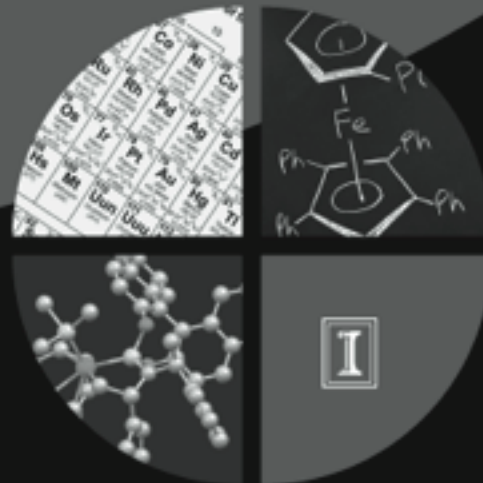


Catalytic Modification of Arenes

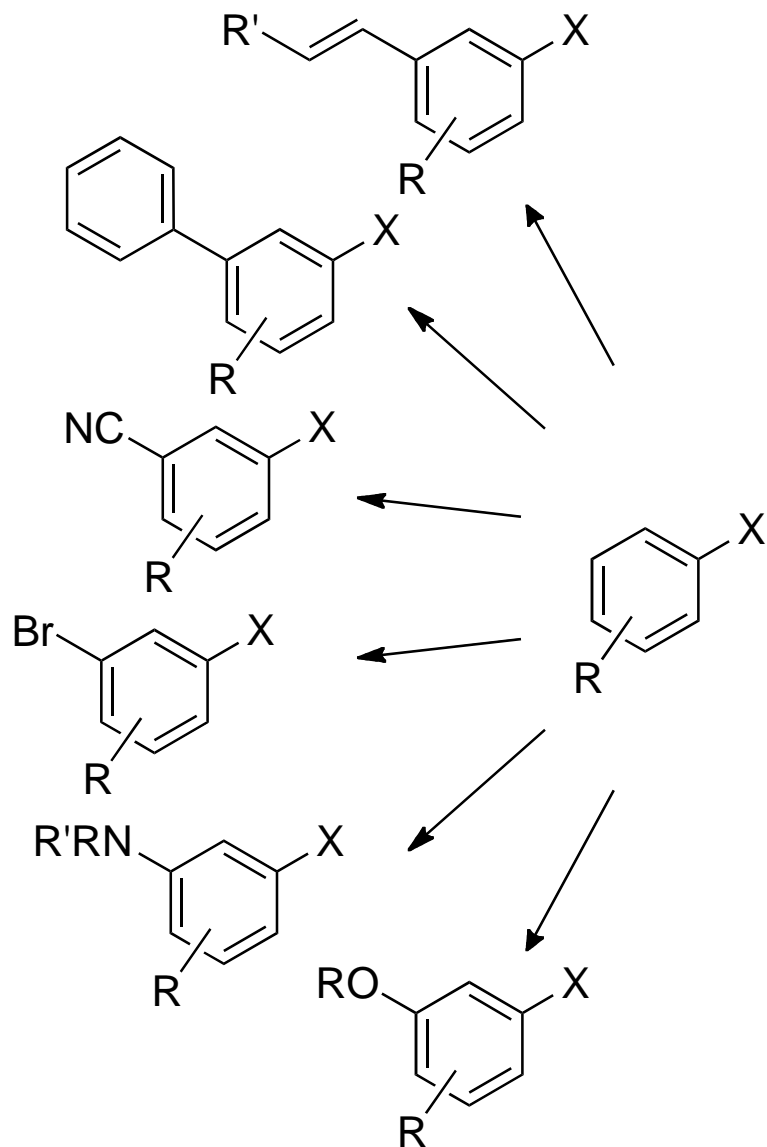
John F. Hartwig

*Department of Chemistry
University of California, Berkeley*

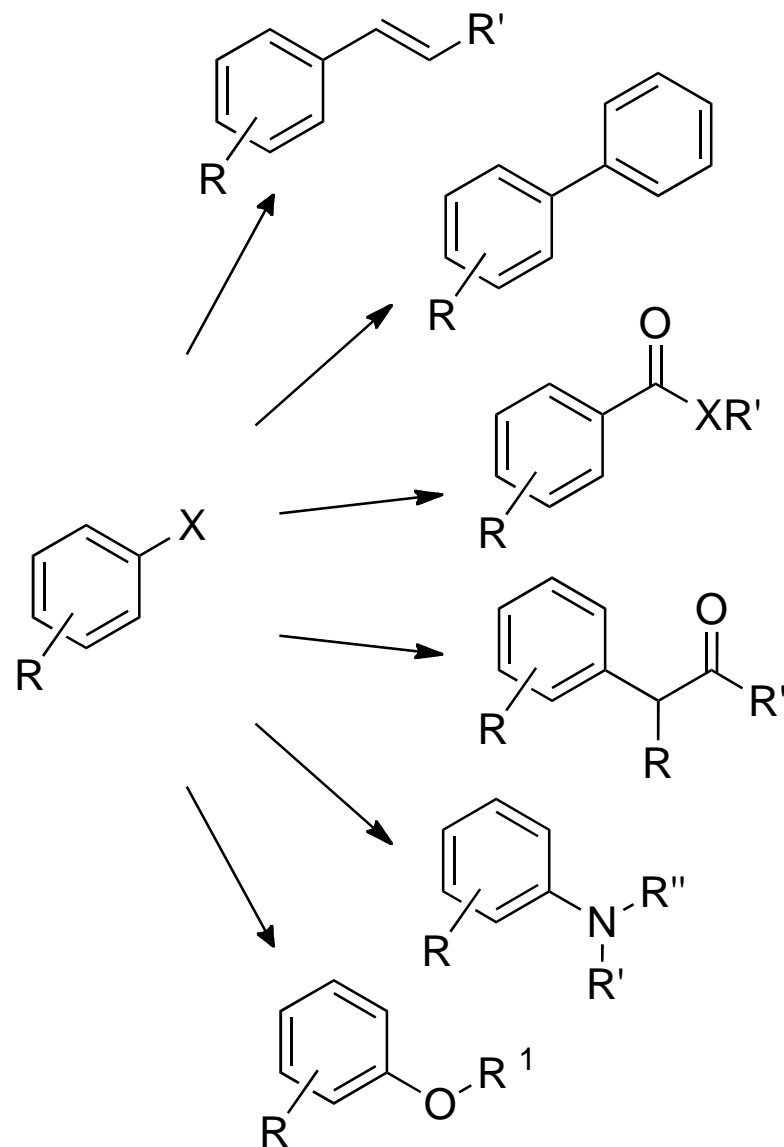


Approaches to Catalytic Modification of Arenes

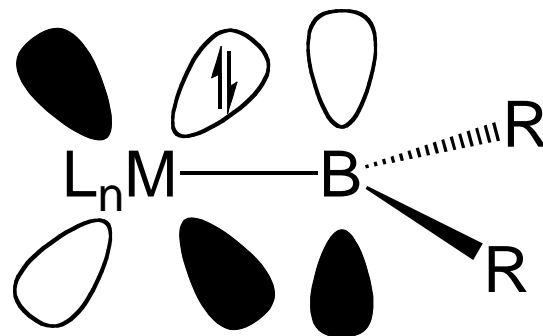
By C-H Bond Functionalization?



By Cross-Coupling Reactions

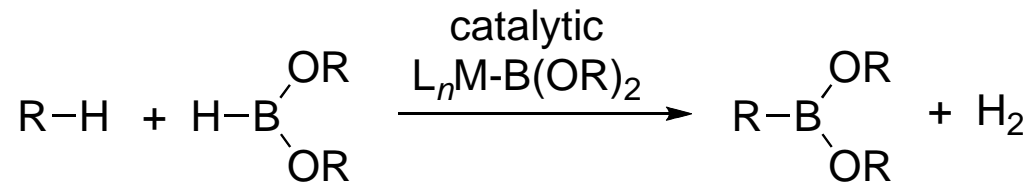


Transition Metal Boryl Complexes

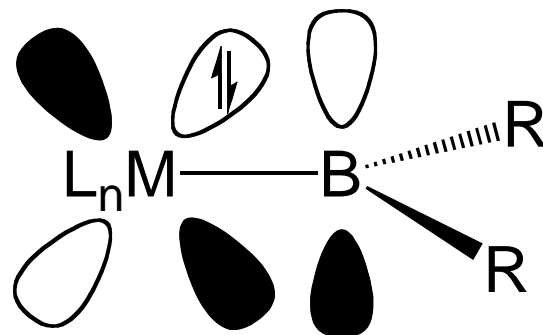


- Potential Properties:
- Lewis Acidic at BR₂
 - M-B pi-bonding
 - Strong σ -Donation from BR₂

These properties lead to selective functionalization of arenes and alkanes:



Transition Metal Boryl Complexes

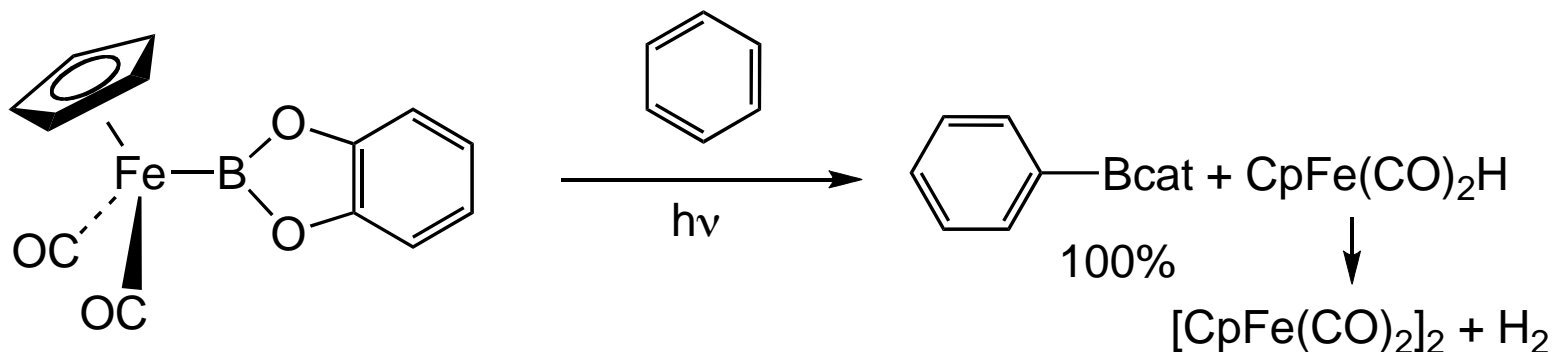


- Potential Properties:
- Lewis Acidic at BR_2
 - M-B pi-bonding
 - Strong σ -Donation from BR_2

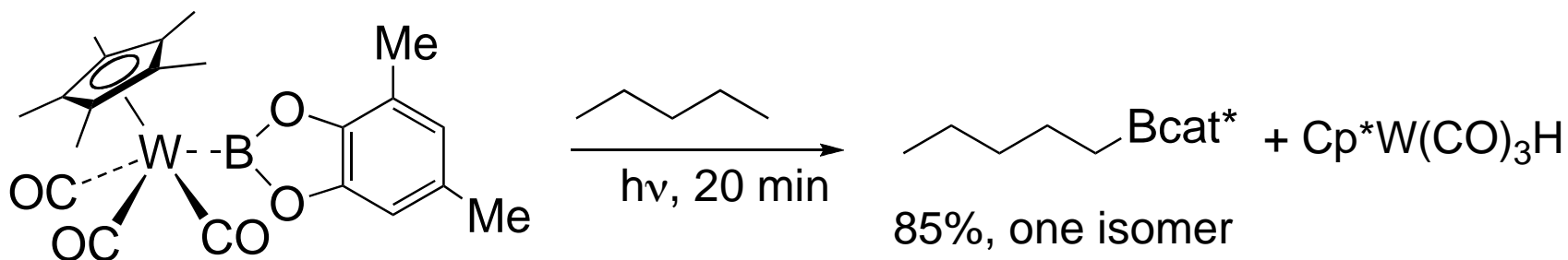
Topics

1. Discovery of C-H borylation of aliphatic C-H Bonds
2. A platform for practical aromatic C-H bond functionalization
3. Use of the mechanistic data to develop new functionalizations
4. Some new C-C coupling reactions

Initial Observations of C-H Bond Functionalization with Metal-Boryl Complexes



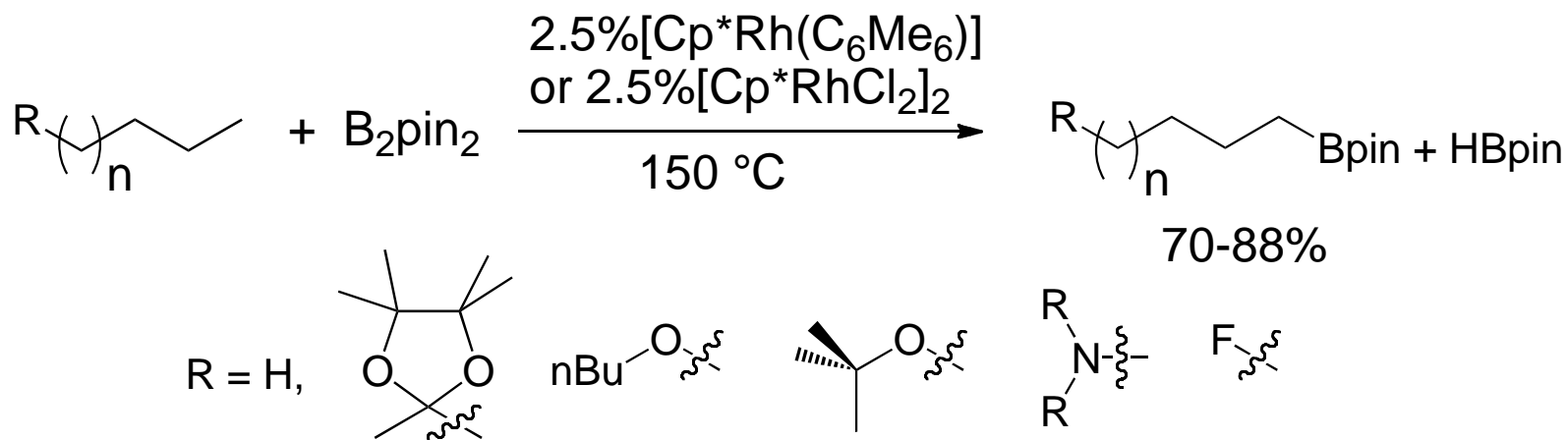
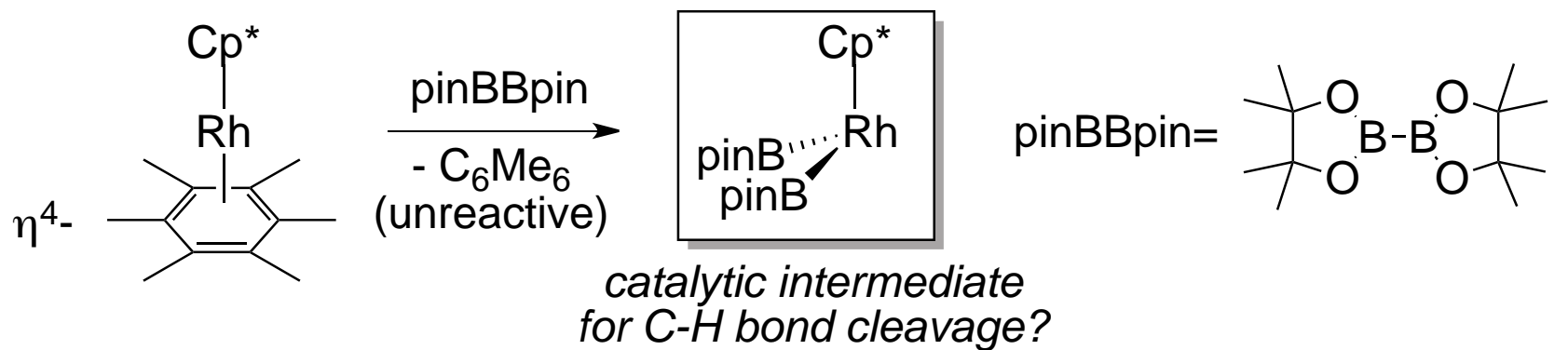
Hartwig, Waltz, He, Muhoro *J. Am. Chem. Soc.* **1995**, *117*, 11357.



