



UNIVERSITY OF  
LIVERPOOL

## A new chemoenzymatic approach to chiral bicyclic [2.2.2]octan-2,5-diones and diene ligands

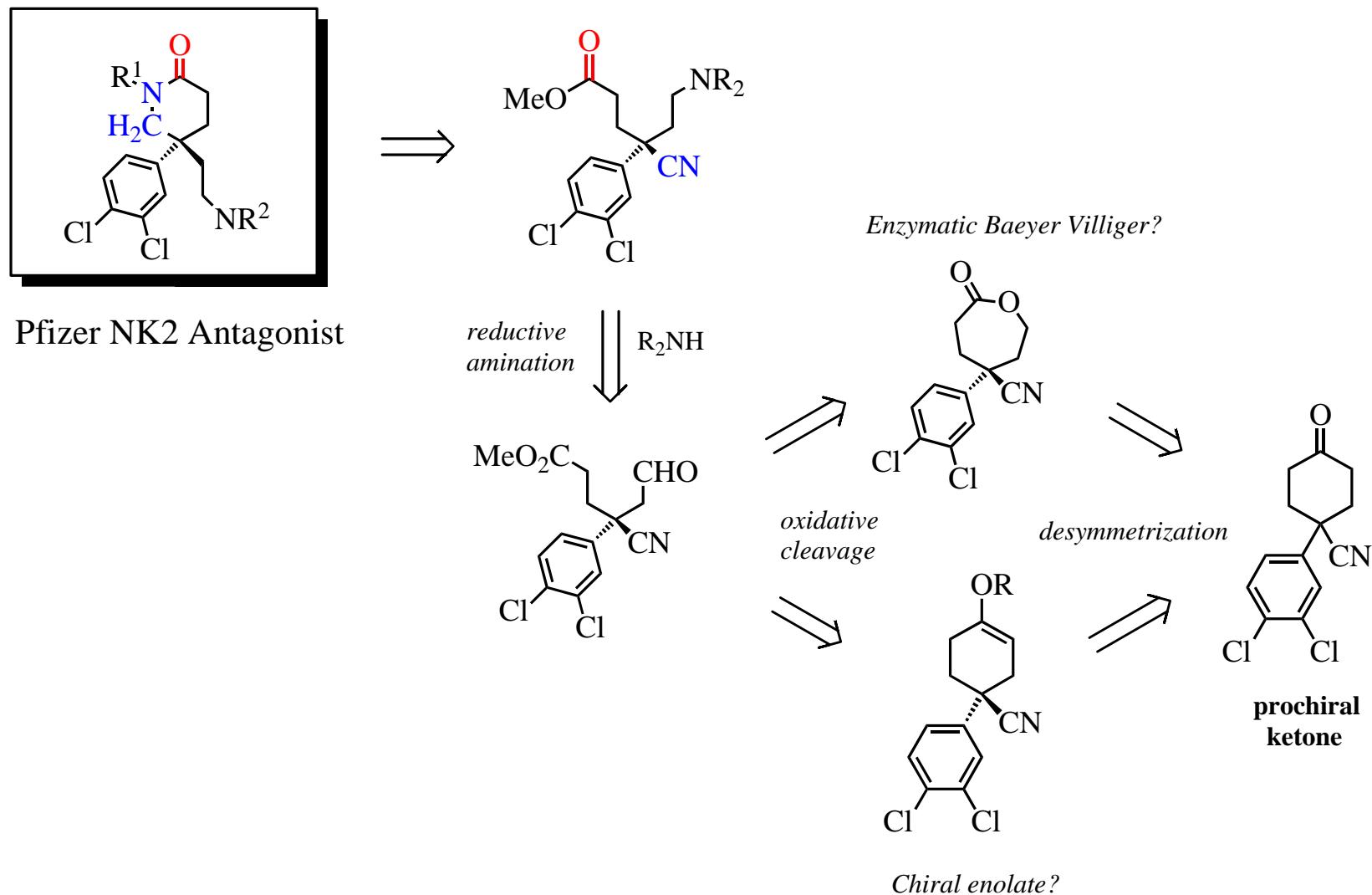
Andrew Carnell

*acarnell@liv.ac.uk*

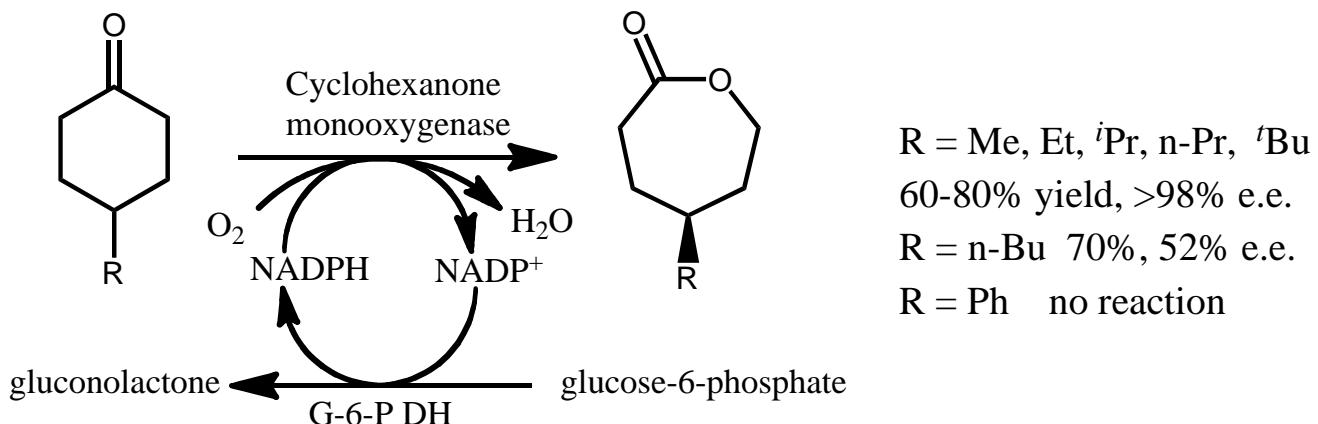
Department of Chemistry, Robert Robinson Laboratories

**Biocatalysis: Challenges for  
Pharmaceuticals & Fine Chemicals**  
**14th October 2010**

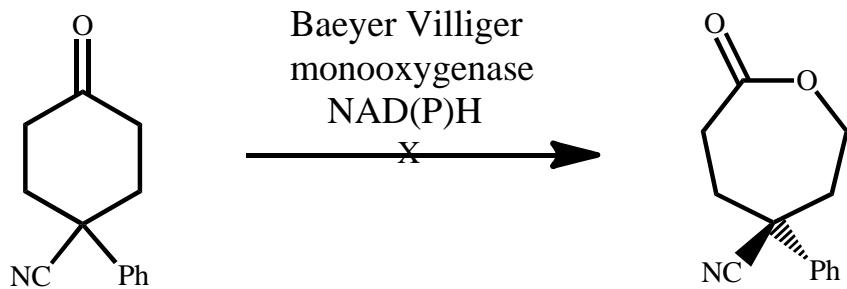
# Retrosynthesis of Pfizer NK-2 Antagonists



# Baeyer Villiger Monooxygenases

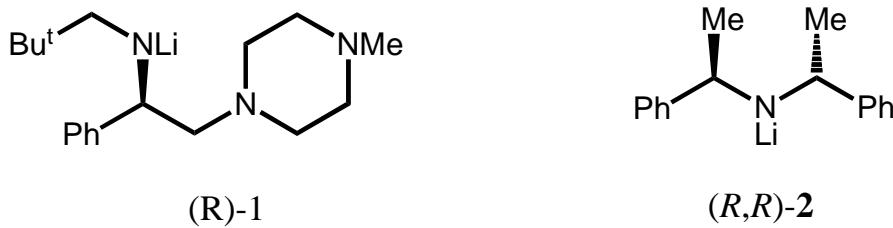
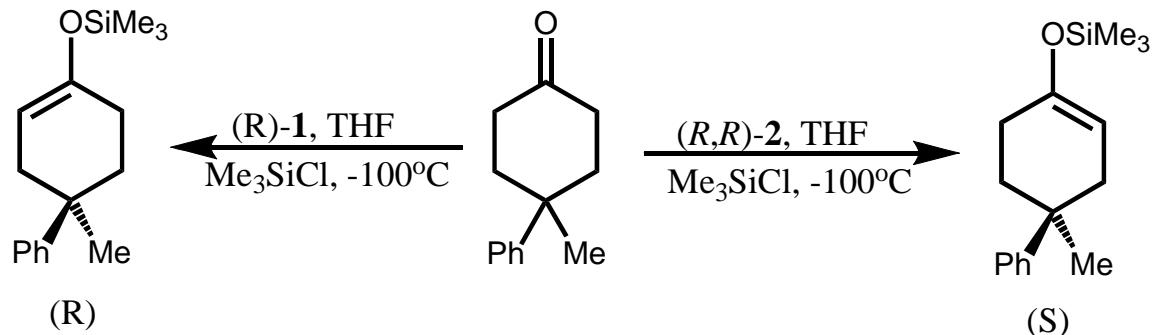


M. J. Tachner and D. J. Black, *J. Am. Chem. Soc.*, 1988, 110, 6892

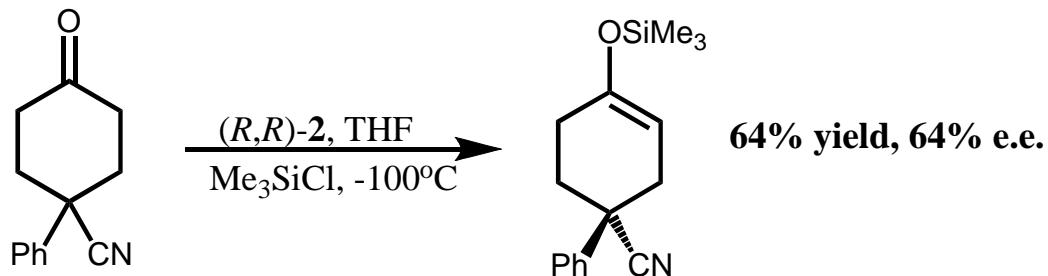


Cyclohexanone Monooxygenase  
Diketocamphane Monooxygenase  
2-oxo- $\Delta^3$ -4,5,5-trimethylcyclopentenylacetylCoA MO

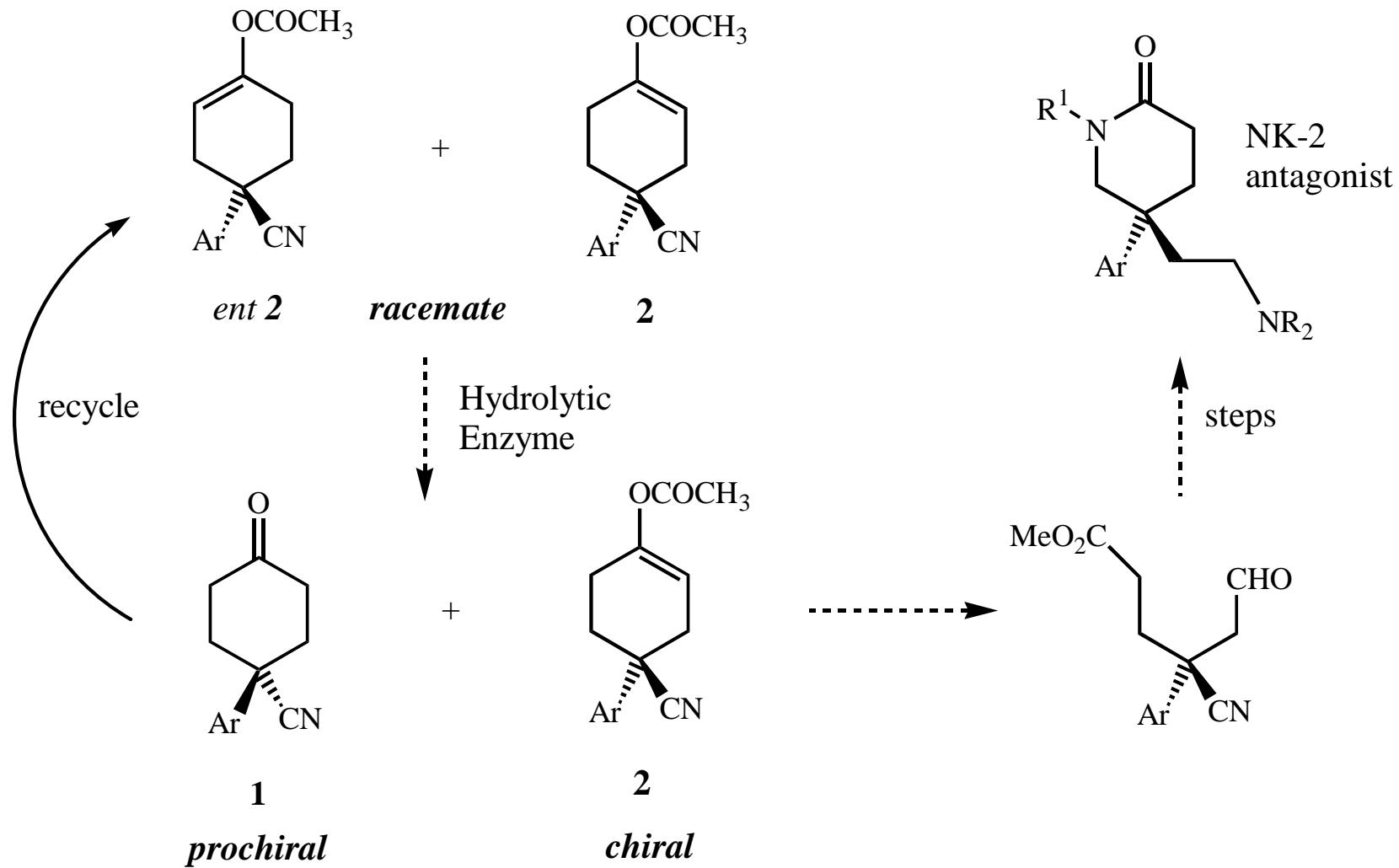
## 4,4-Disubstituted cyclohexanone with chiral lithium amide



