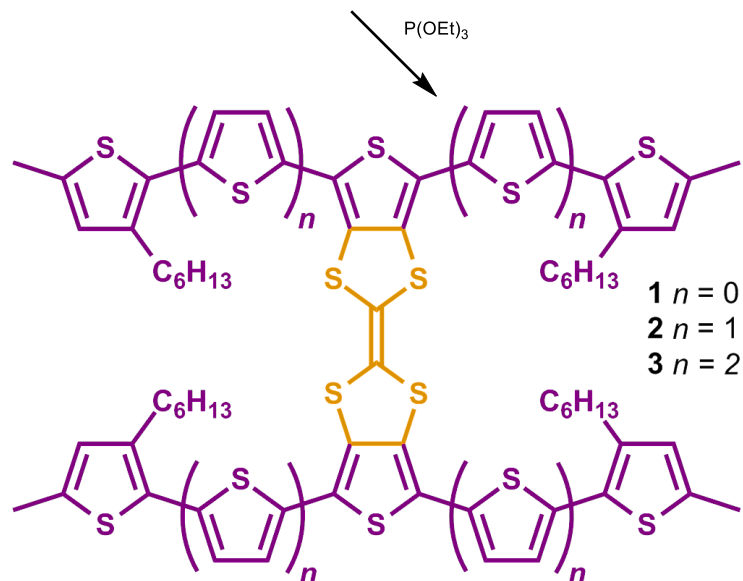


$\mu_h = 4 \pm 2 \times 10^{-2} \text{ cm}^2 \text{ V}^{-1} \text{ s}^{-1}$
air stable devices c.f. non-TTF analogue
J. Mater. Chem. 2012, **22**, 11310.

	1	2	3
λ_{max} (nm)	351	431	461
HOMO-LUMO gap (eV)	2.92	2.45	2.20
$E_{1\text{ox}}$ (V)	+0.39/+0.32	+0.27/+0.21	+0.26/+0.23
$E_{2\text{ox}}$ (V)	+0.86/+0.75	+0.54/+0.48 ^q	+0.66/+0.49 ^q
$E_{3\text{ox}}$ (V)	+1.13/+1.02 ^q	+0.76/+0.71 ^q	+0.97/+0.94 ^q
$E_{4\text{ox}}$ (V)	-	+0.97/+0.89 ^q	-
E_{red} (V)	-2.12 ^{irr}	-2.19 ^{irr}	-1.98 ^{irr}
HOMO (eV) ^a	-5.06	-4.96	-4.95
LUMO (eV) ^a	-2.92	-2.81	-3.00
HOMO-LUMO gap (eV) ^b	2.14	2.15	1.95