Waltz, K.M.; Hartwig, J.F. *Science*, **1997**, *227*, 211.

For formation of two isomers of tolylboronate ester as trace side product from borylation of toluene solvent when forming [Ir(arene)(Bcat)₃], see Nguyen, P.; Blom, H. P.; Westcott, S. A.; Taylor, N. J.; Marder, T. B. *J. Am. Chem. Soc.* **1993**, *115*, 9329.

Catalytic Functionalization of Primary C-H Bonds

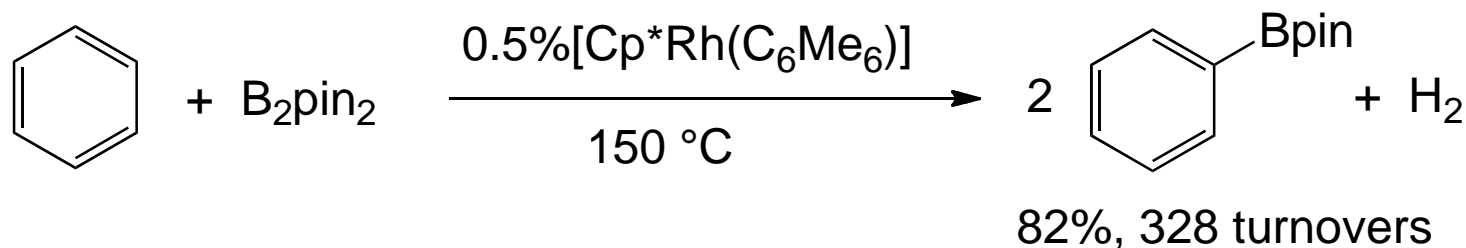
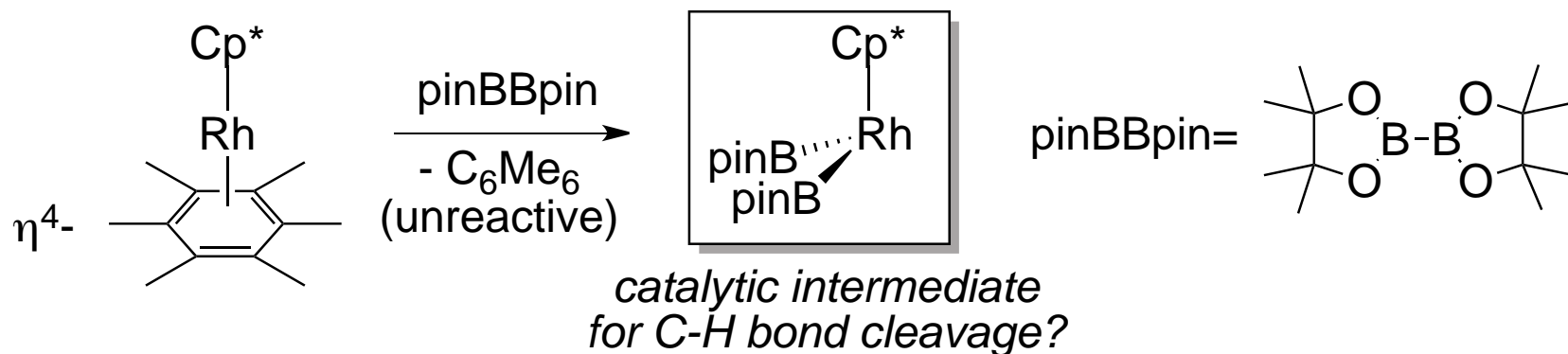


Chen, Schlecht, Semple, Hartwig *Science* **2000**, 287, 1995-8.

Lawrence, Takahashi, Bae, Hartwig *J. Am. Chem. Soc.* **2004**, 126, 15334-15335.

For an the borylation of benzene catalyzed by Cp*Ir(PMe₃)(Ph)(H) with three turnovers see, Iverson, C. N.; Smith, M. R., III *J. Am. Chem. Soc.* **1999**, 121, 7696.

Catalytic Functionalization of Arene C-H Bonds



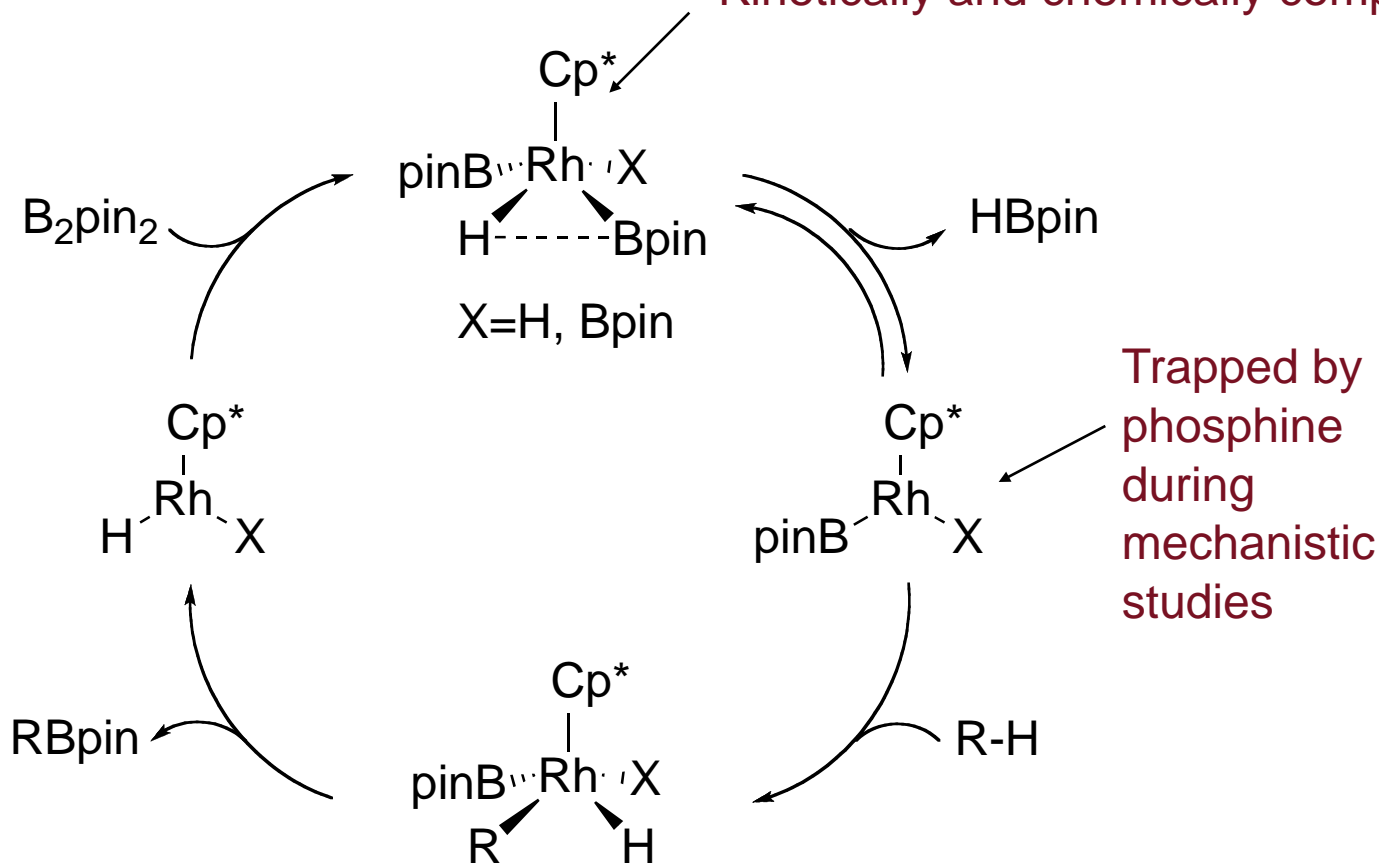
Chen, Schlecht, Semple, Hartwig *Science* **2000**, 287, 1995-8.

Lawrence, Takahashi, Bae, Hartwig *J. Am. Chem. Soc.* **2004**, 126, 15334-15335.

For an the borylation of benzene catalyzed by Cp*Ir(PMe₃)(Ph)(H) with three turnovers see, Iverson, C. N.; Smith, M. R., III *J. Am. Chem. Soc.* **1999**, 121, 7696.

Mechanism of the Rh-Catalyzed Alkane Functionalization

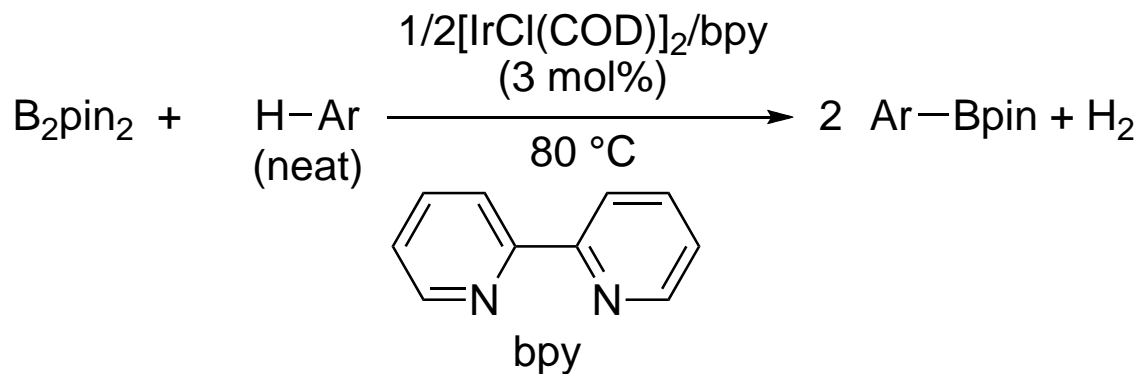
- Isolated and observed in the catalytic system
- Kinetically and chemically competent



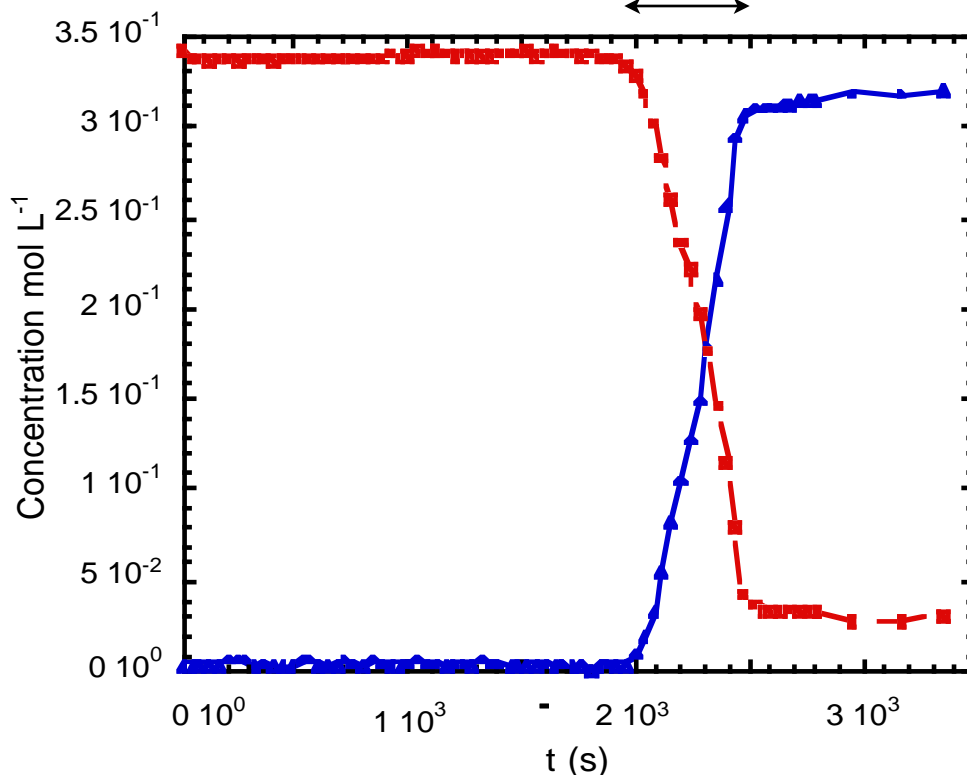
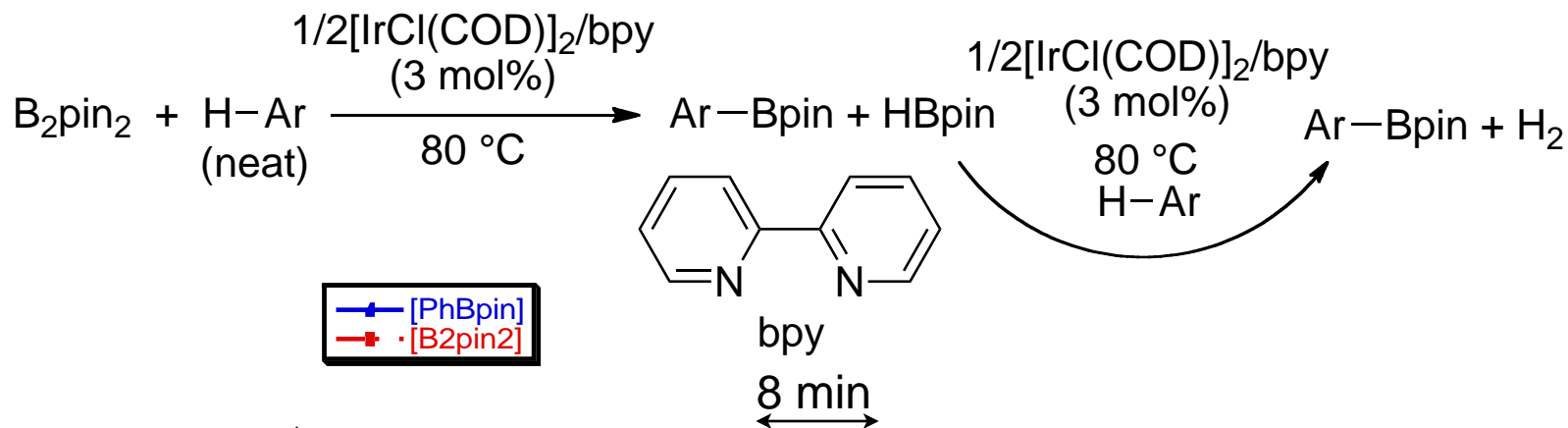
Evidence for this catalytic cycle

How is the C-H bond cleaved and what is the origin of high terminal regioselectivity? Hartwig, Cook, Hanks, Inan, Farn, Webster, Hall, J. Am. Chem. Soc. 2005, 127, 2538-2552

