



Key Elements in Organic Synthesis: Recent Advances in p-Block Chemistry

Recent Advances in Sulfur Chemistry

4th December 2009

Stephen Hilton

Talk Overview

1) Sulfur containing Natural Products

Epidithiodiketopiperazines (ETPs)

2) Sulfur-Carbon Bond Formation

Formation of aryl-S bonds

Benzothiazoles/ Benzothiophenes

Sulfur Electrophiles with Alkynes

Sulfur Radical Chemistry

3) Sulfur Mediated C-C, Bond Formation

Pummerer and Pummerer-type reactions

Ramberg Bäcklund reaction

Sulfoxide directed *ortho*-lithiation

4) Chiral Sulfur Ligands and Catalysts

Ellman type sulfonylimines

5) Rearrangements

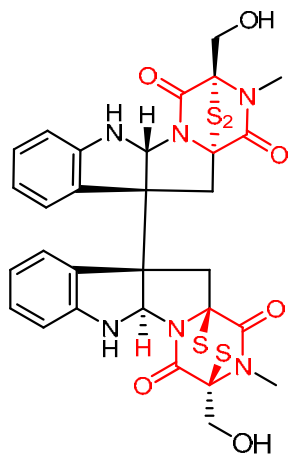


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Sulfur Natural Products

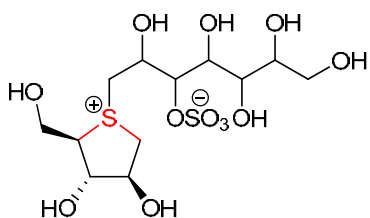


Sulfur Natural Products

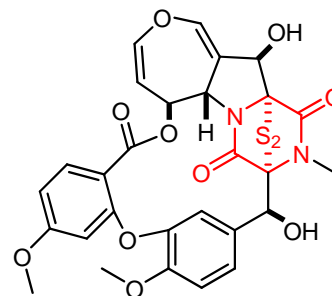


Chaetocin
anti-myeloma

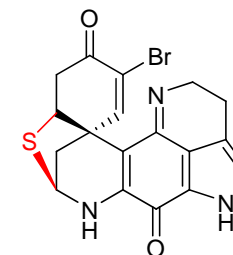
<2007Blood2579>



Kotolanol
Glycosidase Inhibitor
<2009JACS5621>

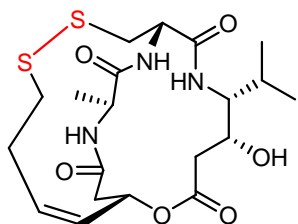


MPC1001
Anticancer
<2009JOC513>



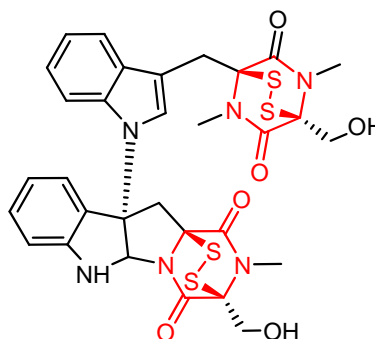
Discorhabin A
Cytotoxic

<2009OL4048>

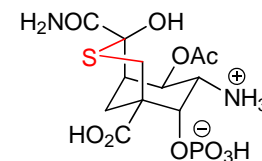


Spiruchostatin A
HDAC inhibitor

<2009TL2970>

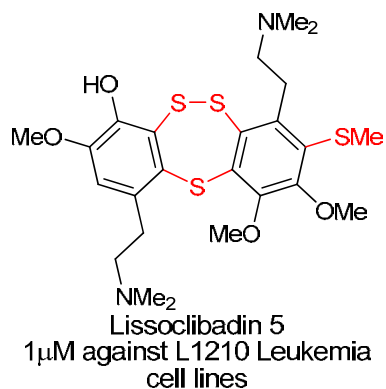


Chetomin
HIF-1 Anti-cancer
50-60 nM
<2004Cancer Cell 33>
<2009JBC26831>



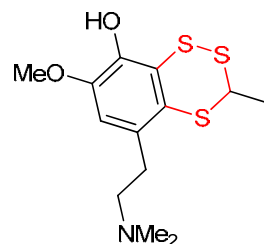
Tagetitoxin
Proposed Structure
Inhibitor of RNA polymerase

<2008OL5477>

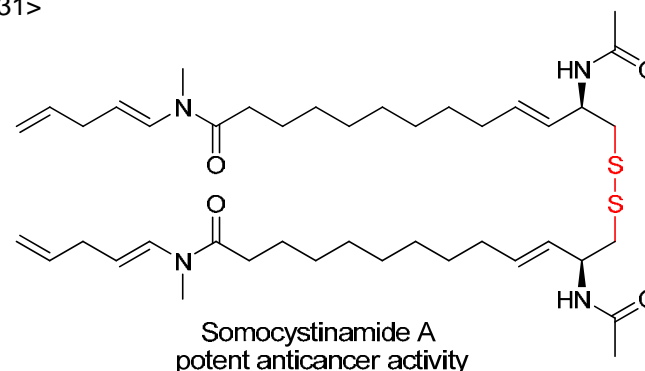


Lissoclibadin 5
1 μ M against L1210 Leukemia
cell lines

<2009T9598>



Lissoclibadin 13
2.2 μ M against L1210 Leukemia
cell lines



Somocystinamide A
potent anticancer activity

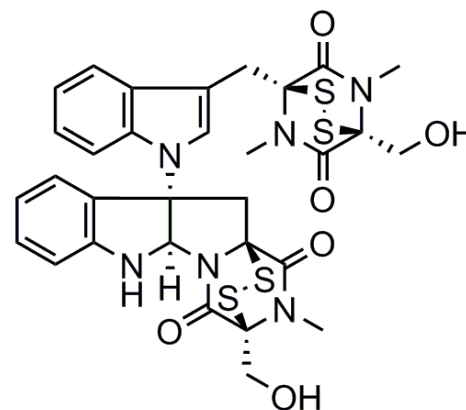
<2008OL4449>
<2009PNAS2313>



Epidithiodiketopiperazines

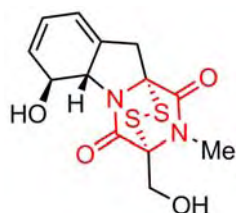
- 2004. Kung et al. identified chetomin from a screen of ~600,000 compounds as the only compound that showed inhibition of HIF-1 α /p300 binding

- Viable drug target
- Chetomin effective *in vitro* and *in vivo*,
 - ↓ HIF mediated gene expression
 - ↓ tumor size in mouse xenograft models



Properties

- Antitumor, antiviral and antibacterial activity.
- Inhibitor of transcription factor NF κ B,
- Inhibitors of farnesyl transferase and HIF.
- catalytic .



Gliotoxin

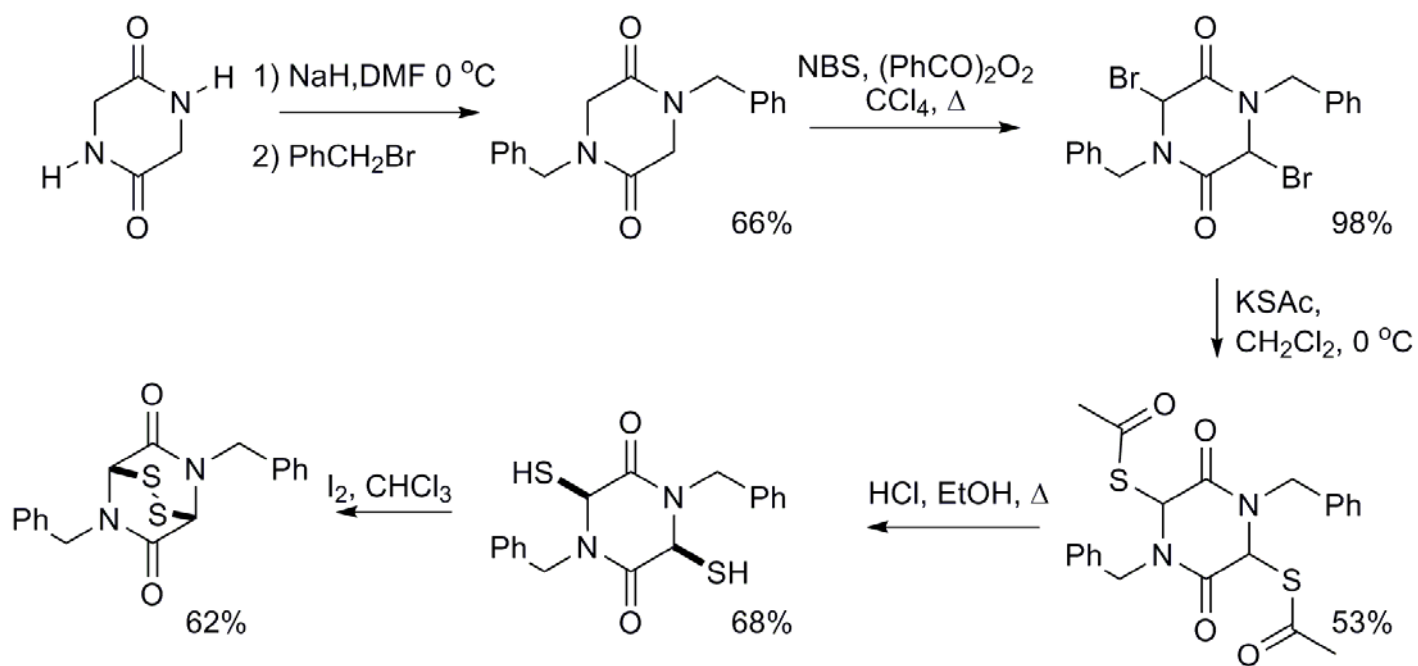
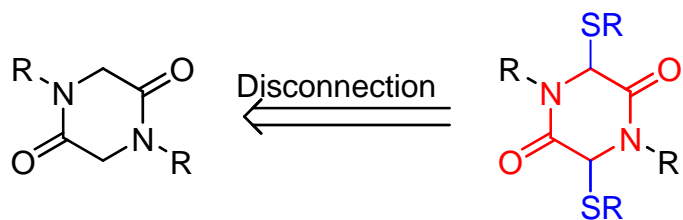
Recent mechanism of action studies of ETP core and Chetomin

Schofield <2009JBC26831>

Kung <2000Nature Medicine 1335> 5
Kung <2004 Cancer Cell 33>

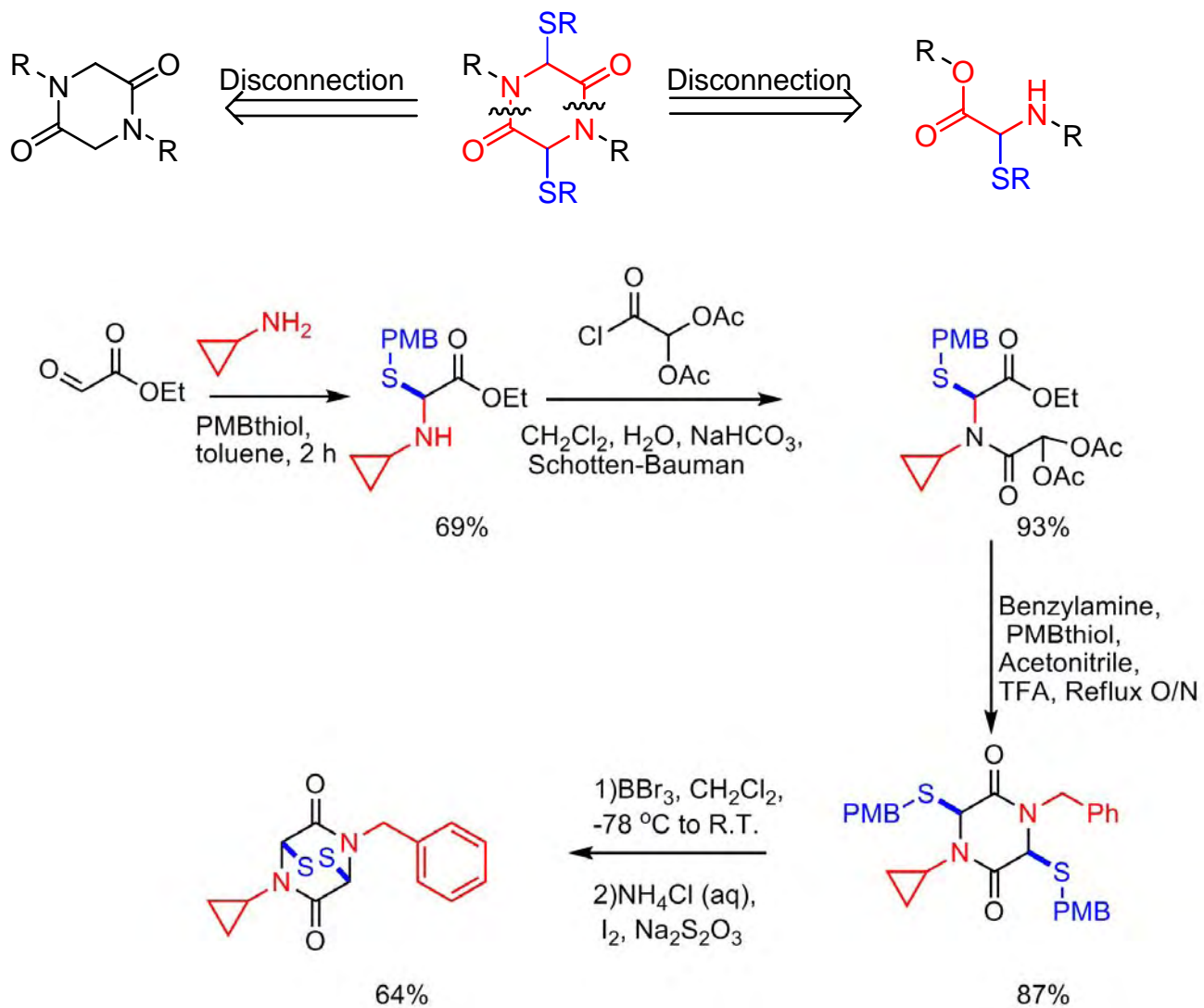


ETP Core Synthesis



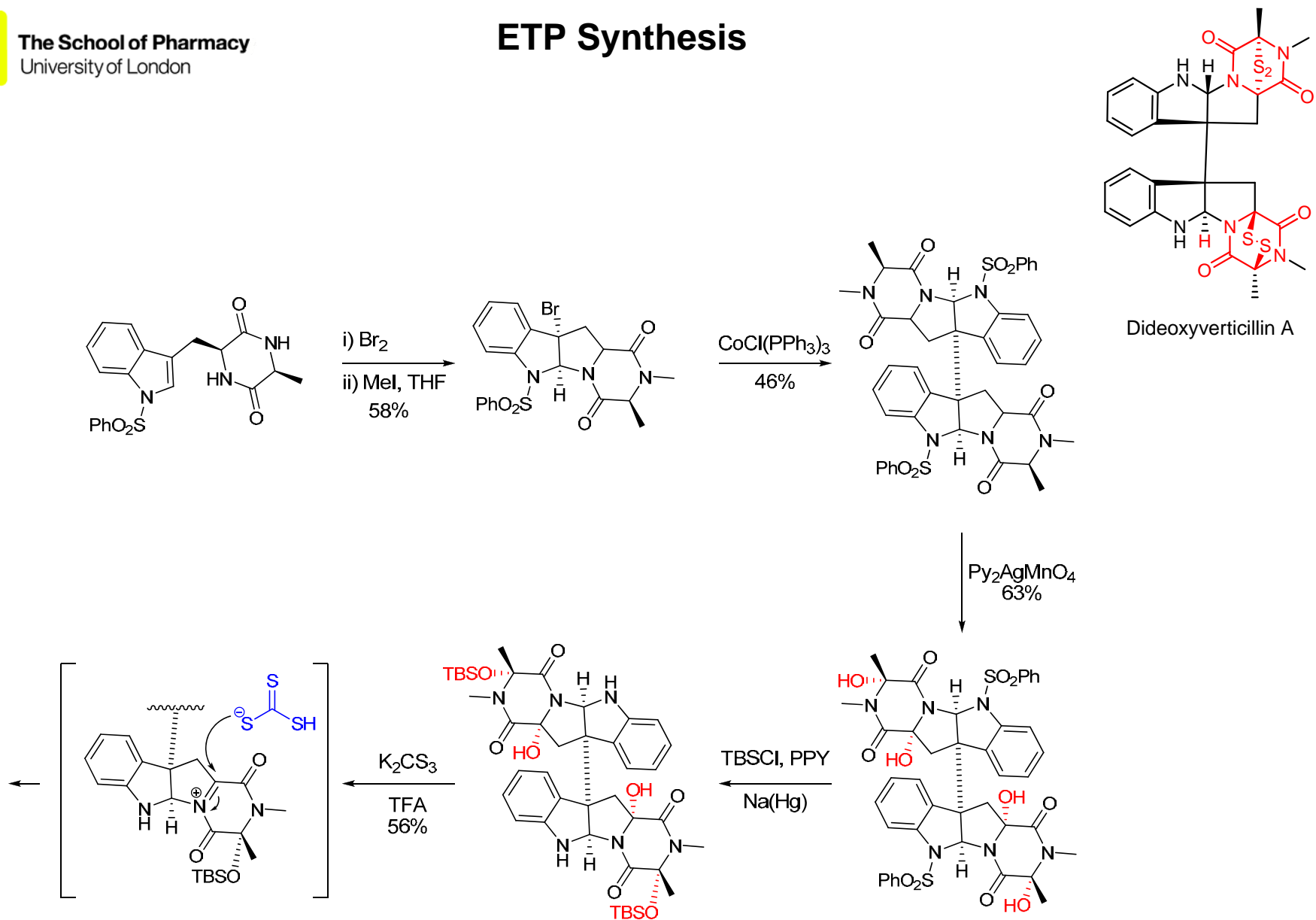


ETP Core Synthesis



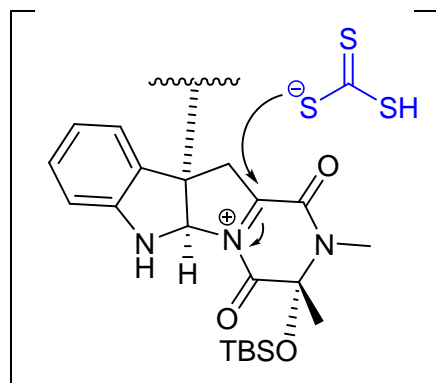


ETP Synthesis

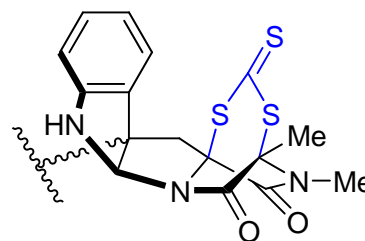




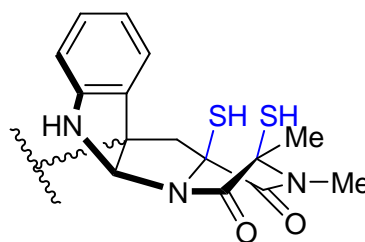
ETP Synthesis



TFA

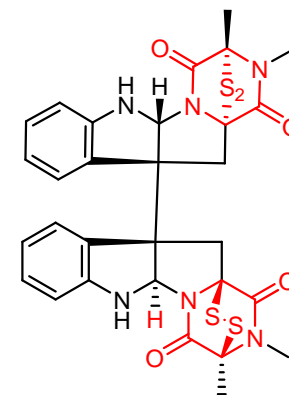
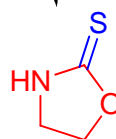
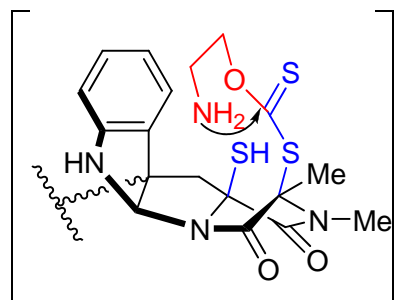
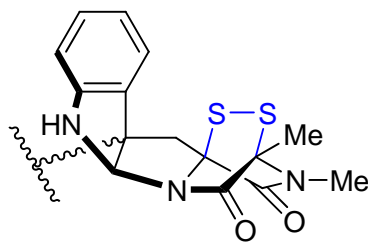


Ethanolamine



KI₃

62%



Dideoxyverticillin A



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Sulfur-Carbon Bond Formation

Formation of Aryl-S bonds

Increasing interest in formation of aryl C-S bond due to increasing prevalence in areas of therapeutic interest, such as Cancer, HIV and Alzheimers disease

Previous work on metal catalysed aryl C-S bond formation has been limited due to the view that metal catalysts would be deactivated by thio-compounds.