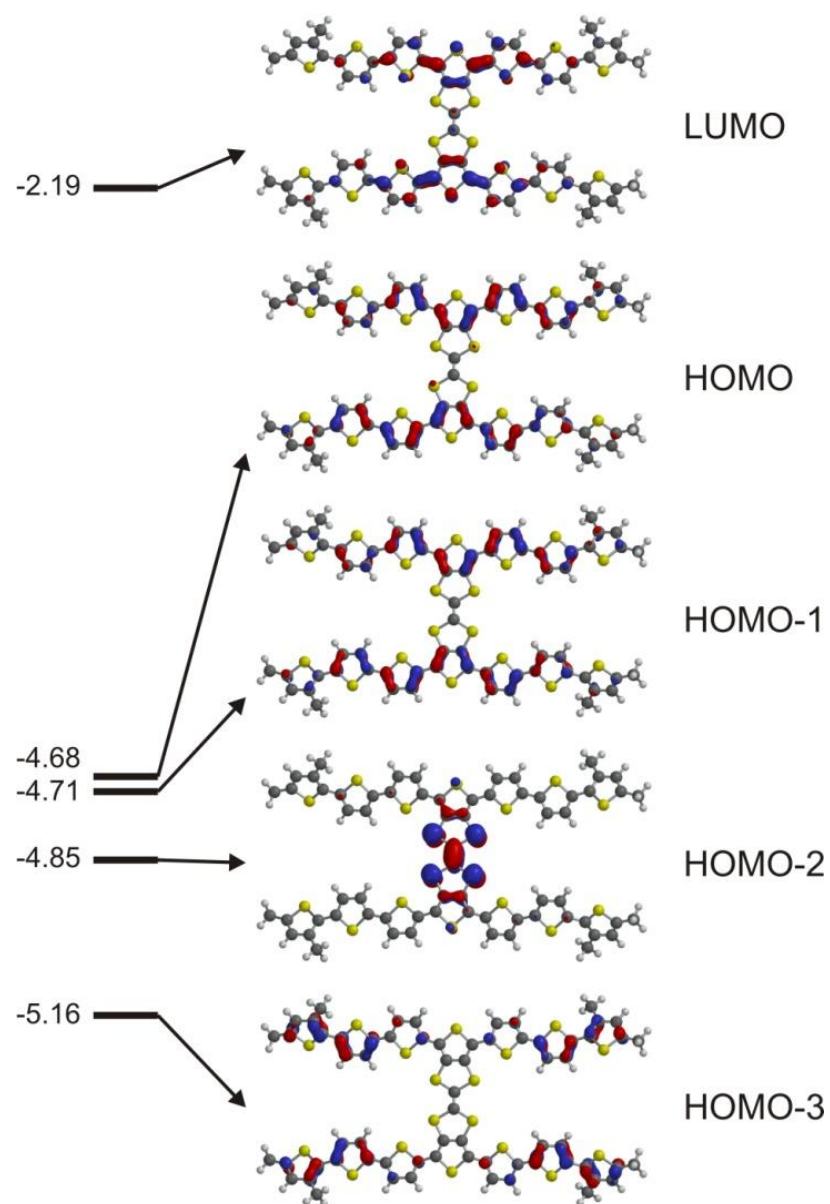
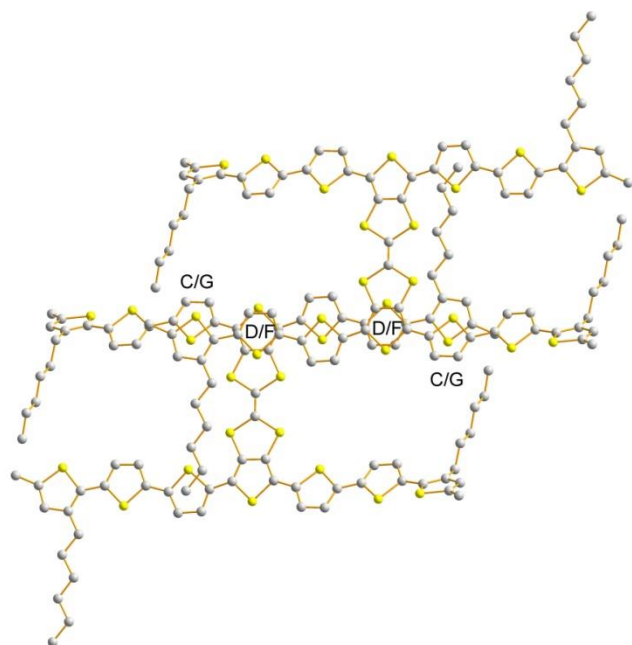
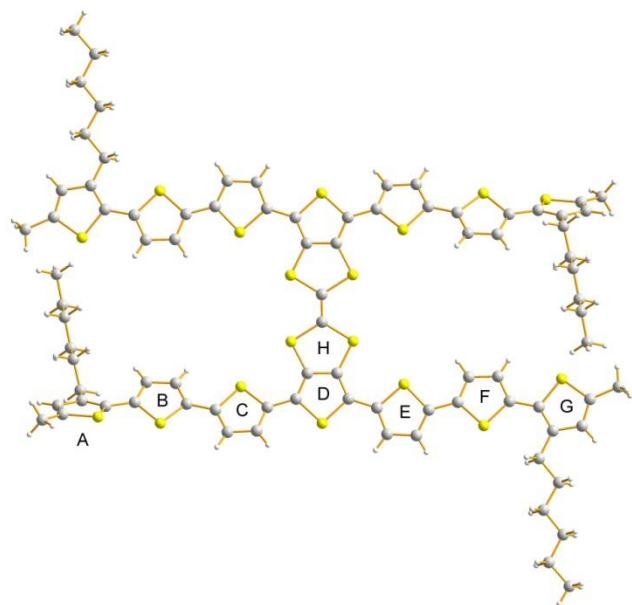