Practical Arene Borylation Catalysts



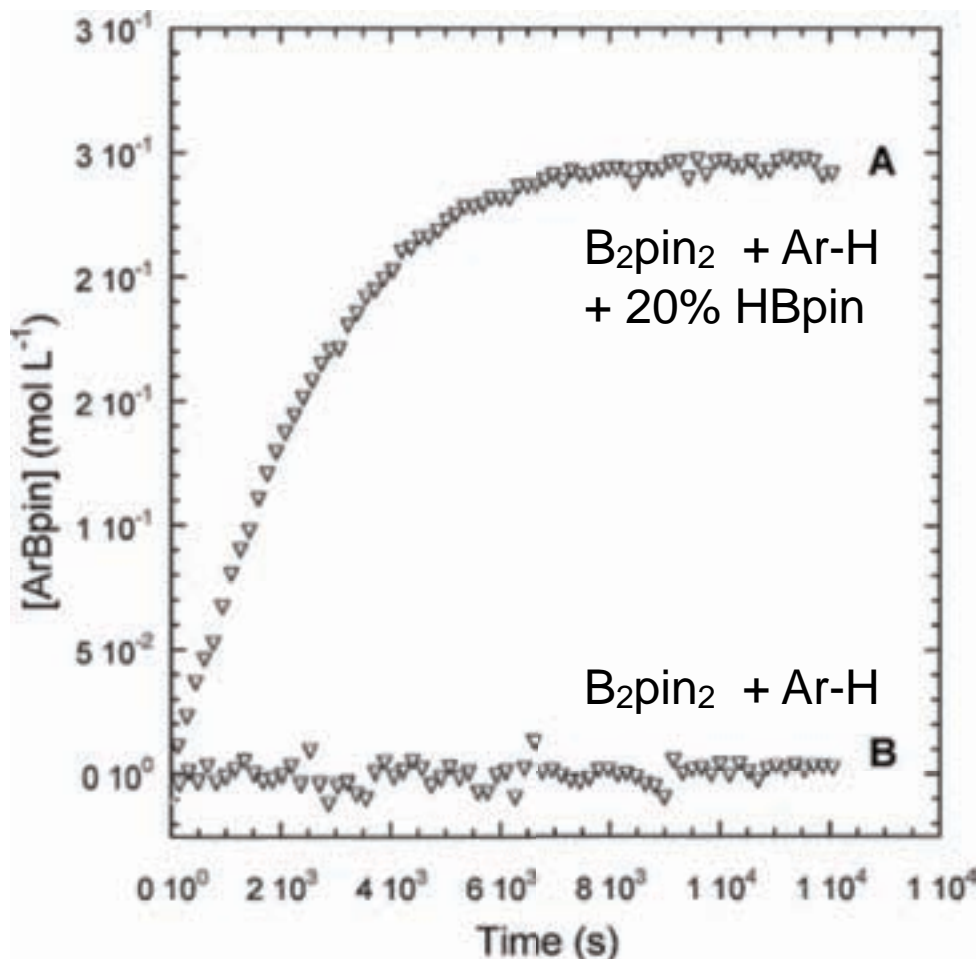
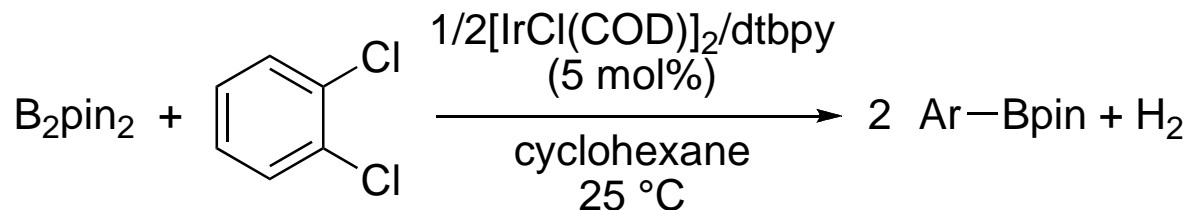
Practical Arene Borylation Catalysts



Origins of the induction period:

- 1) Reduction of COD to COE
- 2) Generation of HBpin from B₂pin₂

Practical Arene Borylation Catalysts



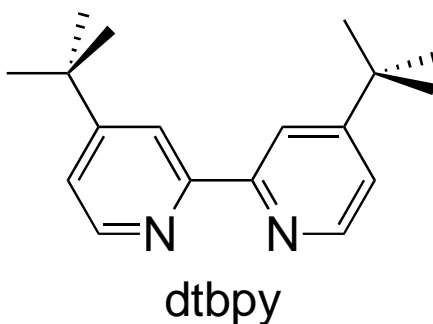
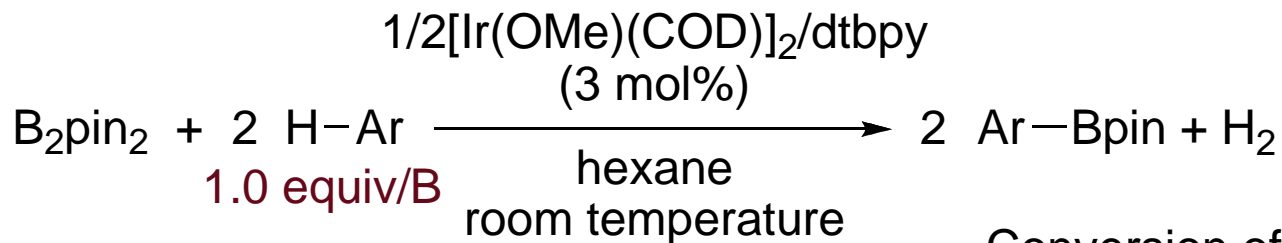
With 0.003 mol % Ir
24,800 turnovers

Origins of the
induction period:

1) Reduction of
COD to COE

2) Generation of
HBpin from B₂pin₂

Highly Active Arene Borylation Catalysts



Conversion of B_2pin_2
and HBpin

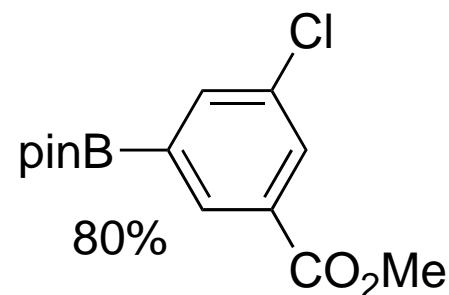
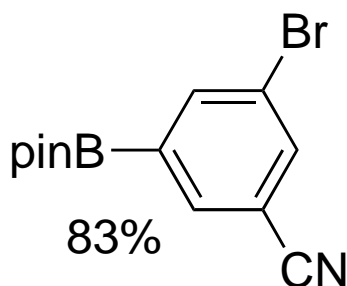
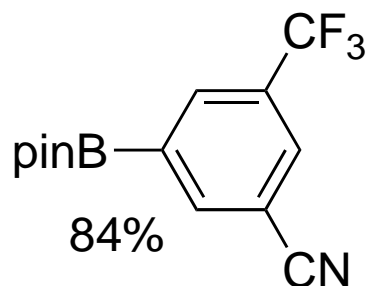
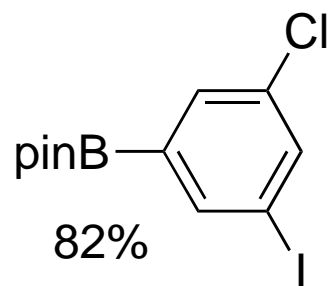
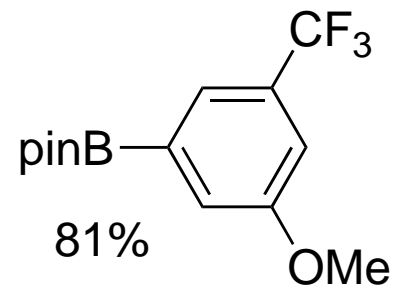
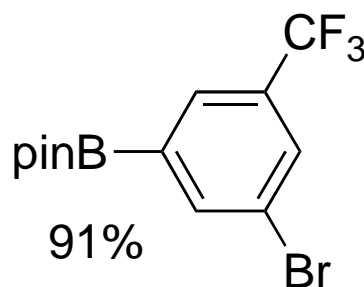
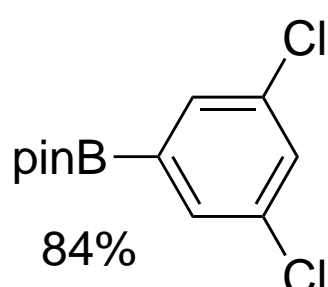
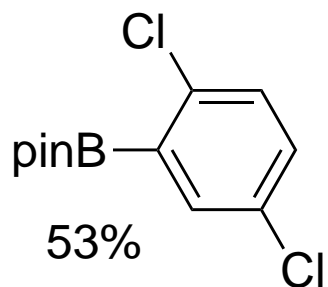
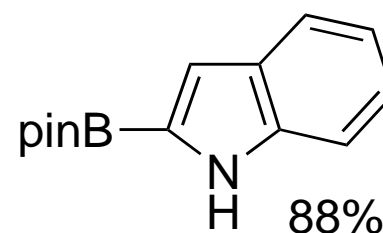
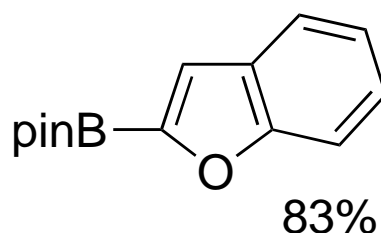
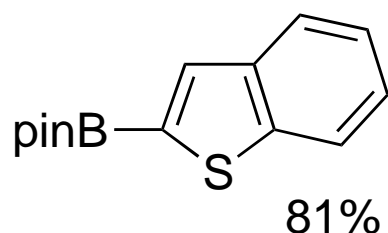
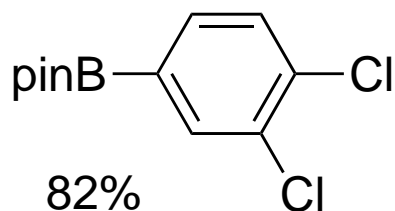
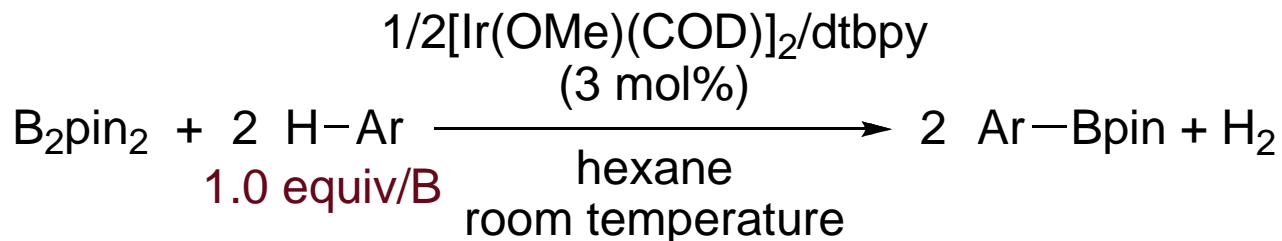
Ishiyama, Takagi, Ishida, Miyaura, Anastasi, Hartwig *JACS* **2002**, *124*, 390.

Ishiyama, Takagi, Hartwig, Miyaura *Angew. Chem. Int. Ed.* **2002**, *41*, 3056.

Boller, Murphy, Hapke, Ishiyama, Miyaura, Hartwig, *JACS* **2005**, *127*, 14263.

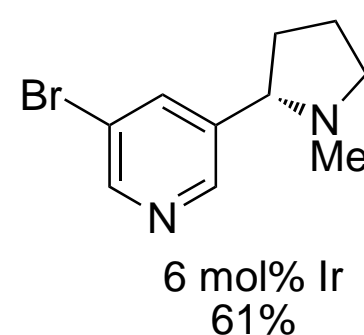
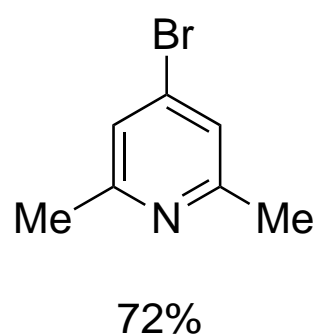
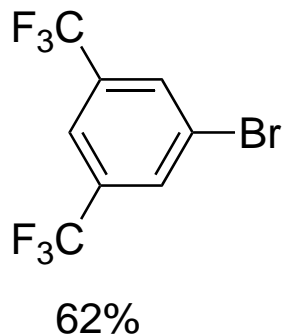
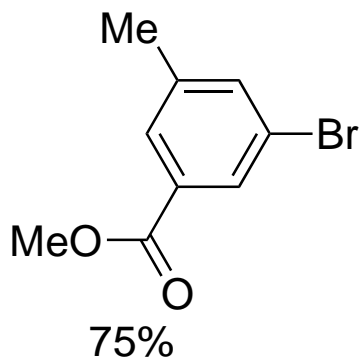
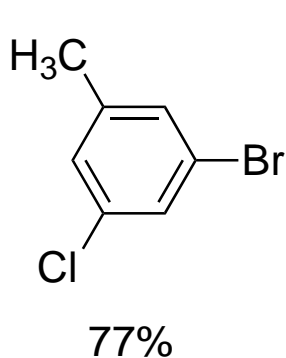
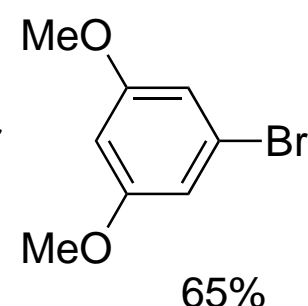
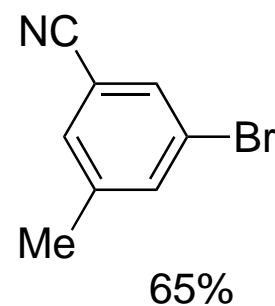
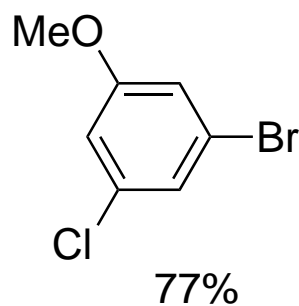
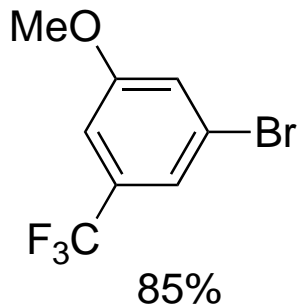
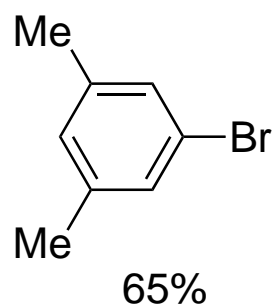
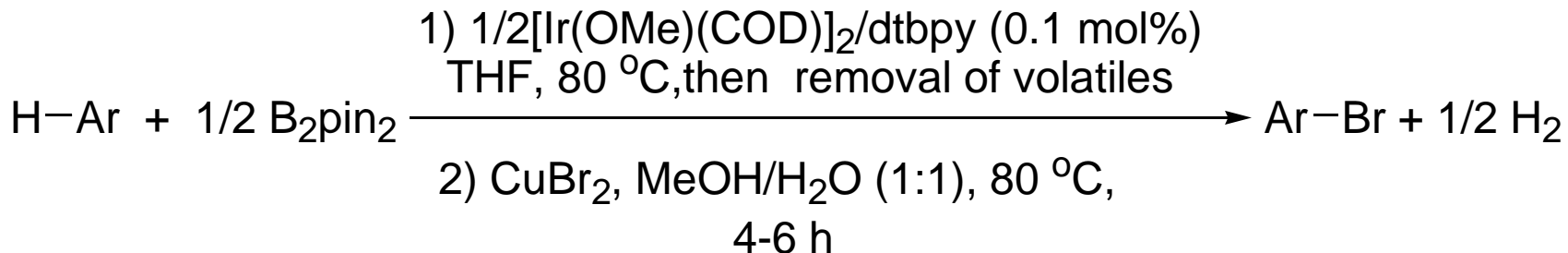
see also with Ir-bisphosphine complexes as catalyst: Cho, J. Y.; Tse, M. K.; Holmes, D.; Maleczka, R. E.; Smith, M. R. *Science* **2002**, *295*, 305-308.