C-H activation has more recently been investigated for the formation of C-S bonds

For recent papers, see the following:

<2009OL1697>

<2008OL5147>

<2009JOC4005>

<2009JOC1663>

<2009JOC459>

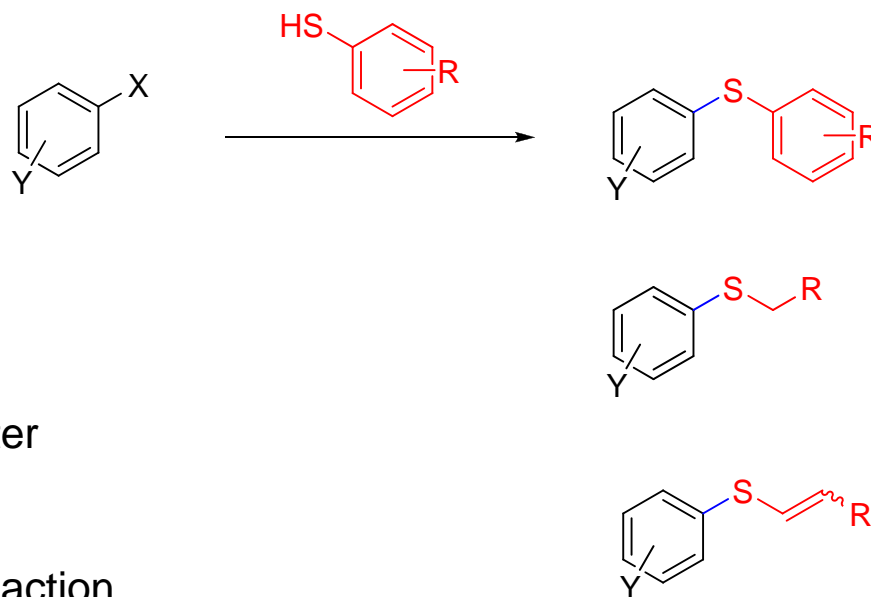
<2009Green Chemistry326> Reaction in water

<2009JOC4005>

<2009JACS7852>

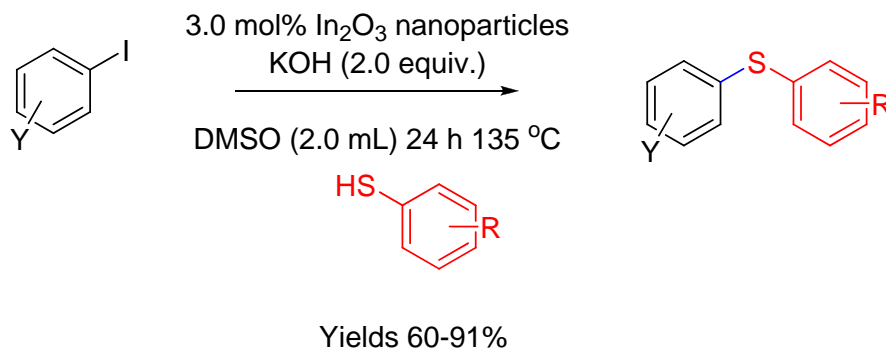
<2009TL3661> - Microwave Ullmann type reaction

Ma<2009OL – ASAP article> Aryl iodide, CuI, Sulfur powder



Formation of Aryl-S bonds

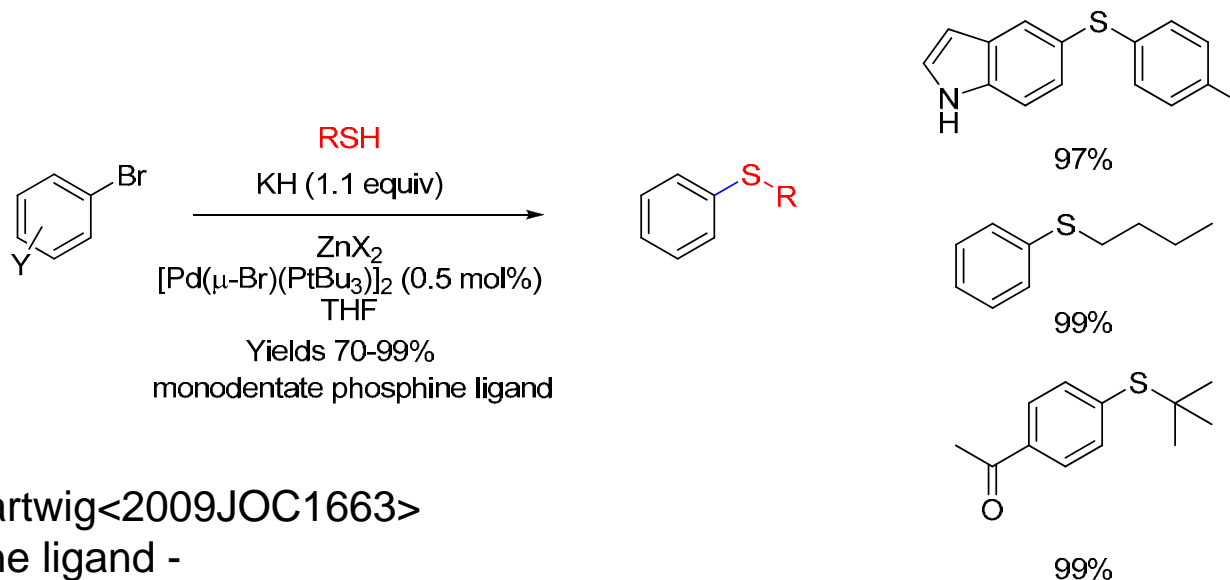
Cross-Coupling of Thiols with Aryl halides under Ligand free Conditions



Aryl thiols only

Rao<2009OL1697>

Zinc-Mediated Pd-catalysed C-S bond formation



See also Hartwig<2009JOC1663>

Bisphosphine ligand -

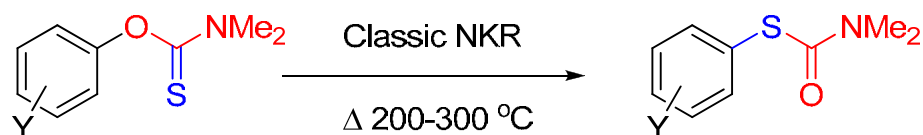
alkyl and aryl thiols tolerated. Yields >80%

Cat loading 0.01-0.5%

Stambuli<2009JOC4005>

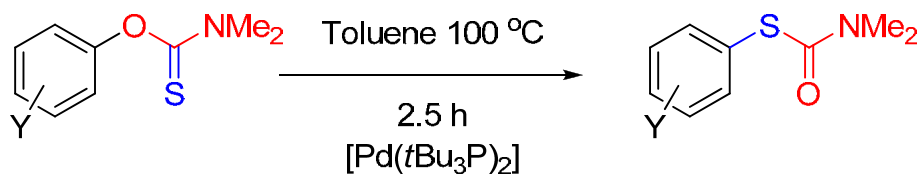
Formation of Aryl-S bonds

Newman-Kwart Rearrangement of O-Aryl Thiocarbamates



Kappe<2009EJOC1321>
<2009OPRD321>.

NKR in flow and microwave chemistry

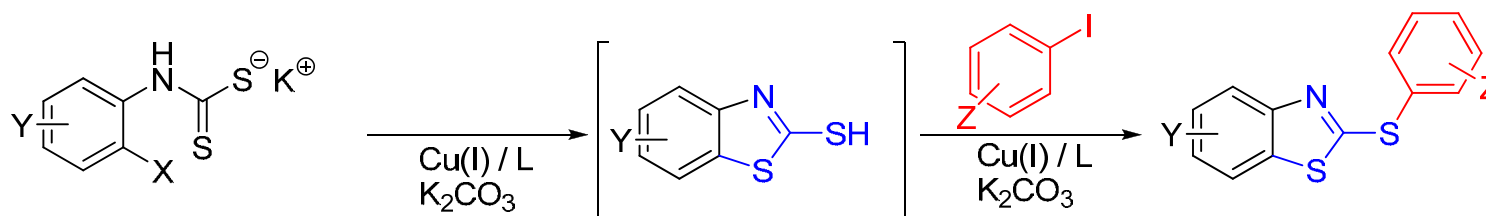
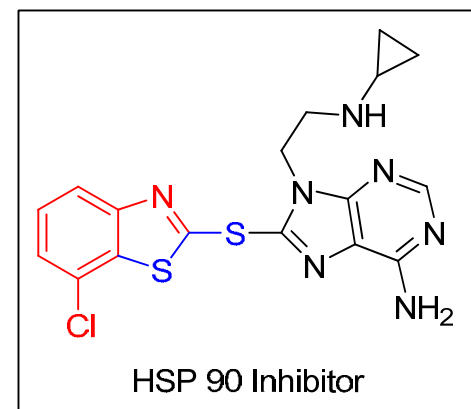
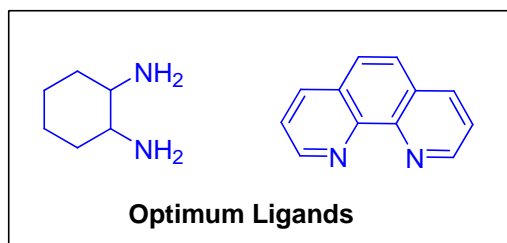


Proposed Pd-Sulfur complex

Entry	Ar	Time	Conv. (%)	Temp.
1	<i>p</i> -NO ₂ C ₆ H ₄	2.5 h	>99	180 °C
2	<i>p</i> -MeOC ₆ H ₄	14 h	92	>295 °C



Arylthiobenzothiazoles and Benzothiazoles

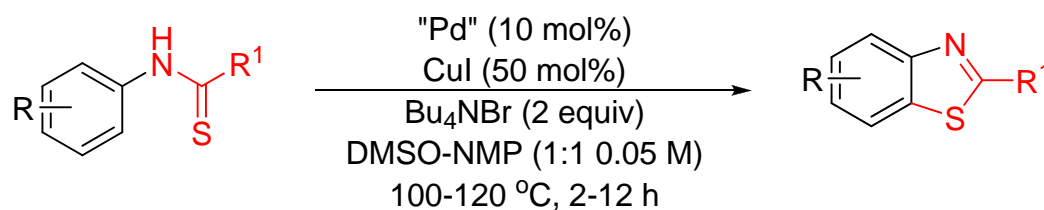


Optimum Ratio = dithiocarbamate:aryl iodide:CuI:ligand:base
1:1:0.05:0.1:1.3

Yields 56-90%

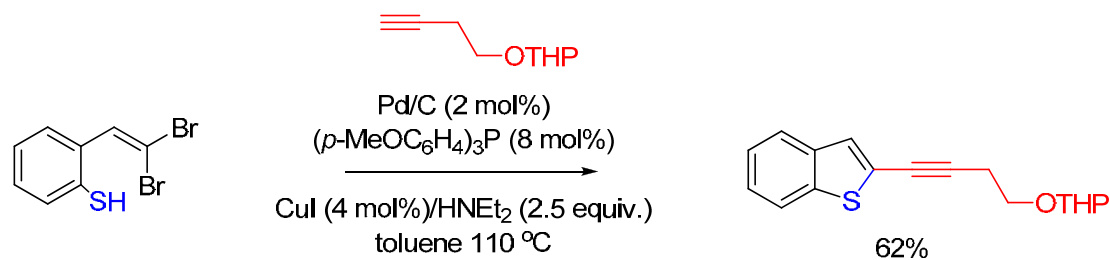
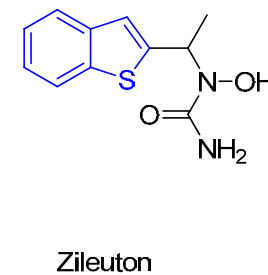
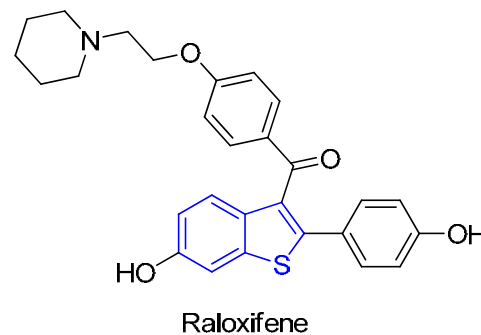
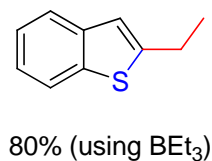
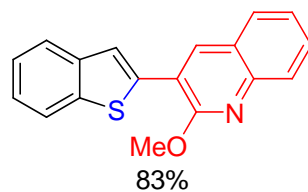
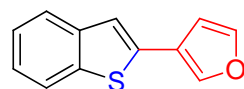
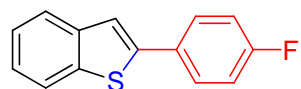
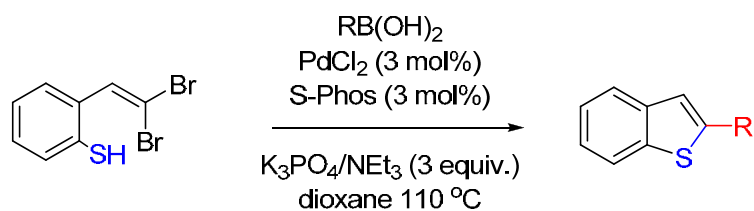
Patel<2009OL4254>

Benzothiazoles *via* C-H activation



Catalysts: PdCl₂, PdCl₂(cod)₂, PdBr₂

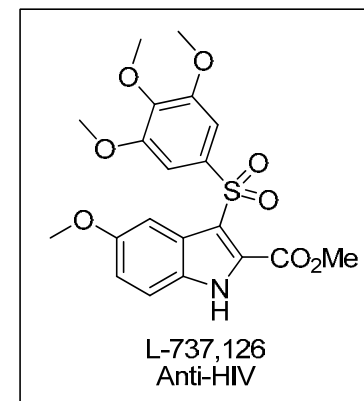
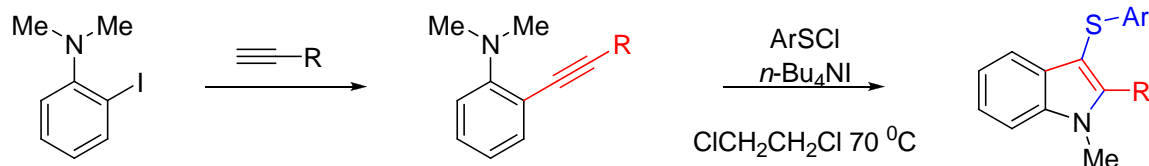
Benzothiophenes via Pd-Catalysed C-S Coupling



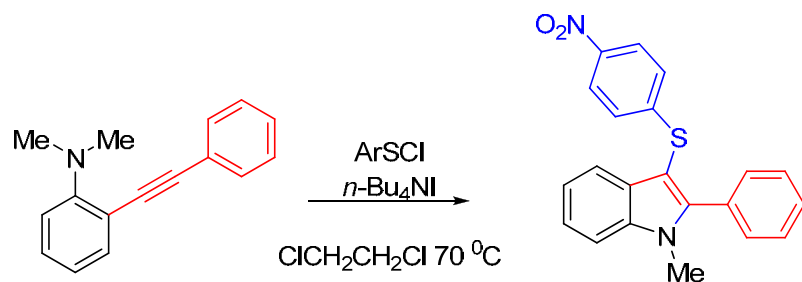
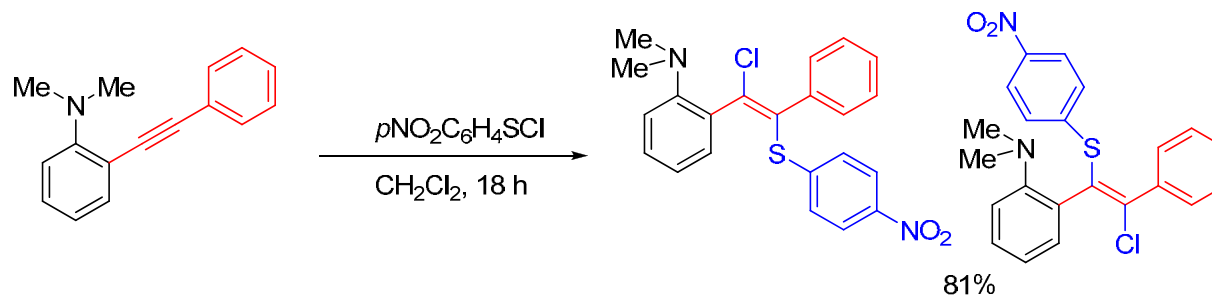
Note: reaction catalysed by Pd/C

Reaction of Sulfur Electrophiles With Alkynes

Synthesis of 3-sulfenylindoles



Reaction in absence of TBAI



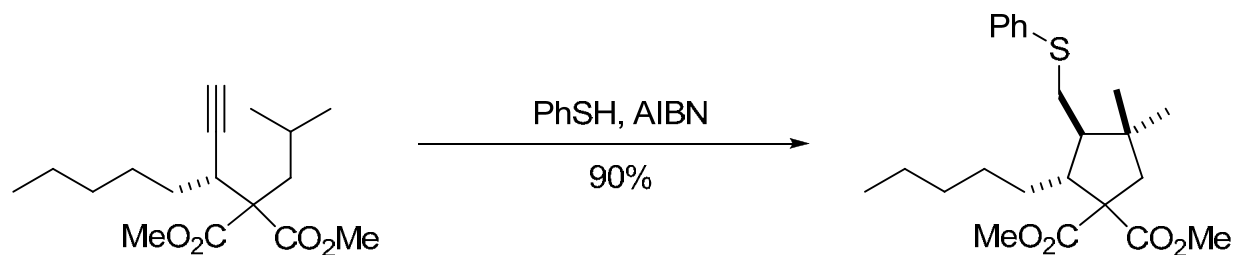
Entry	Additive	Solvent	°C	Time	Yield (%)
1	TBAI (1.0 eq.)	DCM	rt	60 h	86
2	TBAI (1.0 eq.)	1,2-DCE	70	5 h	90

For the synthesis of indoles and 3,1-benzoxazines, dihydrobenzofurans, see:
<2007T8250>

Thiol Mediated Radical Cyclisations

Reaction traditionally carried out with Bu_3SnH or derivatives
Drawbacks: toxicity, purification, cost.

Thiol mediated cyclisations avoid these problems



Renaud<2004OL2563>

For recent papers:

<2008T9799> Review – Majumdar

<2007OL1061>

<2007TL9124>

<2009TL228>

<2007OL4375> - Renaud

<2009OL2651>

<2007TL5265>

<2007TL7031>

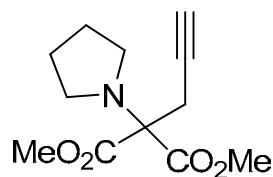
<2009OL3298> - Initiator Free

<2008CSR1603> Review N-centred radicals, Xanthate - Zard



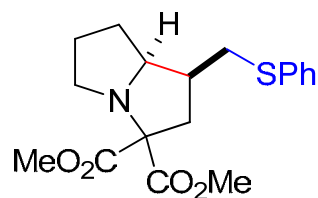
Thiol Mediated Radical Cyclisations

Thiophenol-mediated 1,5-hydrogen transfer for the synthesis of indolizidenes and related compounds

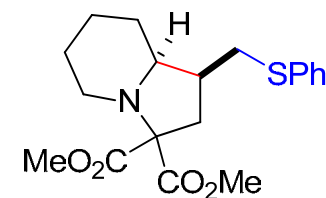


PhSH (2 equiv.)
AIBN (2 equiv.)

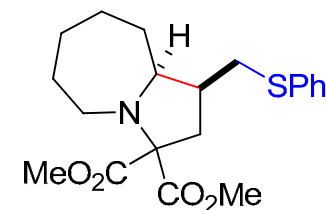
*t*BuOH, reflux



55%

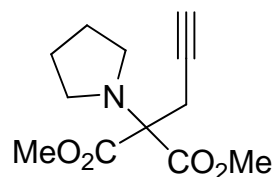


90%



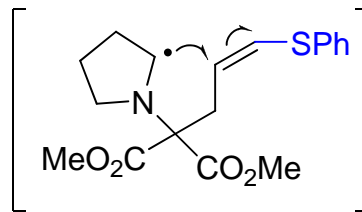
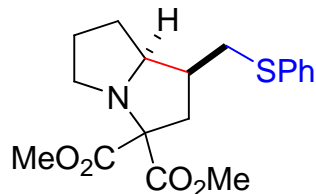
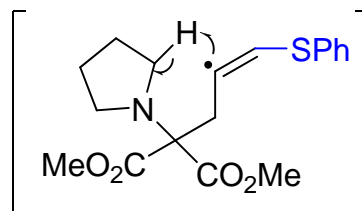
92%

Mechanism of cyclisation



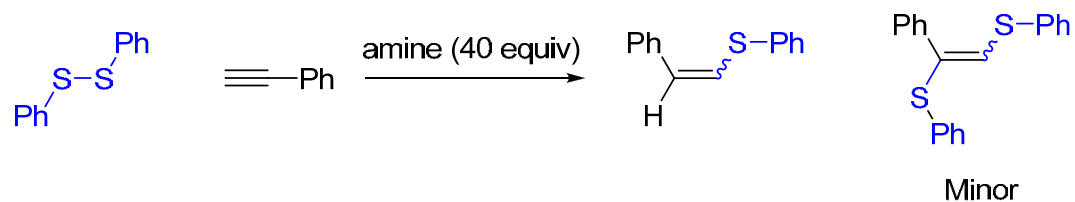
PhSH (2 equiv.)
AIBN (2 equiv.)