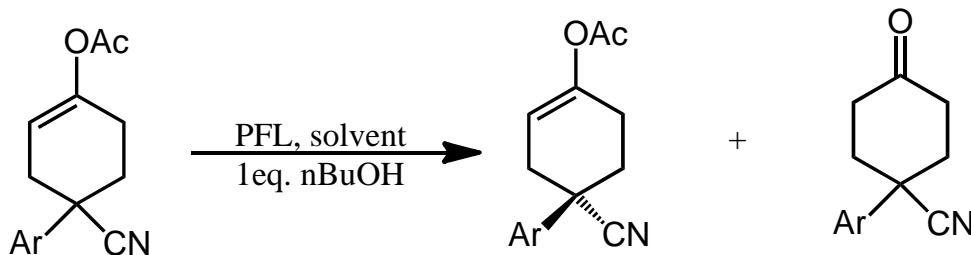
T. Honda, N. Kimura, M. Tsubuki, *Tetrahedron: Asymmetry*, 1993, 4, 21



# Hydrolytic enzymes to desymmetrise the prochiral ketone

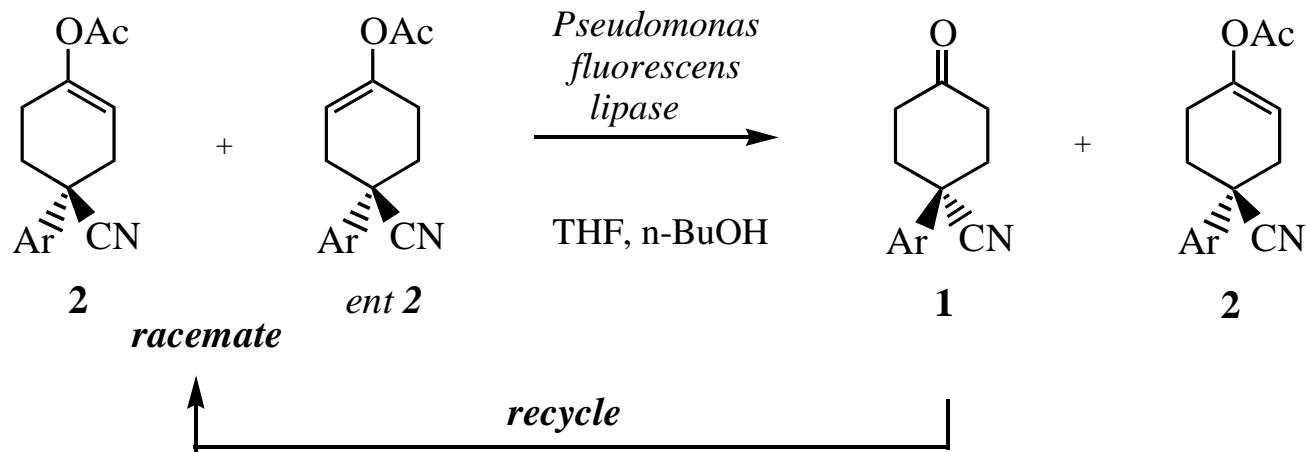


## Biotransformation of 4-aryl-4-cyano enol esters



Ar	solvent time (h)	enol ester yield (%)	enol ester E.e. (%)	ketone (%)
	THF, 8.5h	30%	100%	70%
	THF, 3.5h	38%	100%	62%
	Toluene, 2h	48%	61%	52%
	THF, 22h	29%	96%	71%
	Toluene, 6h	30%	94%	70%
	THF, 6h	38%	78%	62%

# Continuous Process for Deracemization without Work-up



Each cycle:

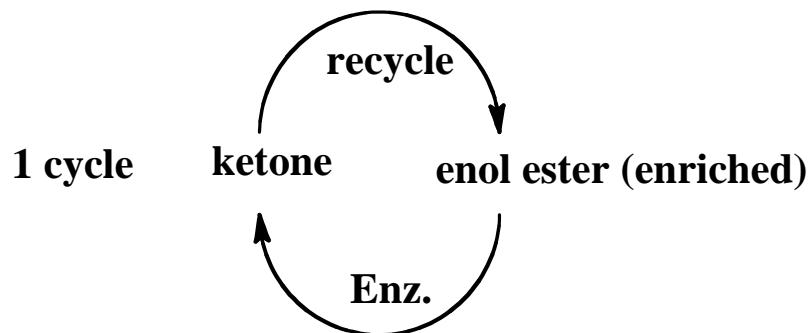
1. remove enzyme
2.  $^t\text{BuOK}$ , isopropenyl acetate (3eq.)
3. Stir with Dowex  $\text{H}^+$  resin
4. remove resin
5. add Enzyme + nBuOH

Enzyme tolerates  $^t\text{BuOH}$  and acetone

## Recycling the ketone

For a kinetic resolution  $E = \frac{\ln[(1-c)(1-ee(S))]}{\ln[(1-c)(1+ee(S))]}$        $E.e \text{ (max) for cyclic reaction} = \frac{(E - 1)}{(E + 1)}$

i.e. for  $E = 13$ ,  $e.e.(max) = 85.7\%$  for  $100\%$  yield

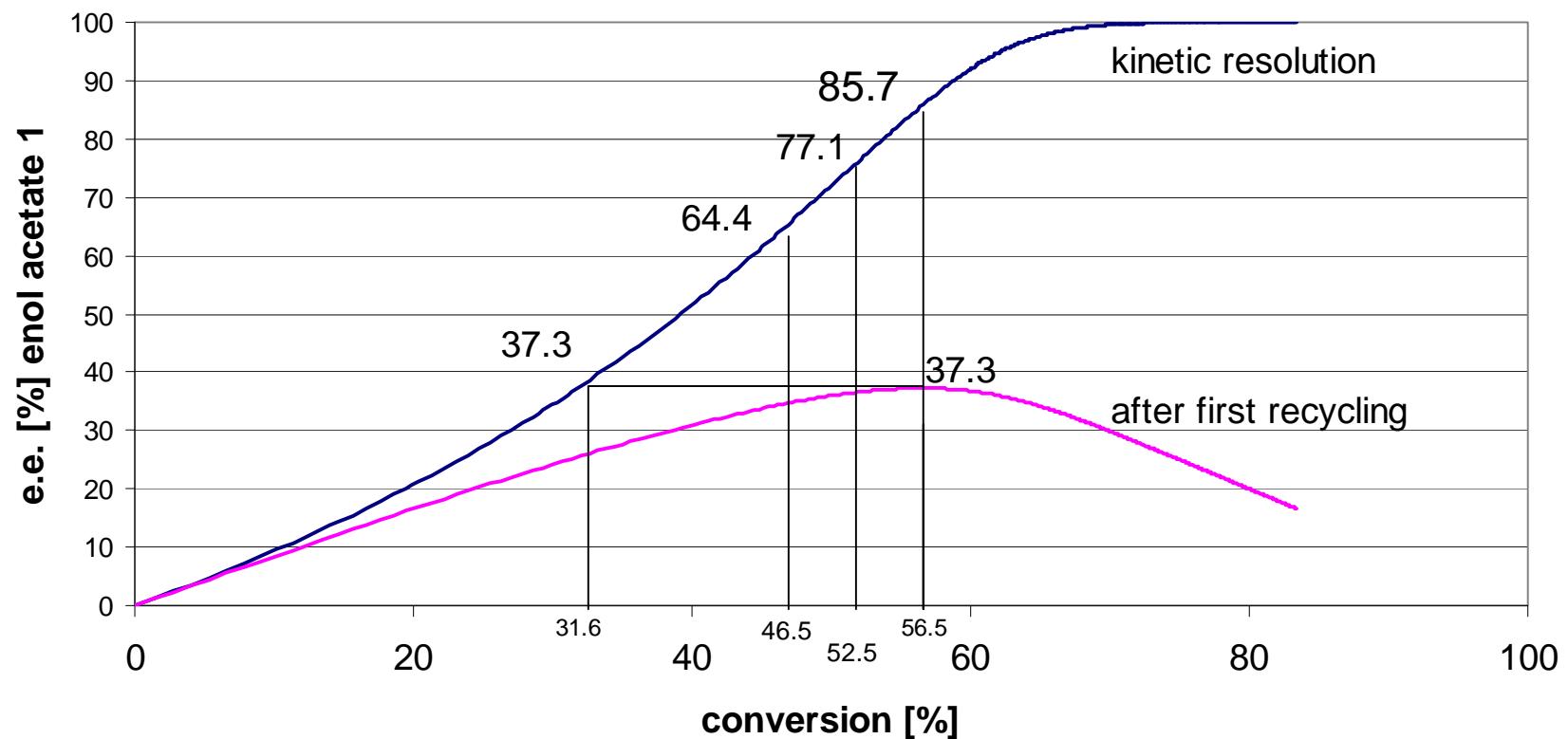


See: W. Kroutil and K. Faber, *Tetrahedron:Asymmetry*, 1998, 9, 2901

Theoretical conversions and e.e.'s for deracemisation  
for E = 13

Cycle	% e.e. <b>1</b> (start)	% e.e. <b>1</b> (end)	% Conversion to ketone <b>2</b> needed
1	0	85.7	56.5
2	37.3	85.7	24.9
3	64.3	85.7	10
4	77.1	85.7	4
5	82.3		

**Enantiomeric excess (e.e.) of enol acetate **1** after recycling of the ketone **2** versus conversion for E=13.**

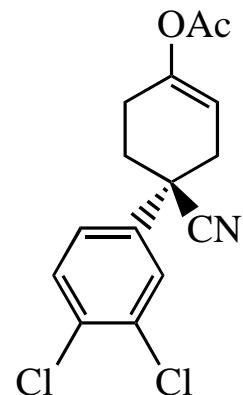


G. Allan, A.J. Carnell and W. Kroutil, *Tetrahedron Lett.*, 2001, 42, 5959-5962.

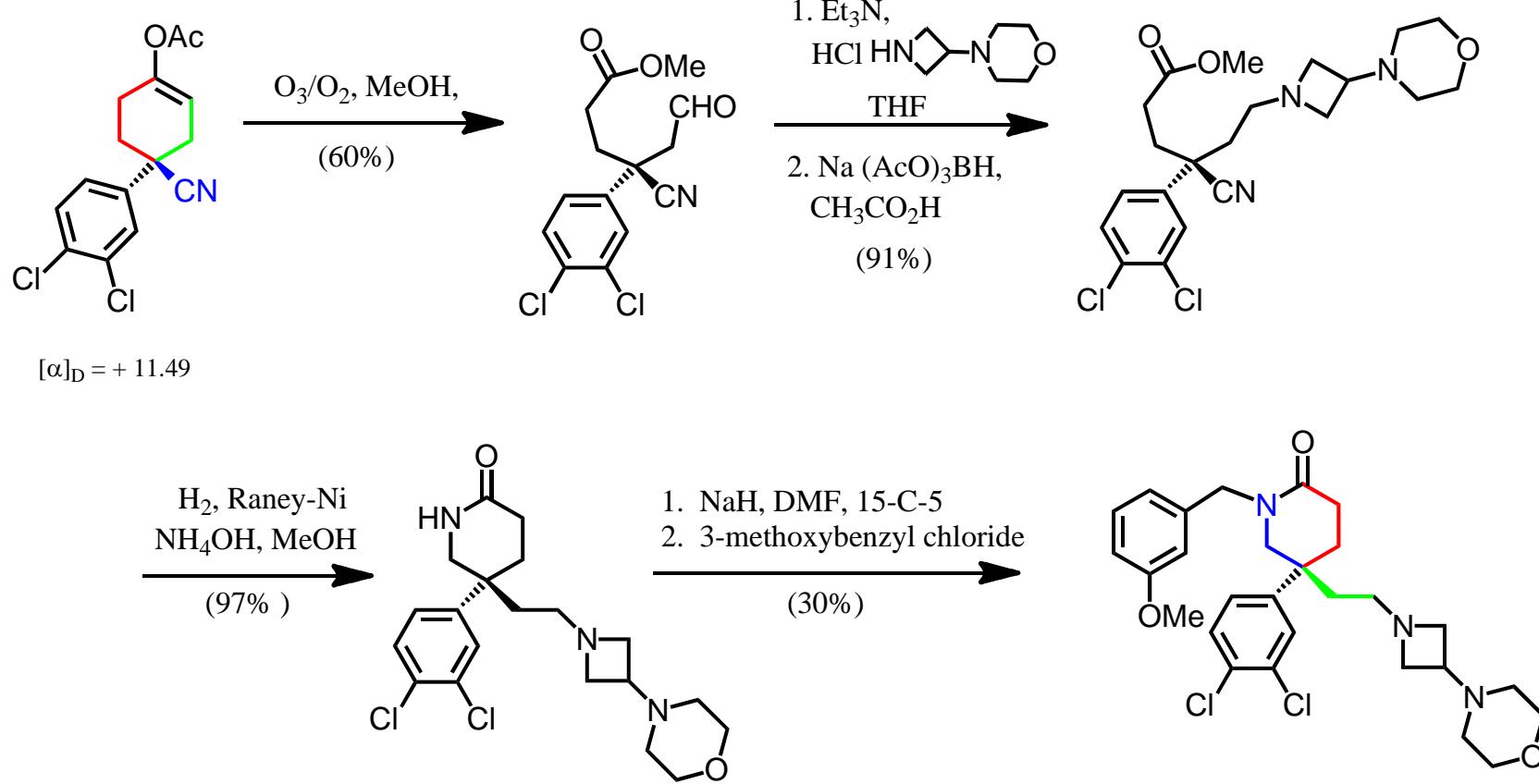
## Deracemisation of enol ester **1** over 4 cycles