Highly Active Arene Borylation Catalysts

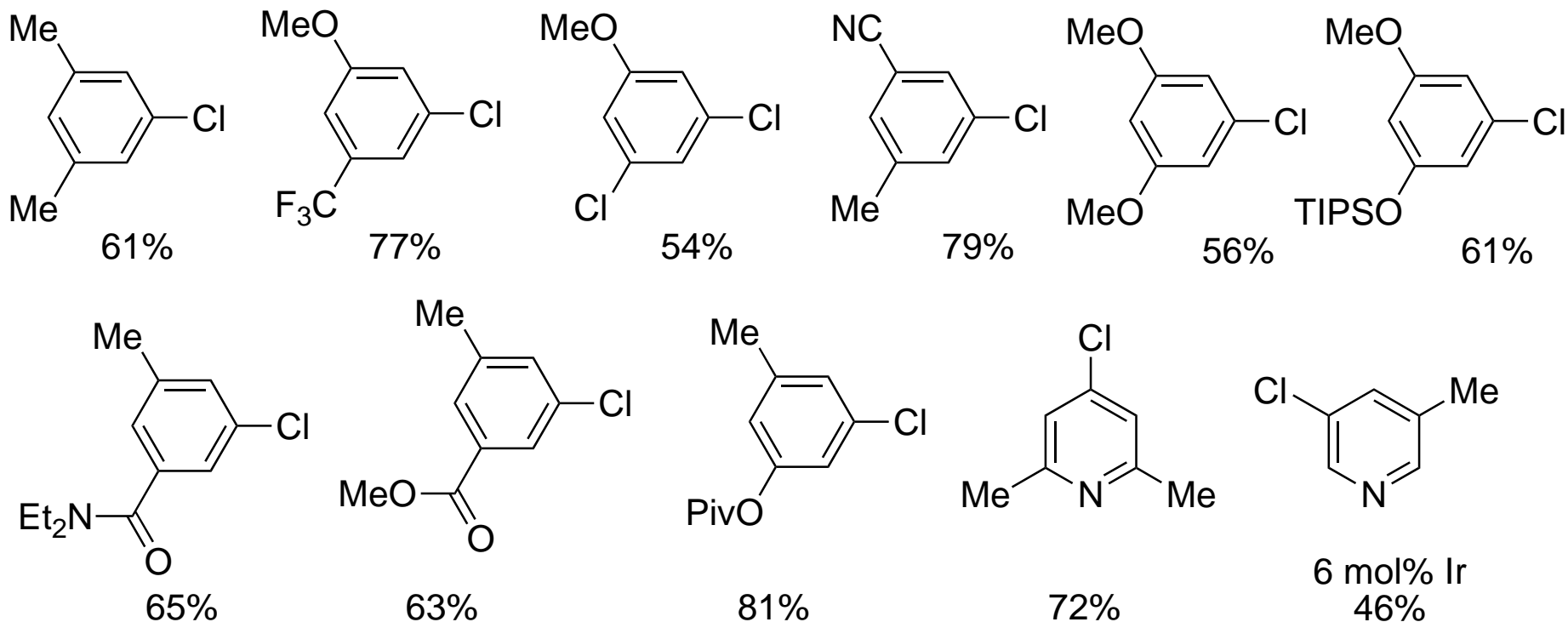
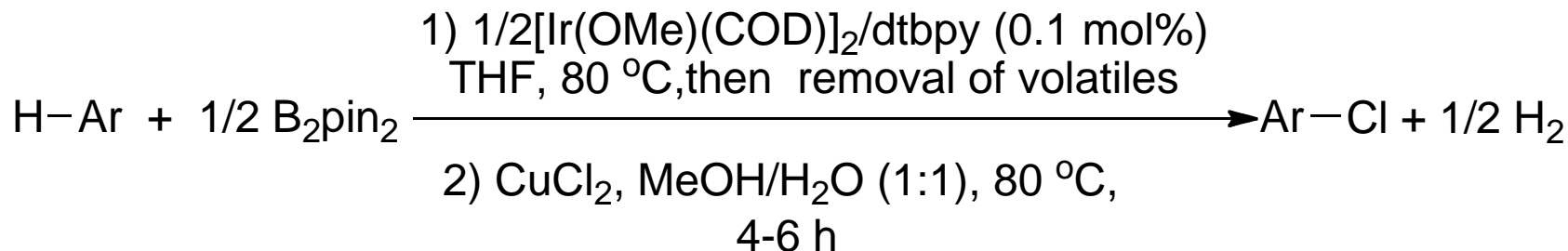


Ishiyama, Takagi, Hartwig, Miyaura *Angew. Chem. Int. Ed.* **2002**, 41, 3056.

Meta-Halogenation of Arenes



Meta-Halogenation of Arenes



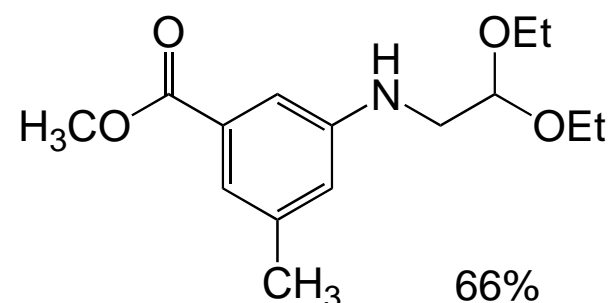
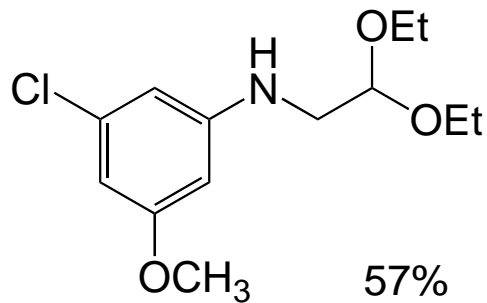
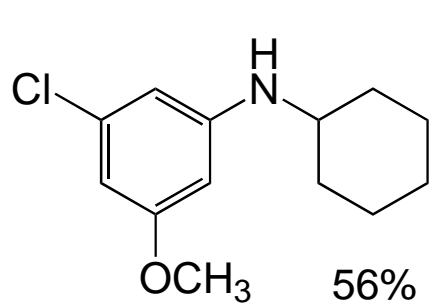
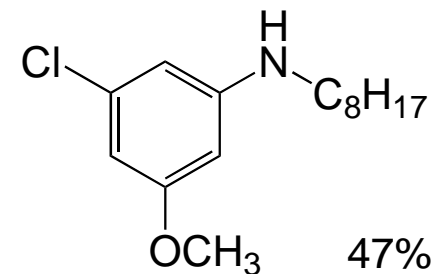
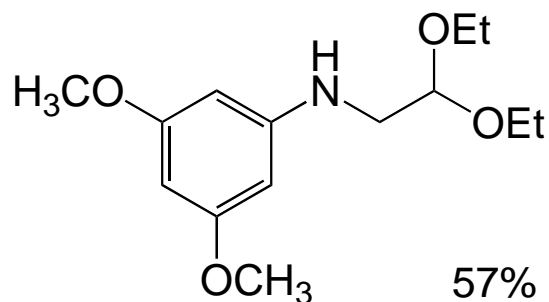
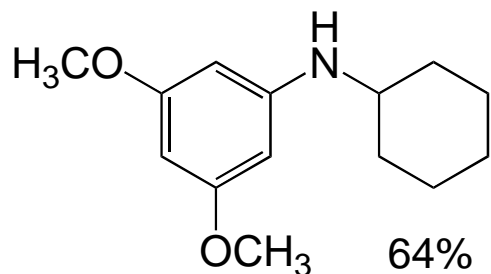
Meta-Amination of Arenes

1) $1/2$ B₂pin₂ $1/2$ [Ir(OMe)(COD)]₂/dtbpy (0.1 mol%)
THF, 80 °C, then removal of volatiles

H—Ar

Ar—NHR

2) 2 equiv H₂NR, 1 equiv KF,
10 mol % Cu(OAc)₂·H₂O,
MS 4 A, CH₃CN, O₂, 80 °C



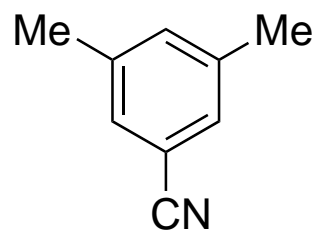
Meta-Cyanation of Arenes

1) 1/2 B₂pin₂, 1/2[Ir(OMe)(COD)]₂/dtbpy (0.1 mol%)
THF, 80 °C, the removal of volatiles

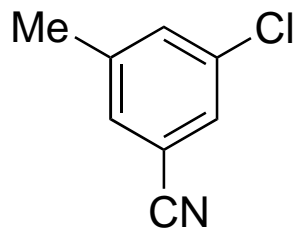
H-Ar

2) 1.5 equiv Cu(NO₃)₂•3H₂O, 3 equiv Zn(CN)₂,
1 equiv CsF
MeOH/H₂O (5:1), 90 °C

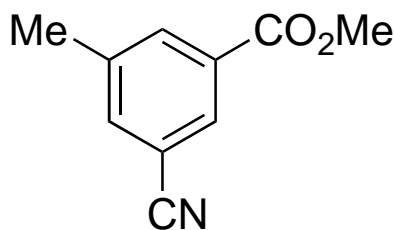
Ar-CN



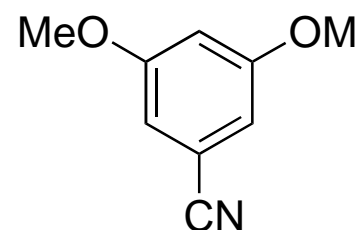
61%



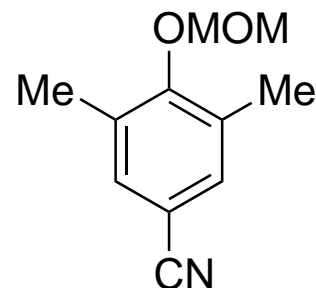
65%



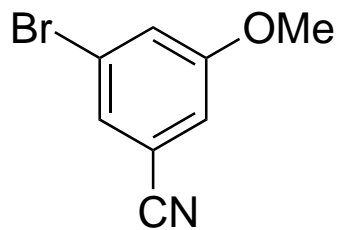
59%



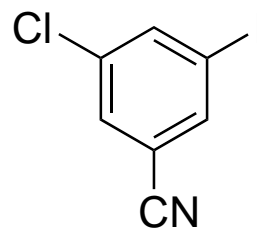
53%



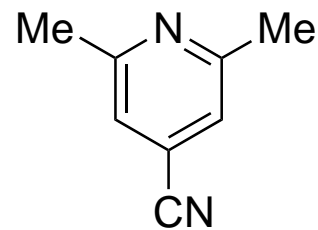
51%



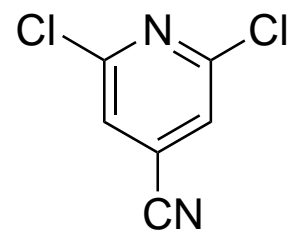
58%



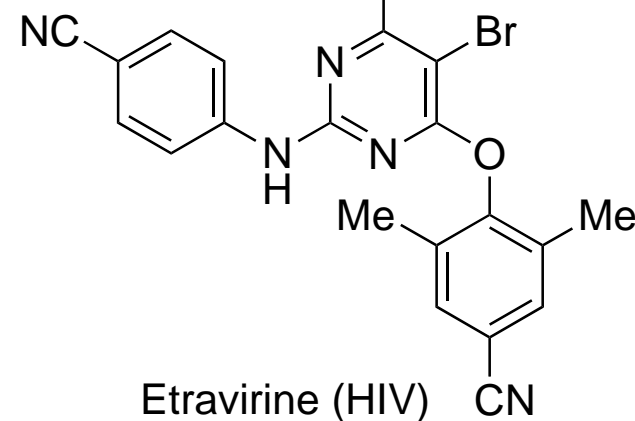
57%



46%



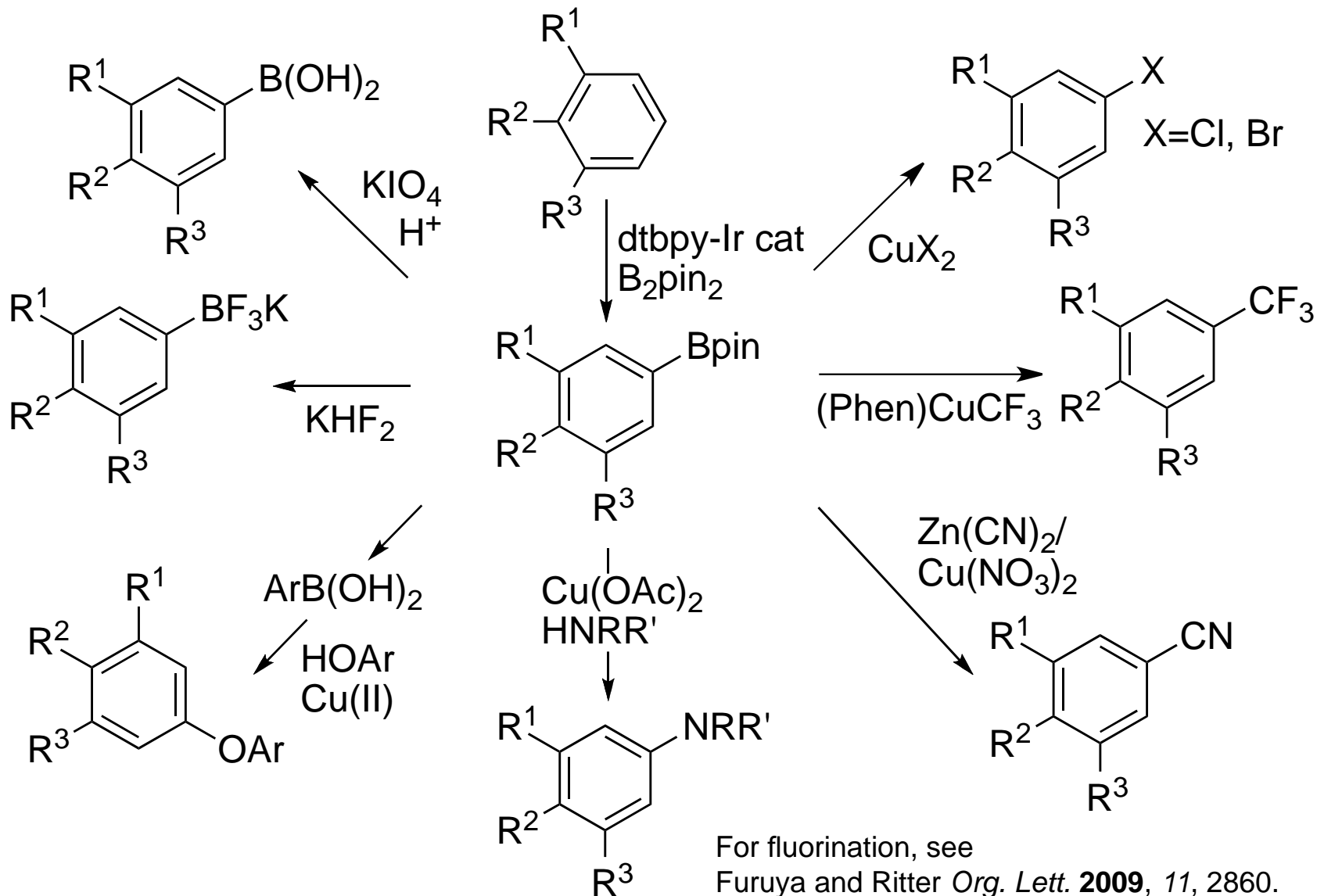
58%



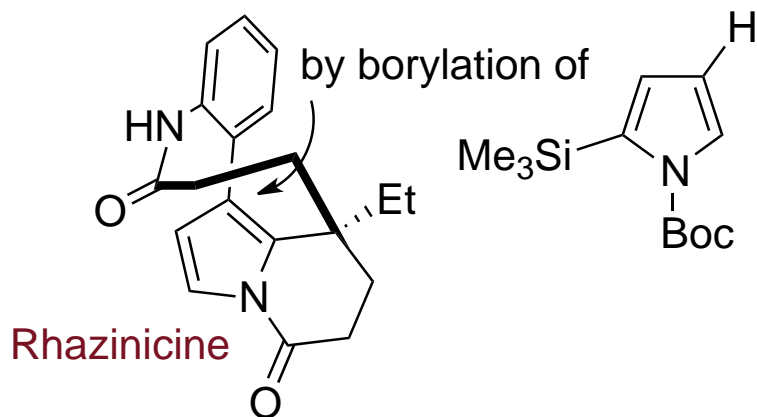
Etravirine (HIV)