nT-TTF

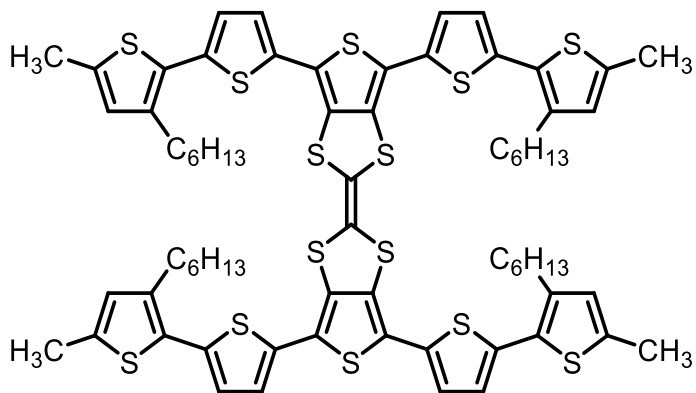


J. Mater. Chem. C, 2014, **2**, 2674

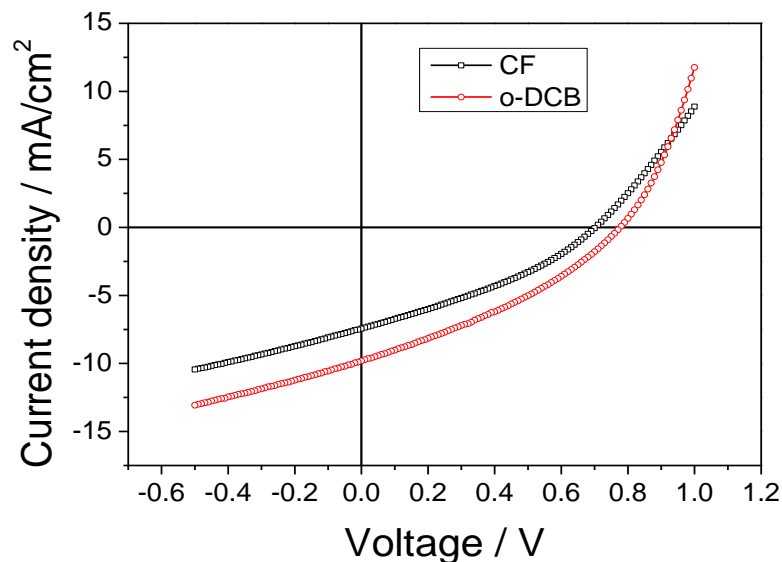
	1	2	3
λ_{max} (nm)	351	431	461
HOMO-LUMO gap (eV)	2.92	2.45	2.20
$E_{1\text{ox}}$ (V)	+0.39/+0.32	+0.27/+0.21	+0.26/+0.23
$E_{2\text{ox}}$ (V)	+0.86/+0.75	+0.54/+0.48 ^q	+0.66/+0.49 ^q
$E_{3\text{ox}}$ (V)	+1.13/+1.02 ^q	+0.76/+0.71 ^q	+0.97/+0.94 ^q
$E_{4\text{ox}}$ (V)	-	+0.97/+0.89 ^q	-
E_{red} (V)	-2.12 ^{irr}	-2.19 ^{irr}	-1.98 ^{irr}
HOMO (eV) ^a	-5.06	-4.96	-4.95
LUMO (eV) ^a	-2.92	-2.81	-3.00
HOMO-LUMO gap (eV) ^b	2.14	2.15	1.95