*t*BuOH, reflux



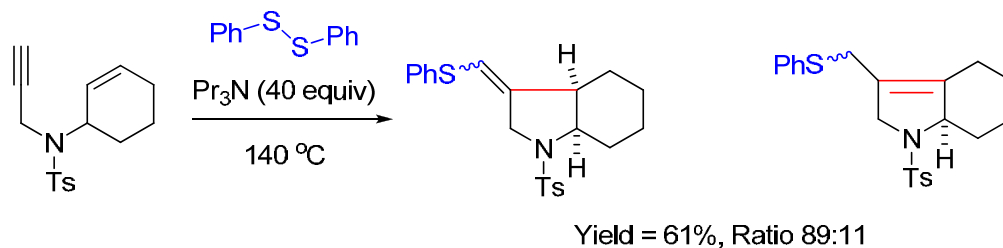
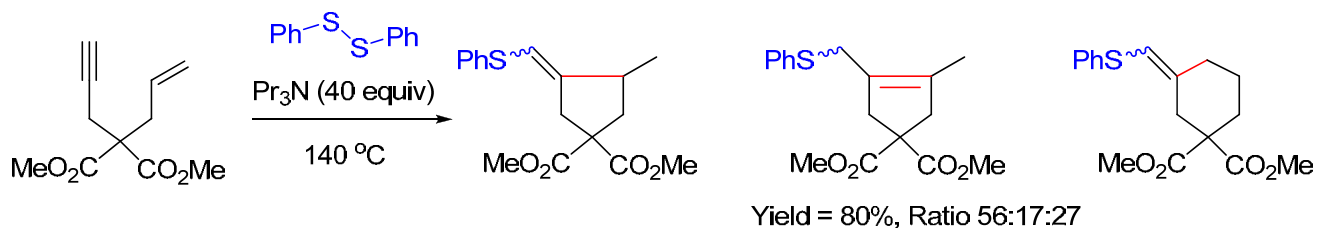
Thiol Mediated Radical Cyclisations

Amine-mediated single electron transfer

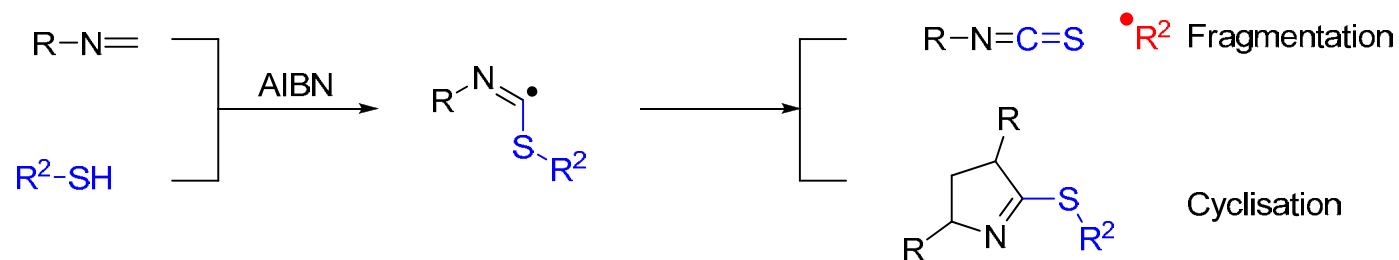


Entry	amine	Time	Temp.	Yield (%)
1	NEt ₃	8 h	90 °C	74
2	Pr ₃ N	3 h	140 °C	83
3	Pyridine	24 h	115 °C	No reaction
4	<i>i</i> PrNEt ₂	4.5 h	125 °C	69

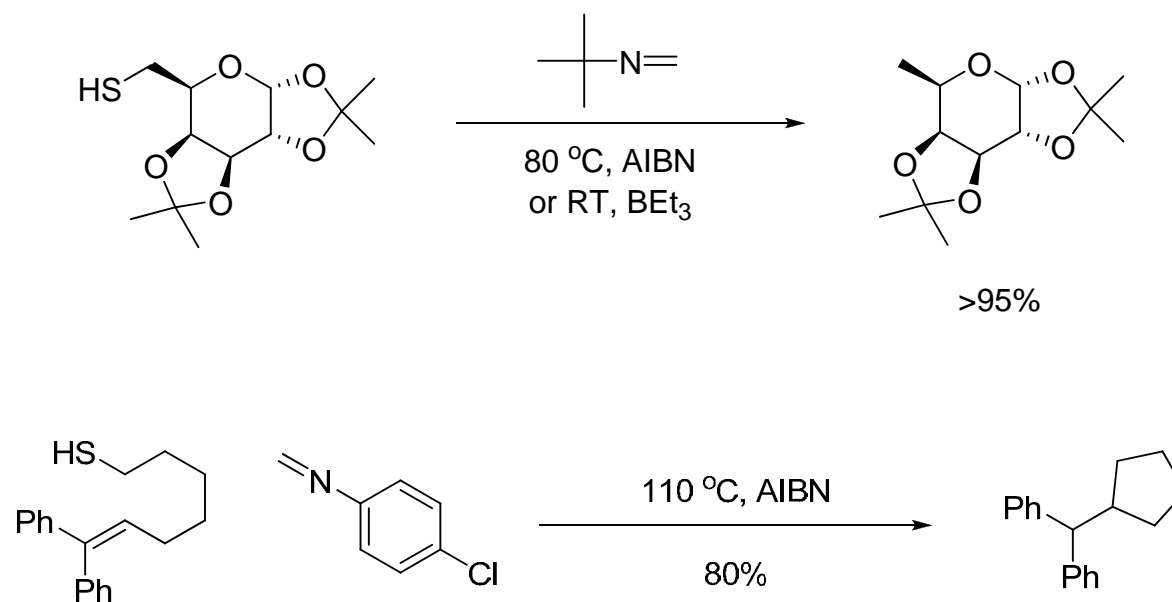
Addition of 2 equiv.
water increases yield



Thiol Radicals and Isocyanides

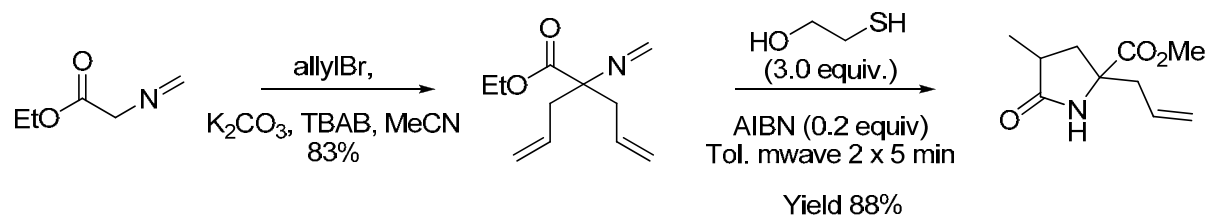
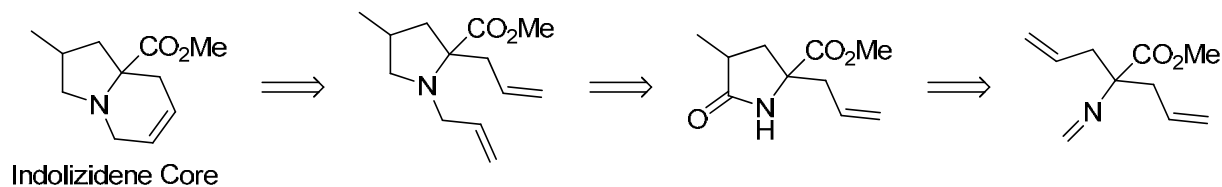


Barton-McCombie-type reaction

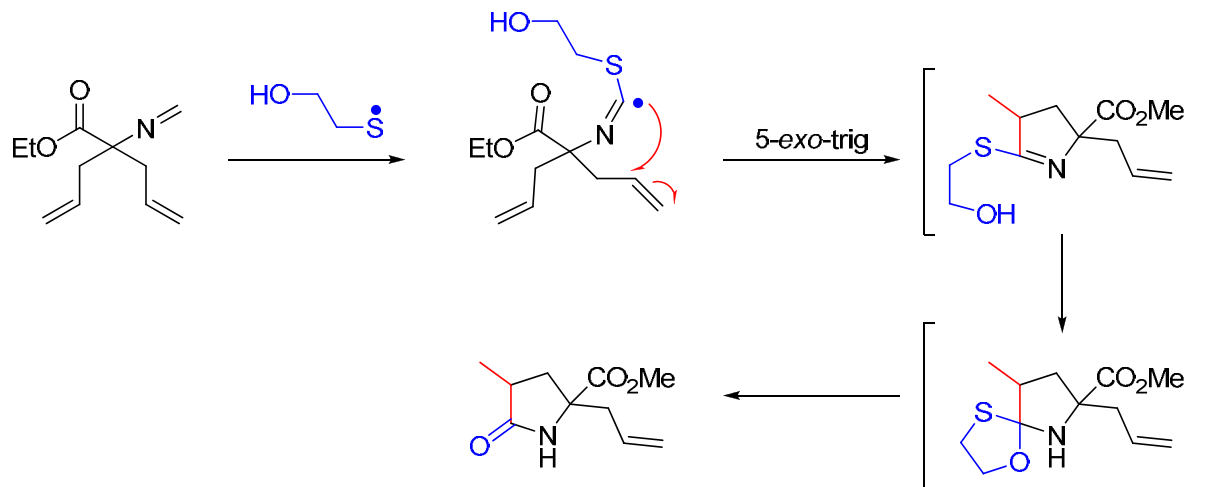




Thiol Radicals and Isocyanides



Can also be carried out on a solid support: <2004TL8541>



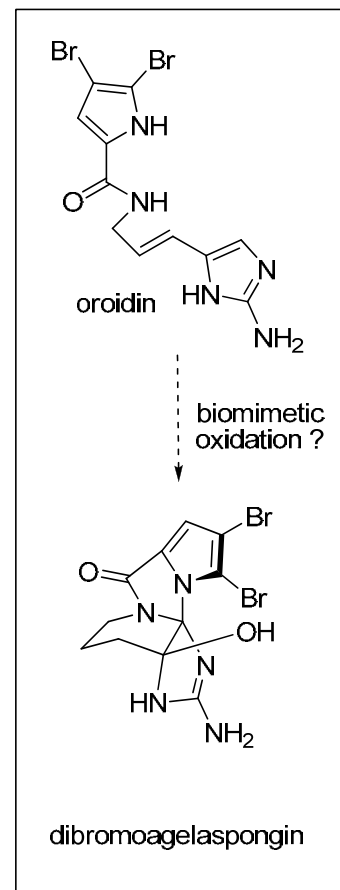
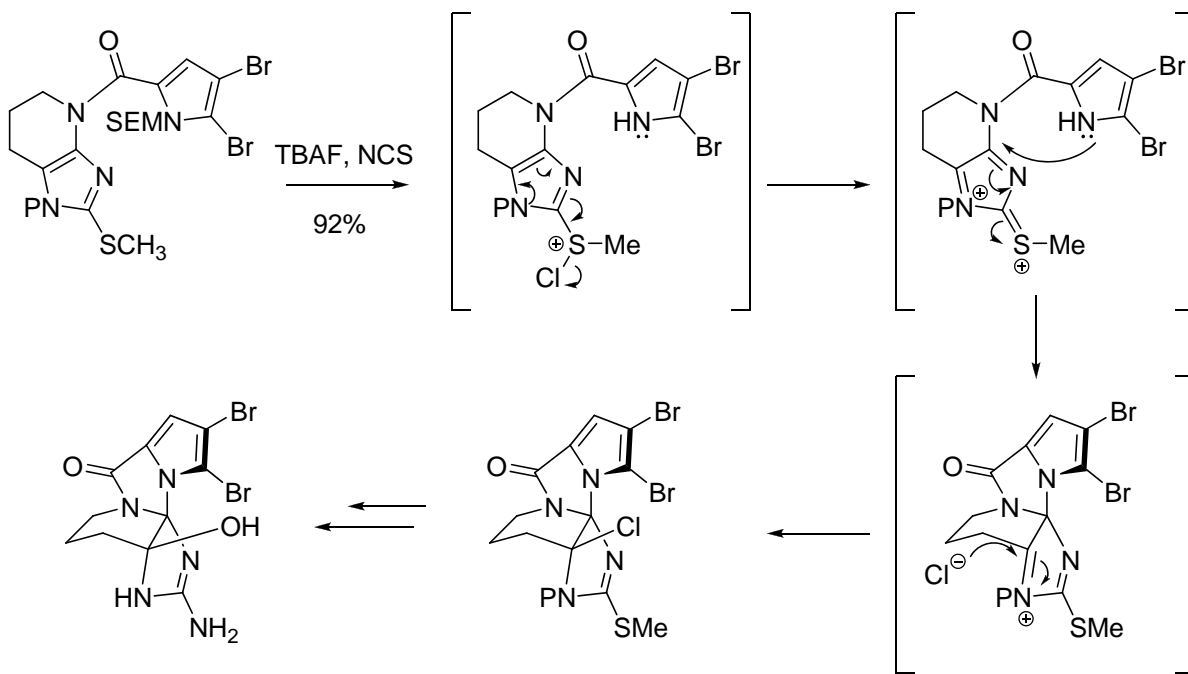
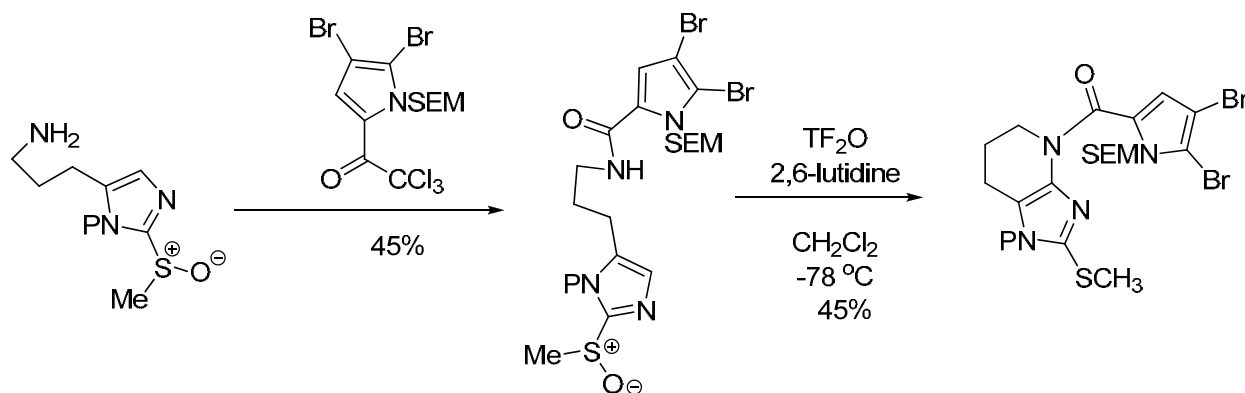


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Sulfur Mediated C-C bond Formation

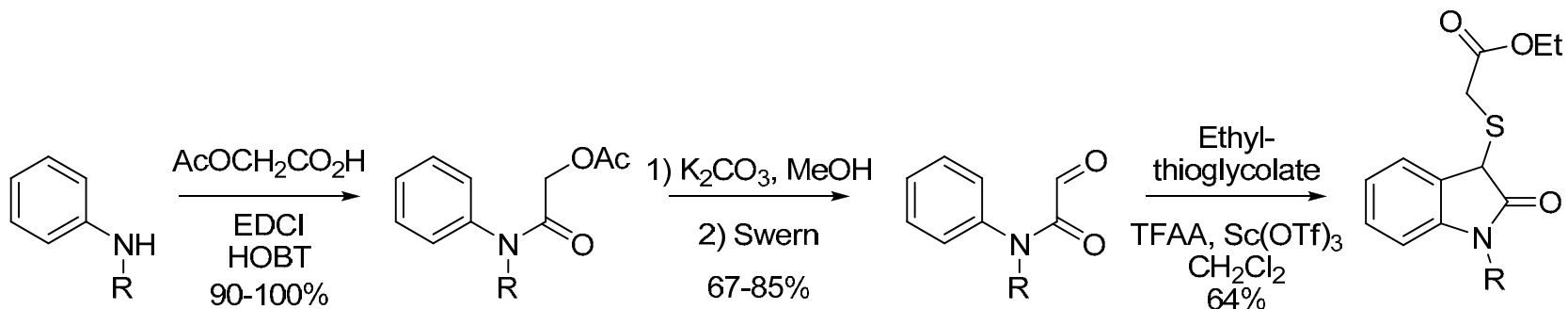


Reviews: Feldman<2006T5003> see also Pellissier<2006T1619>



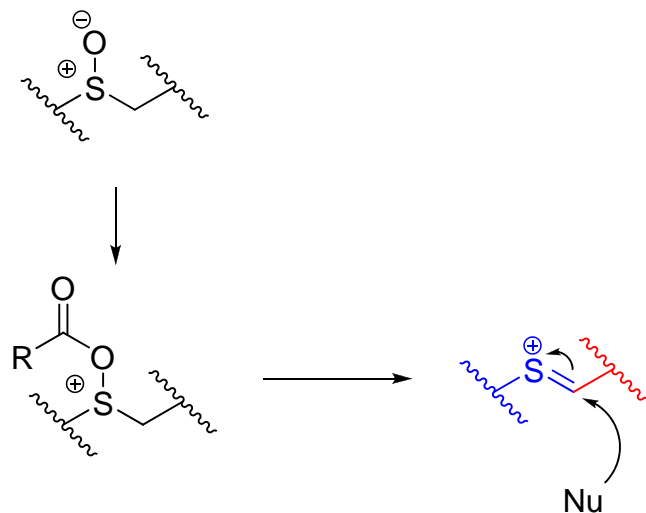


Pummerer and Pummerer-type Reactions

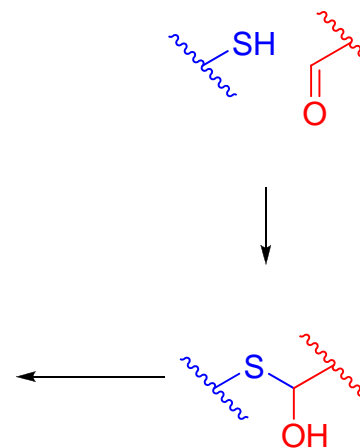


Reaction proceeds *via* a Pummerer-type reaction of the hemithioacetal

Pummerer Reaction



Connective Pummerer-type Reaction

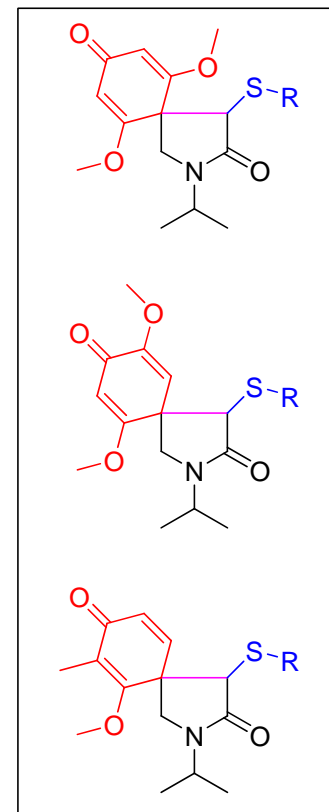
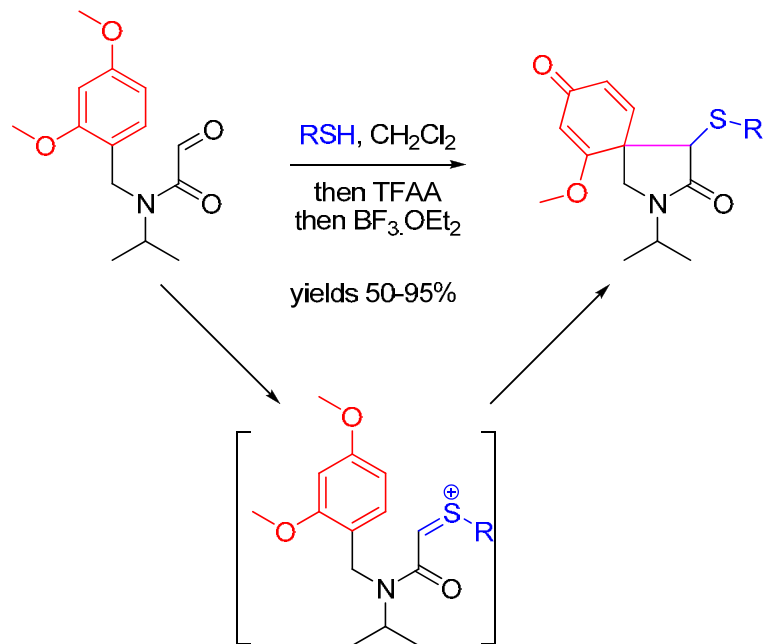