Cycle	% e.e. residual enol ester <b>1</b>	% conversion to ketone <b>2</b>
1	92	61
2	86	28
3	87	16
4	>99	12

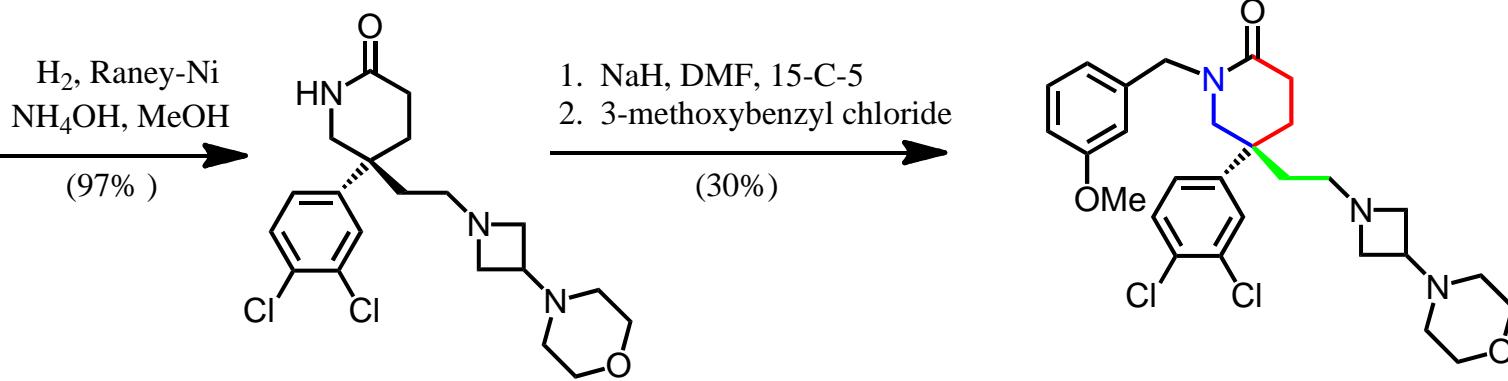
Starting from 400 mg of the racemic material over 4 cycles 330 mg (82%) of the enol ester **1** in >99% e.e was isolated.



# Synthesis of NK-2 antagonists Using Enol Ester

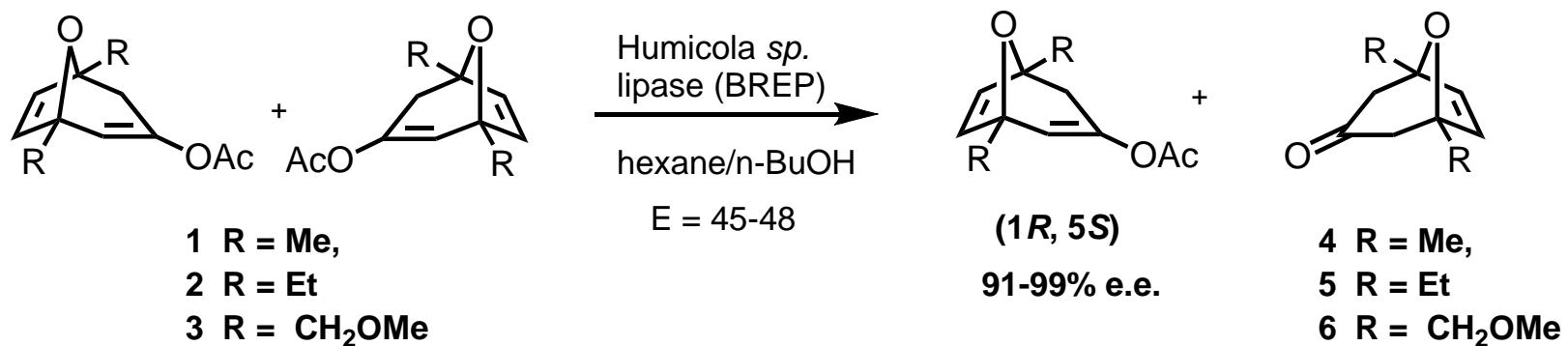


$[\alpha]_D = +11.49$



A.J. Carnell, M. L. Escudero Hernandez, A. Pettman and J.F. Bickley, *Tetrahedron Lett.*, 2000, 41, 6929  
G. Allan, A. J. Carnell, M. L. Escudero Hernandez and A. Pettman, *Tetrahedron*, 2001, 57, 81932.

# Resolution of 8-oxabicyclic enol esters

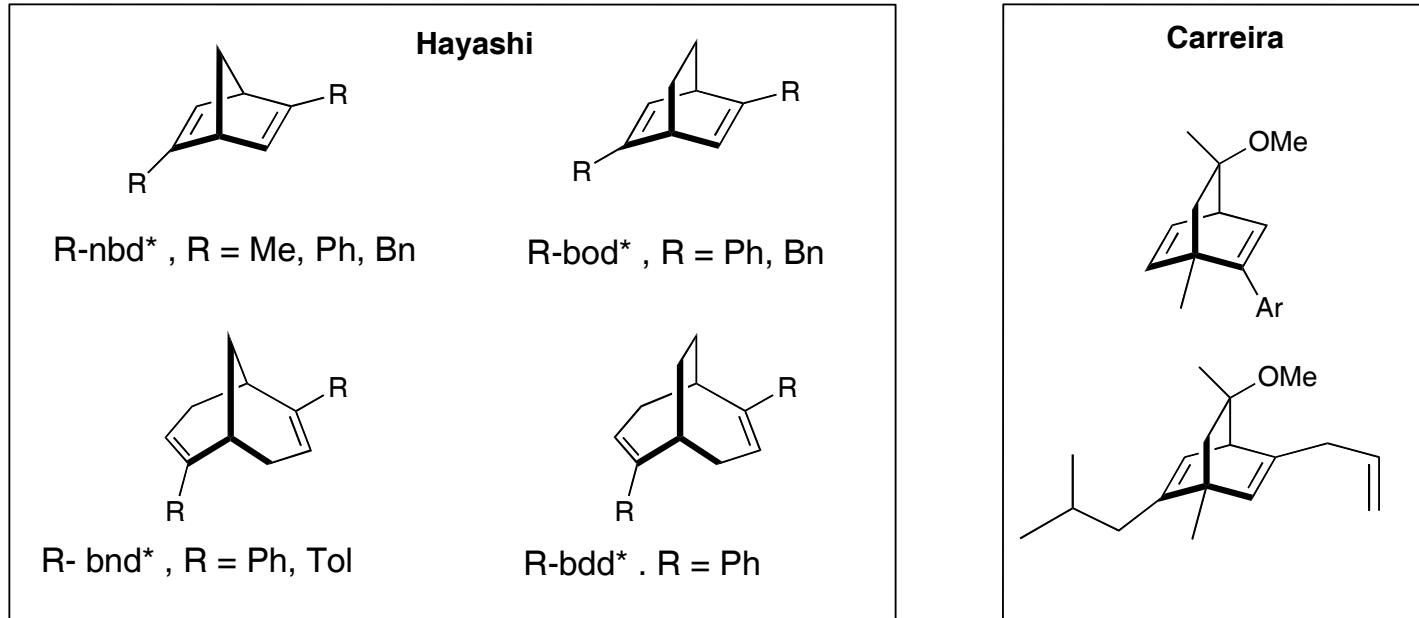


BREP = butanol rinsed enzyme preparation

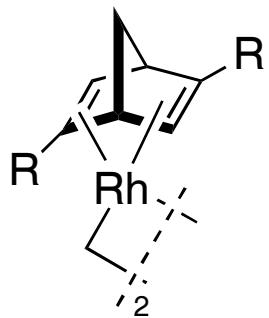
- see Halling *et al.*, J.Chem. Soc., *Chem. Commun.*, 1998, 7, 841

A.J. Carnell, S.A. Swain and J.F. Bickley, *Tetrahedron Lett.*, **1999**, *40*, 8633.

# Chiral diene ligands for asymmetric catalysis

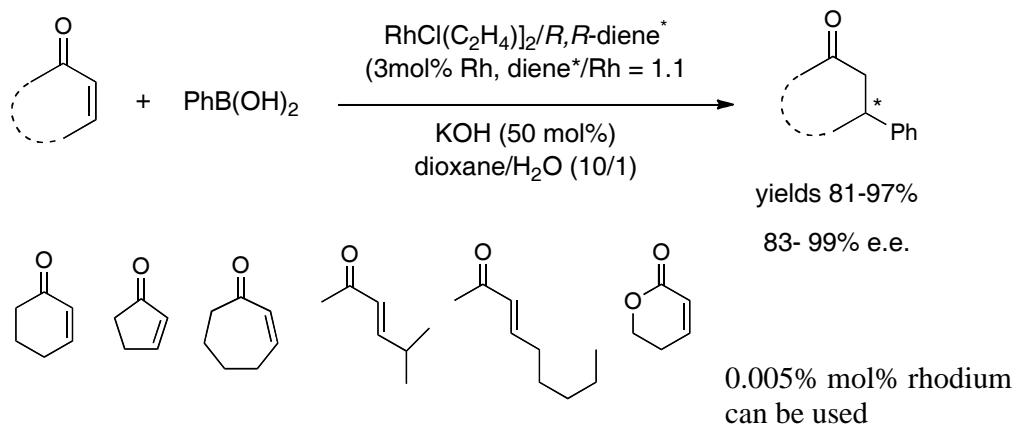


Catalyst:



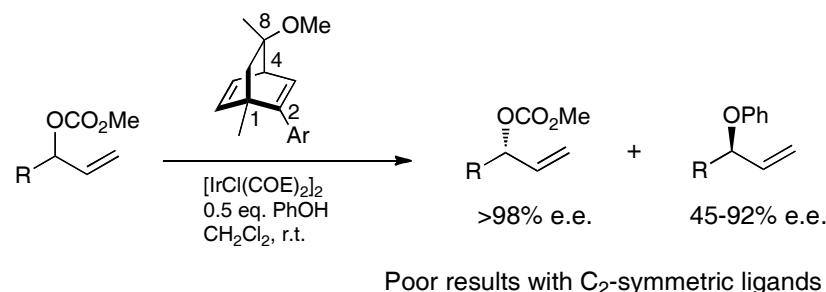
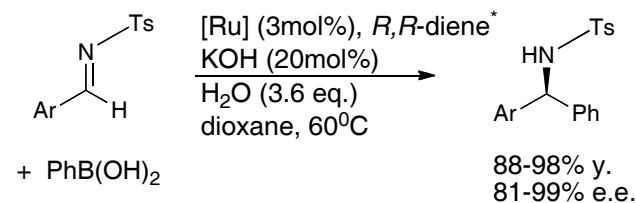
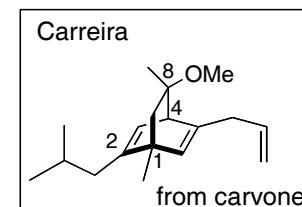
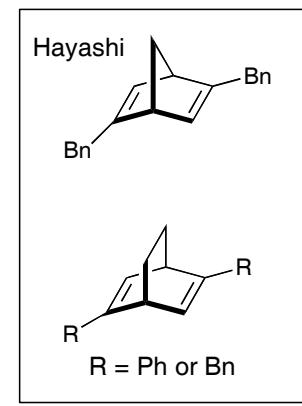
C. Defieber, H. Grutzmacher, E. M. Carreira, *Angew. Chem. Int. Ed.*, 2008, 47, 4482.