5T-TTF



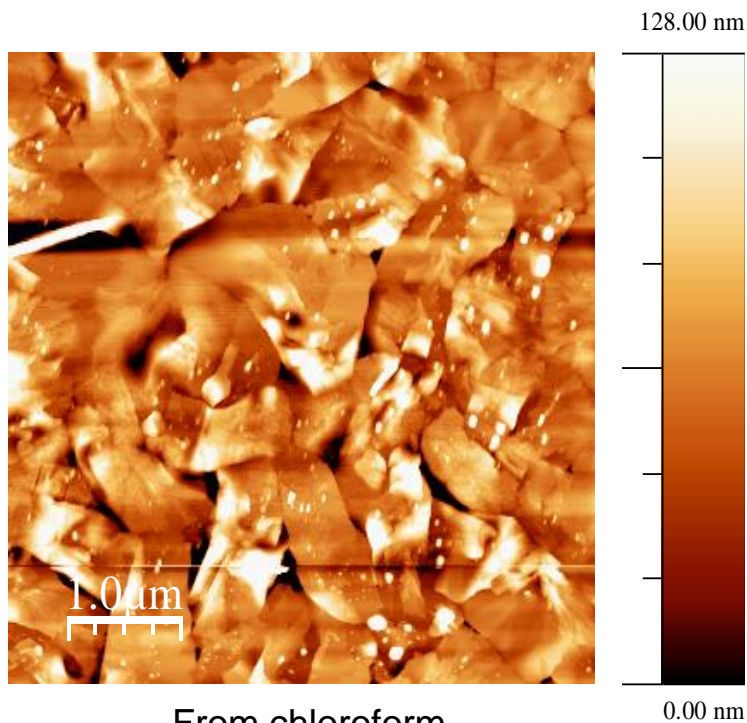
$E_g = 2.45 \text{ eV}$



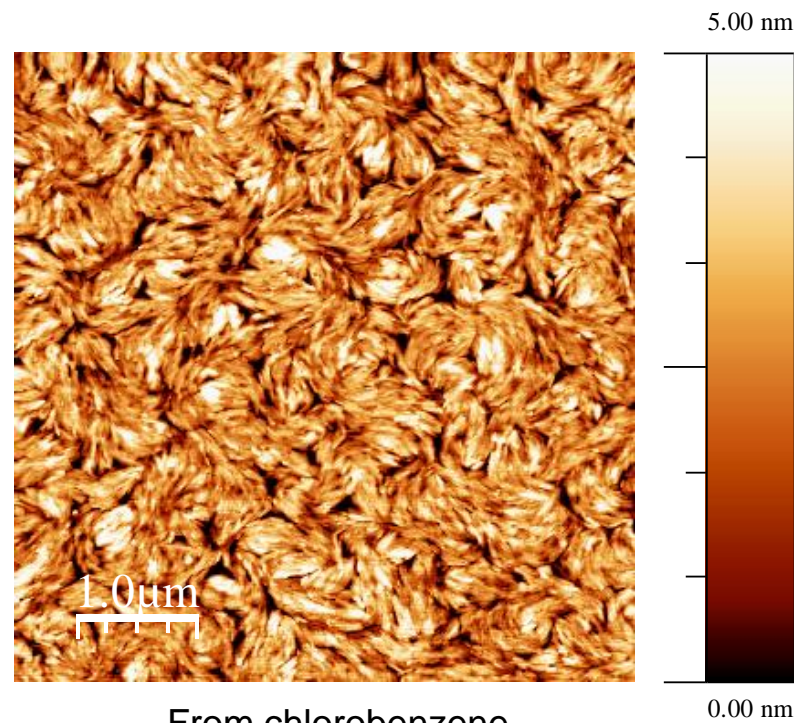
I/V curves of glass/ITO/PEDOT-PSS/**5T-TTF**: PC₇₁BM
(1:4)/Ca/Al configured photovoltaic cell