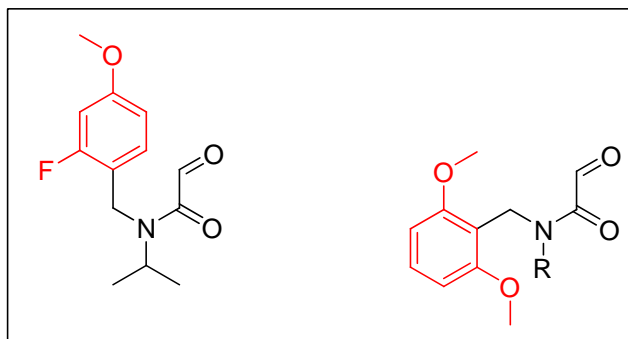
Pummerer and Pummerer-type Reactions

Synthesis of Azospirocyclic cyclohexadienones

Reaction proceeds *via* the same Pummerer-type reaction of the hemithioacetal



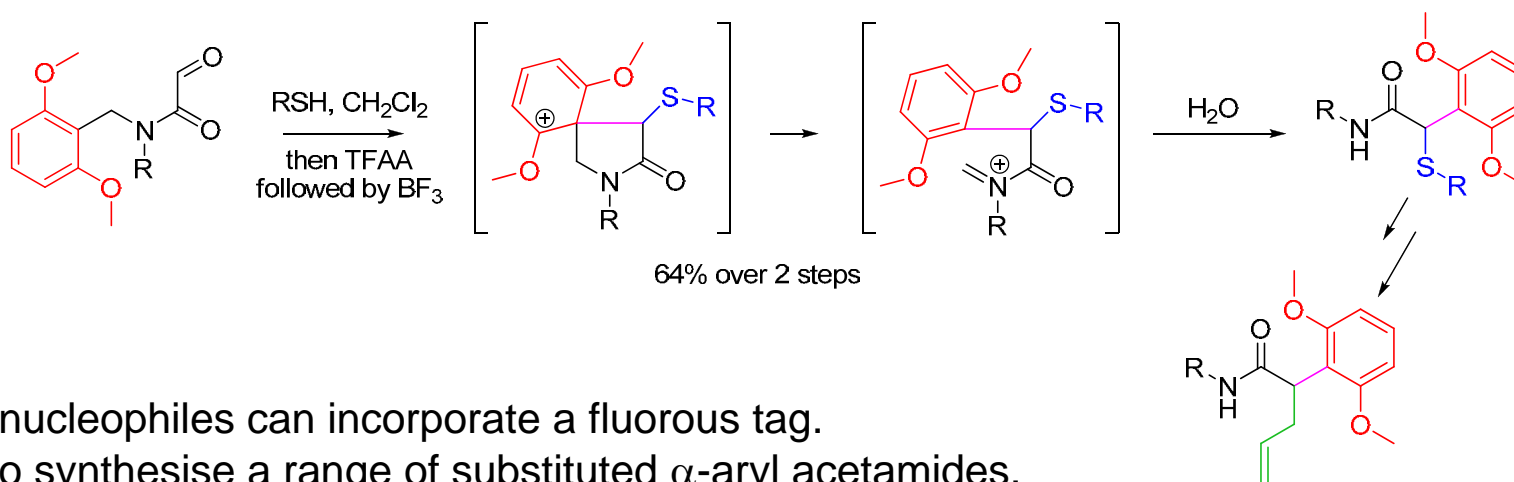
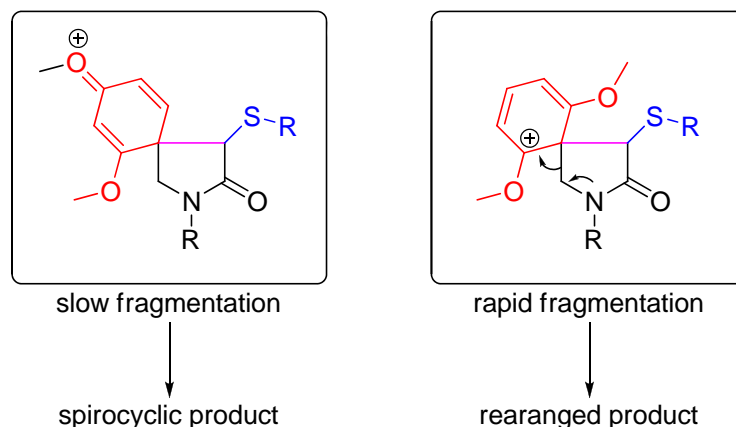
However, reaction does not work for all benzylic substrates:





Pummerer and Pummerer-type Reactions

Reaction proceeds *via* the same Pummerer-type reaction of the hemithioacetal.
However, the outcome of the reaction depends heavily on the intermediate:

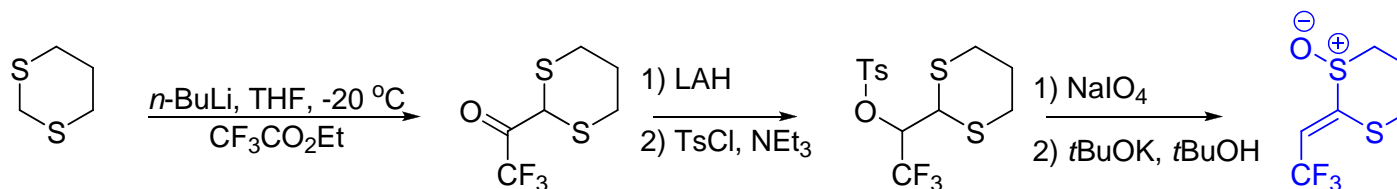


Sulfur nucleophiles can incorporate a fluororous tag.
Used to synthesise a range of substituted α -aryl acetamides.

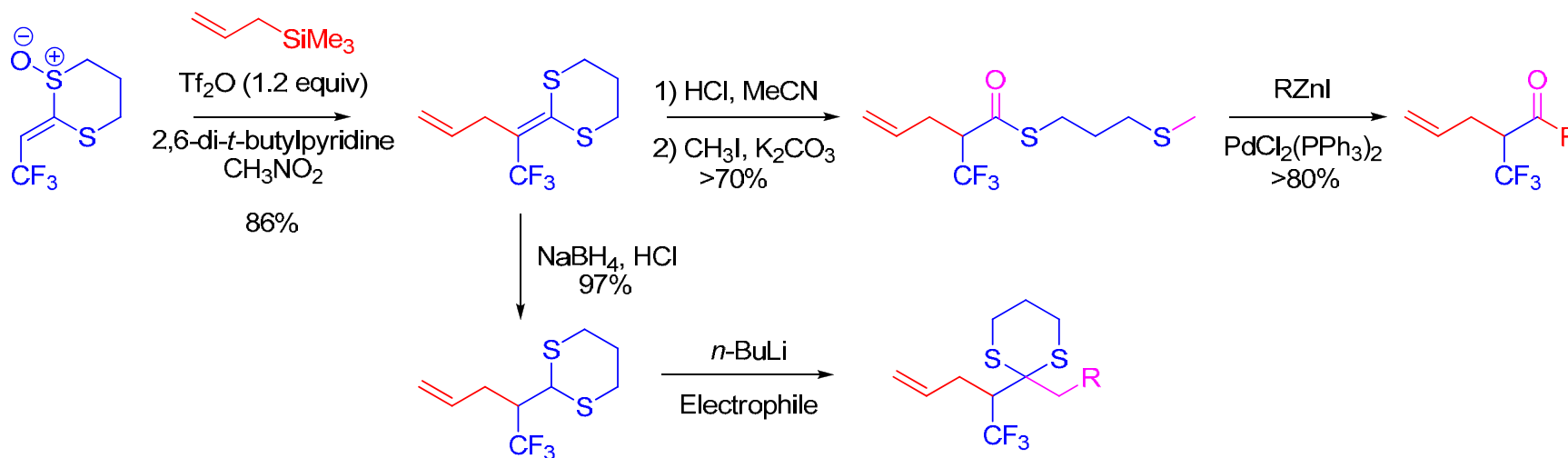


Pummerer and Pummerer-type reactions Trifluoromethylation

Dithianes as trifluoromethylketene equivalents:
provides for a facile way to introduce the CF₃ group

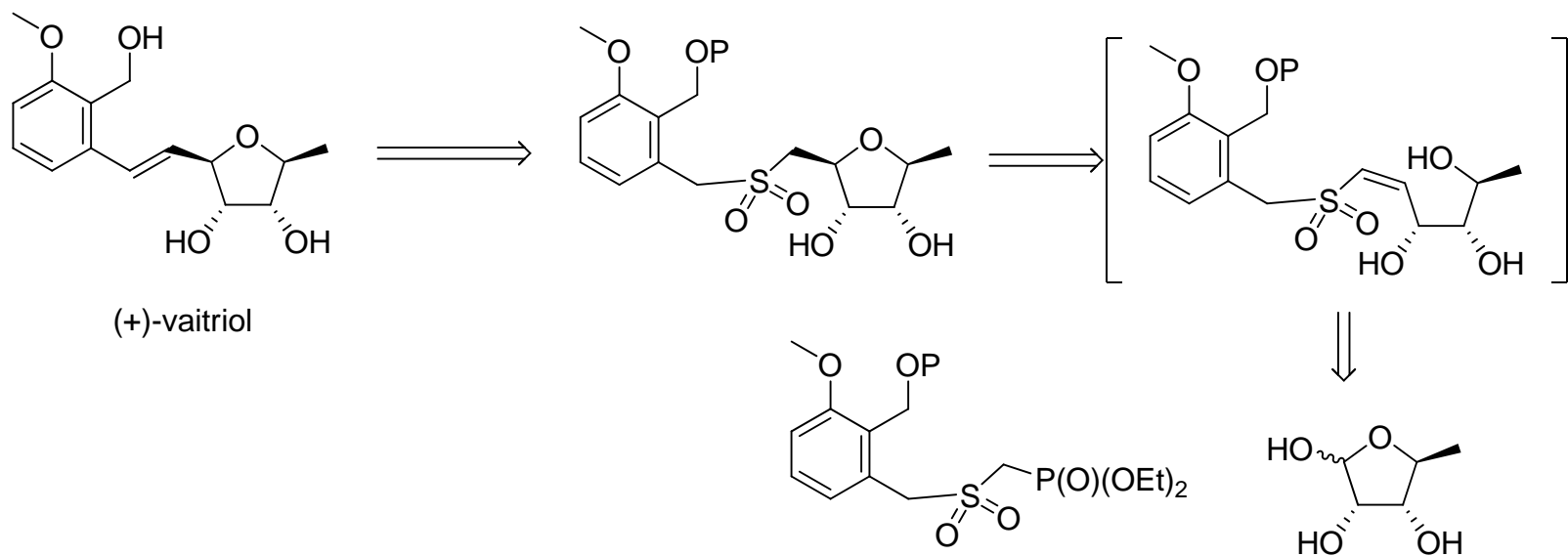


Products are versatile and make use of the dithiane

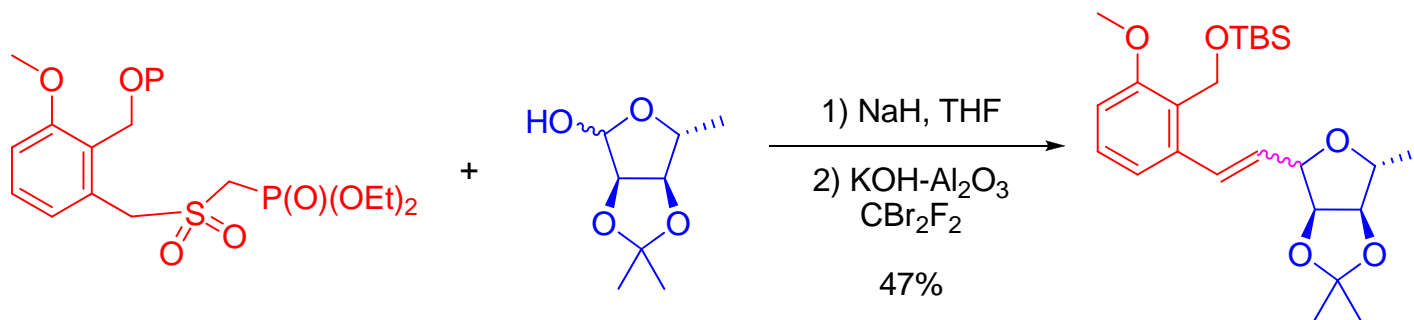




Ramberg-Bäcklund Reaction



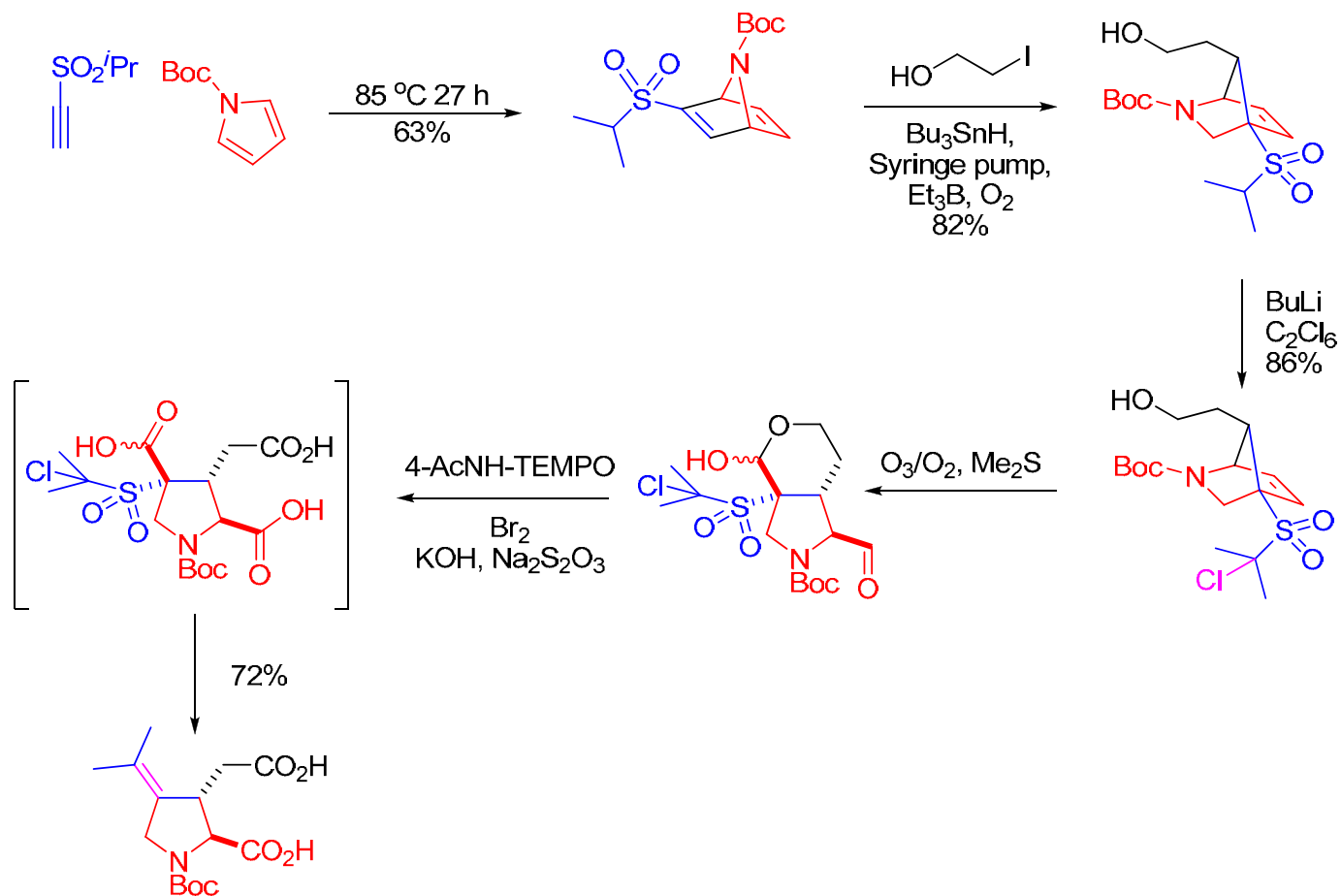
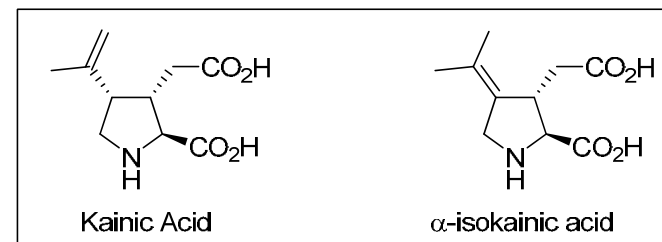
Utilising a domino HWE/conjugate addition/Ramberg-Bäcklund sequence



See also <2009JOC2271>

Ramberg-Bäcklund Reaction

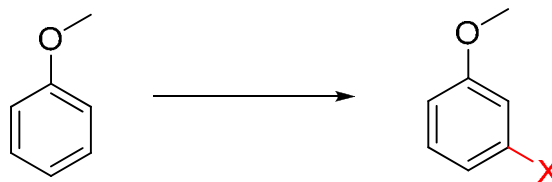
Decarboxylative Ramberg-Bäcklund sequence



Sulfoxide Directed *ortho*-Lithiation

Method for *meta*-substitution of arenes

Synthesis of substituted arenes and heterocycles using sulfoxides as directing groups for lithiation and also as removable auxiliaries

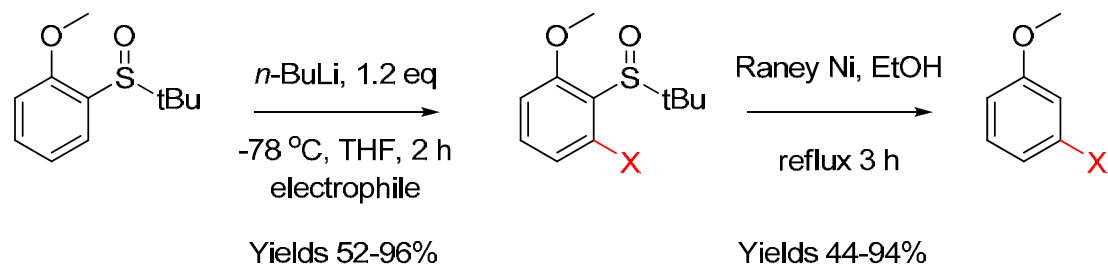


For directed *ortho*-metallation see also:
<2004ACIE888>, <2007JOC3199>

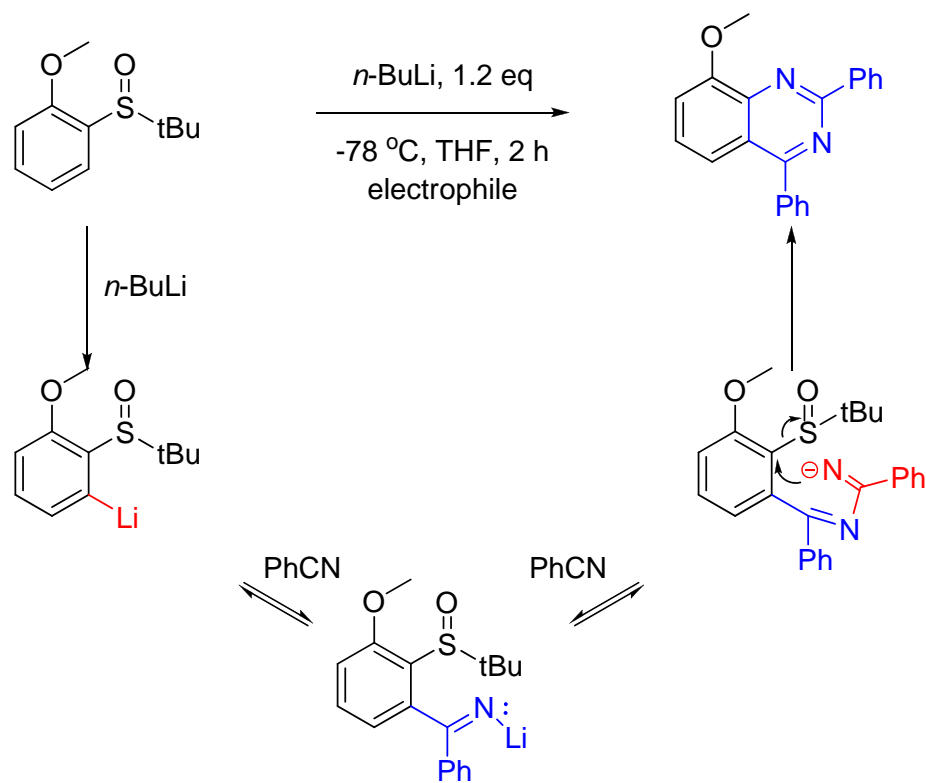
Brown<2008OBC1215>

Sulfoxide Directed *ortho*-Lithiation

Method for *meta*-substitution of arenes



Formation of quinazolines



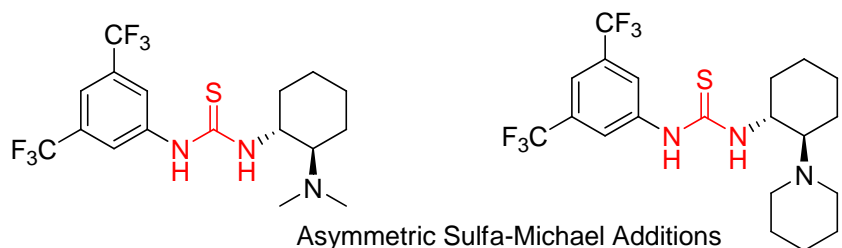


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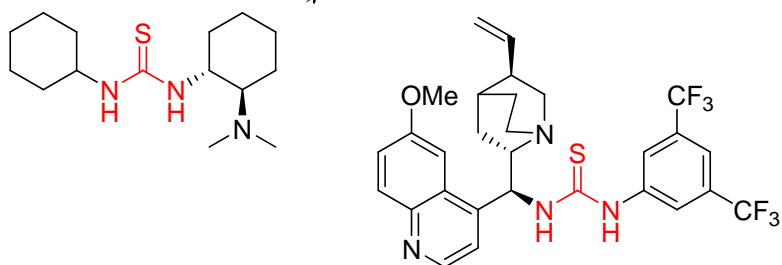
Chiral Sulfur Ligands and Catalysts