# Rh-catalysed asymmetric reactions



**BINAP or chiral phosphoramidite less effective**

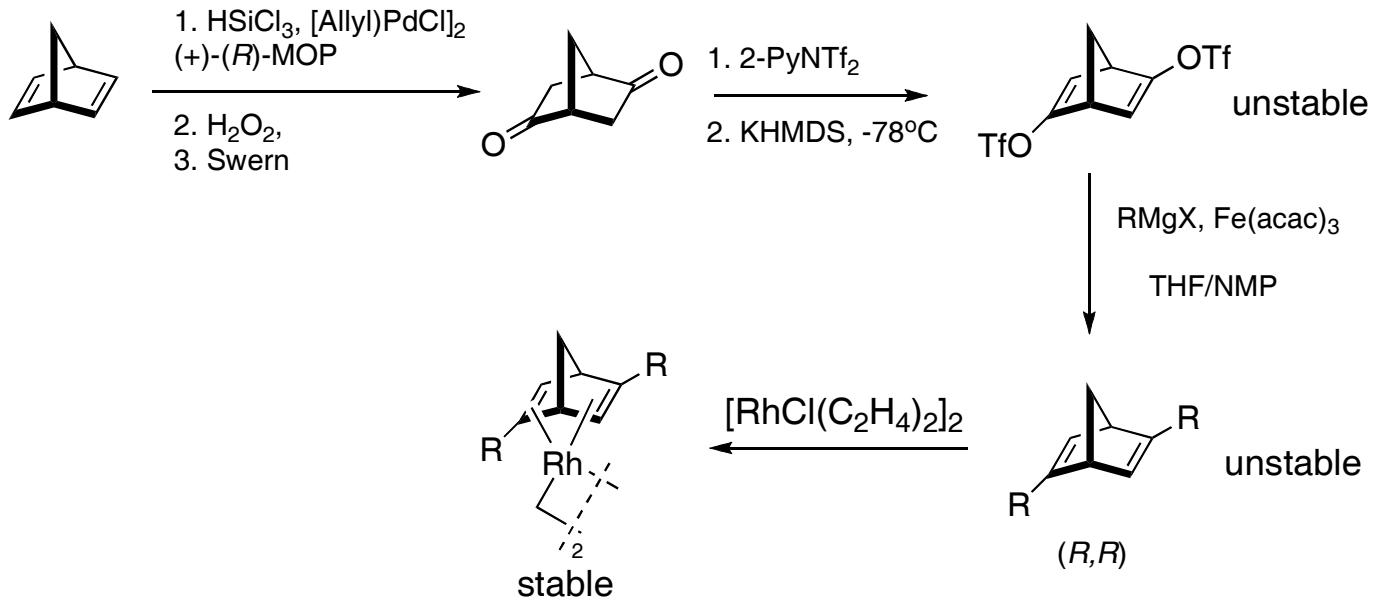
Hayashi *et al.* *J. Org. Chem.*, 2005, 70, 2503-2508; *Tetrahedron: Asymmetry*, 2005, 16, 1673-1679; Guillaume B.-G.; Hayashi, T. *J. Org. Chem.*, 2006, 71, 8957-8960; Carreira E.M *et al.*, *Org. Lett.* 2004, 6, 3873



Hayashi *et al.* *Tetrahedron: Asymmetry*, 2005, 16, 1673; *J. Org. Chem.*, 2006, 71, 8957.

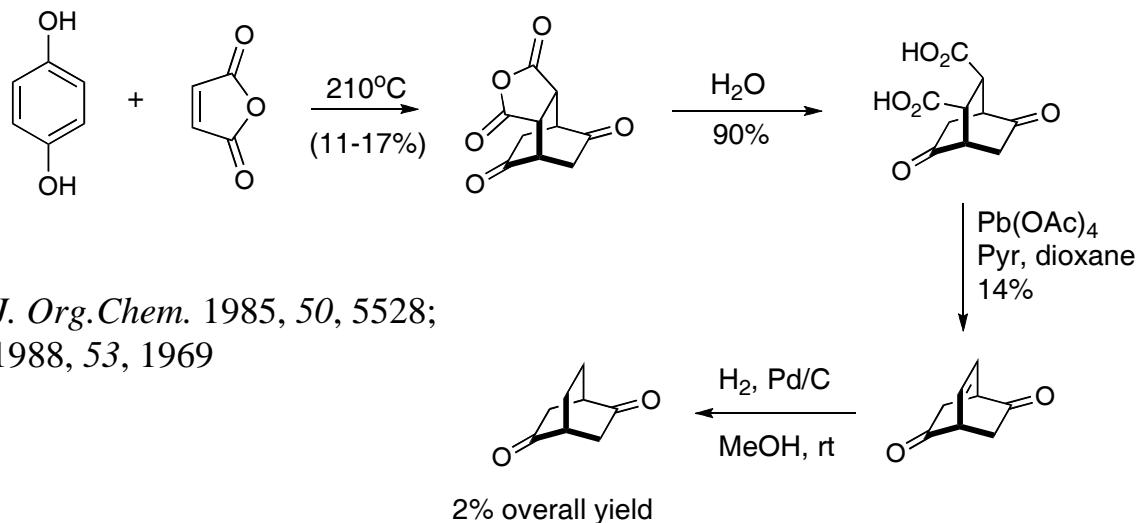
Carreira *et al.* *J. Am. Chem. Soc.* 2004, 126, 1628-1629

# Chiral Synthesis of [2.2.1] Bicyclic Diene Ligand

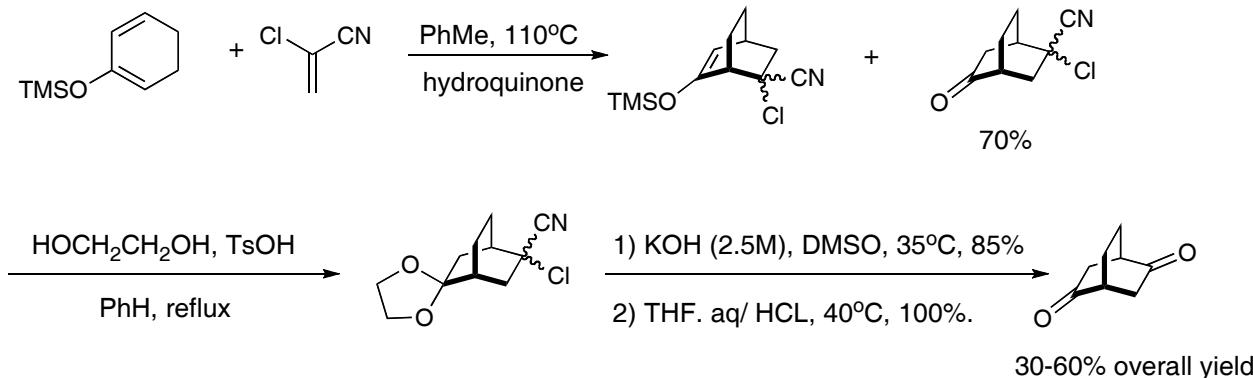


Guillaume B.-G.; Hayashi, T. *J. Org. Chem.*, 2006, 71, 8957.

# Synthesis of bicyclic [2.2.2] diketone

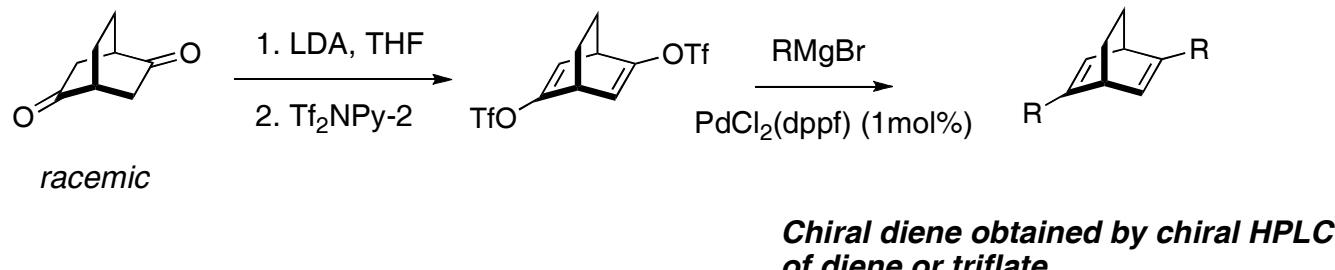
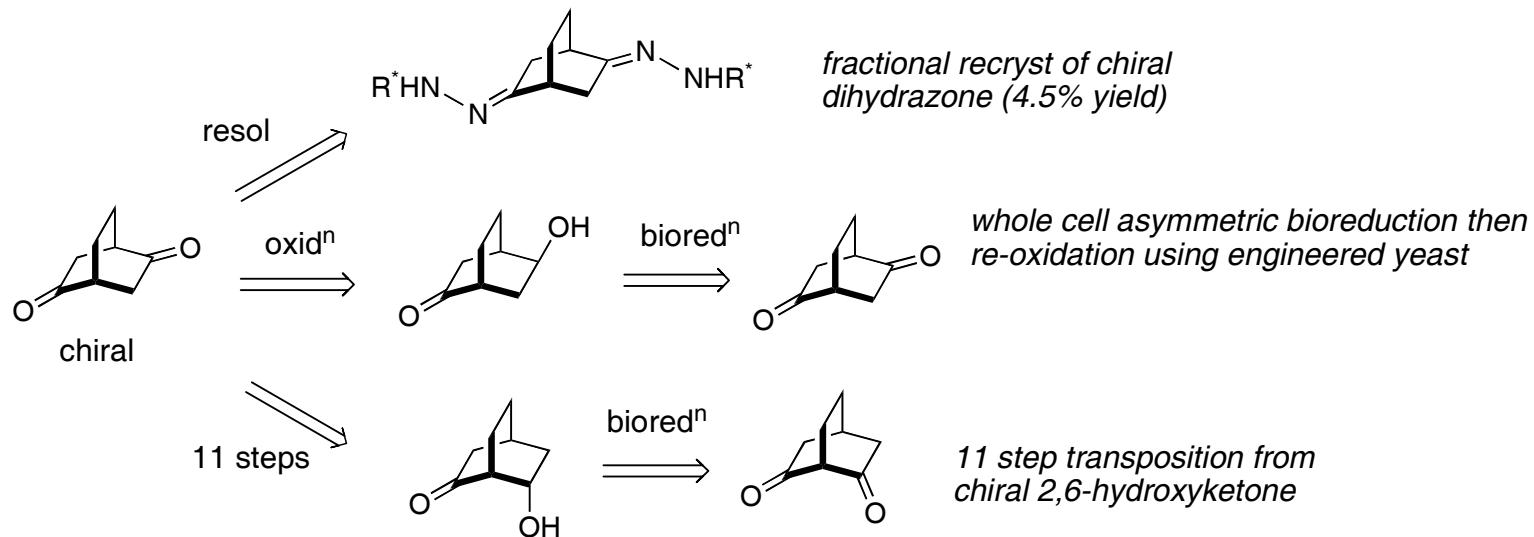


Paquette *et al.* *J. Org. Chem.* 1985, 50, 5528;  
*J. Org. Chem.* 1988, 53, 1969



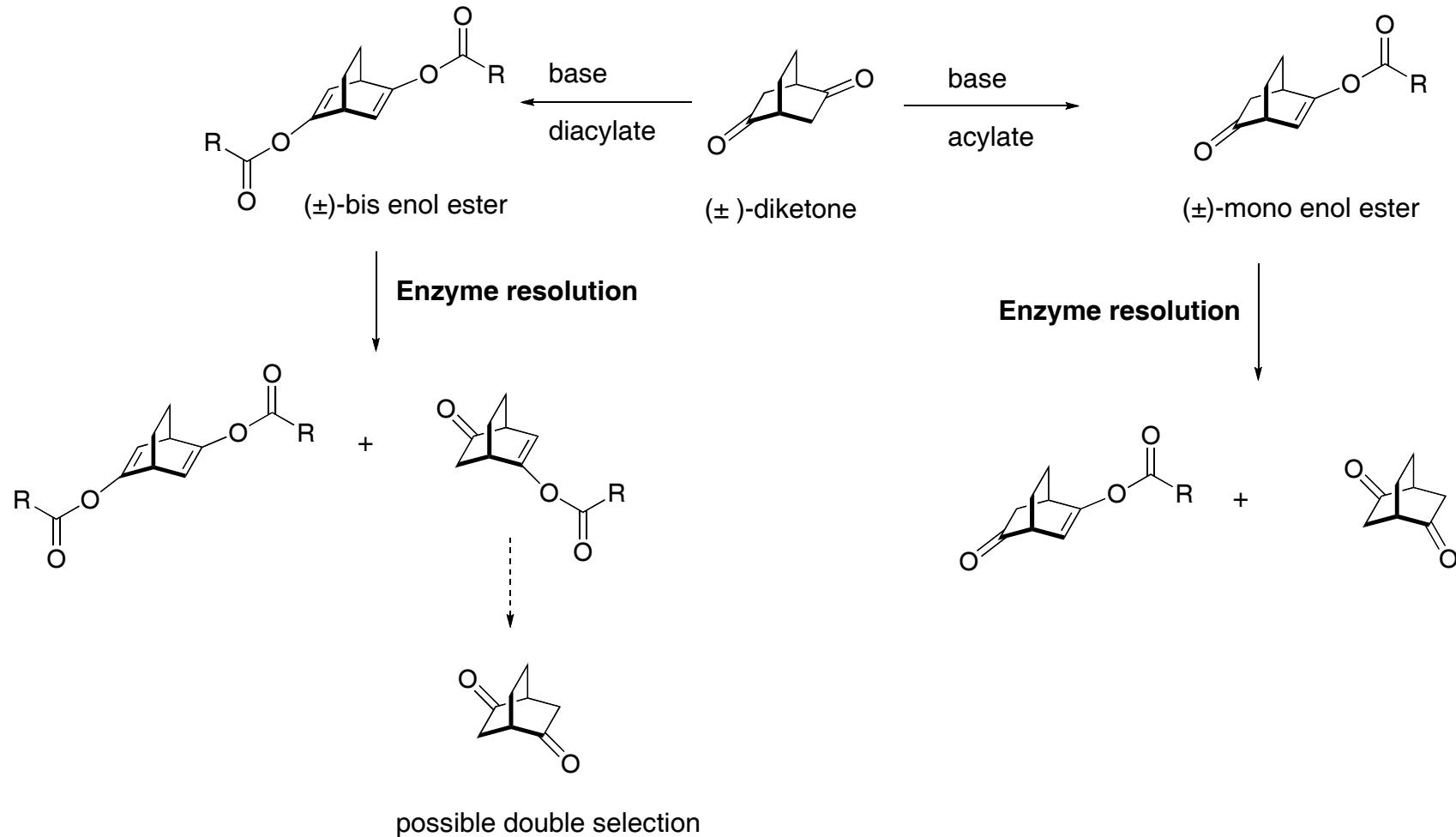
N. H. Werstiuk *et al.* *Can. J. Chem.* 1992, 70, 974. .