	I _{sc}	V _{oc}	FF	PCE (%)
From o-dichlorobenzene	9.81	0.78	0.33	2.5
From chloroform	7.44	0.70	0.33	1.7

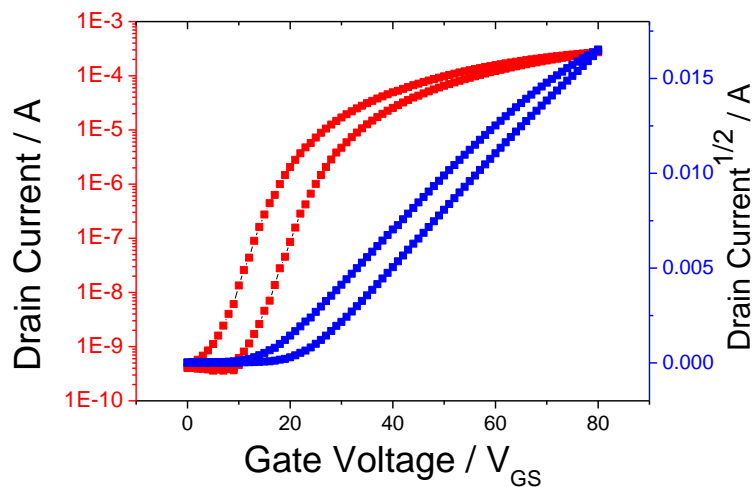
5T-TTF



From chloroform

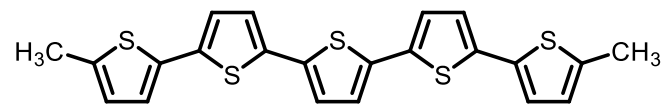


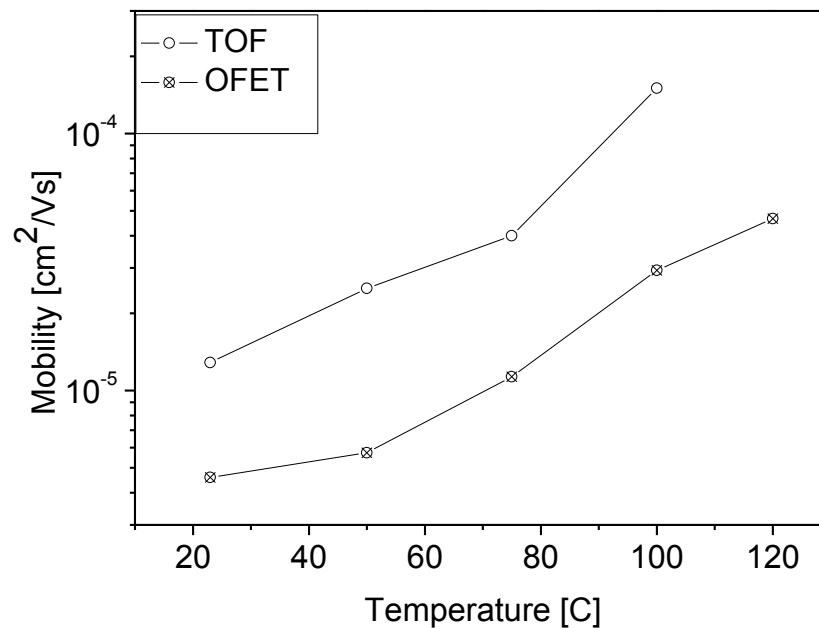
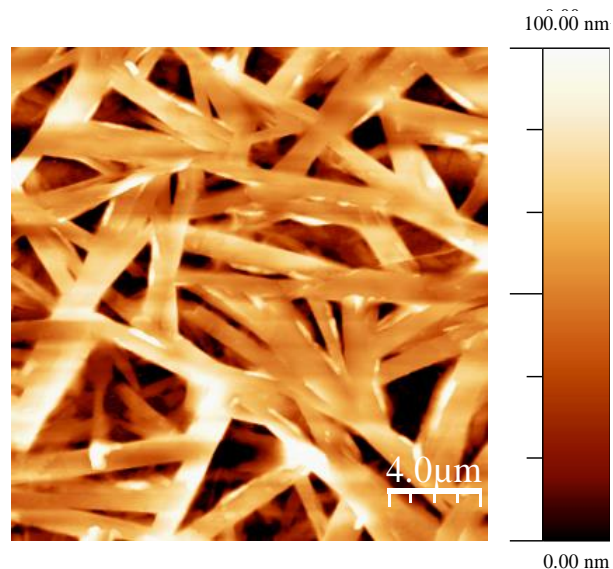
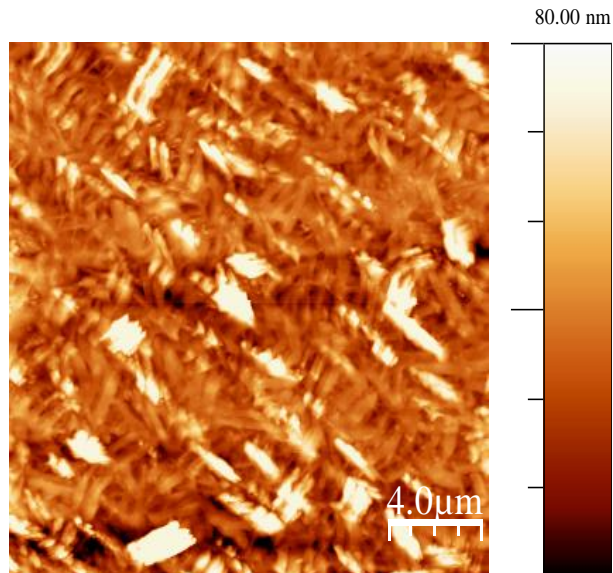
From chlorobenzene