Chiral Sulfur Ligands and Catalysts

Reviews: Schulz<2007CR5133> see also Pellissier<2007T1297>,
Sone<2008Current Organic Synthesis305>
Aggarwal<2007CR5841>



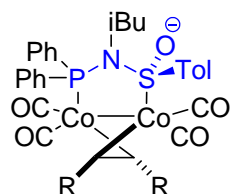
Asymmetric Sulfa-Michael Additions
to α,β -unsaturated sulfonates



Enders<2009EJOC1665>

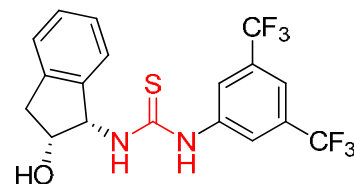
Dixon<2008OL1389>
Enantioselective Michael addition

Asymmetric Pauson-Khand Reaction



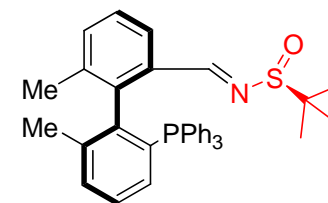
Verdaguer<2009OL4346>
<2007ACIE5020>

Enantioselective addition
of indoles to nitroalkenes



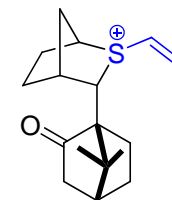
Seidel<2008JACS16464>

Ligand in Palladium-catalysed
addition to *N*-benzylisatin



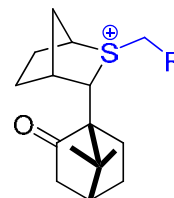
Qin<2009JOC283>

Epoxy-Annulations
of α -amido ketones



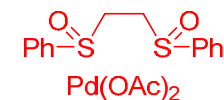
Aggarwal<2008OL1501>

Epoxidation/Cyclopropanation
Aziridination
Reaction with Boranes <2007JACS14632>



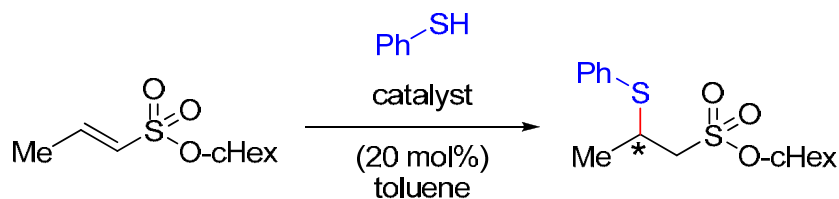
Aggarwal<2008OBC1185>
<2008CC120>, <2006CEJ568>
<2006T11297>
Substituted pyrrolidines /
piperidines <2006CC2156>

Intramolecular
C-H oxidation



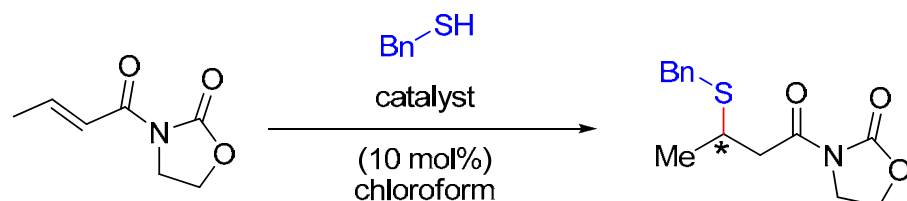
White<2009NC547>

Chiral Sulfur Ligands and Catalysts



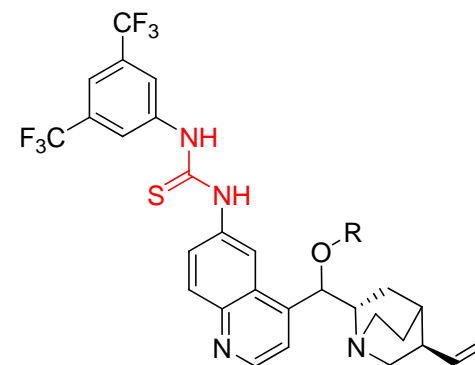
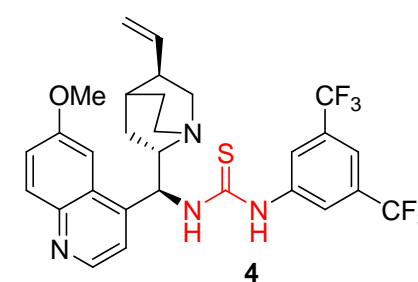
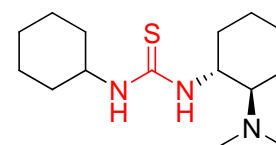
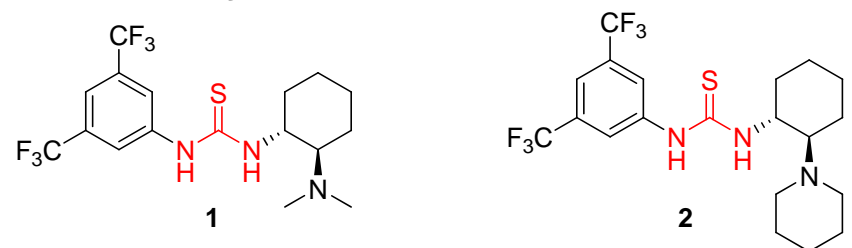
Enders<2009EJOC1665>

Catalyst 4 (Takemoto<2003JACS12675>)
yields 40-94% ee's of 31-64%



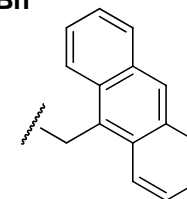
Entry	Catalyst	Time	Conv. (%)	ee (%)
1	4	24 h	95	45
2	5	5 h	>99	68
3	6	5 h	>99	85

Yields 84-99%, ee's 85-96%



5, R = Bn

6, R =

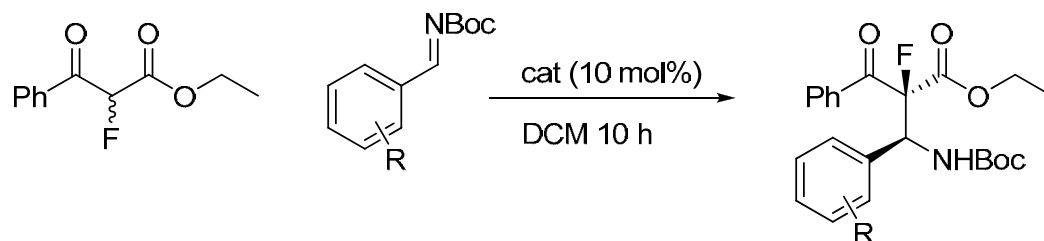


34

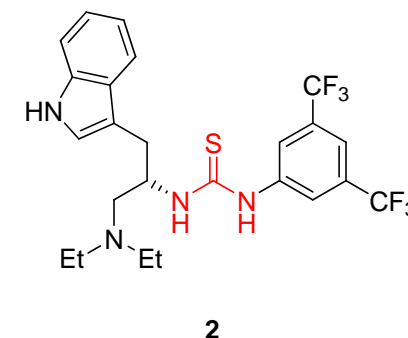
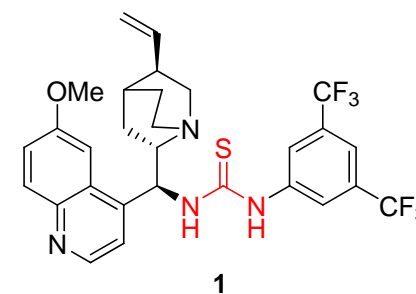
Deng<2009JACS418>



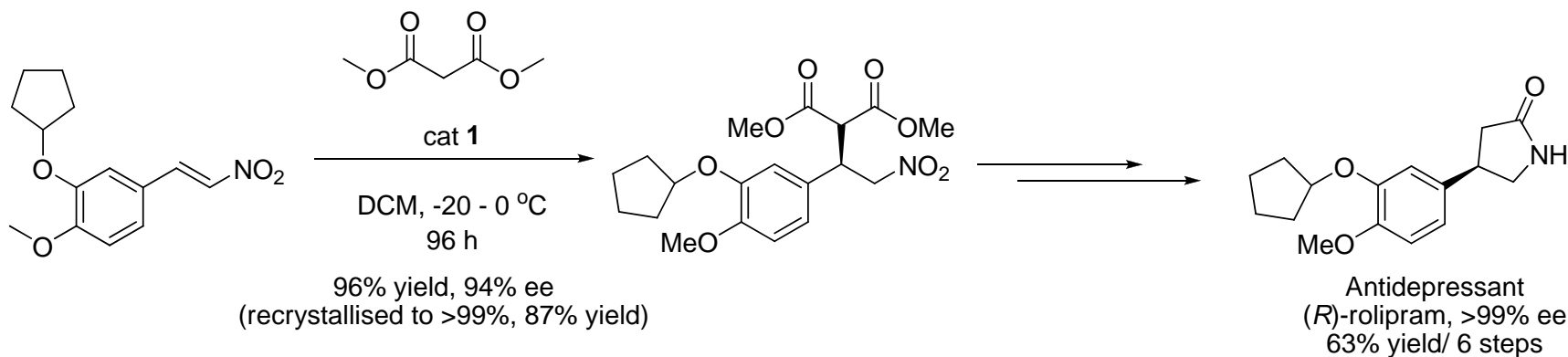
Chiral Sulfur Ligands and Catalysts



Catalyst 2 gave best ee's and yield
Yields 70-96%
ee's 81-97%

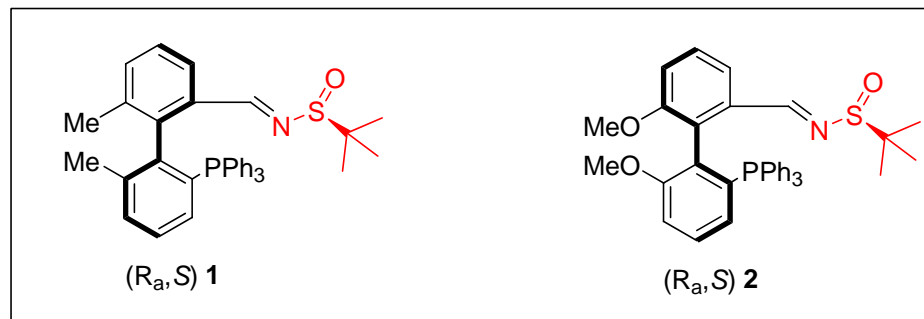


Lu<2009ACIE7604>



See also Ellman<2009JACS8754> for similar additions to nitroalkenes
and Seidel<2008JACS16464>

Chiral Sulfur Ligands and Catalysts



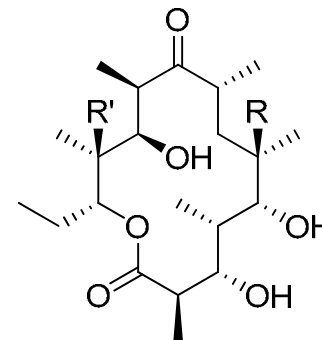
Entry	Ligand	Product	Yield (%)	ee (%)
1	(R_a, S) 1	R = Ph	63	67 (S)
2	(S_a, S) 1	R = Ph	48	62 (R)
3	(R_a, S) 2	R = Ph	40	60 (S)
4	(S_a, S) 2	R = Ph	30	57 (R)

Ligands are related to those of Ellman<2003OL545>

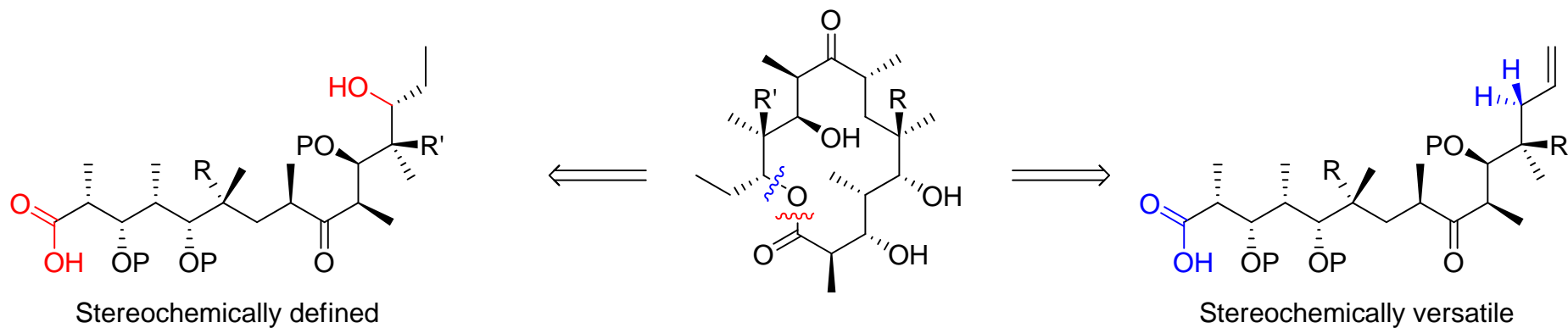


Chiral Sulfur Ligands and Catalysts

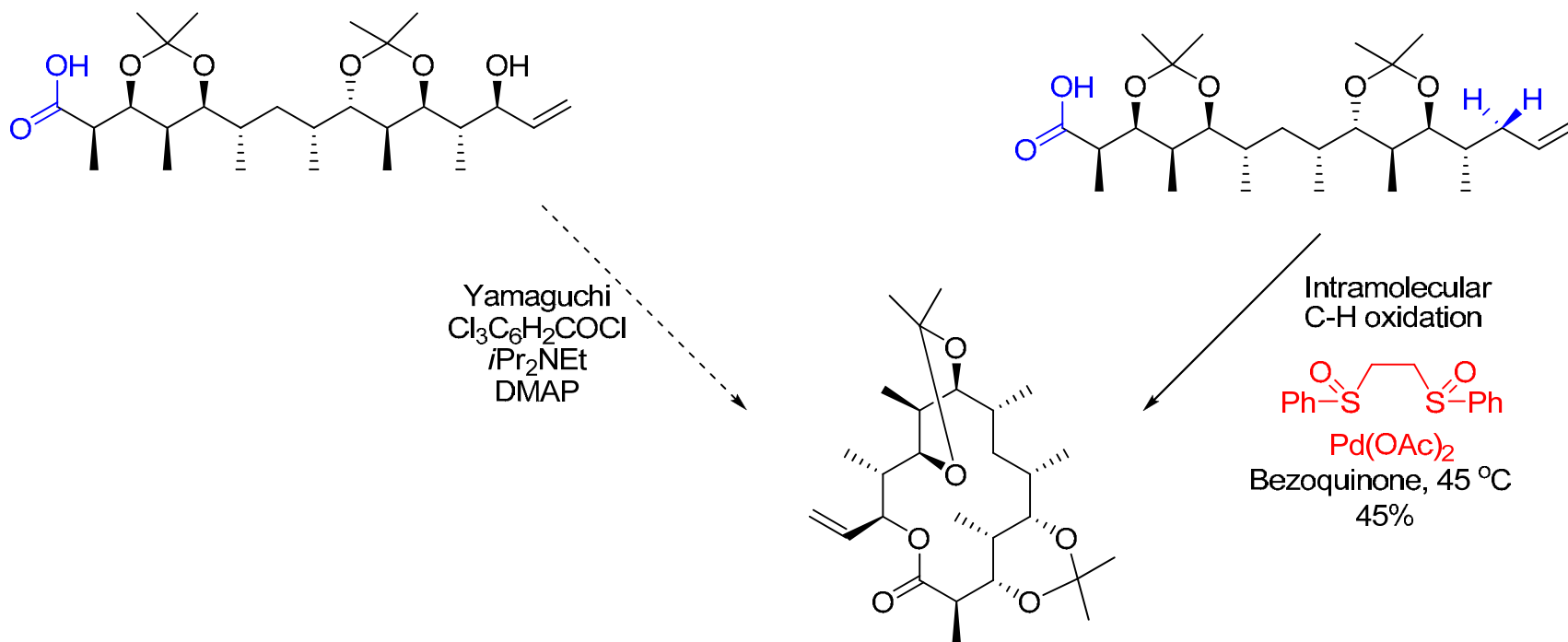
Polyketide macrolides present a number of synthetic challenges including: stereocontrol, and macrocyclisation



6-Deoxyerythronolide B, R = R' = H
Erythronolide B, R = OH, R' = H
Erythronolide A, R = R' = OH



Chiral Sulfur Ligands and Catalysts



Yamaguchi macrolactonisation failed to produce the macrolide. Use of C-H oxidation leads to activation with the palladium catalyst facilitating selective ring formation

Ellman Type Sulfonylimines



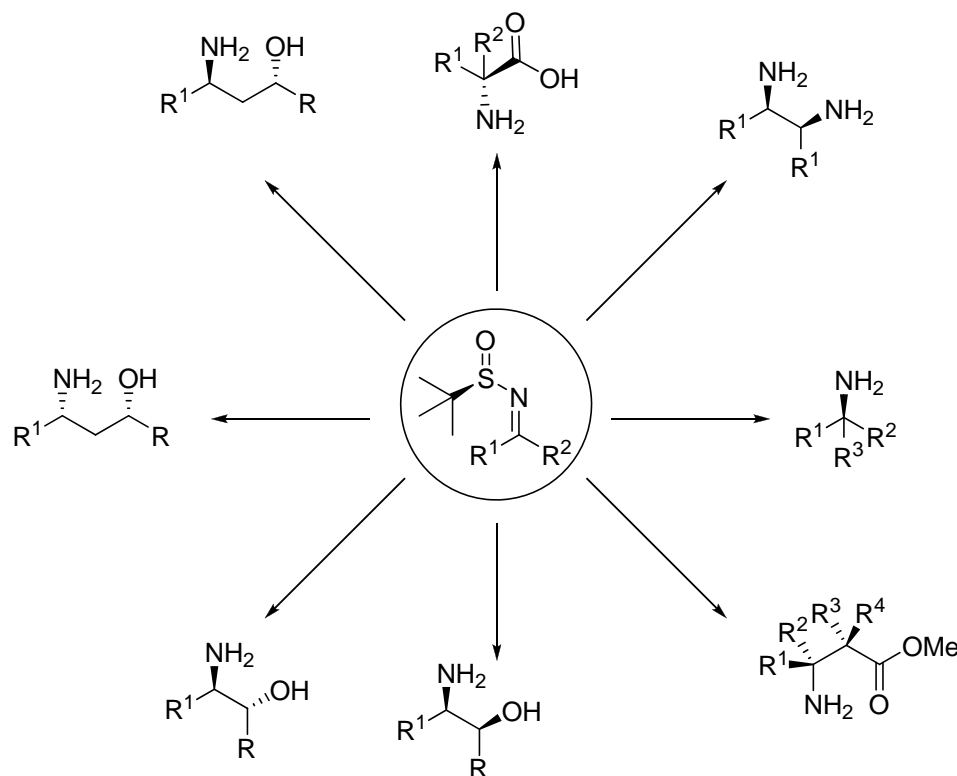
tert-butanesulfonylimines - chiral amine equivalents
-related to Davis' *para*-toluenesulfonylimines <2006JOC8993><1998CSR13>

For reviews of the Ellman reaction see: Ellman<2002ACR984>, Chemla<2009CSR1162>, Lin<2009ACR831>, Stockman<2006T8869>.