# Accessing the Chiral Hayashi [2.2.2] Diene

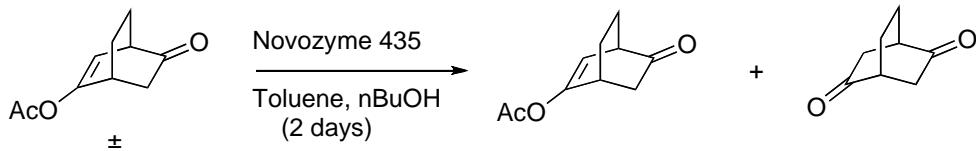


- Otomaru *et al.*, *J. Org. Chem.*, 2005, 70, 2503  
 Friberg *et al.*, *Org. Biomol. Chem.*, 2006, 4, 2304.  
 Almqvist *et al.*, *J. Org. Chem.*, 1996, 61, 3794.

## Options for enzyme resolution of [2.2.2] bicyclic diketone via enol esters



# Enzyme Screening



Entry	Enzyme	% Conversion	ee% enoester (enoester left%)	E
1	<i>Canadida Rugosa</i> (Chirazyme L-3, Europa Lipase AY)	0	-- (100)	
2	<i>Pseudomonas fluorescens</i> (PFL, Amano AK30)	0	-- (100)	
3	<b><i>Canadida antartica</i> (CAL-B, Novozyme 435)</b>	11.8	12.4 (88.2)	<b>17</b>
4	<i>Pseudomonas cepacia</i> (PCL, Lipase PS Amano)	0	-- (100)	
5	Lipase Amano 10	0	-- (100)	
6	<i>Penicillium camembertii</i> (Lipase G Amano 50)	0	-- (100)	
7	<i>Rhizopus oryzae</i> (Lipase F AP-15)	0	-- (100)	
8	<i>Mucor Meihei</i> (Lipase M Amano, Europa lipase 15(RS))	0	-- (100)	
9	<b>Chirazyme L-2 (Cal-B)</b>	11.3	14.8 (88.7)	<b>16</b>
10	Europa Lipase 20	3.5	5.0 (98.5)	

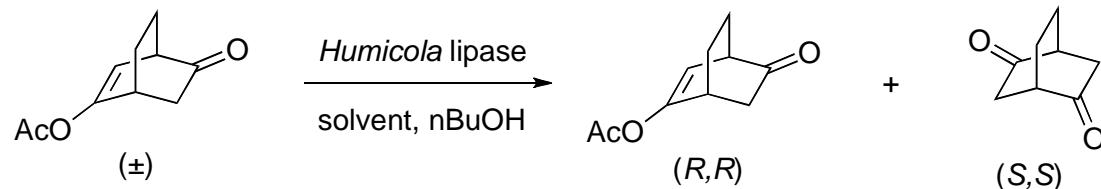
Reactions in Buffer or hexane less selective

## Silica-absorbed *Humicola* Lipase Solvent Screen

entry	Solvents	Conversion	E.E.	E value
1	Hexane	28	26	6
2	Toluene	No reaction	-	-
3	TBME	41	43	6
4	Et <sub>2</sub> O	27	13	2
5	DME	No reaction	-	-
6	Acetone	No reaction	-	-

entry	Solvents	Conversion	E.E.	E value
1	Hexane	28	26	6
2	Cyclohexane	31	28	5
3	Heptane	29	27	6
4	Pentane	31	37	15
5	Petroether	27	26	5

# *Humicola* Lipase - support screening



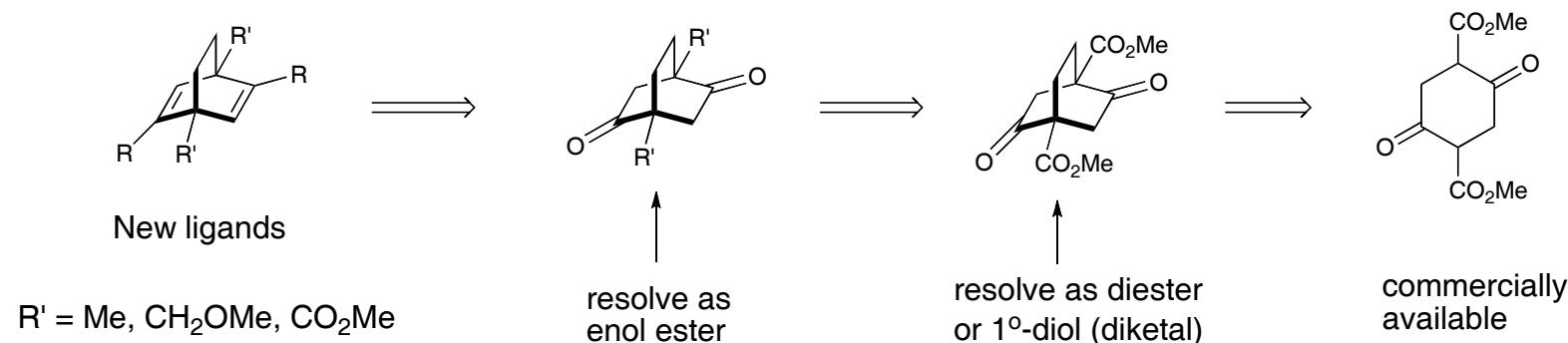
Entry	Solvent	Enzyme Form	E Value
1	toluene	Silica adsorbed	No reaction
2	TBME	Silica absorbed	6
3	Et <sub>2</sub> O	Silica absorbed	2
4	Hexane/cyclohexane/heptane/petroleum ether	Silica absorbed	5/6
5	Pentane	Silica absorbed	15
6	Pentane	lyophilised	6
7	Pentane	Protein coated microcrystals	10
8	Pentane	Sol gel entrapped	20
9	Pentane	GS01-immobilised	23
10	Pentane	GS02- immobilised	40*
<b>11</b>	<b>Pentane</b>	<b>GS03-immobilised</b>	<b>270</b>
12	Pentane	Accurel	177
13	Pentane	Accurel (dried)	33
14	Pentane	Eupergit	4

**Increasing  
Hydrophobicity  
Of GS0X  
surface**

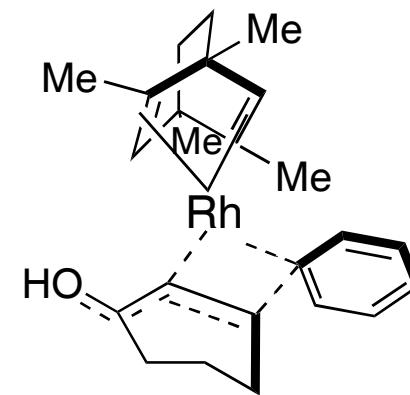
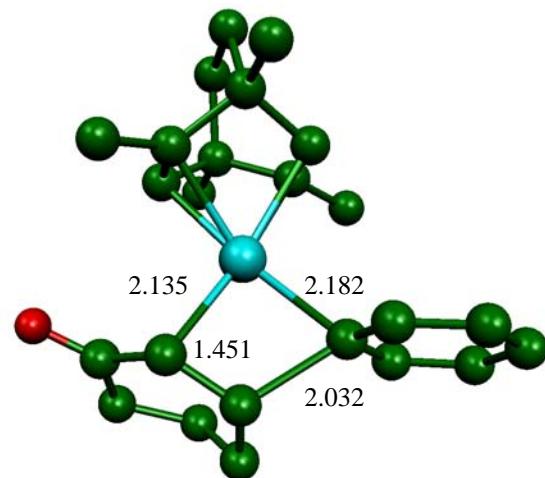
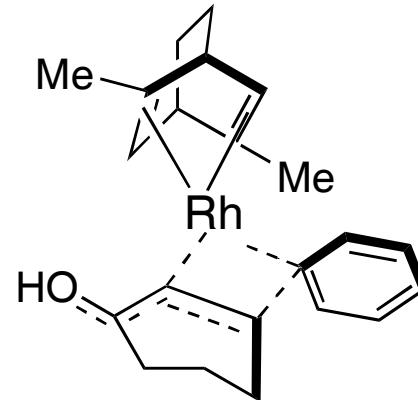
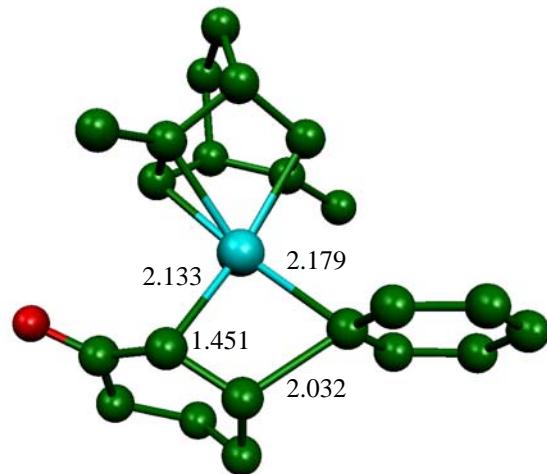
## GS0X modified silica supports from PhosphonicS Ltd

*Luo and Carnell, J. Org. Chem., 2010, 75, 2057-2060*

# New Range of C<sub>2</sub>-Symmetric Chiral Diene Ligands

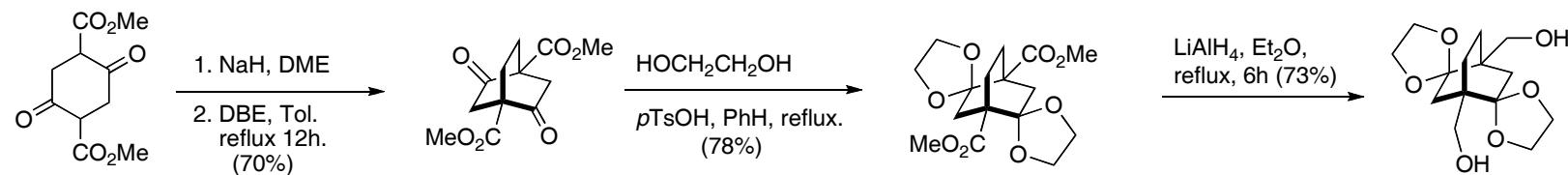


## Transition states for conjugate additions

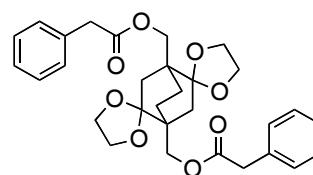


DFT calculations using PCGAMESS predict similar reaction rates and e.e.'s  
N. Berry, A. J. Carnell, Y. Luo unpublished results