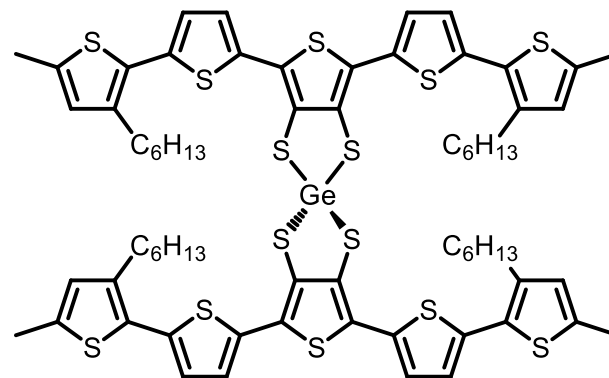
Hole mobility = $8.6 \times 10^{-3} \text{ cm}^2/\text{Vs}$ (OFET)
(TOF gave $1 \times 10^{-5} \text{ cm}^2/\text{Vs}$)

cf. hole mobility for end-capped
quinquithiophene = $9 \times 10^{-4} \text{ cm}^2/\text{Vs}$

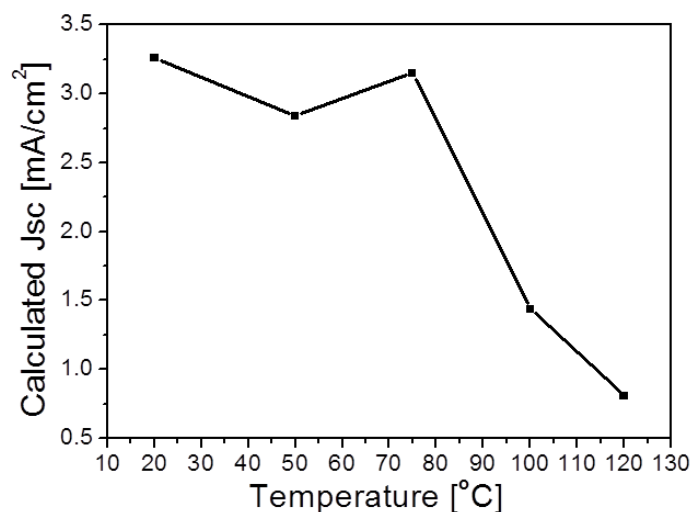
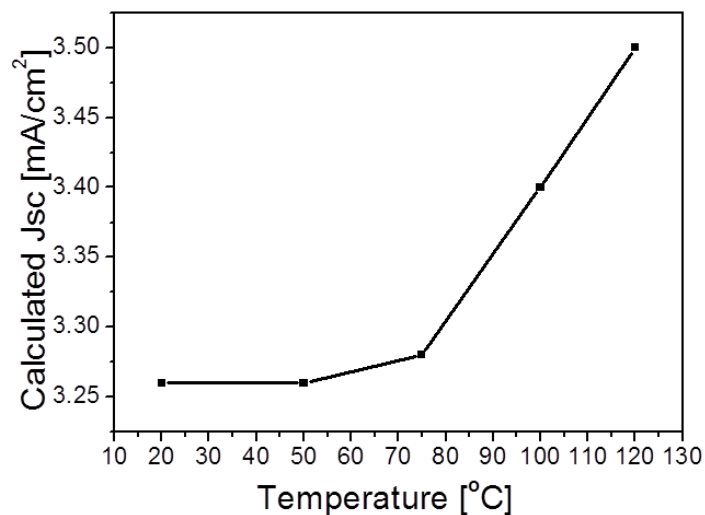




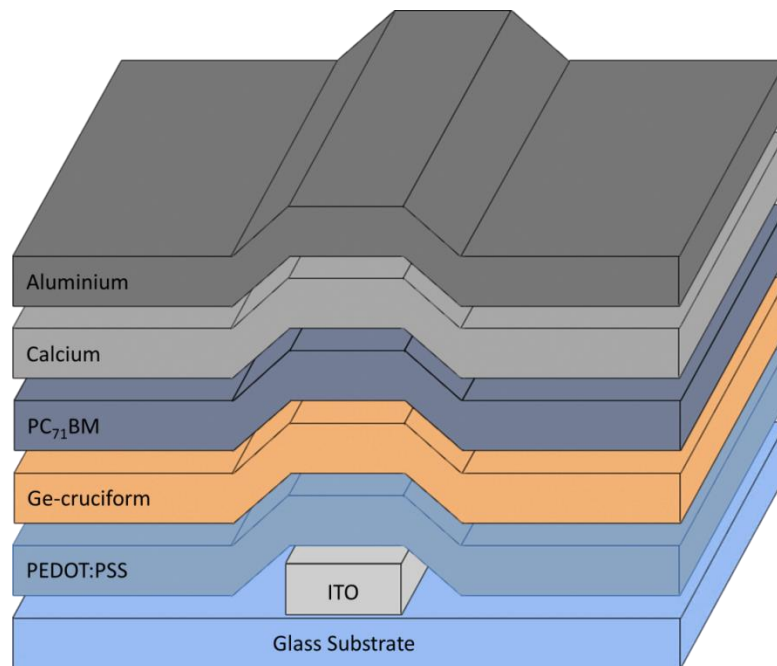
Temperature dependent mobility of step-annealed **Ge-cruciform** measured by time of flight and organic field effect transistor methods.



Tapping mode AFM images of **Ge-cruciform**: (top) annealed at 120 °C for 20 minutes after subsequent annealing at 50, 75, 100 for 20 minutes; (bottom) annealing straight to 120 °C for 20 minutes.



Short-circuit current calculated for a single device measured at RT, then step-annealed at 50, 75, 100 and 120 °C with measurements taken between each temperature step of 20 minutes (top) and several individual devices annealed straight to a given temperature for 20 minutes (bottom).



Device structure of a planar heterojunction used to determine the effect of annealing on short circuit current density.