CN

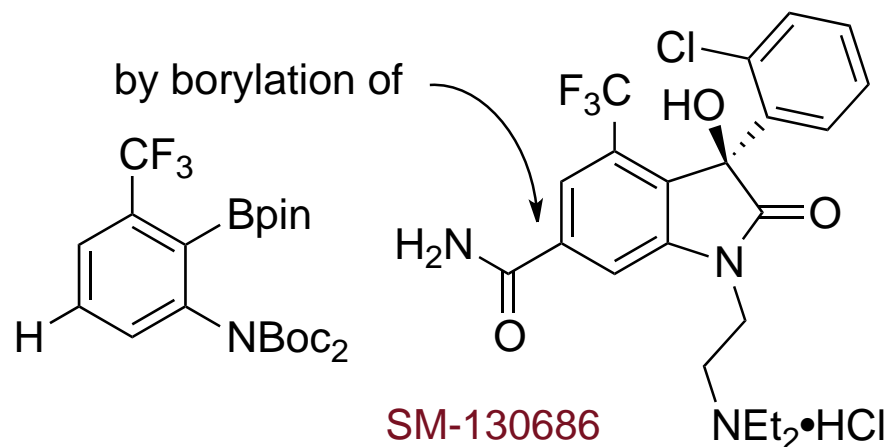
A General meta-Functionalization of 1,3, and 1,2,3-Substituted Arenes



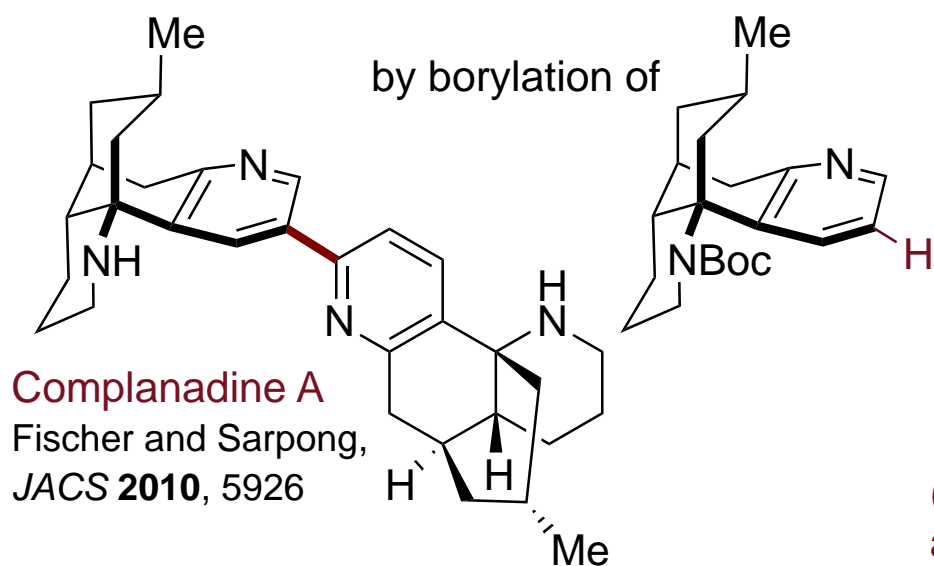
[Ir(COD)X]/dtbpy Catalyst in Total Synthesis



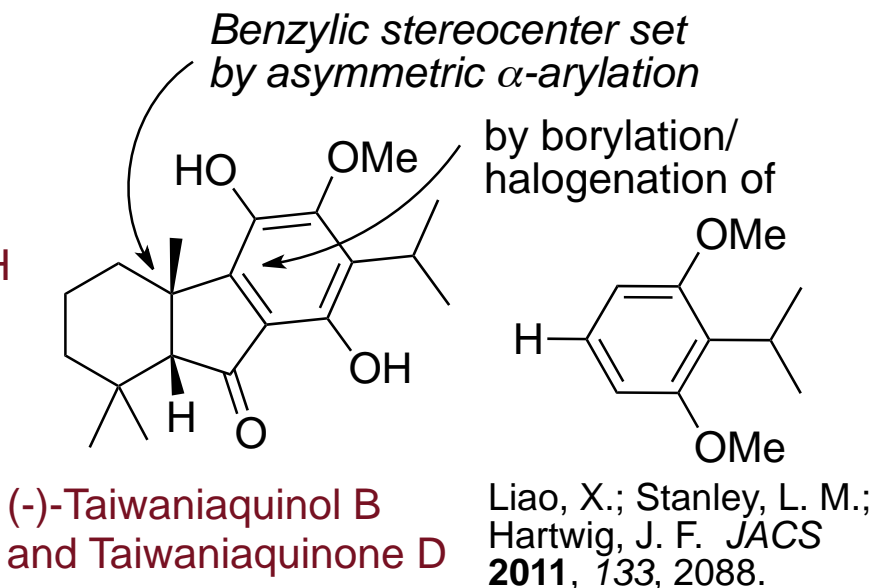
Beck, E. M.; Hatley, R.; Gaunt, M. J.
Angew. Chem. Int. Ed. **2008**, 47, 3004-3007.



Tomita, D.; Yamatsugu, K.; Kanai, M.; Shibasaki, M.
J. Am. Chem. Soc. **2009**, 131, 6946-7.

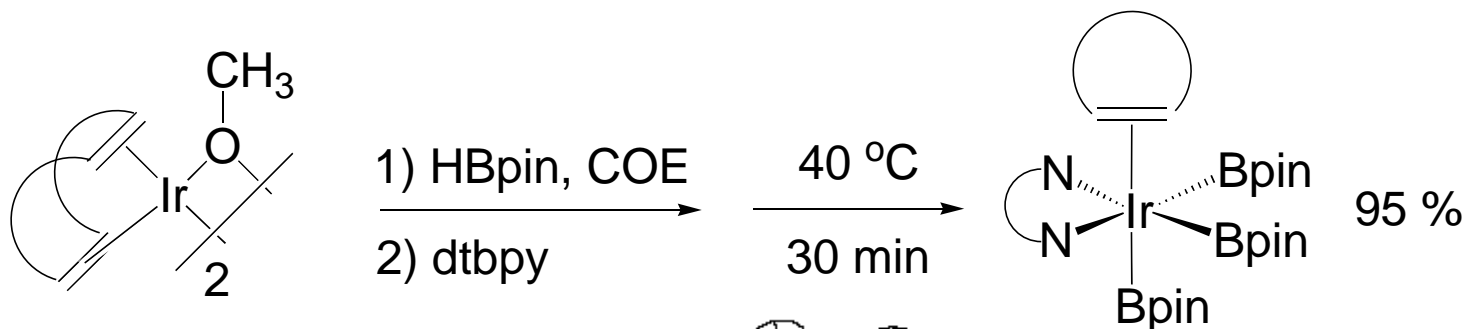


Fischer and Sarpong,
JACS **2010**, 5926



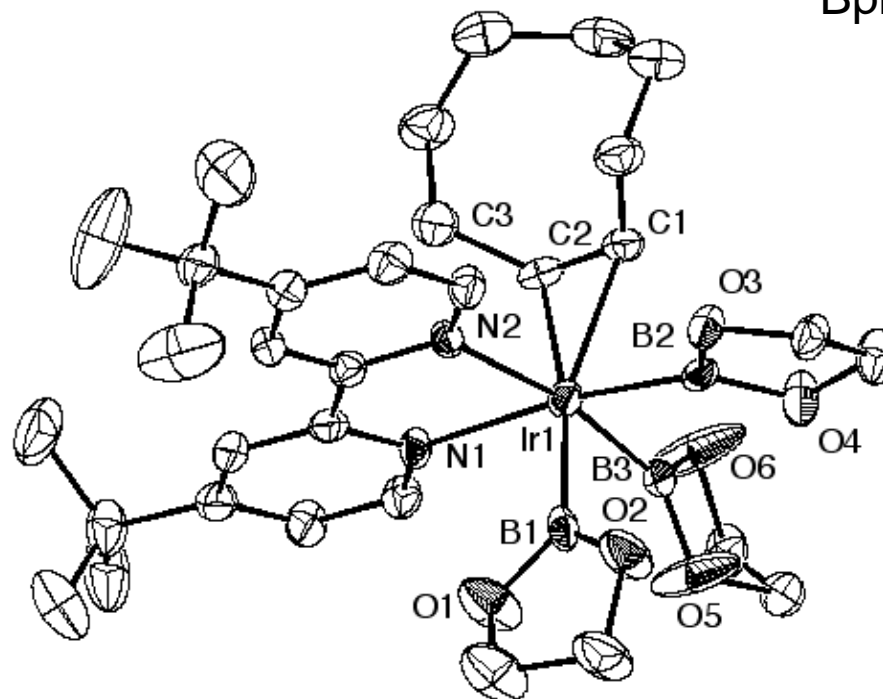
Liao, X.; Stanley, L. M.;
Hartwig, J. F. *JACS*
2011, 133, 2088.

Intermediate in the Functionalization of Arenes by Bpy-Ir



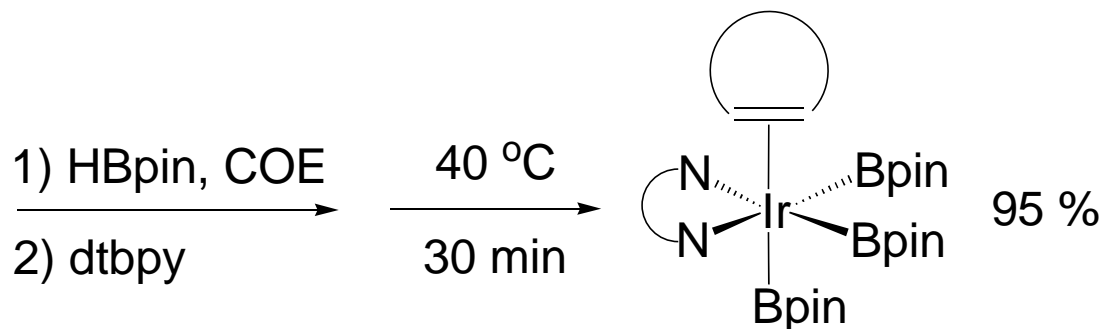
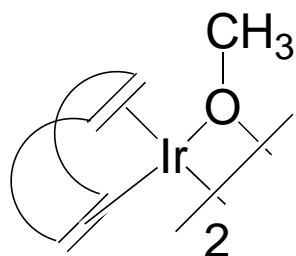
Ishiyama, Takagi,
Ishida, Miyaura,
Anastasi, Hartwig,
JACS **2002**, *124*,
390.

Boller, Murphy,
Hapke, Ishiyama,
Miyaura, Hartwig
JACS **2005**, *127*,
14263



Ir-N1=2.177 Å, Ir-
N2=2.221 Å, Ir-
B1=2.055 Å Ir-
B2=2.057 Å, Ir-
B3=2.027 Å, Ir-
C1=2.308 Å Ir-
C2=2.318 Å

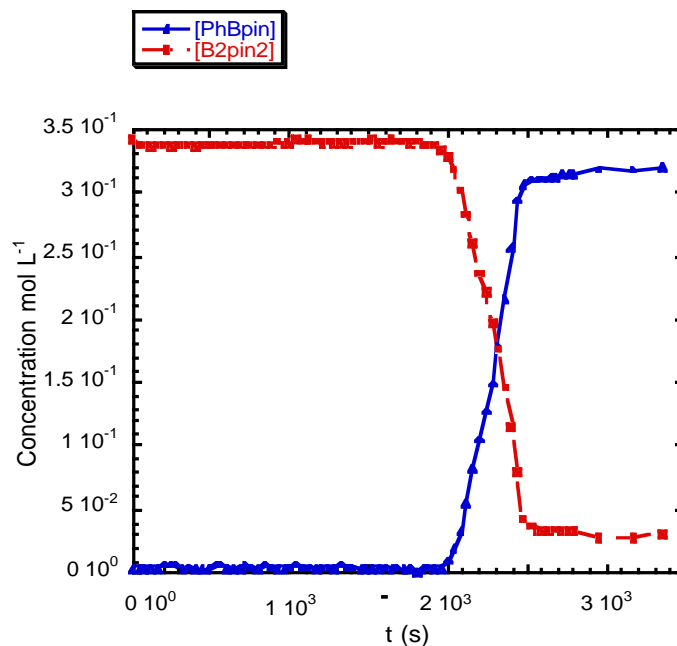
Intermediate in the Functionalization of Arenes by Bpy-Ir



Ishiyama, Takagi,
Ishida, Miyaura,
Anastasi, Hartwig,
JACS **2002**, *124*,
390.

Boller, Murphy,
Hapke, Ishiyama,
Miyaura, Hartwig
JACS **2005**, *127*,
14263

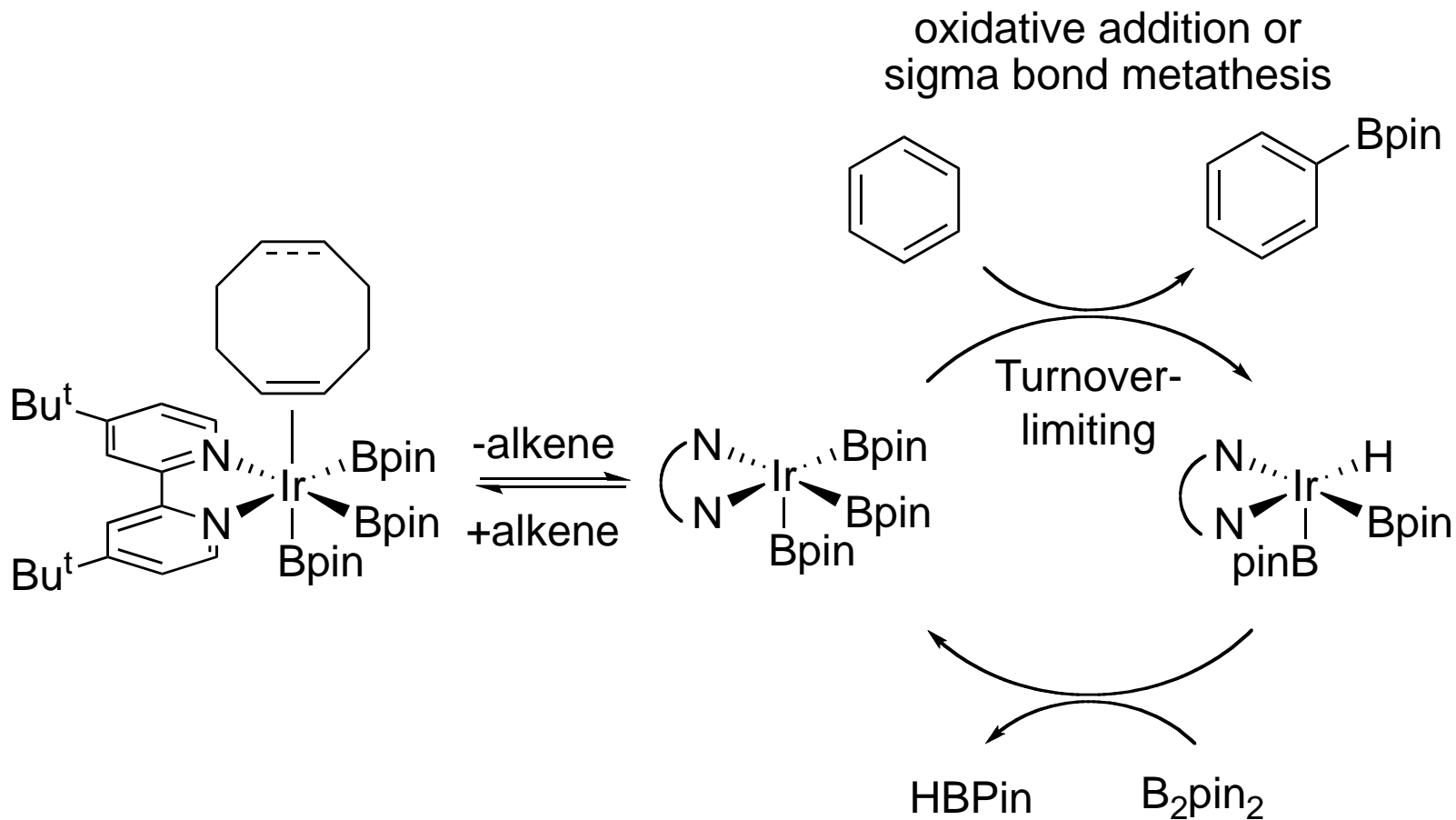
Recall that HBpin eliminated this induction period:



Origins of the induction period:

- 1) Reduction of COD to COE
- 2) Generation of HBpin from B₂pin₂

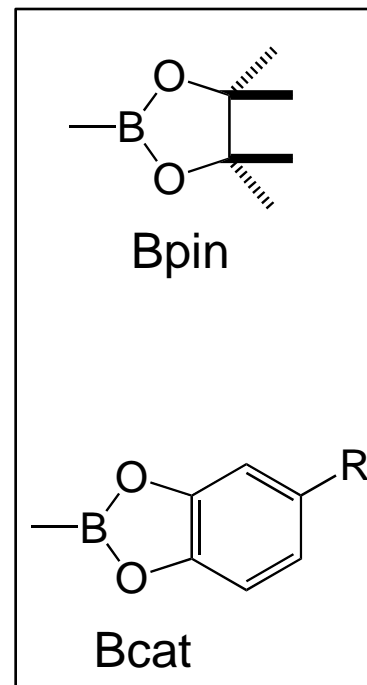
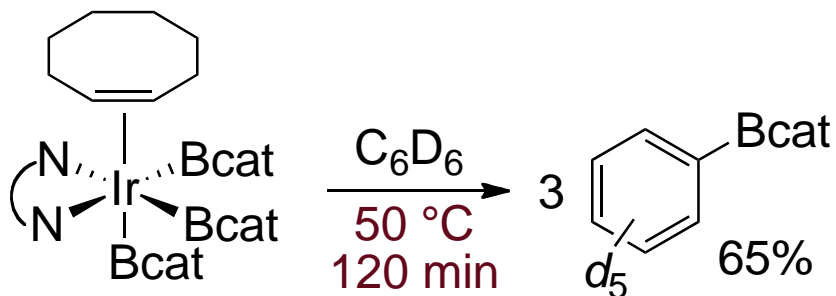
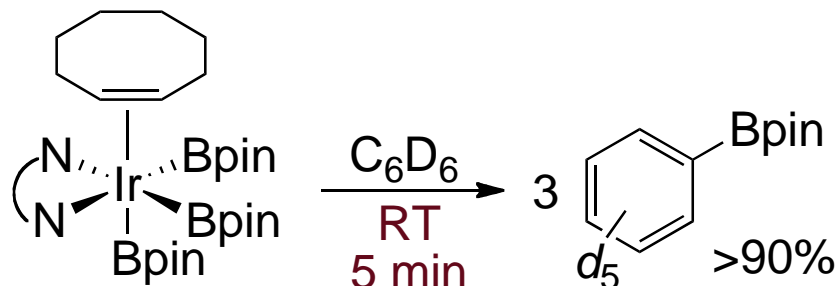
Mechanism for Arene Borylation by Ir(III) Complexes



Ishiyama, Takagi, Ishida, Miyaura, Anastasi, Hartwig, *JACS* **2002**, *124*, 390
Boller, Murphy, Hapke, Ishiyama, Miyaura, Hartwig *JACS* **2005**, *127*, 14263.

Electronic Effects on C-H Borylation

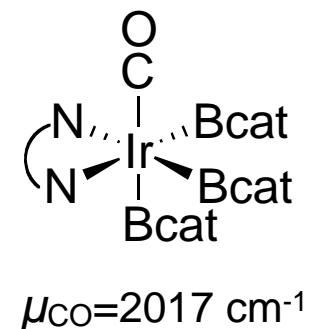
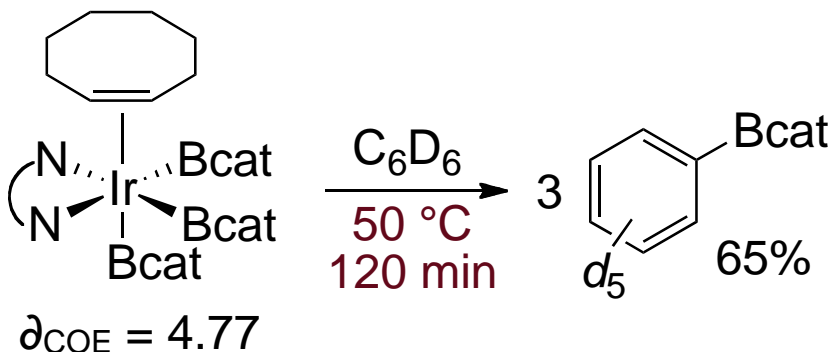
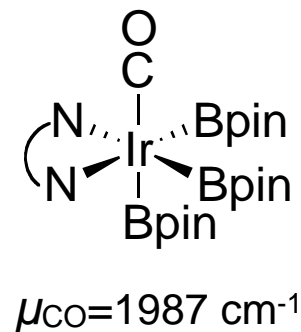
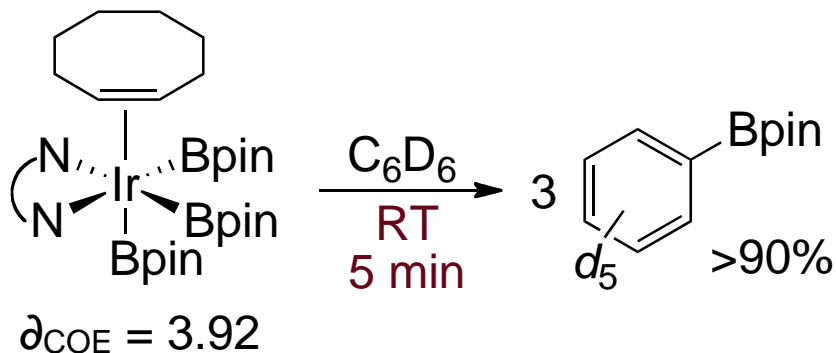
Effect of pinacolate or catecholate on the electron density at Ir



Ishiyama, Takagi, Ishida, Miyaura, Anastasi, Hartwig, *JACS* **2002**, 124, 390.
Boller, Murphy, Hapke, Ishiyama, Miyaura, Hartwig *JACS* **2005**, 127, 14263
Liskey, C.W.; Wei, C.S.; Pahls, D.R.; Hartwig, J.F. *Chem. Commun.*, **2009**, 5603
web theme issue: 'Selective Catalysis for Organic Synthesis'

Electronic Effects on C-H Borylation

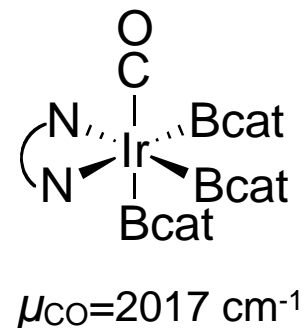
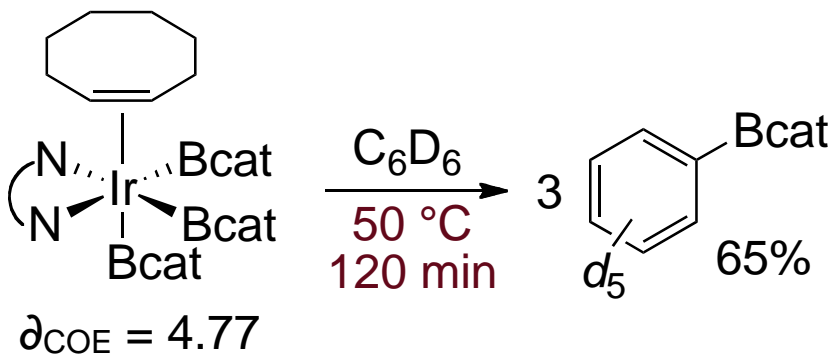
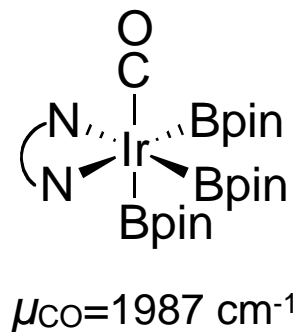
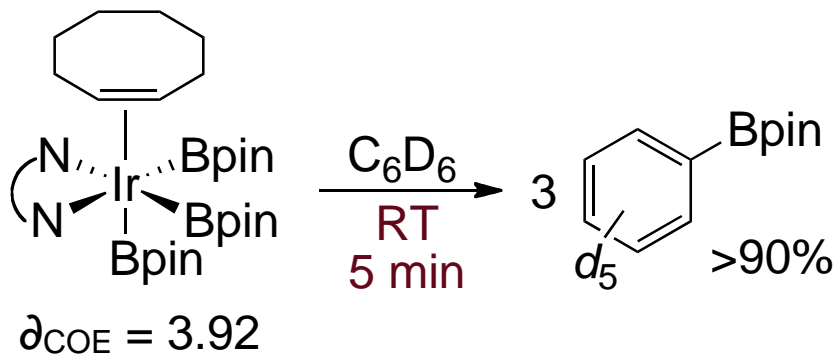
Effect of pinacolate or catecholate on the electron density at Ir



Ishiyama, Takagi, Ishida, Miyaura, Anastasi, Hartwig, *JACS* **2002**, 124, 390.
Boller, Murphy, Hapke, Ishiyama, Miyaura, Hartwig *JACS* **2005**, 127, 14263
Liskey, C.W.; Wei, C.S.; Pahls, D.R.; Hartwig, J.F. *Chem. Commun.*, **2009**, 5603
web theme issue: 'Selective Catalysis for Organic Synthesis'

Electronic Effects on C-H Borylation

Effect of pinacolate or catecholate on the electron density at Ir



Therefore, although the C-H cleavage can occur by σ -bond metathesis...

- This C-H bond cleavage is favored for more electron-rich Ir centers.
- Strong σ -donation by the boryl group is important for C-H bond cleavage.

Developing More Active Catalysts

Electronic properties of ligands used for Ir-catalyzed arene borylation:

$\nu(\text{CO})$ Values for $\text{L}_2\text{M}(\text{CO})_m$ complexes:

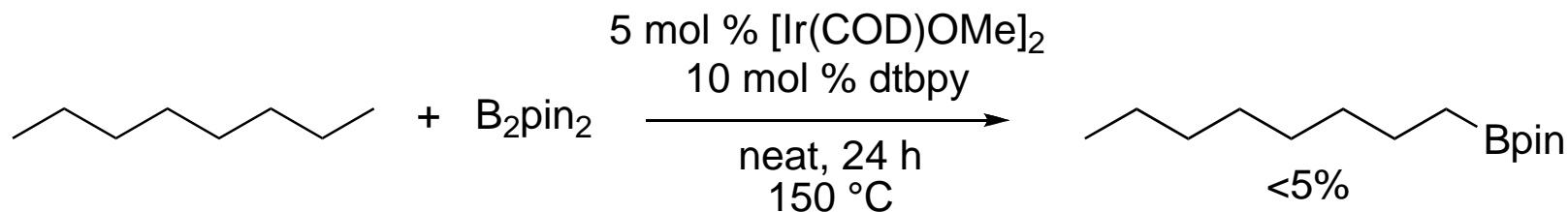
L_2	$\nu(\text{CO})$ for $\text{L}_2\text{Ni}(\text{CO})_2$		reference	for $\text{L}_2\text{Mo}(\text{CO})_4$
2 PPh ₃	2000	1940	<i>a,b,c</i>	highest ν = 2023
2 PMe ₃	1990	1926	<i>c</i>	2019
bpy	1978	1904	<i>a,c</i>	2017
phen	1977-1980	1897-1915	<i>a,c</i>	-
4,4'-Me ₂ py	1973	1893	<i>d</i>	2014
3,4,7,8-Me ₄ -phen	1967-1970	1875-1993	<i>predicted</i>	-

^a Plankey, B. J.; Rund, J. V. *Inorg. Chem.* **1979**, *18*, 957. ^b Meriwether, L. S.; Fiene, M. L. *J. Am. Chem. Soc.* **1959**, *81*, 4200. ^c Tolman, C. A. *J. Am. Chem. Soc.* **1970**, *92*, 2956. ^d Christensen, P. A.; Hamnett, A.; Higgins, S. J.; Timney, J. A. *J. Electroanal. Chem.* **1995**, *395*, 195. ^d Sieler, J.; Than, N.-N.; Benedix, R.; Dinjus, E.; Walther, D. *Z. Anorg. Allg. Chem.* **1985**, *522*, 131.

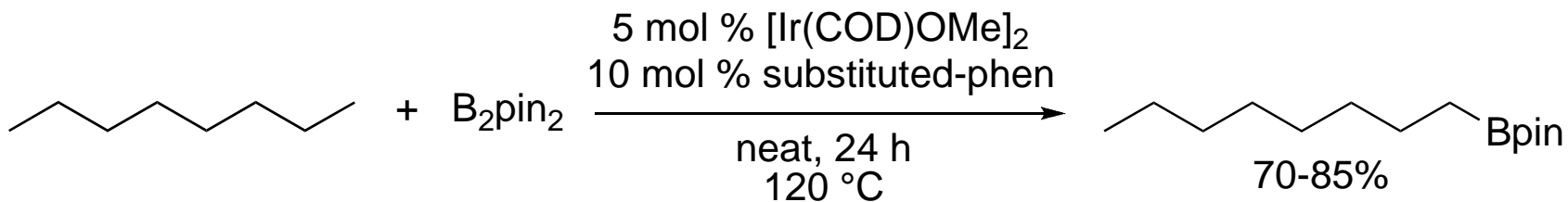
**Among this series, tetramethylphenanthroline is the most donating...
and the one to generate a catalyst for alkane borylation.**

Developing More Active Catalysts

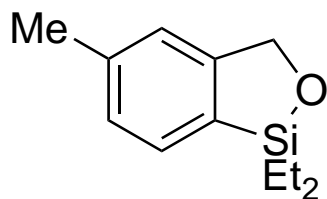
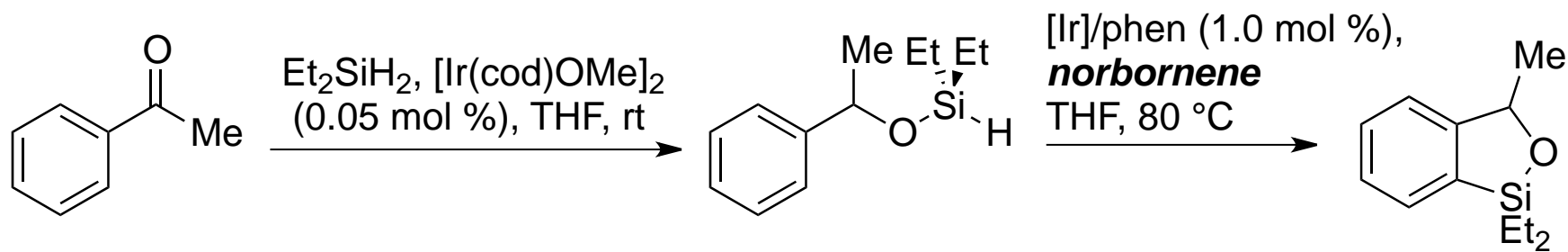
The arene borylation catalyst does not functionalize aliphatic C-H bonds



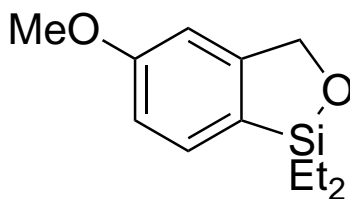
But the catalysts containing substituted phenanthrolines do...