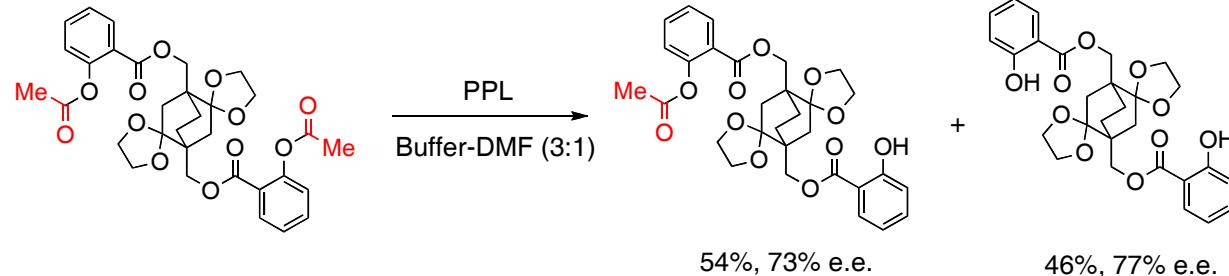
# Resolution of Diol Diketal



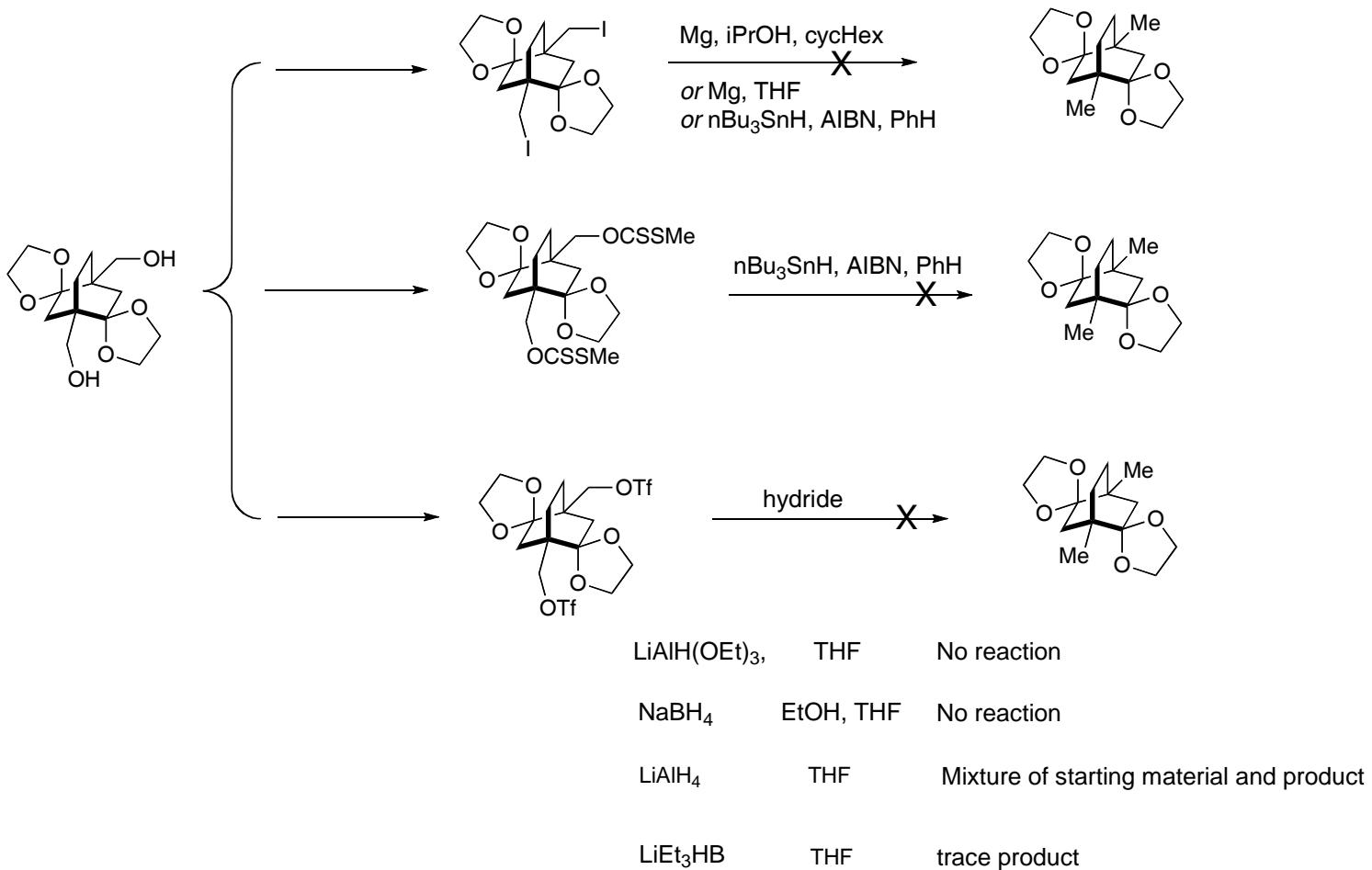
Substrates - Introduction of chromophore into hydrolysable group



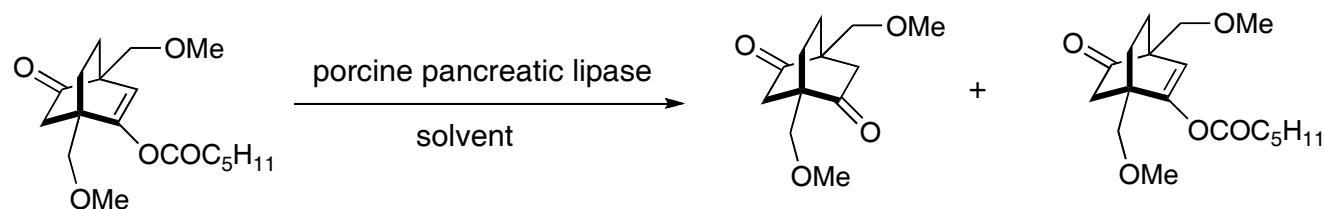
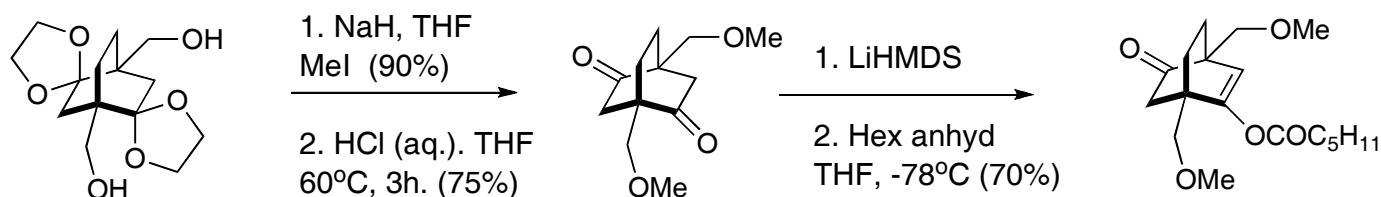
Penicillin acylase/buffer - no reaction



# Attempted Deoxygenation



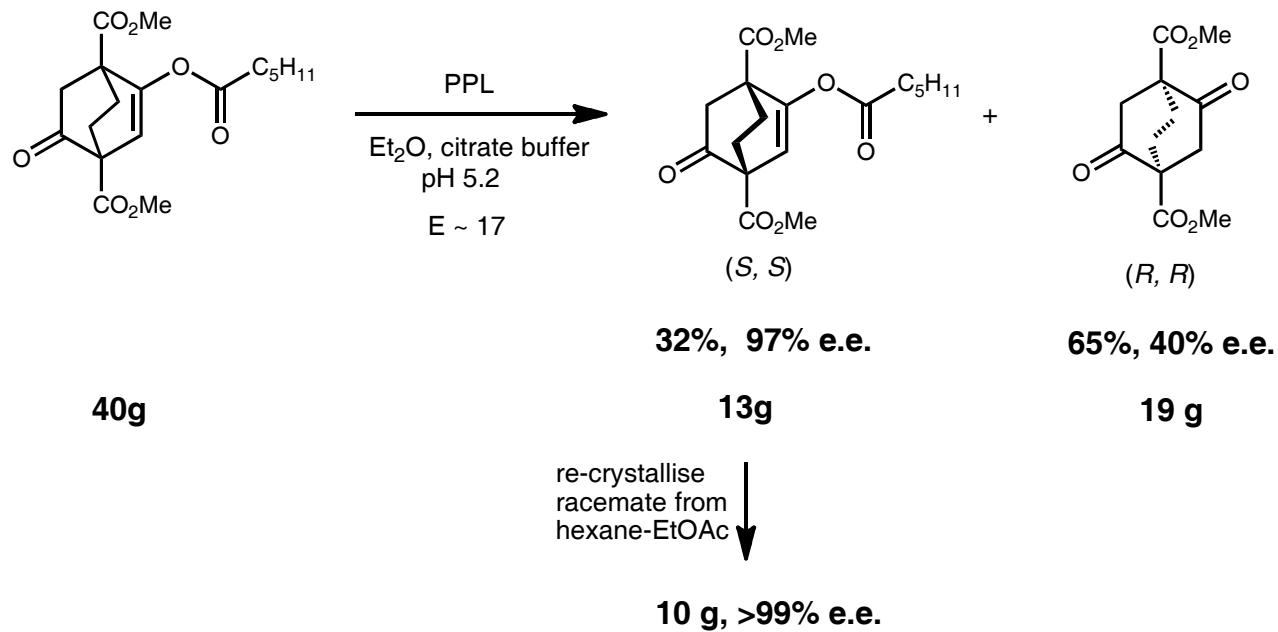
# Resolution of Methoxy Ether Enol Hexanoate



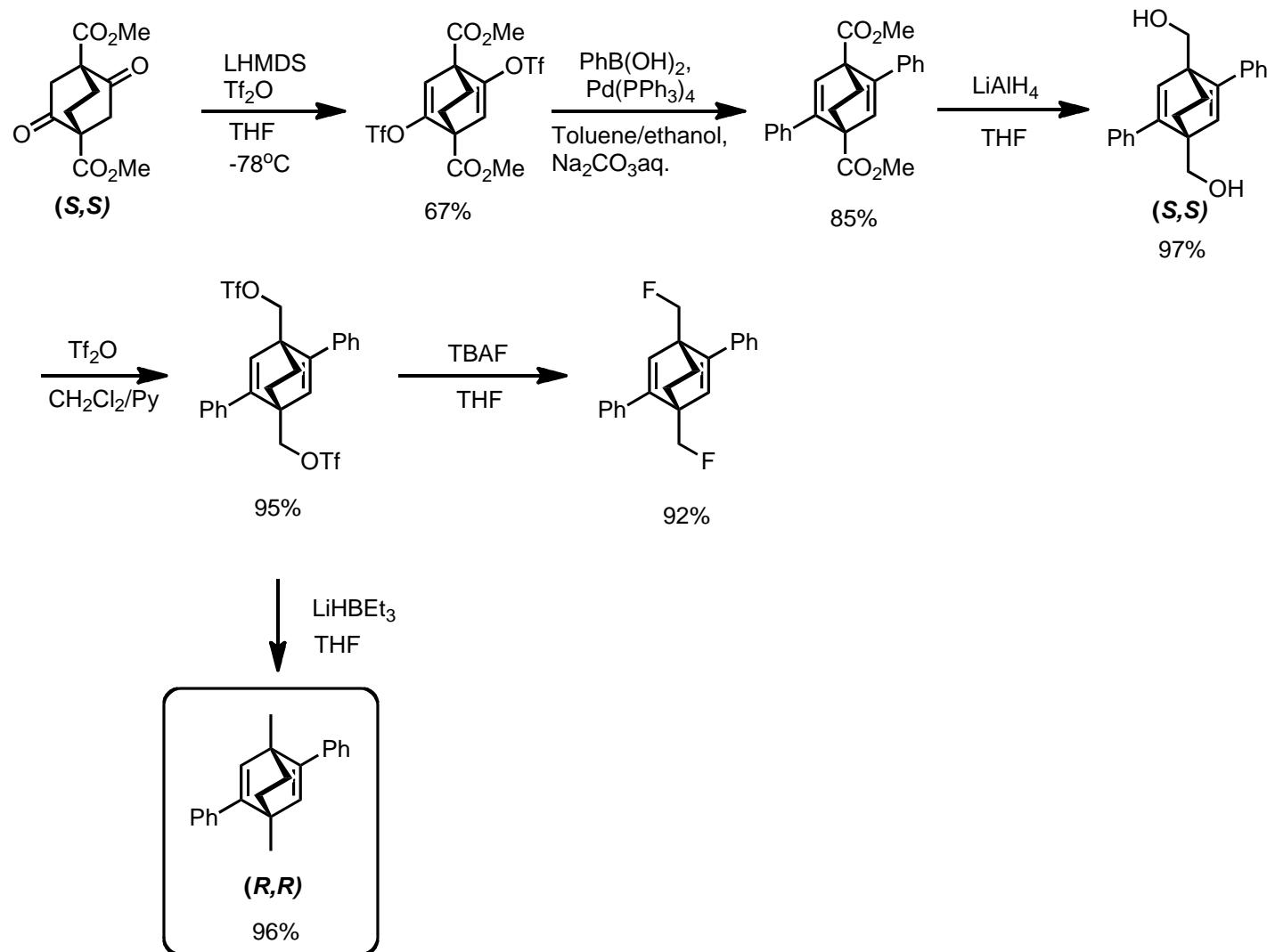
Solvent		Buffer-co-solvent (4:1)			
Toluene	Buffer (pH 7.5)	DME	DMF	DMSO	
E value	nr	4-7	1	25-30	2

**...too many steps!**

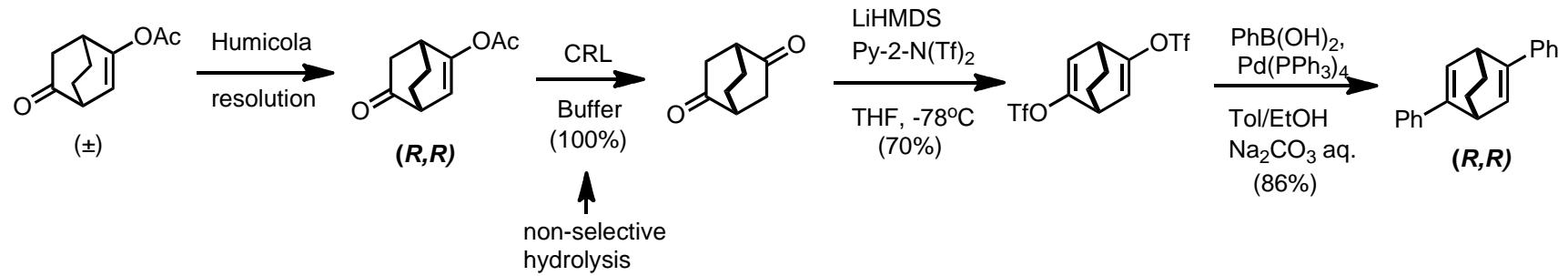
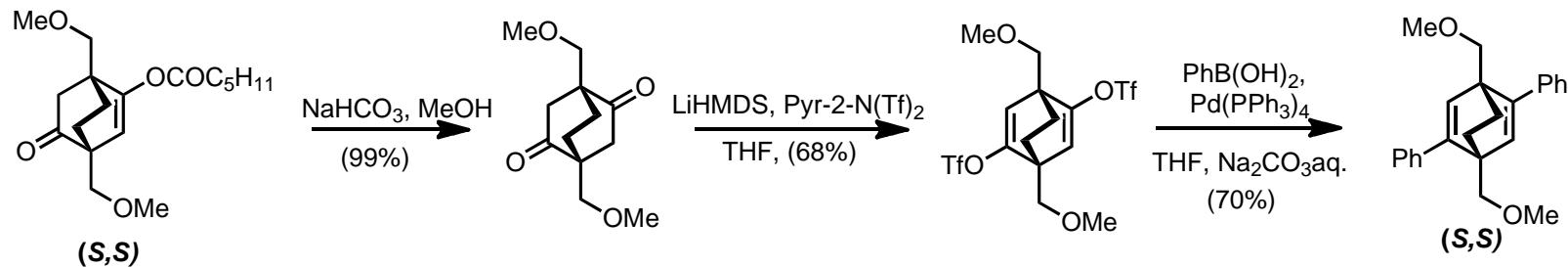
# Resolution of Dimethyl Ester Enol Hexanoate