Ellman Type Sulfonylimines

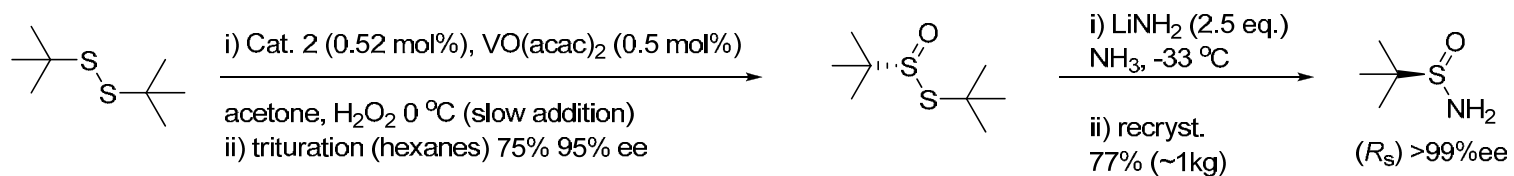
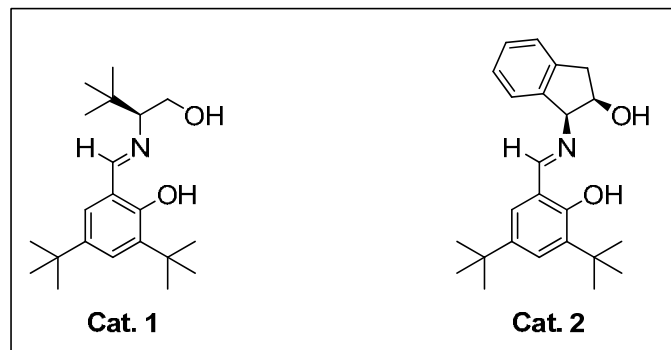


tert-butanesulfonylimines - chiral amine equivalents
-related to Davis' *para*-toluenesulfonylimines <2006JOC8993><1998CSR13>

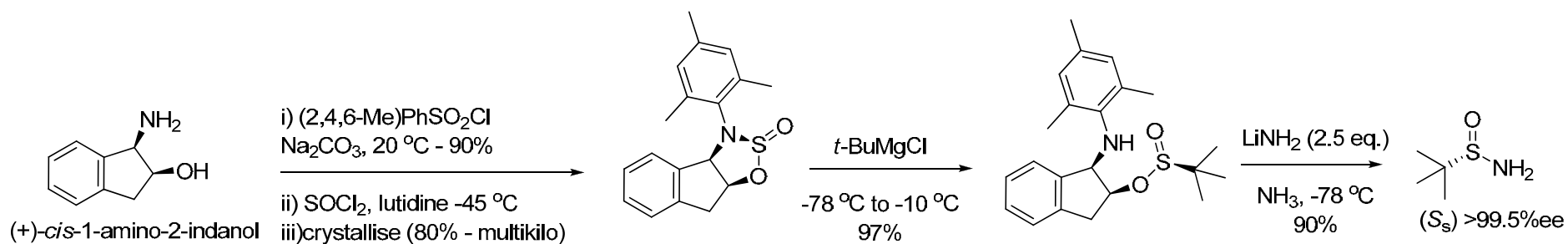


For reviews of Ellman sulfonylimines see: Ellman<2002ACR984>, Chemla<2009CSR1162>, Lin<2009ACR831>, Stockman<2006T8869>.

Ellman Type Sulfonylimines

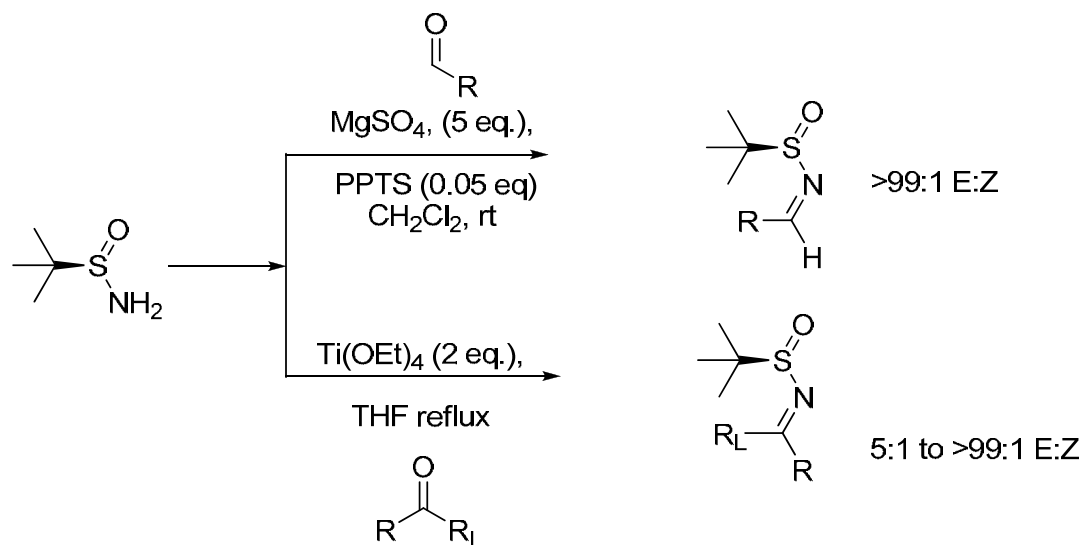


Ellman<2005OS157>

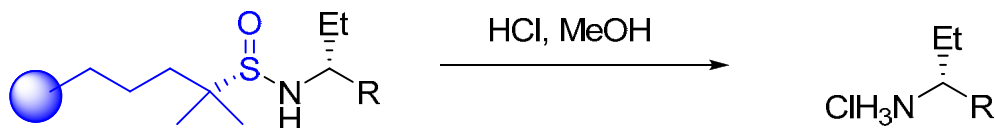


Ellman Type Sulfonylimines

Formation of aldimines and ketimines



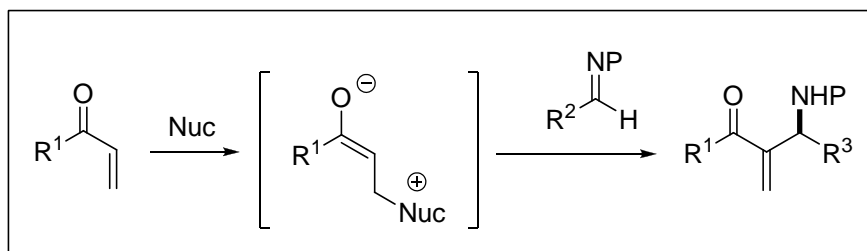
Deprotection is straightforward under acidic conditions:



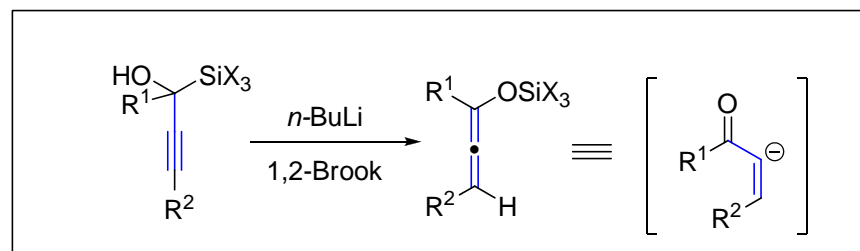
Recycling of *tert*-butanesulfinyl group: Ellman<2009JOC2646>, Aggarwal <2009TL3482>

Ellman Type Sulfonylimines

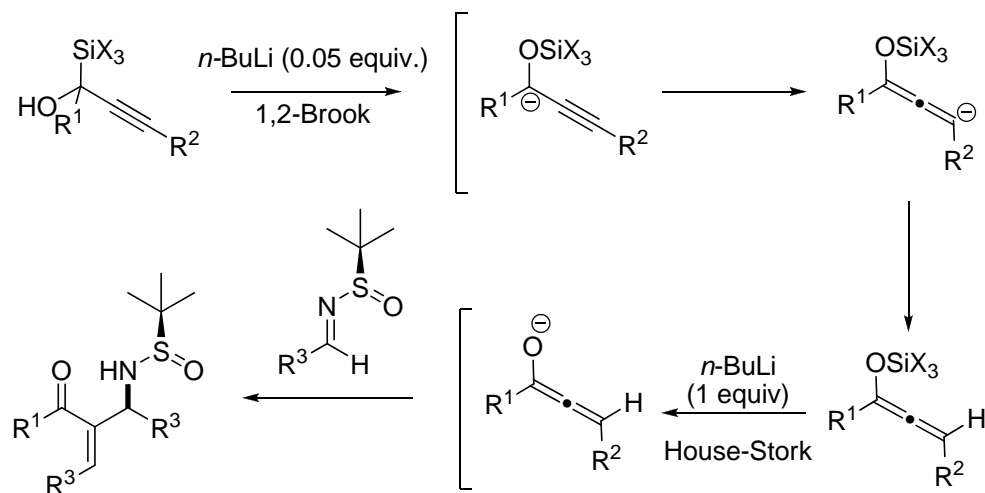
aza-Morita-Baylis-Hillman reaction



Analogous use of lithium allenolates—
synthesised *in-situ* via initial Brook rearrangement



yields 70-94%
d.r. 8:1 to 20:1

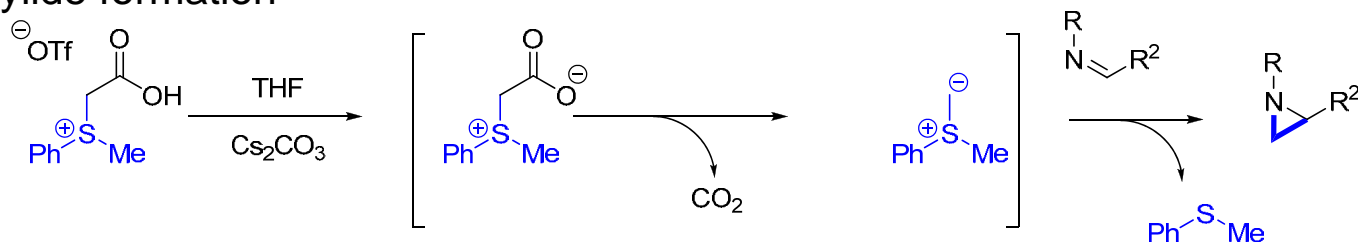


See also Chemla <2009OL931> and Viso <2008OL4775> for related reactions

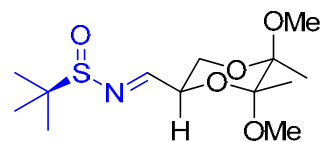


Ellman Type Sulfonylimines Aziridine Formation

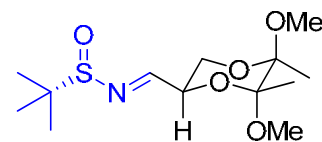
Sulfur ylide formation



Combined Ellman reagent with Ley butanediactal to synthesise aziridines

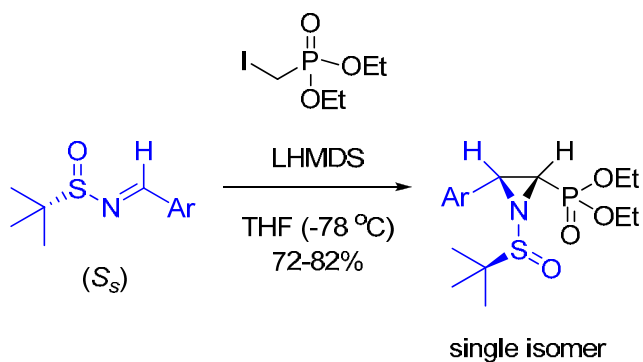


Yield 92% dr >95:5

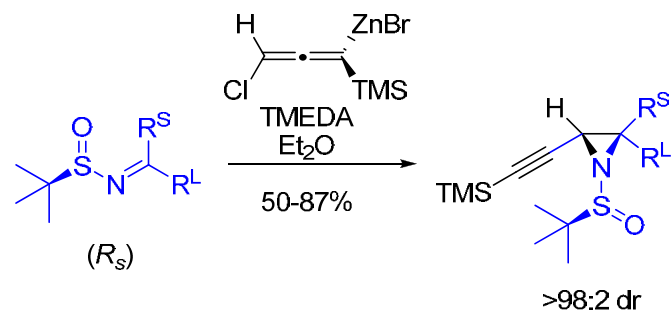


Yield 88% dr >78:22

Forbes<2009CC1882>



Davis<2006JOC6894>



Chemla<2004JOC8244>

For additional examples: Sweeney<2009EJOC4911>



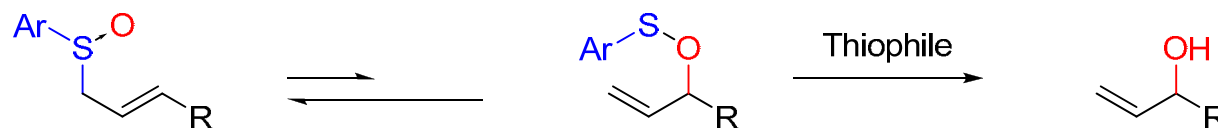
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Rearrangements

Rearrangements

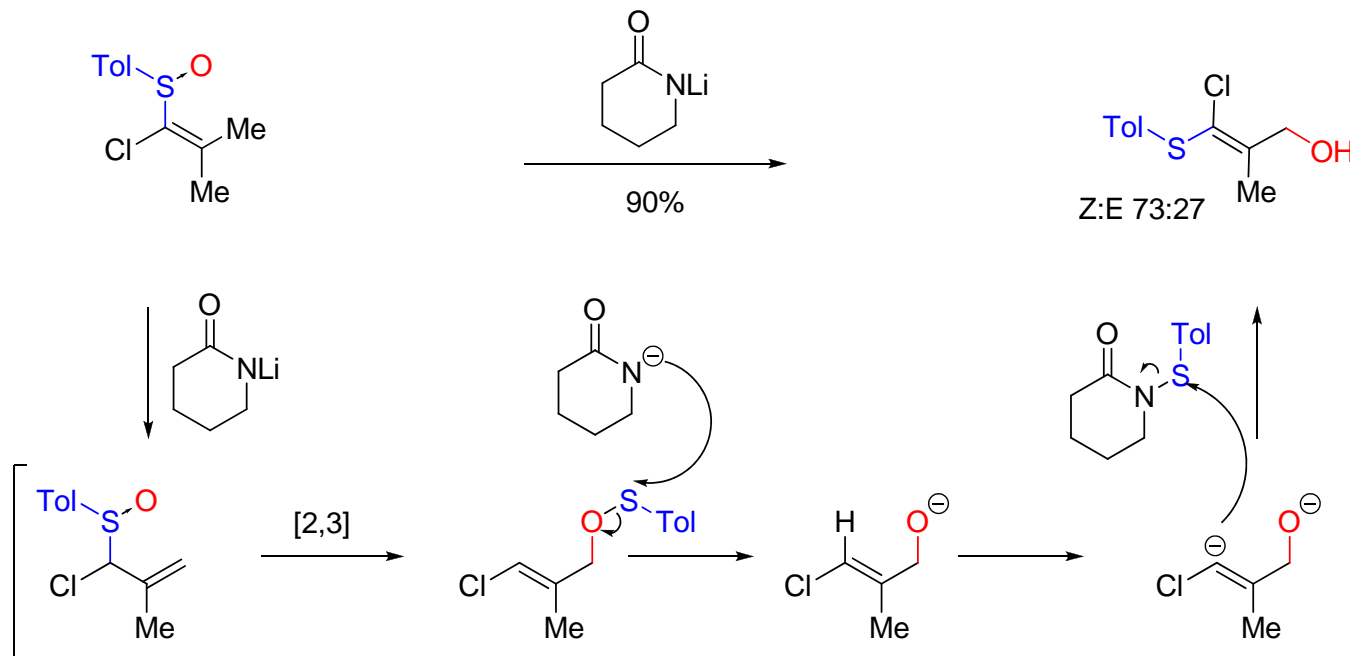
[2,3] Mislow-Braverman-Evans Rearrangement

Classical [2,3] Mislow-Braverman-Evans rearrangements results in formation of an alcohol with loss of sulfur



Satoh's approach results in retention of the sulfur atom on the original carbon

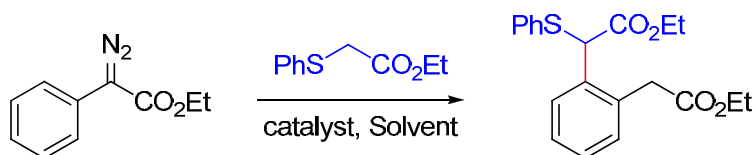
Satoh<2006TL1981>



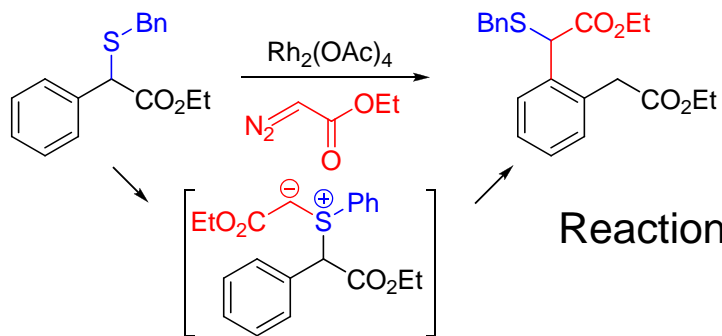
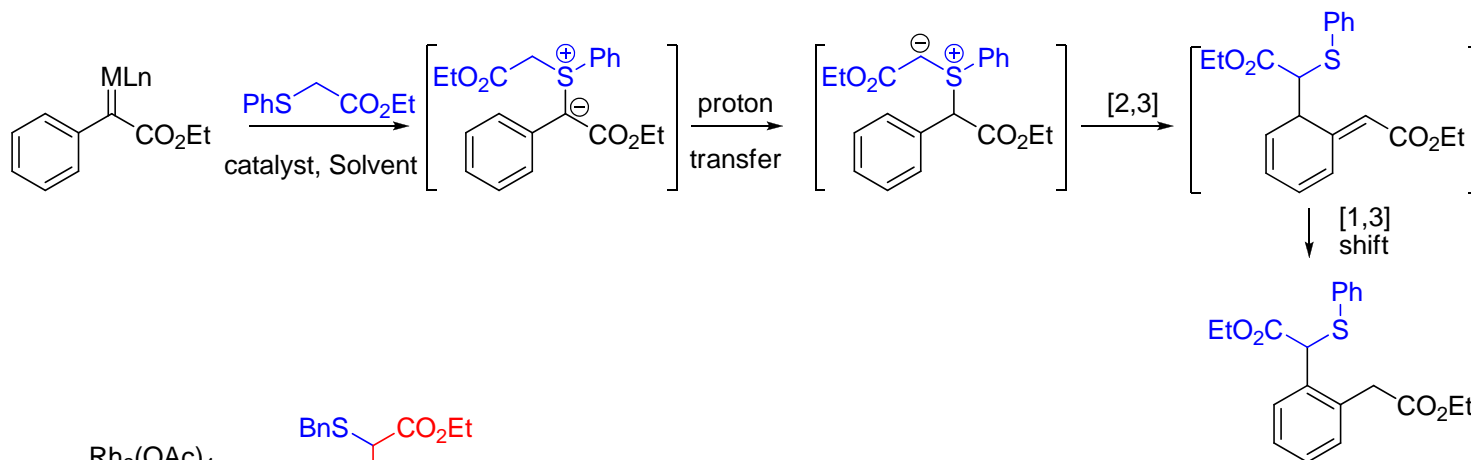
Reaction was carried out with 5 equivalents of base (LDA produced complex mixtures)

Rearrangements [2,3] Thia-Sommelet-Hauser

Synthesis of substituted arenes



Entry	Catalyst	Solvent	Time	Yield (%)
1	Rh ₂ (OAc) ₄	DCM	2 h	51
2	Rh ₂ (OAc) ₄	Tol.	2 h	67
3	Rh ₂ (O ₂ CCF ₃) ₄	Tol.	0.5 h	87



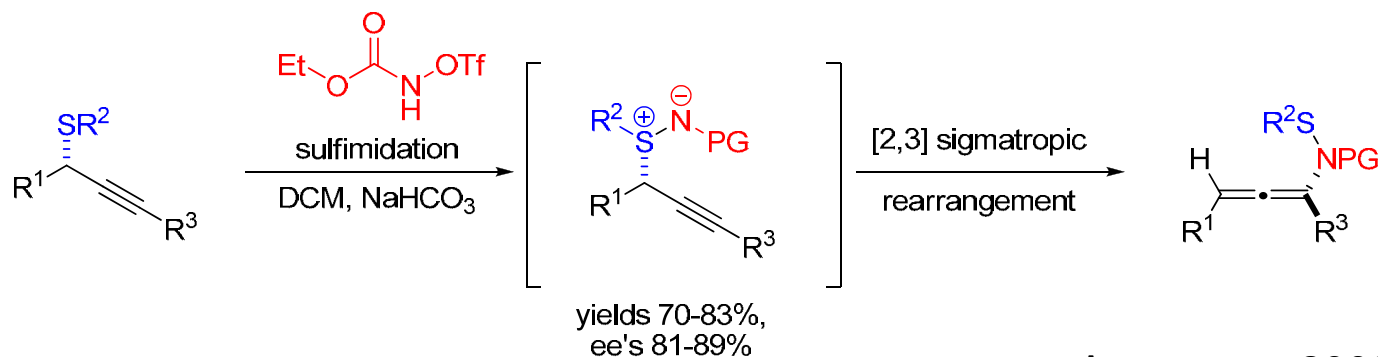
Reaction can easily be carried out in “reverse” with ethyl diazoacetate