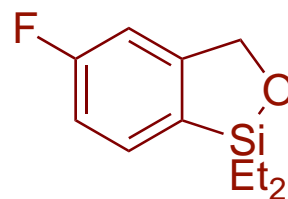
Intramolecular Silylation of Aromatic C-H Bonds



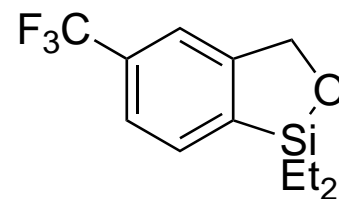
58%
>20:1 selectivity



88%
>20:1 selectivity



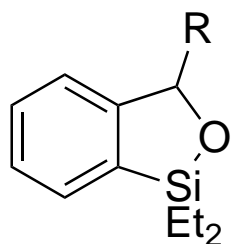
73%
1:1 selectivity



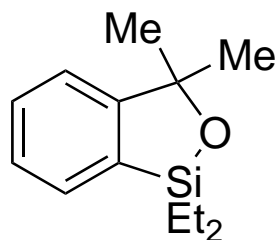
55%
>20:1 selectivity

Regioselectivity appears to be controlled by steric effects.

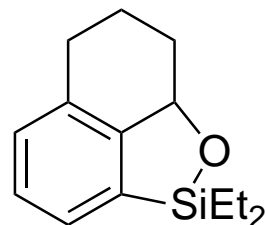
Intramolecular Silylation of Aromatic C-H Bonds



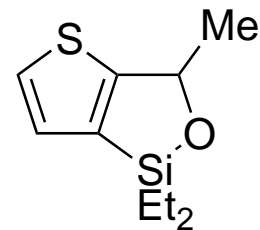
R=Et, 63%
R=Pr, 65%



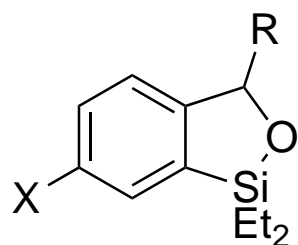
70%



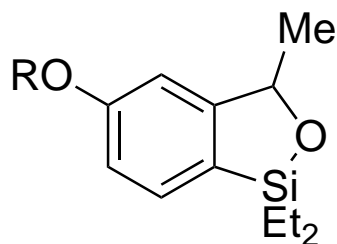
71%



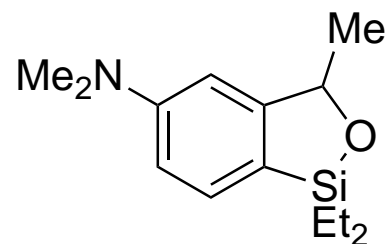
61%



R=H, X=I, 86%
R=Me, X=Br, 75%



R=TBS, 86%
R=Piv, 83%

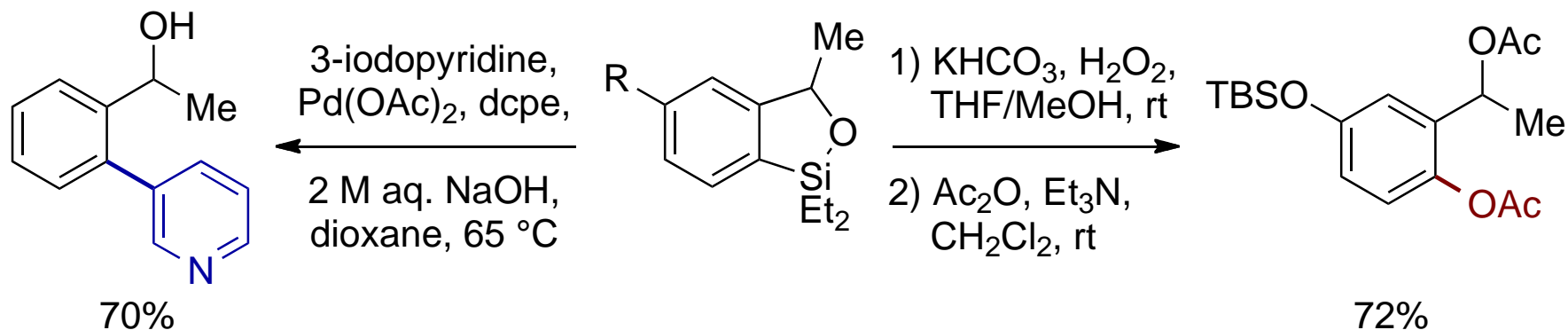


82%

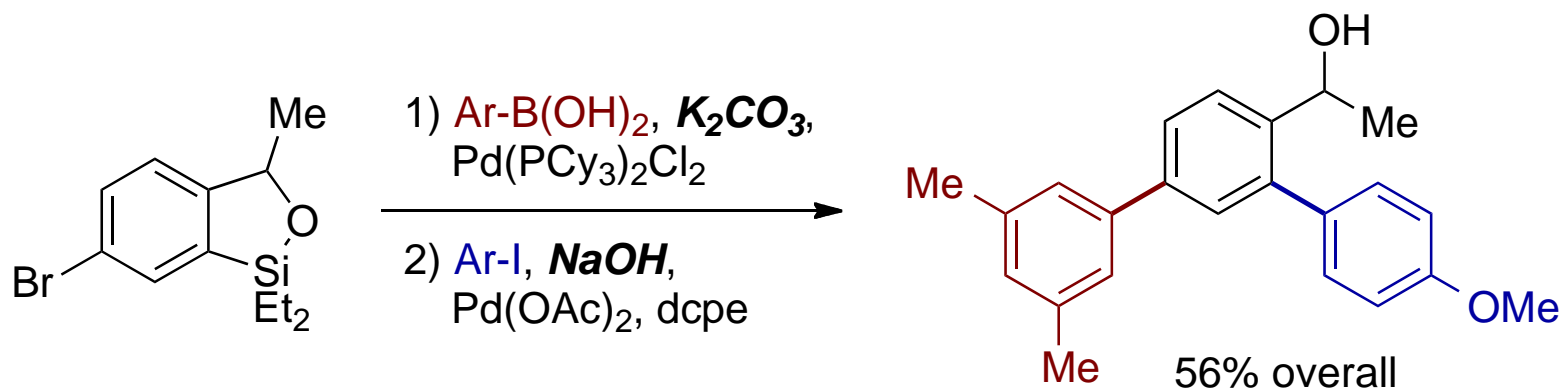
*Regioselectivity appears to be controlled by steric effects.
...and functional group tolerance looks high.*

Transformations of the Siloles from Directed Silylation

Cross coupling and oxidation can be conducted on the oxasiloles

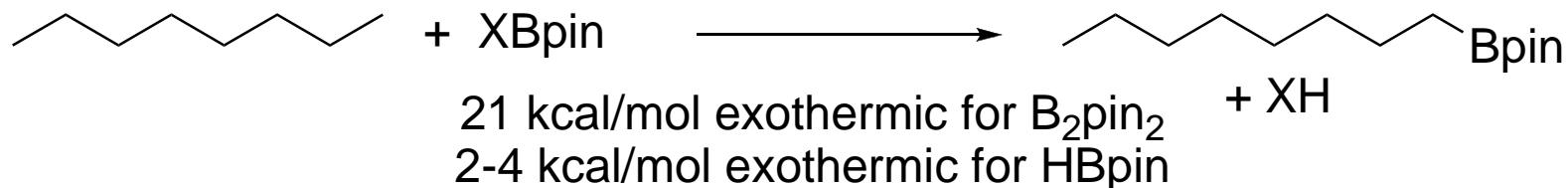


Orthogonal Coupling of Halogen-Containing Oxasiloles

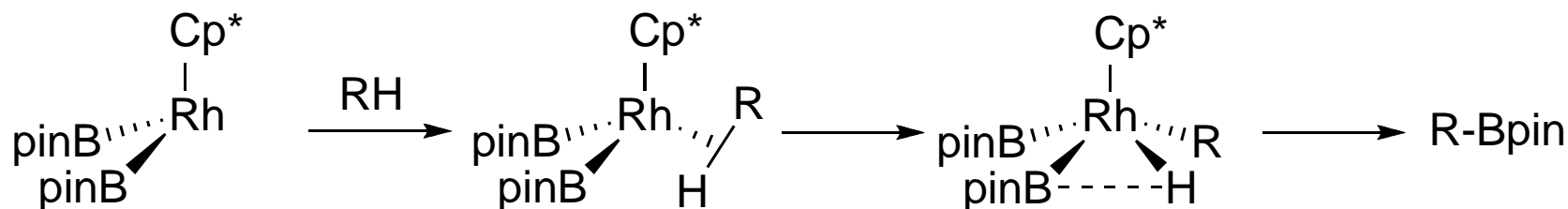


Why C-H Activation with Boron Reagents?

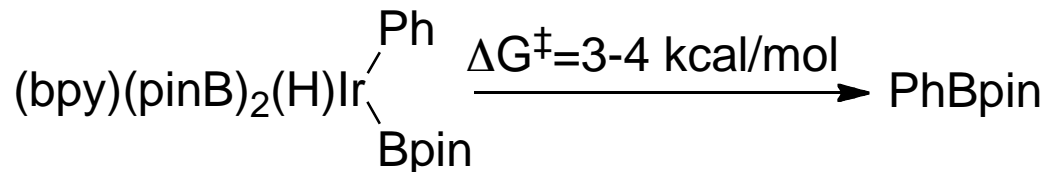
- Strong B-C bond makes the C-H bond functionalization favored thermodynamically



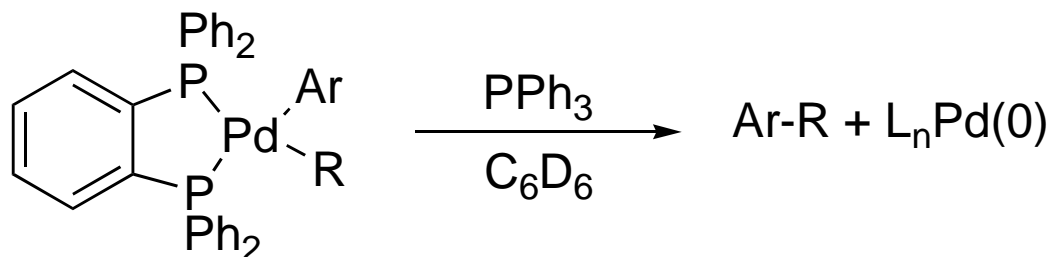
- Participation of the p-orbital and the strong σ -donation by an “anionic” boryl ligand leads to a low barrier for C-H bond cleavage



- Reversal of bond polarity leads to rapid formation of B-C bonds by metathesis or reductive elimination



Electronic Effect on C-C Reductive Elimination



Taft parameter

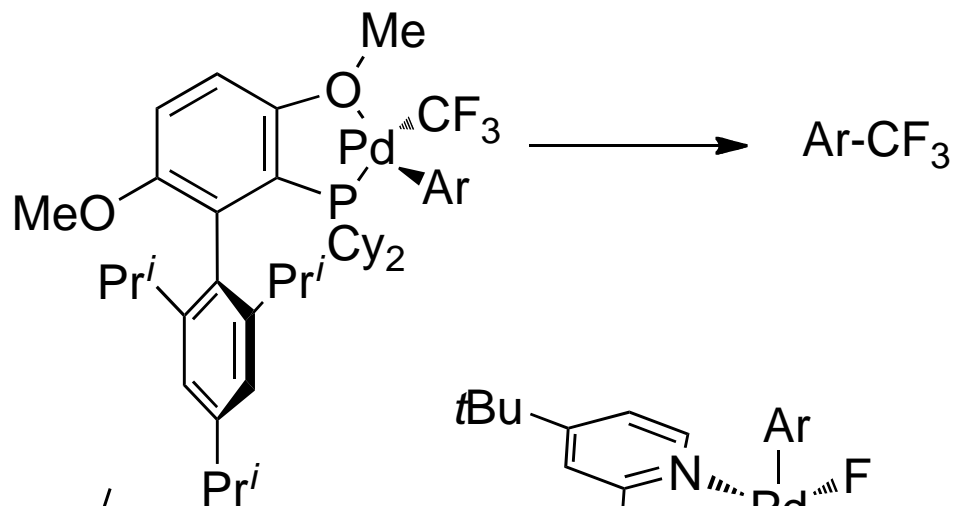
R=	σ^*	$k_{\text{rel}}(110\text{ }^\circ\text{C})$
Me	0.00	>600
CH ₂ Ph	0.22	>250
CH ₂ C(O)Ar	0.60	31
CH ₂ CF ₃	0.92	1.7
CH ₂ CN	1.30	1
CF ₃	2.60	no rxn
CH(CO ₂ Me) ₂		no rxn

Culkin, D.A.; Hartwig, J.F. *J. Am. Chem. Soc.* **2001**, 123, 5816-5817.

Culkin, D.A.; Hartwig, J.F. *J. Am. Chem. Soc.* **2002**, 124, 9330-9331.

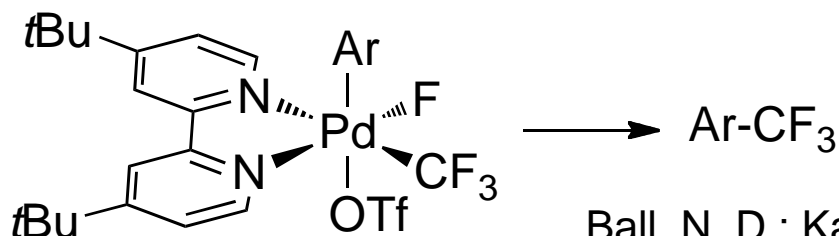
Culkin, D.A., Hartwig, J.F. *Organometallics*, **2004**, 23, 3398-3416.

Reductive Elimination of Trifluoroarenes

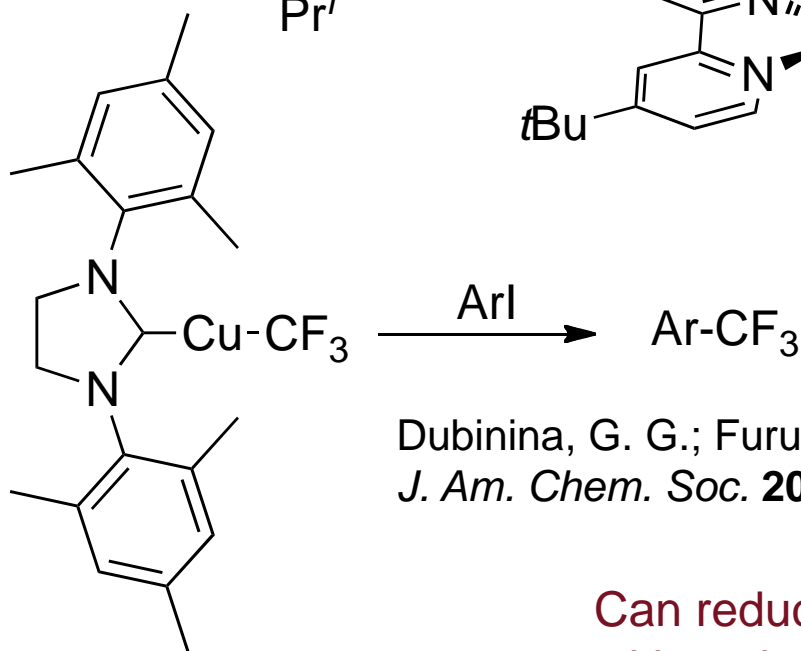


Cho, E. J.; Senecal, T. D.; Kinzel, T.; Zhang, Y.; Watson, D. A.; Buchwald, S. L. *Science* **2010**, 328, 1679.

For Xantphos-Pd see:
Grushin, V. V.; Marshall, W. J. *J. Am. Chem. Soc.* **2006**, 128, 12644.