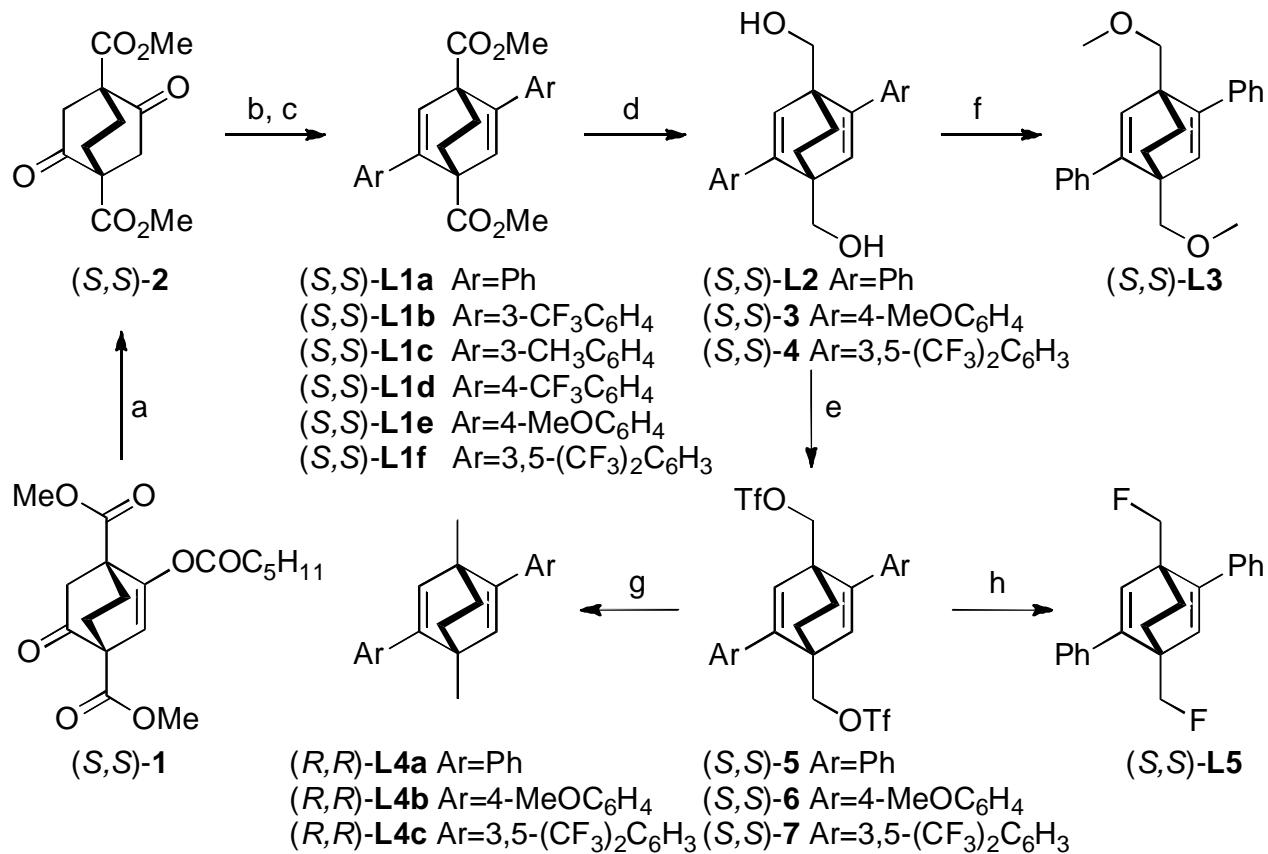
## Diene Ligand Synthesis



# Further Ligand Synthesis

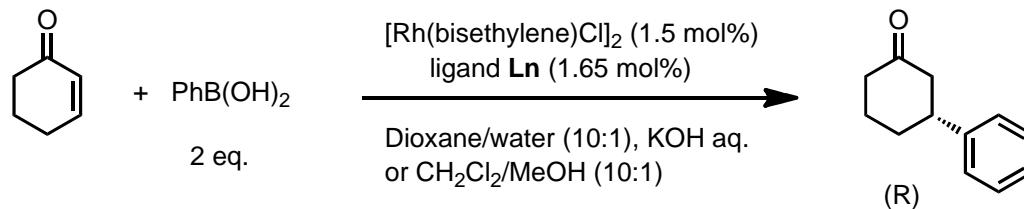


# Ligand Synthesis



a)  $\text{Na}_2\text{CO}_3$ ,  $\text{MeOH}$  (99%); b) LHMDS,  $\text{Tf}_2\text{O}$ ,  $\text{THF}$ ,  $-78^\circ\text{C}$  (70%); c)  $\text{ArB}(\text{OH})_2$ ,  $\text{Pd}(\text{PPh}_3)_4$ ,  $\text{Tol/EtOH/aq. Na}_2\text{CO}_3$ , rt. (95%)  
d)  $\text{LiAlH}_4$ ,  $\text{THF}$ , (99%); e)  $\text{Tf}_2\text{O}$ ,  $\text{CH}_2\text{Cl}_2/\text{Pyr}$ ,  $-78^\circ\text{C} - \text{rt}$  (99%); f)  $\text{NaH}$ ,  $\text{MeI}$ ,  $\text{THF}$  (95%); g)  $\text{LiHBET}_3$ ,  $\text{THF}$ , rt (99%);  
h)  $\text{TBAF}$ ,  $\text{THF}$ ,  $40^\circ\text{C}$  (95%).

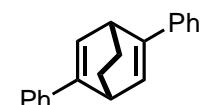
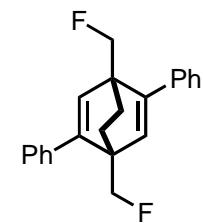
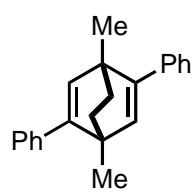
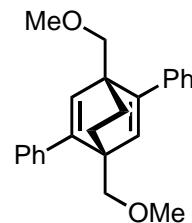
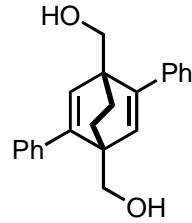
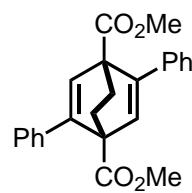
# Catalytic Asymmetric Conjugate Addition



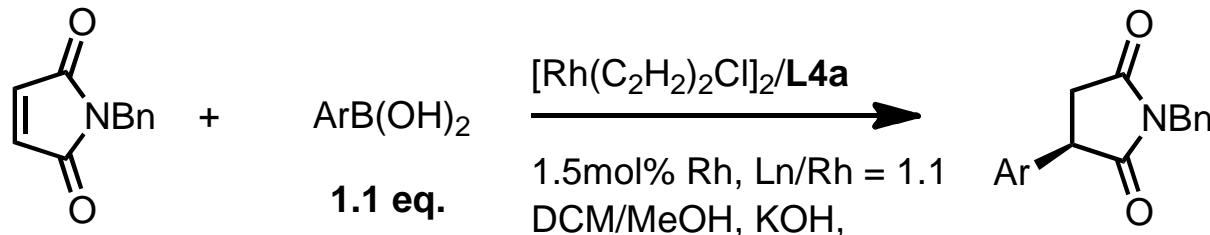
Entry <sup>[a]</sup>	<b>Ln</b>	Ligand		Yield (%)	e.e. (%)
		1,4-R	2,5-Aryl		
1	<b>L1a</b>	$\text{CO}_2\text{Me}$	Ph	95	89
2	<b>L2</b>	$\text{CH}_2\text{OH}$	Ph	94	97
3	<b>L3</b>	$\text{CH}_2\text{OMe}$	Ph	95	88
4	<b>L4a</b>	Me	Ph	96	98
5 <sup>[b]</sup>	<b>L4a</b>	Me	Ph	98	99
6	<b>L5</b>	$\text{CH}_2\text{F}$	Ph	96	98
7 <sup>[c]</sup>	<b>L6</b>	H	Ph	97	96

← 1.2 eq.  $\text{PhB(OH)}_2$  with **L4a**

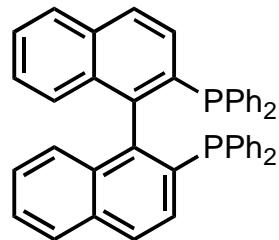
## Ligands **Ln**



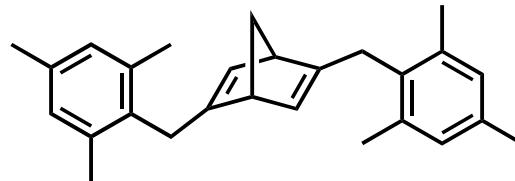
# Asymmetric Conjugate Addition to N-benzyl maleimide



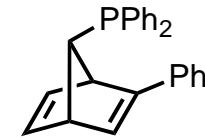
9 examples  
92-99% yield  
91-99% e.e.



70%, 58% e.e.



88%, 69% e.e.

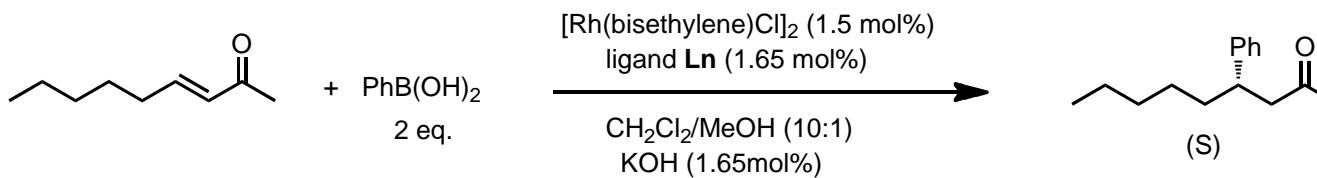


98%, 93% e.e.

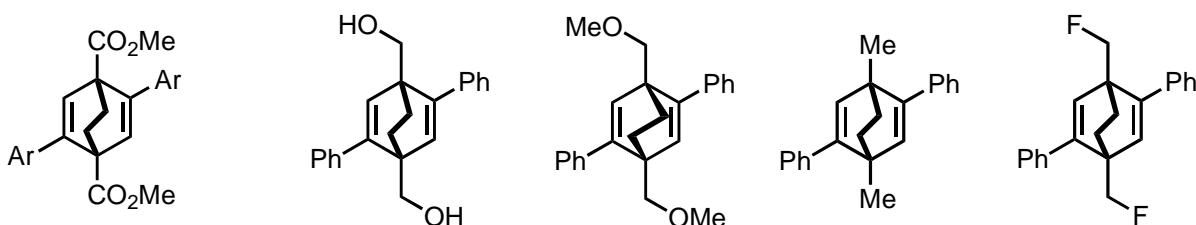
all used 3 eq. of  $\text{PhB}(\text{OH})_2$

Luo and Carnell, *Angew. Chem. Int. Ed.*, 2010, 49, 2750-2754

# Acyclic substrate ligand optimisation

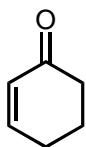
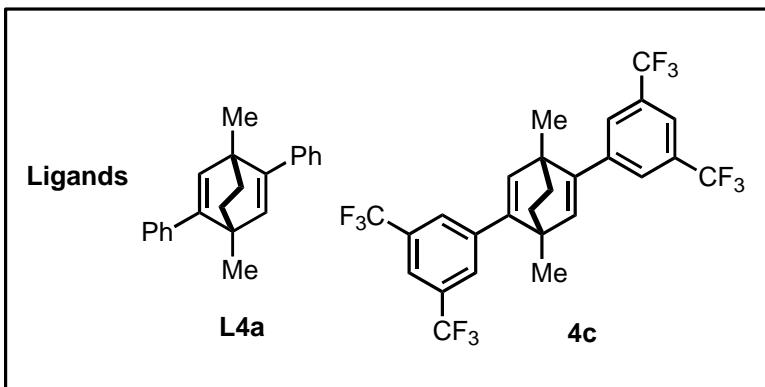
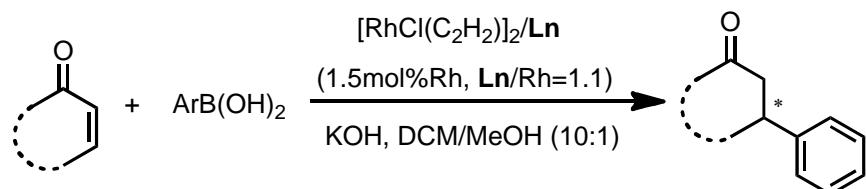


Entry	<b>Ln</b>	Ligand		Yield (%)	e.e. (%)	1.2 eq. PhB(OH) <sub>2</sub> with <b>L4a</b>
		1,4-R	2,5-Aryl			
1	<b>L4a</b>	Me	Ph	99	52	←
2	<b>L2</b>	CH <sub>2</sub> OH	Ph	98	57	
3	<b>L3</b>	CH <sub>2</sub> OMe	Ph	98	71	
4	<b>L5</b>	CH <sub>2</sub> F	Ph	95	67	
5	<b>L1a</b>	CO <sub>2</sub> Me	Ph	95	75	
6	<b>L1b</b>	CO <sub>2</sub> Me	3-CF <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	93	83	
7	<b>L1c</b>	CO <sub>2</sub> Me	3-CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	98	73	
8	<b>L1d</b>	CO <sub>2</sub> Me	4-CF <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	95	82	
9	<b>L1e</b>	CO <sub>2</sub> Me	4-MeOC <sub>6</sub> H <sub>4</sub>	99	74	
10	<b>L1f</b>	CO <sub>2</sub> Me	3,5-(CF <sub>3</sub> ) <sub>2</sub> C <sub>6</sub> H <sub>3</sub>	43	89	

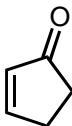


- L1a** Ar=Ph
- L1b** Ar=3-CF<sub>3</sub>C<sub>6</sub>H<sub>4</sub>
- L1c** Ar=3-CH<sub>3</sub>C<sub>6</sub>H<sub>4</sub>
- L1d** Ar=4-CF<sub>3</sub>C<sub>6</sub>H<sub>4</sub>
- L1e** Ar=4-MeOC<sub>6</sub>H<sub>4</sub>
- L1f** Ar=3,5-(CF<sub>3</sub>)<sub>2</sub>C<sub>6</sub>H<sub>3</sub>

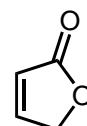
# Tuning electron density of diene ligand for acyclic enones



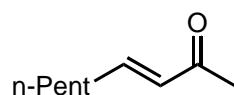
**4a** 98%, 99% e.e.  
**4c** 92%, 98% e.e.