Rearrangements

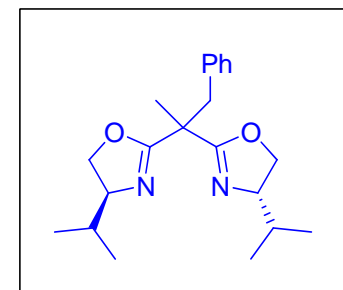
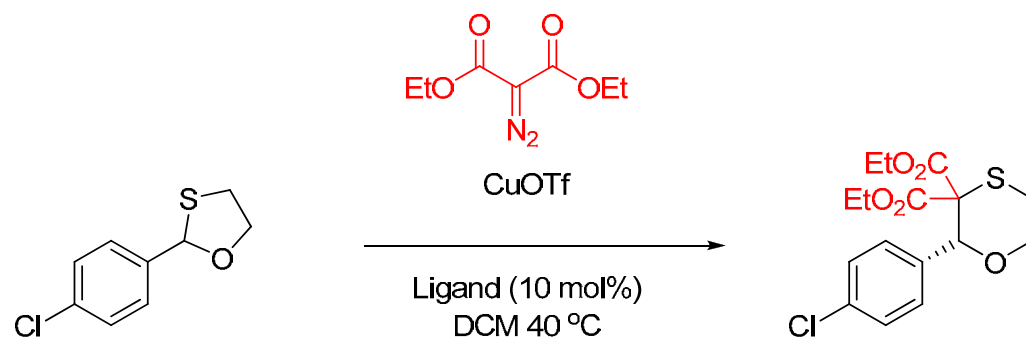
[1,2] Stevens and sulfimide [2,3] rearrangements

Synthesis of Allenamides *via* sulfimide [2,3]-sigmatropic rearrangement



Armstrong<2009OL1547>

Asymmetric [1,2]-Stevens Rearrangement of Sulfur Ylides



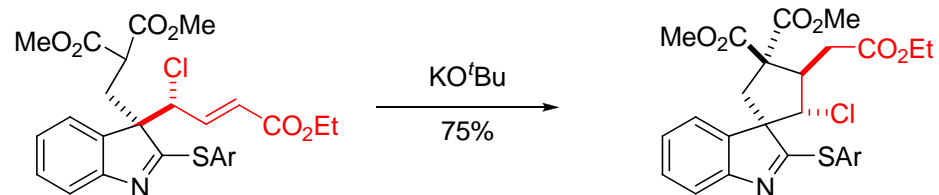
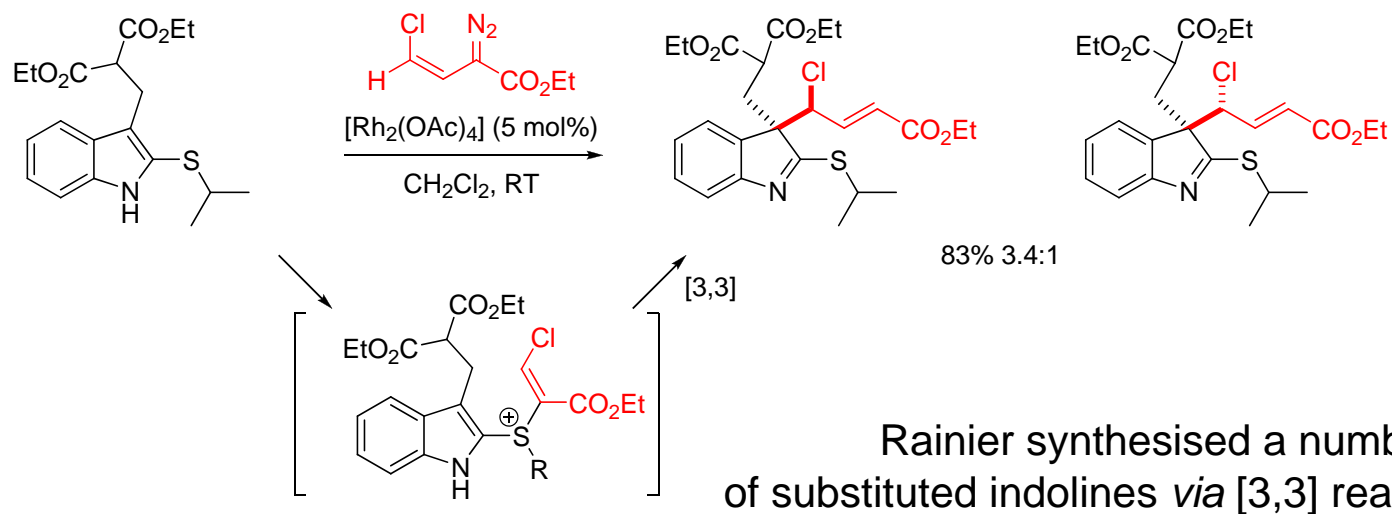
Tang and co-workers investigated a range of bisoxazoline ligands

Yields 60-98% ee's 50-90%

Tang<2009ASC309>

Rearrangements

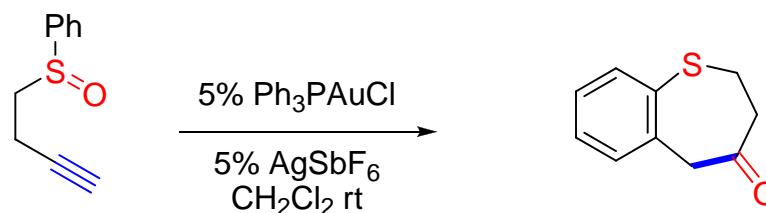
[3,3] Sulfonium Ylide rearrangements



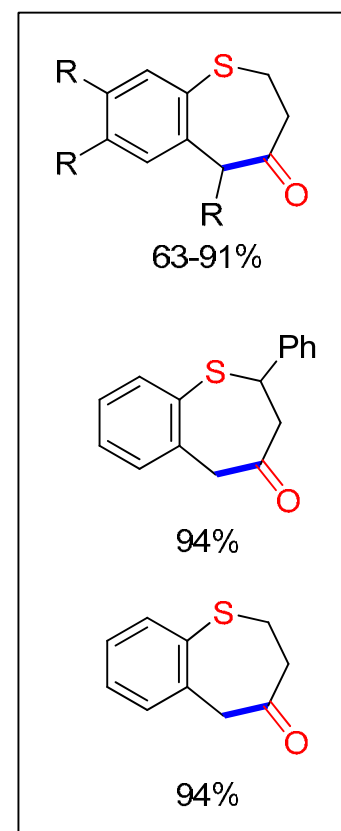
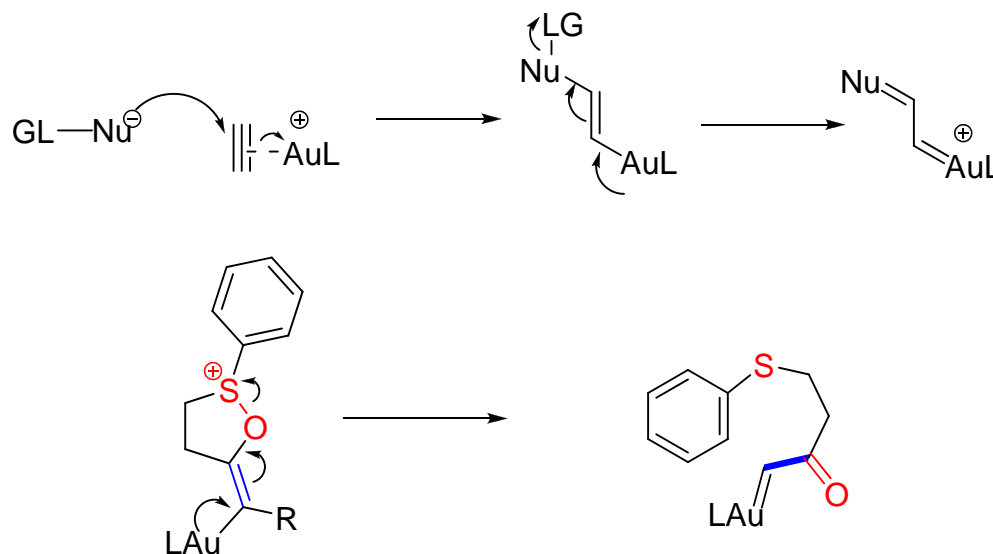
Products are readily converted to spirocyclopentanes

Rearrangements of Alkynyl Sulfoxides Catalysed by Gold (I) Complexes

Reaction of sulfoxides with gold-carbenoid species:



Reaction proceeds via rearrangement and reaction of the metal carbenoid:

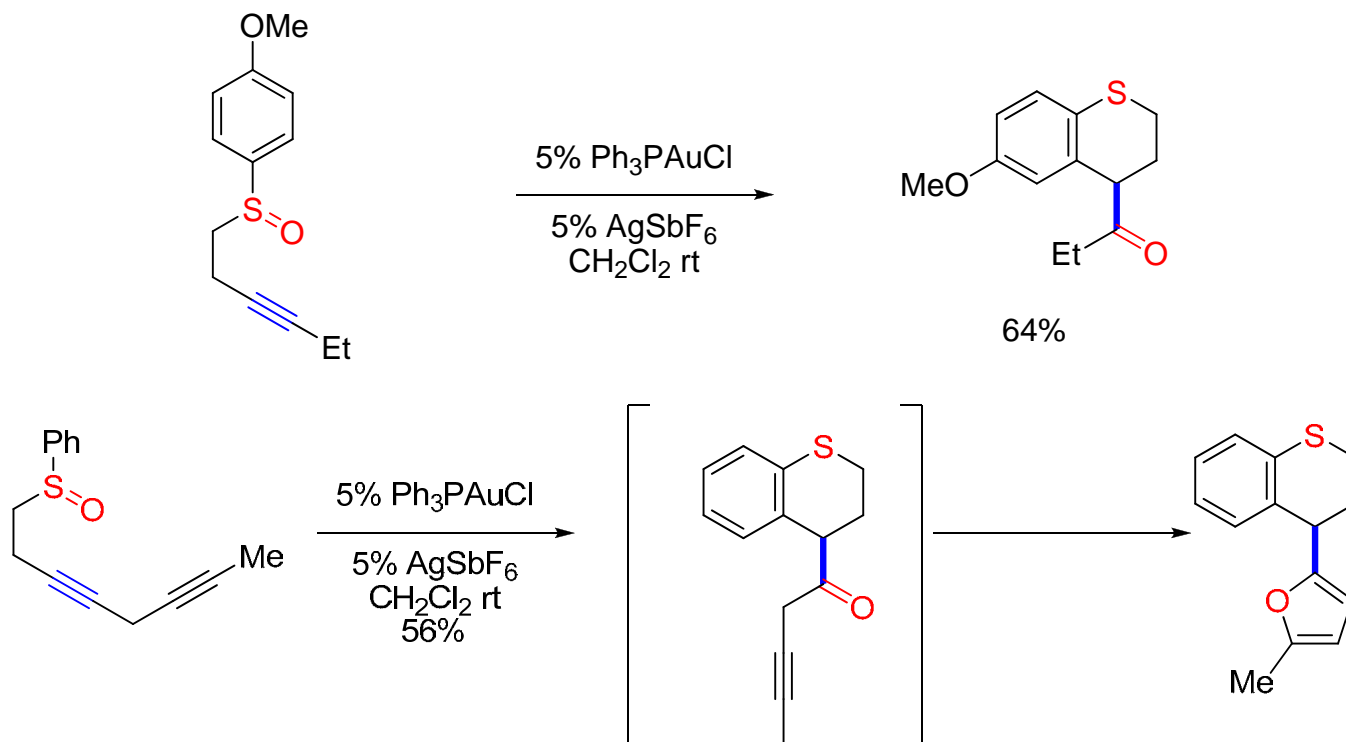


For related reactions see also Zhang<2007ACIE5156>
and reactions of allenes Krause<2006ACIE1897>

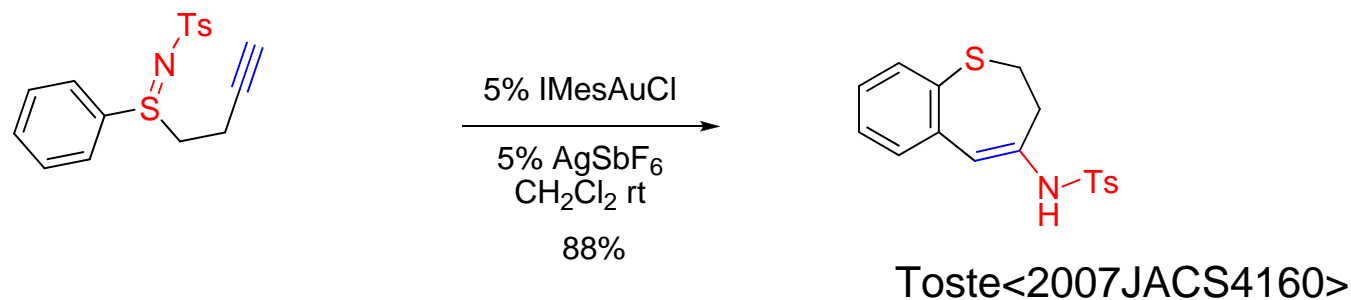


Rearrangements of Alkynyl Sulfoxides Catalysed by Gold (I) Complexes

Reaction of sulfoxides with gold-carbenoid species:

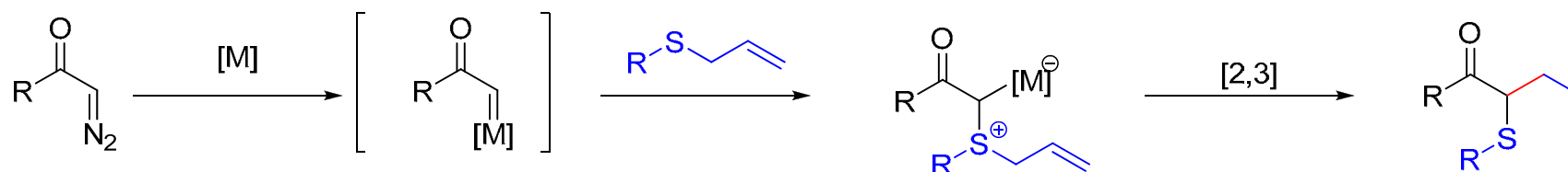


Reaction also works with sulfimines:

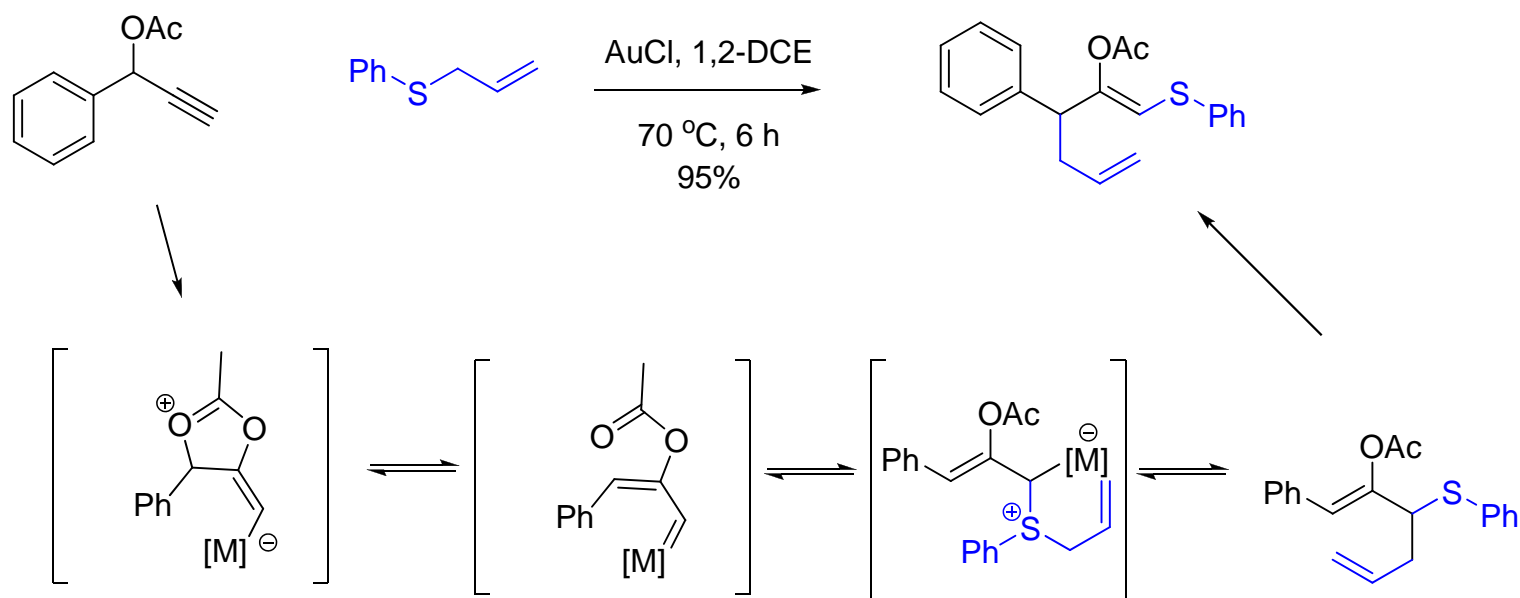


Rearrangements of Alkynyl Sulfoxides Catalysed by Gold (I) Complexes

Reaction of allyl sulfides with gold-carbenoid species:



Doyle-Kirmse reaction – anticipated product:

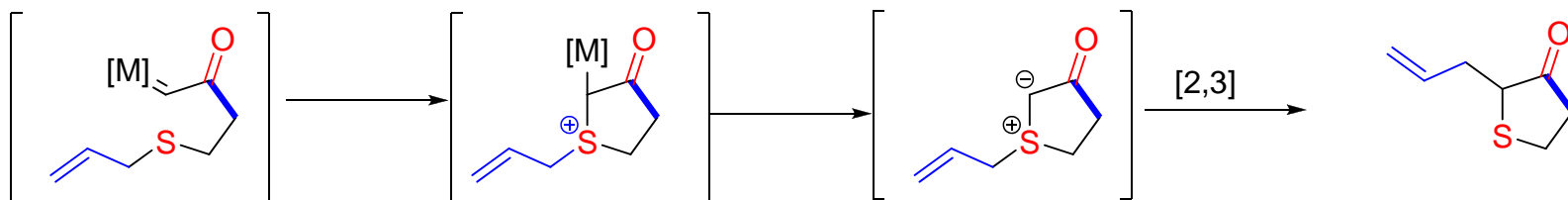
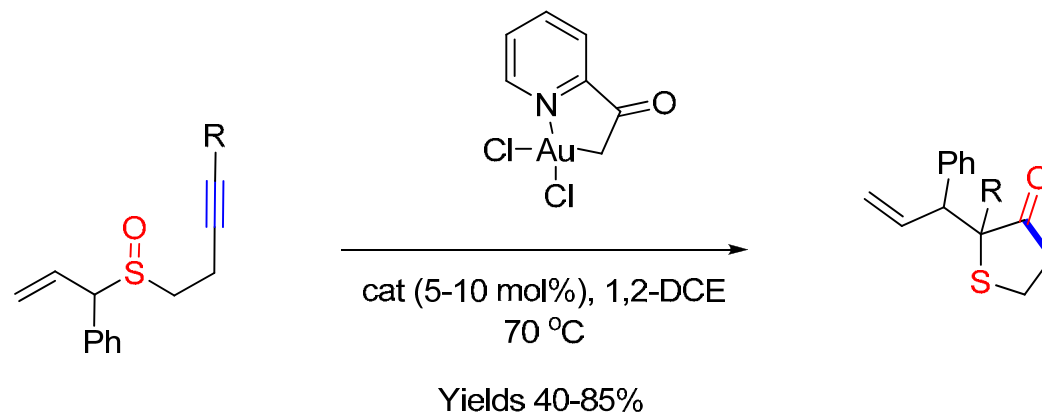


Rearrangements of Alkynyl Sulfoxides Catalysed by Gold (I) Complexes

Reaction of sulfoxides with gold-carbenoid species:



Reaction of sulfur with the metal carbenoid: Toste<2007JACS4160>





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Thanks

Thanks to the organising Committee

Thank you for listening



Journal Abbreviations

ACA	Aldrichimica Acta
ASC	Advanced Synthesis and Catalysis
ACIE	Angewandte Chemie International Edition
ACR	Accounts of Chemical Research
CC	Chemical Communications
CR	Chemical Reviews
CSR	Chemical Society Reviews
EJOC	European Journal of Organic Chemistry
JACS	Journal of the American Chemical Society
JBC	Journal of Biological Chemistry
JOC	Journal of Organic Chemistry
NC	Nature Chemistry
OBC	Organic and Biomolecular Chemistry
OL	Organic Letters
OPRD	Organic Process Research and Development
OS	Organic Syntheses
PNAS	Proceedings of the Natural Academy of Sciences
SL	Synlett
T	Tetrahedron
TL	Tetrahedron Letters

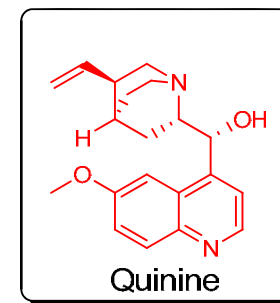
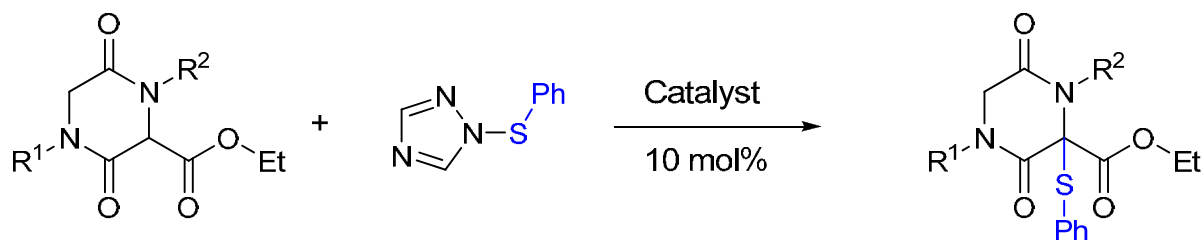


The School of Pharmacy
University of London

Additional Slides

ETP Core Synthesis

Stable sulfur electrophile instead of sulfenyl chloride

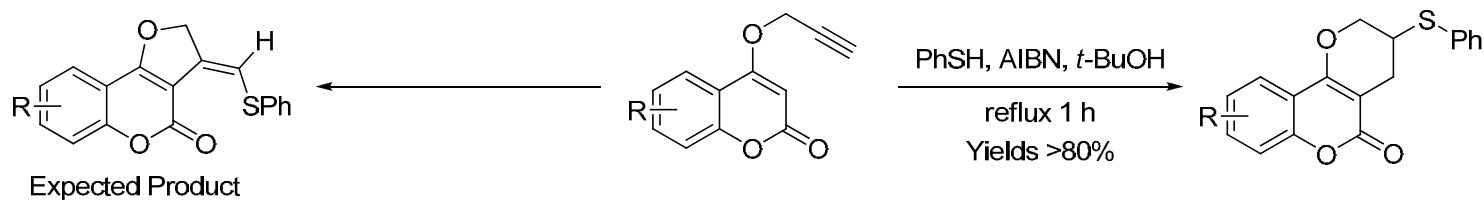


R ₁	R ₂	Solvent	Temp.	Equiv.	Time	Yield (%)
Et	Et	Tol.	Rt	3.0	60 h	77
Et	Et	Tol.	Rt	8.0	48 h	77
Bn	Bn	Tol.	Rt	3.0	60 h	75

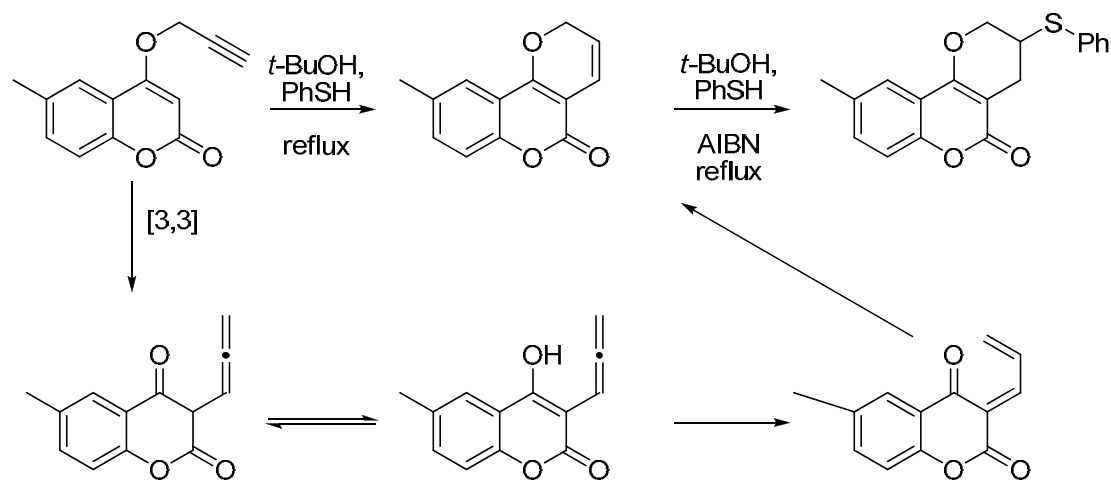


Thiol Radicals and Rearrangements

Thiophenol catalysed Claisen rearrangement of coumarin derivatives



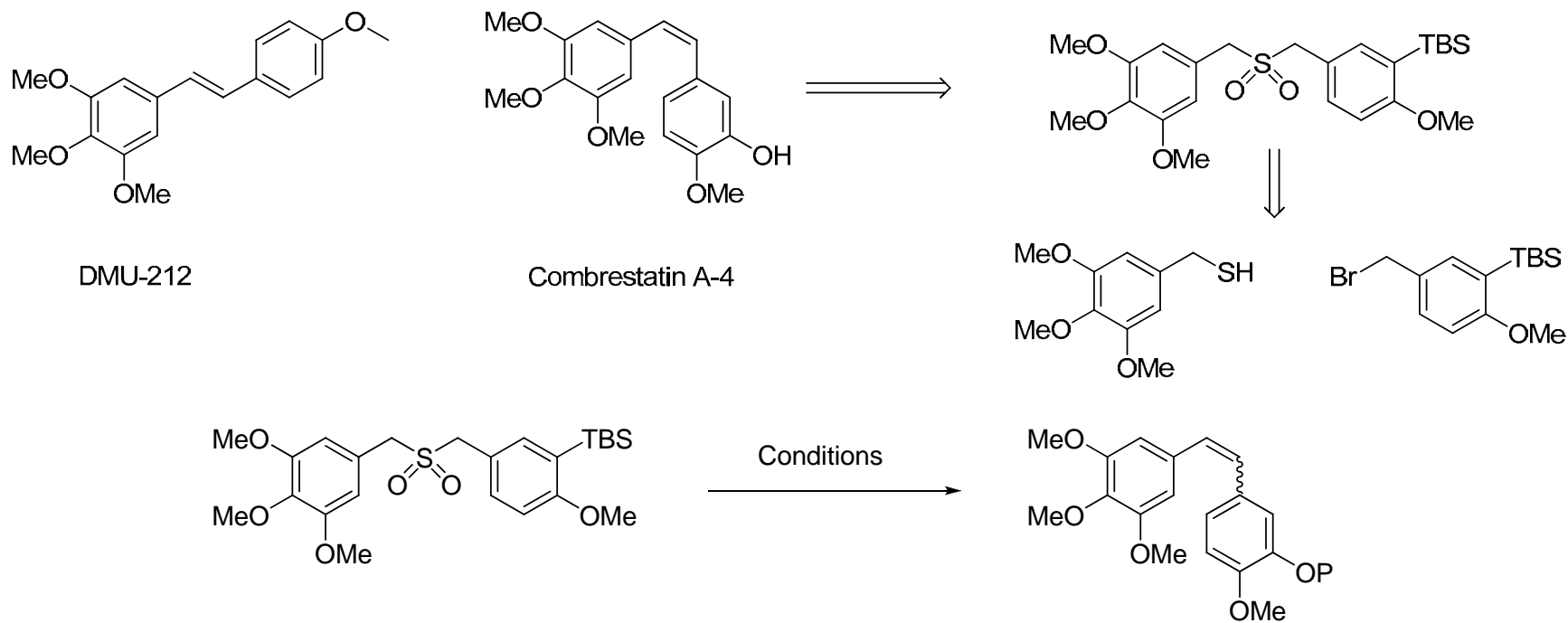
In absence of Thiophenol no rearrangement takes place



Similar reaction observed with indole annulated sulfur heterocycles <2007TL7031>

Ramberg-Bäcklund Reaction

Synthesis of stilbenoid anti-cancer agents combrestatin A-4 and DMU-212

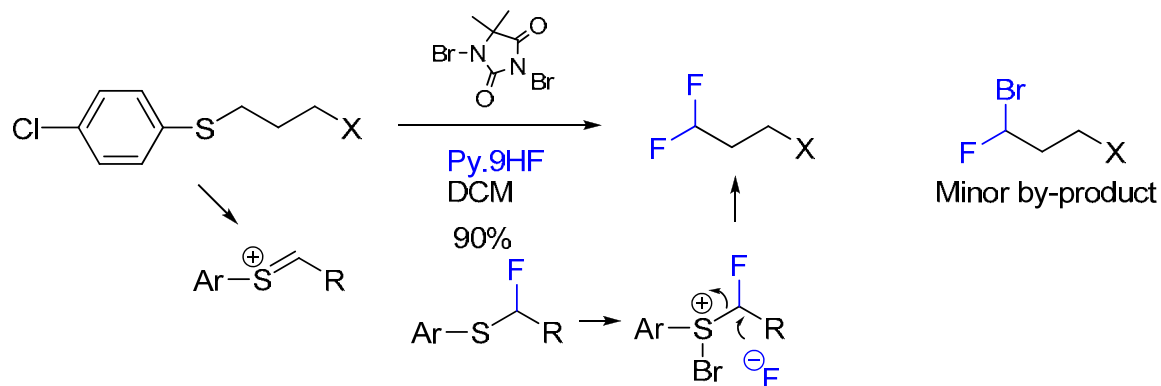


Entry	Conditions	<i>E:Z</i>	Yield (%)	Conditions
1	CF ₂ Br ₂ , <i>t</i> BuOH, KOH-Al ₂ O ₃ , 0 °C to rt 12 h	90:10	81	Chan 1994CC1771
2	C ₂ F ₄ Br ₂ , <i>t</i> BuOH, KOH-Al ₂ O ₃ , reflux 12 h	85:15	72	Franck 1999OL2149
3	CCl ₄ , <i>t</i> BuOH, KOH, H ₂ O	47:53	69	Meyers 1969JACS7510



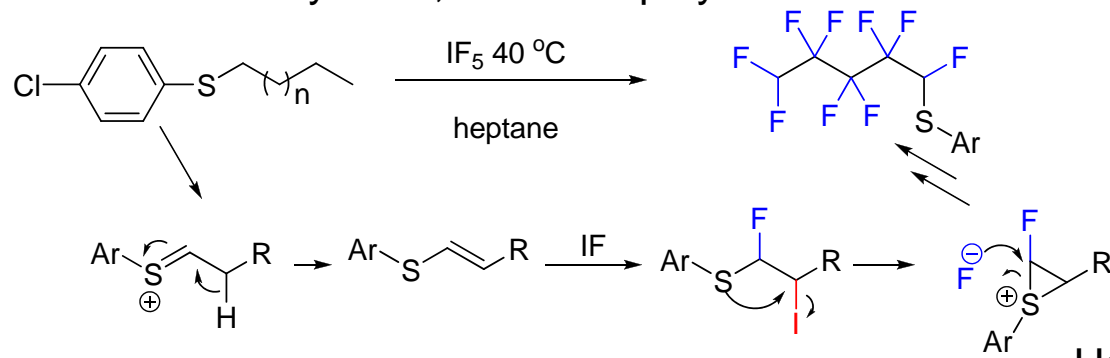
Pummerer and Pummerer-type reactions Fluorination

Oxidative desulfurisation procedure to produce geminal 1,1-difluoroalkanes



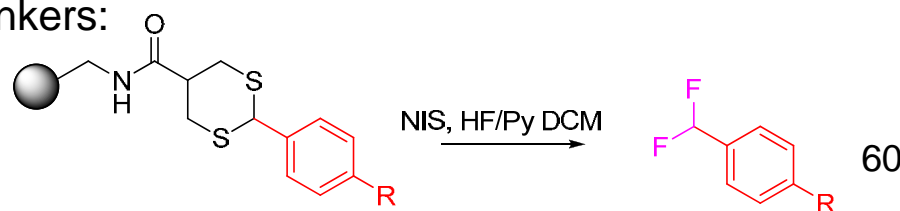
Haufe<2008SL106>

Similar work by Hara, results in polyfluorinated alkanes



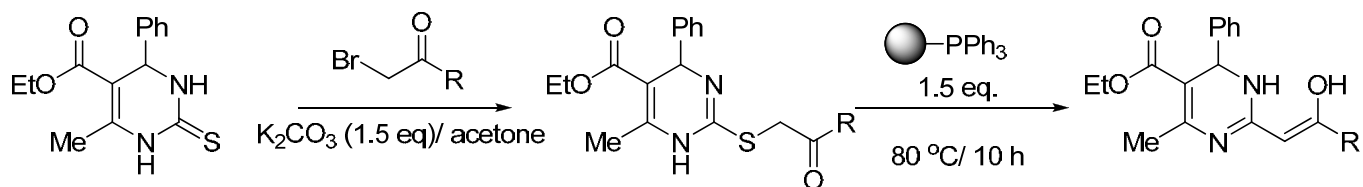
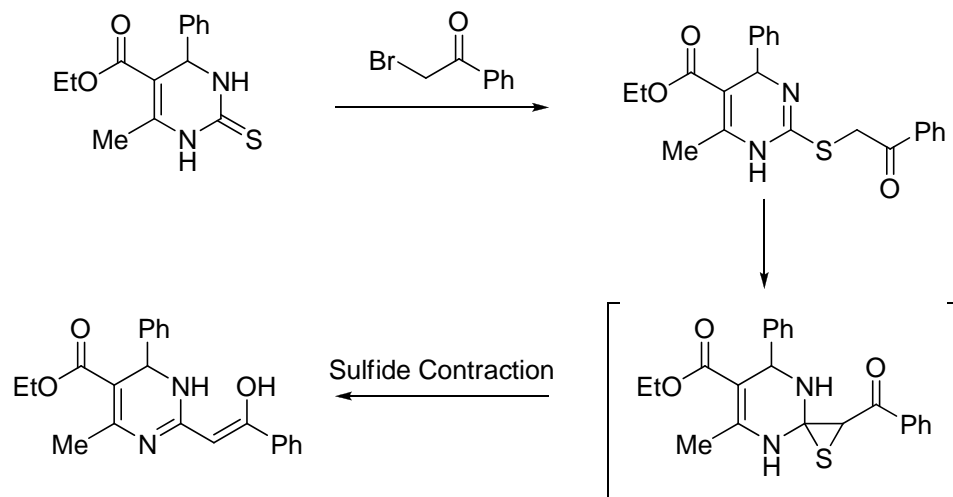
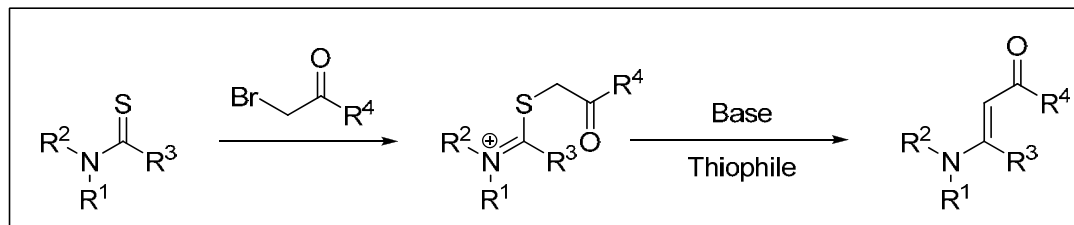
Hara<2004T11445>

See also fluorinating cleavage of solid phase linkers:
<2008ACIE8120>





Eschenmoser Coupling reaction

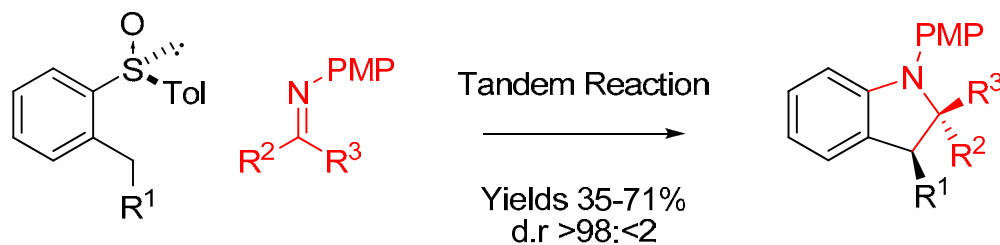


One-pot, two step procedure

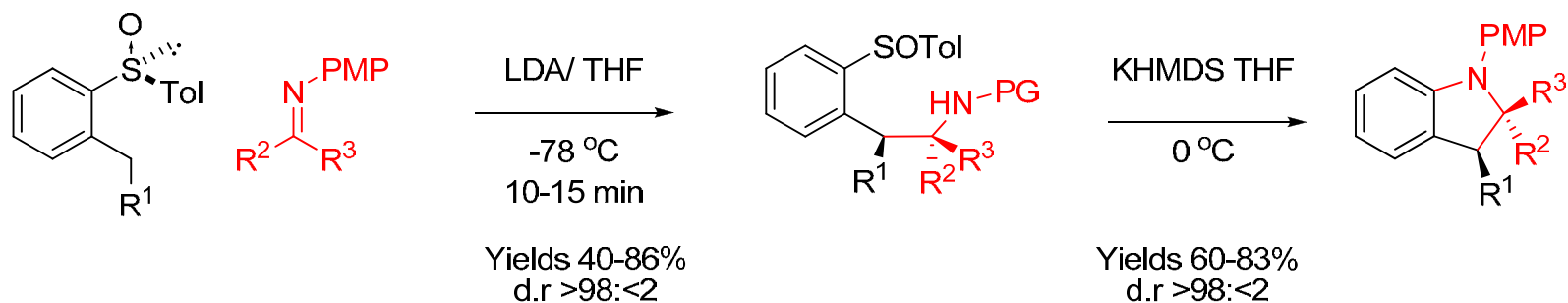


Ellman-type reaction

Optically pure Indolines from chiral tolyl-sulfinylalkylbenzenes



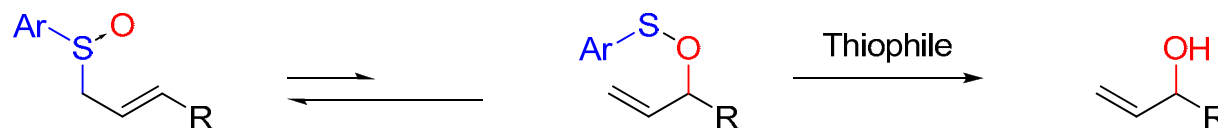
Tandem process proceeds at room temperature



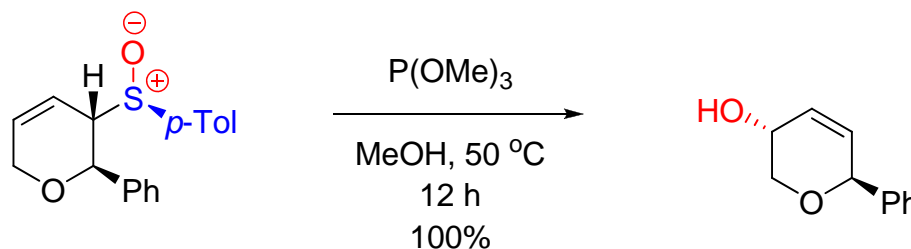
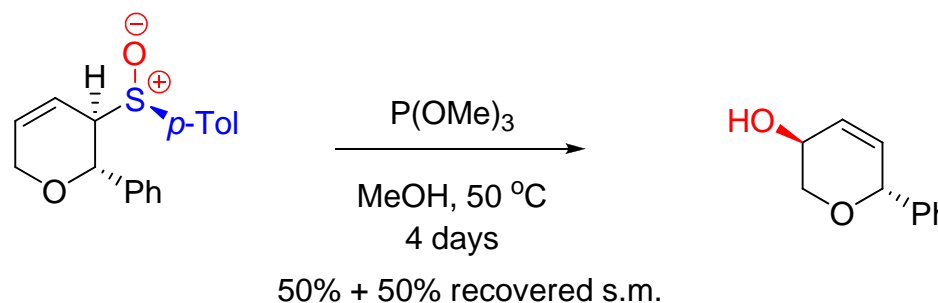
Rearrangements

[2,3] Mislow-Braverman-Evans Rearrangement

Classical [2,3] Mislow-Braverman-Evans rearrangements results in formation of alcohol with loss of sulfur



Satoh<2006TL1981>



Detailed investigation into the effects of substituents on the pyran ring and how they affect the rearrangement outcome.