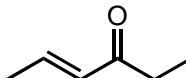
**4a** 100%, 99% e.e.  
**4c** 80%, 96% e.e.



**4a** 95%, 98% e.e.  
**4c** 76%, 92% e.e.

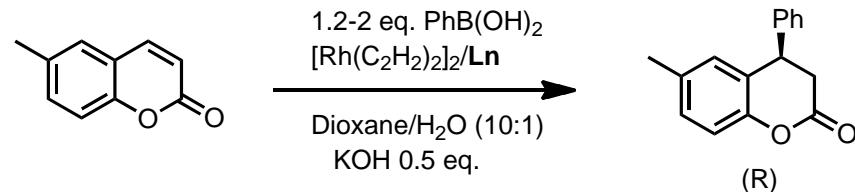


**4a** 99%, 52% e.e.  
**4c** 99%, 97% e.e.



**4a** 99%, 67% e.e.  
**4c** 99%, 95% e.e.

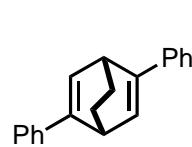
## Asymmetric conjugate addition to 6-methylcoumarin



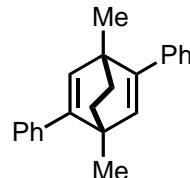
Entry	Ligand	temp. (°C)	Time (hr)	conv. (%)	Yield (%)	e.e. (%)
1	<b>L6</b>	30	24	<5	--	n.d.
2	<b>L6</b>	50	6	40	<b>32</b>	98
3	<b>L4a</b>	30	6	55	<b>50</b>	98
4	<b>L4a</b>	50	6	85	<b>72</b>	98
5	<b>L4b</b>	30	3	100	<b>95</b>	98

← 1.2 eq. PhB(OH)<sub>2</sub>  
25 mol% KOH

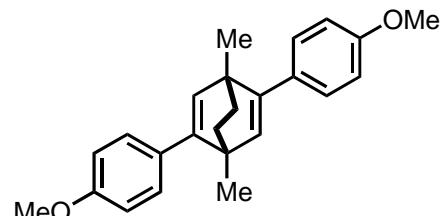
### Ligands



**L6** Hayashi Ligand

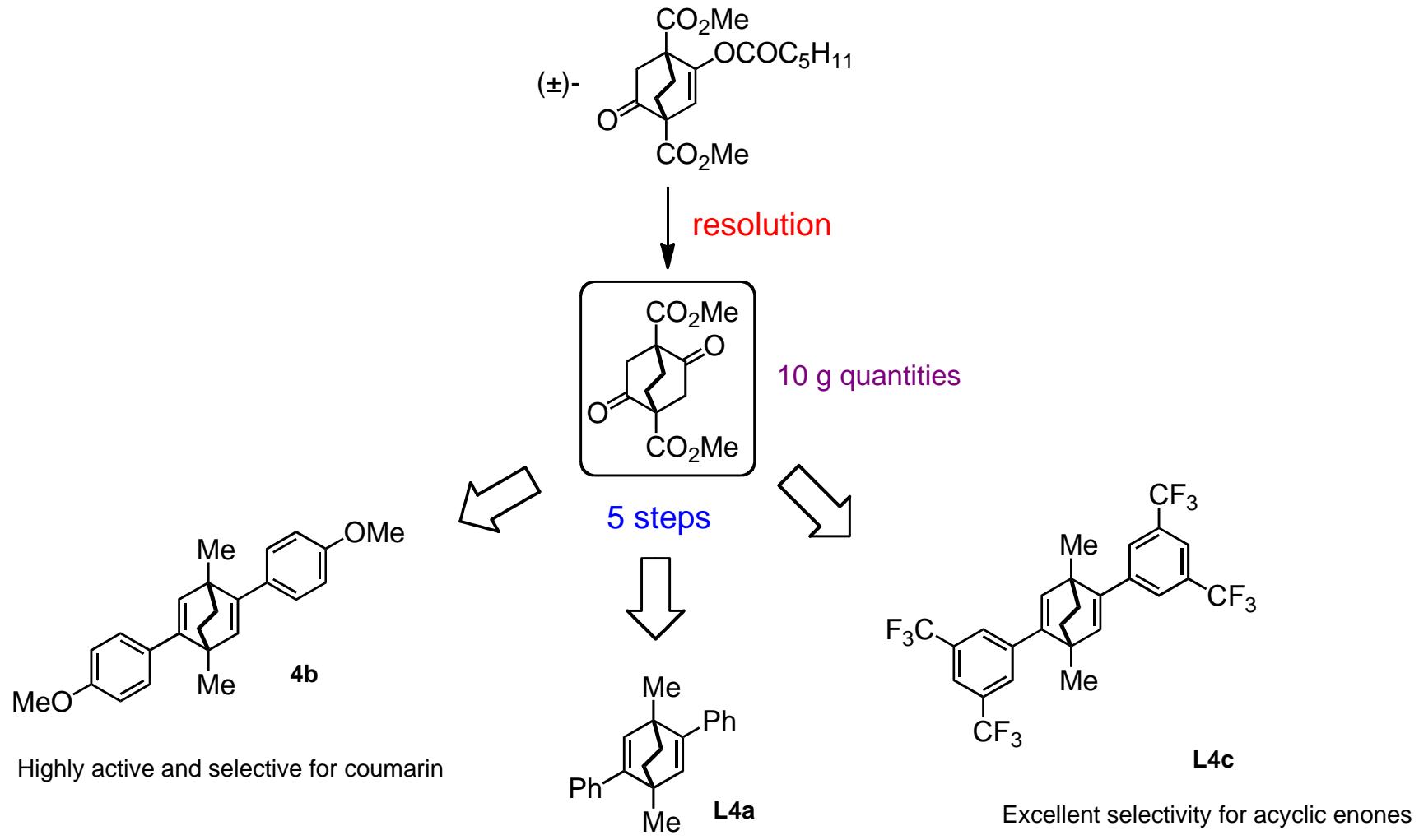


**L4a**



**4b**

# Chemoenzymatic Synthesis of New Bicyclic Ligands



Faster than Hayashi's  
and others (1hr at rt vs 3hrs at 30°C)  
minimal deboronation - only 1.2 eq.  $\text{PhB(OH)}_2$  required cf  
2 eq. or more with Hayashi diene

## Acknowledgements

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Graham Allan

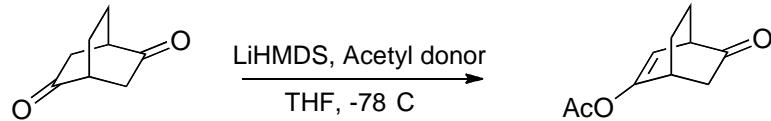
Alan Pettman

John Whittall

PhosphonicS ltd

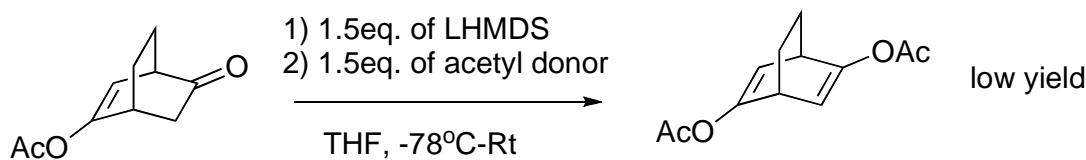
Funding: Pfizer, Dorothy Hodgkin Postgraduate Award (Y.L.), EPSRC (DTAs), BBSRC (PDRA to G.A.).

# Synthesis of Enol Ester

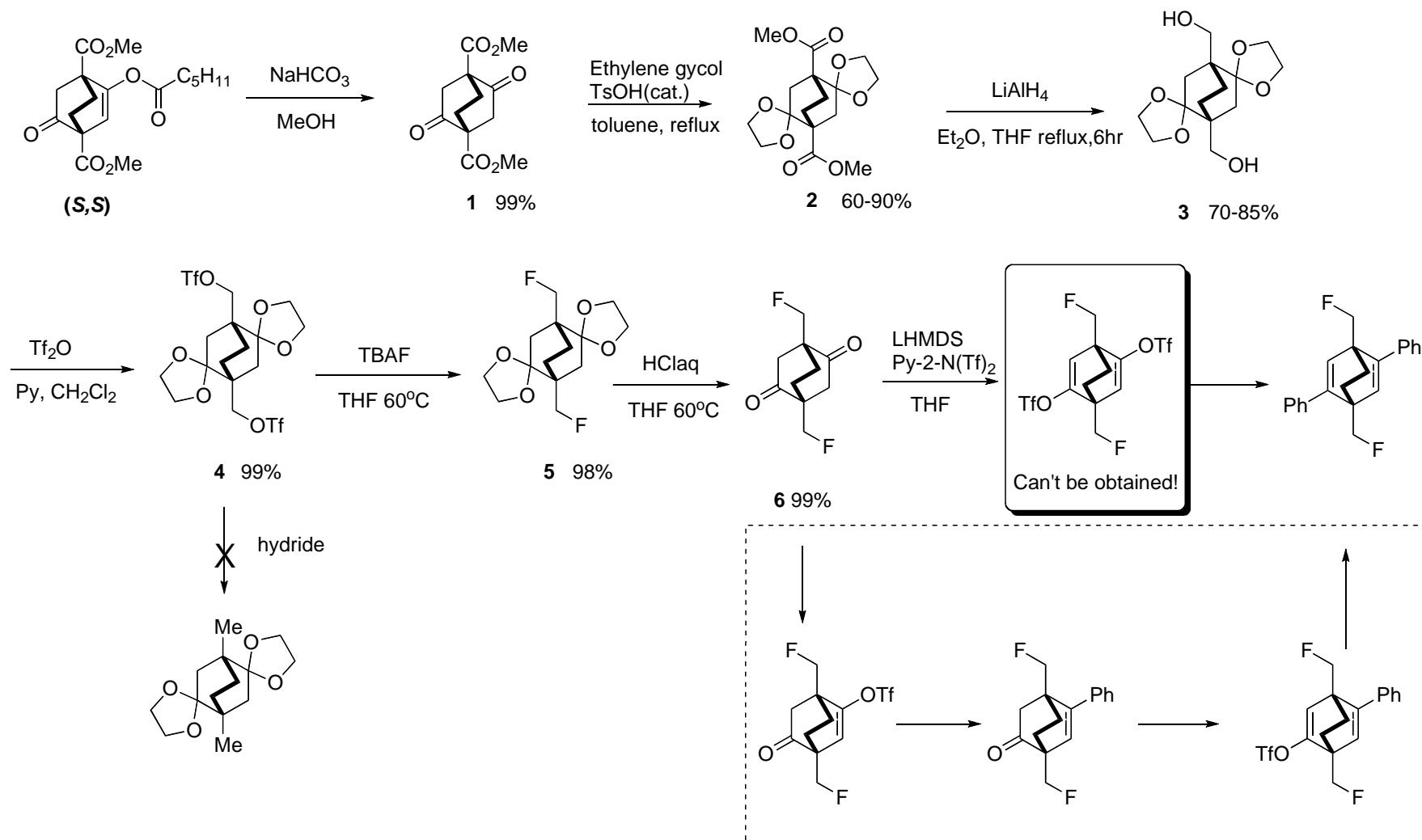


Conditions screening for synthesis of enolacetate

Base	Quantity	Time	Acetyl donor	yield	Temperature
LHMDS	3eq.	12hr	Acetyl chloride	39.7	-78
LHMDS	3eq.	48hr	Acetyl chloride	42.9	-78
LDA	3eq.	12hr	Acetyl chloride	7.9	-78
LHMDS	3eq.	48hr	Acetyl chloride	47.6	-78 - rt
LDA	3eq.	48hr	Acetic anhydride	12.7	-78 - rt
LHMDS	3eq.	48hr	Acetic anhydride	47.6	-78 - rt
<b>LHMDS</b>	<b>1.5eq.</b>	<b>48hr</b>	<b>Acetyl chloride</b>	<b>90</b>	<b>-78 - rt</b>
LHMDS	1.5eq.	48hr	Acetic anhydride	88	-78 - rt



# Diene Ligand Synthesis



# X-ray Structure of Rh Complex with ligand L4a