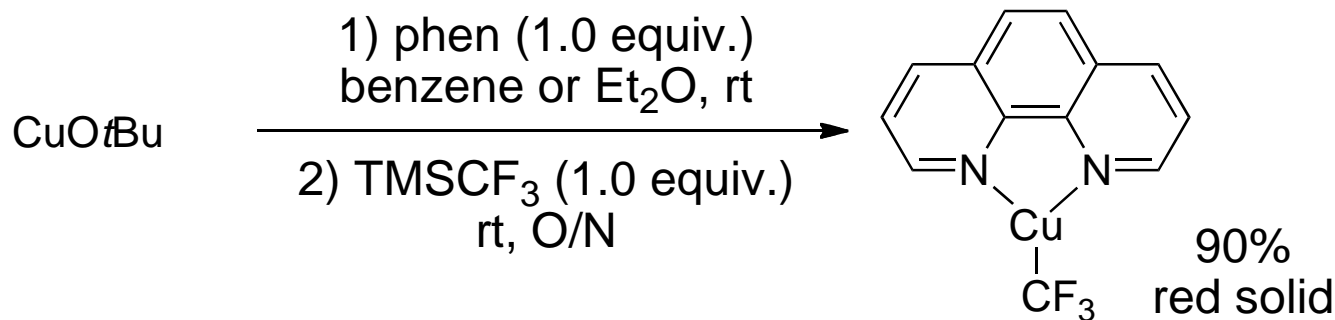
Ball, N. D.; Kampf, J. W.; Sanford, M. S. *J. Am. Chem. Soc.* **2010**, 132, 2878.



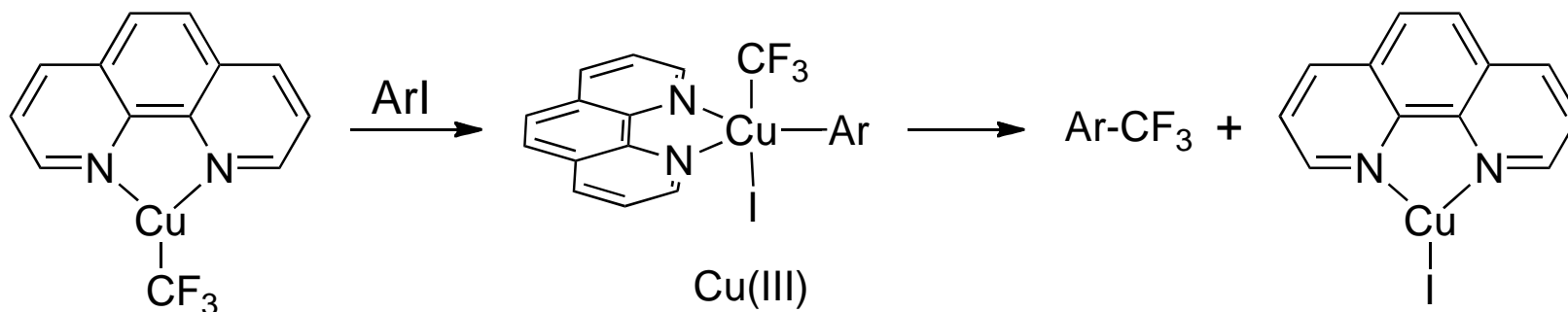
Dubinina, G. G.; Furutachi, H.; Vicic, D. A. *J. Am. Chem. Soc.* **2008**, 130, 8600.

Can reductive elimination of trifluoroarenes occur with an inexpensive metal, ligand and reagent?

Synthesis and Reactivity of (phen)CuCF₃



If an aryl-Cu(III) intermediate is generated, reductive elimination should be faster than from palladium(II).



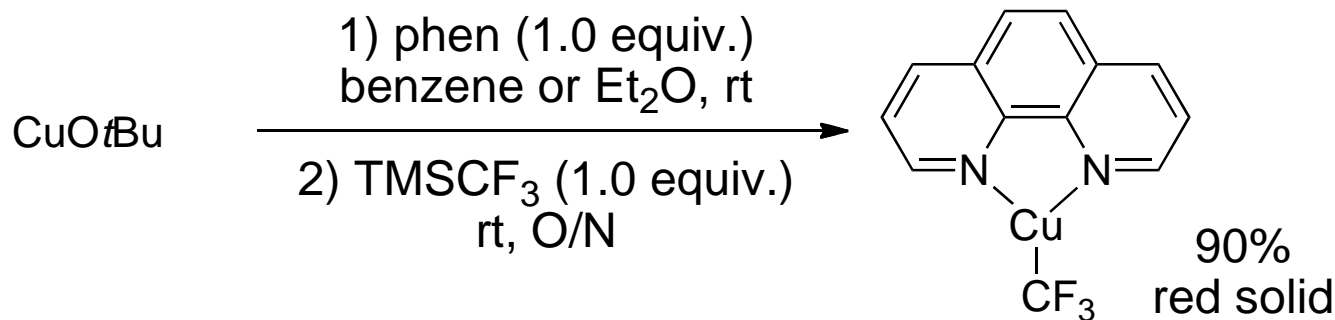
Reductive elimination from Cu(III) occurs with a low barrier:

Tye, J. W.; Weng, Z.; Johns, A. M.; Incarvito, C. D.; Hartwig, J. F. *J. Am. Chem. Soc.* **2008**, 130, 9971.

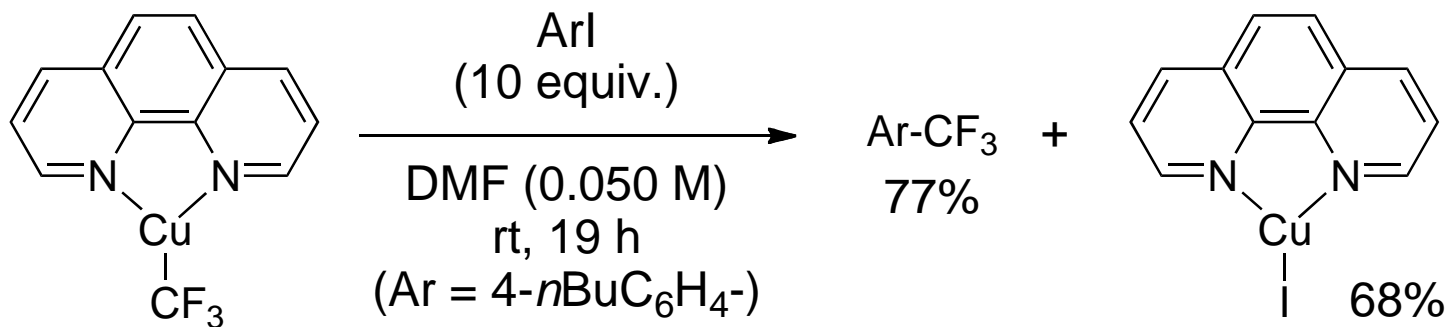
Zhang, S.-L.; Liu, L.; Fu, Y.; Guo, Q.-X. *Organometallics* **2007**, 26, 4546.

Tye, Jesse W.; Weng, Z.; Giri, R.; Hartwig, John F. *Angew. Chem. Int. Ed.* **2010**, 49, 2185.

Synthesis and Reactivity of (phen)CuCF₃



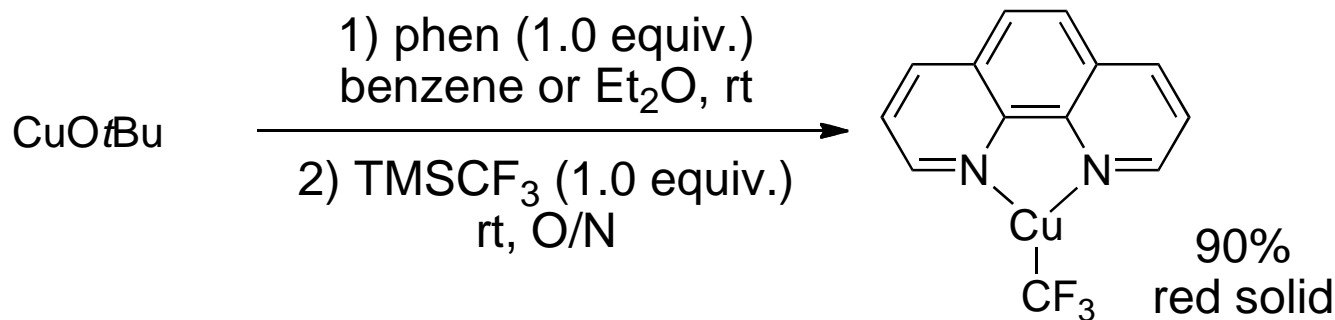
In the lab...



What is the stability of (phen)Cu-CF₃ and what is the scope of its reactions with ArX?

Hiroyuki Morimoto

Stability of (phen)CuCF₃



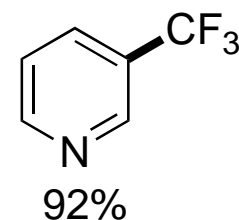
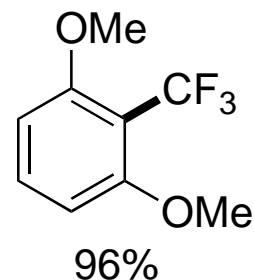
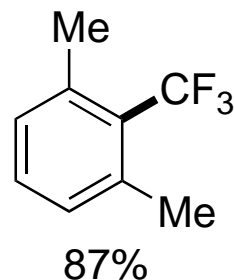
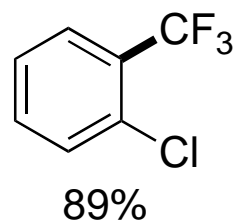
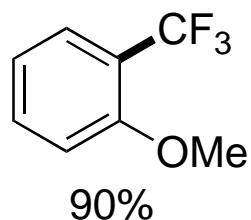
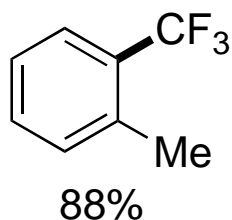
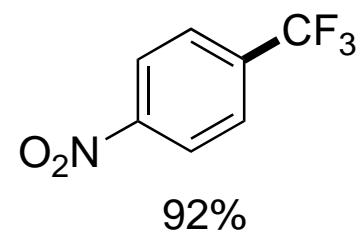
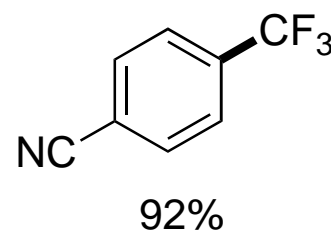
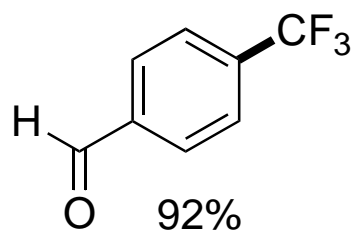
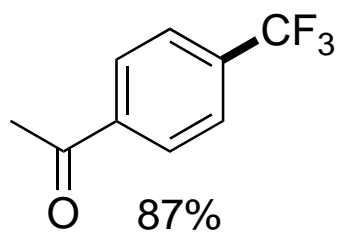
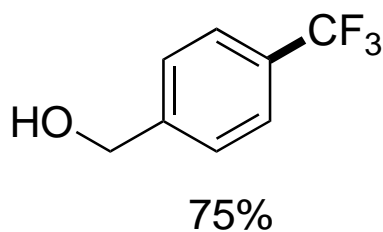
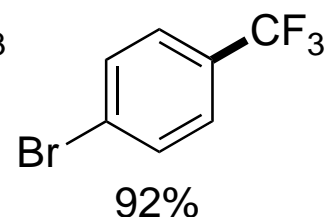
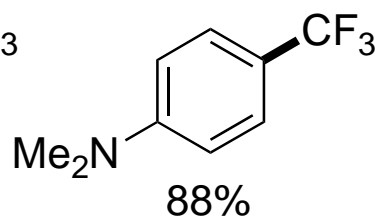
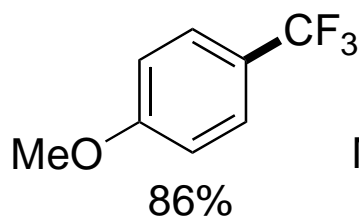
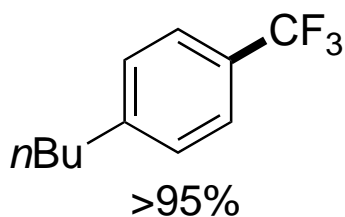
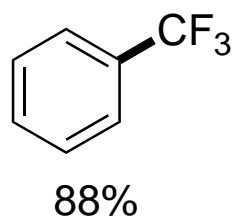
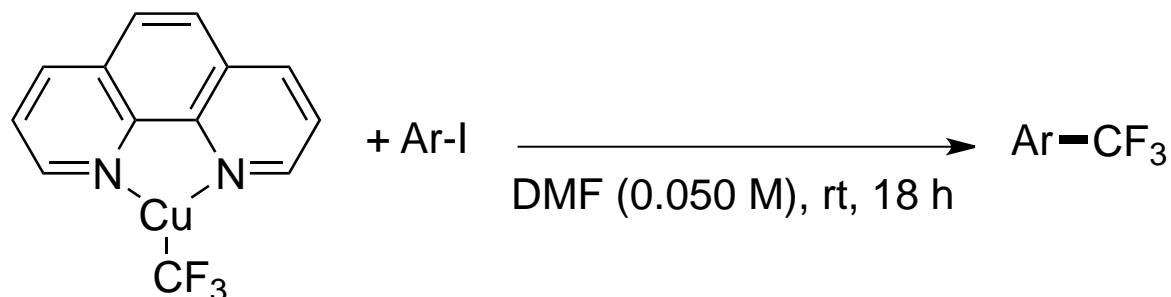
Available from Aldrich
(agreement with Catylix)

Stability:

conditions	solid state	in solution (0.05 M in DMF)
under N ₂	stable at rt > 1 month	stable at rt for 1 day decomposed at 50 °C
in air	decomposed < 1 day	not examined

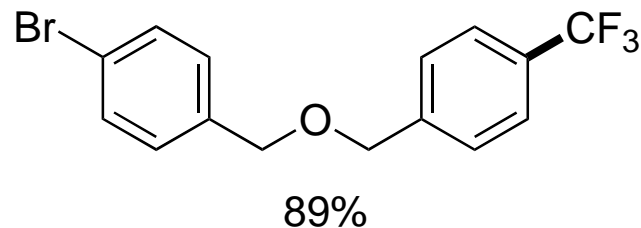
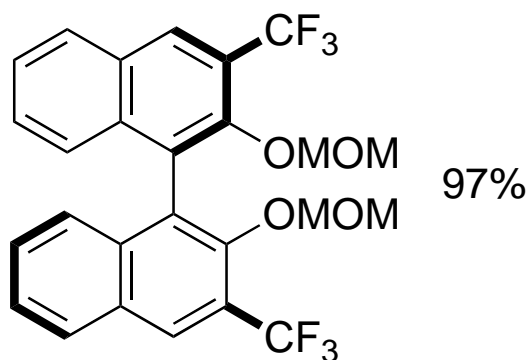
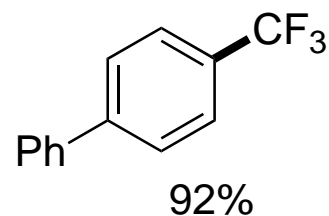
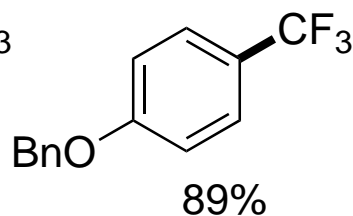
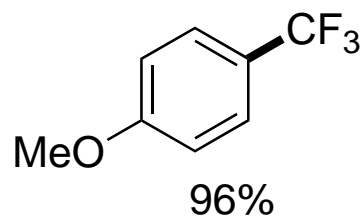
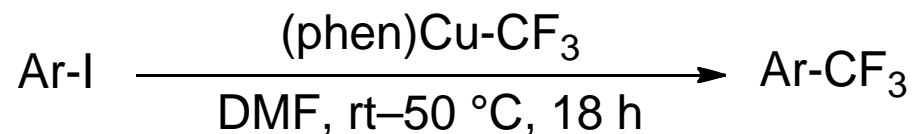
Hiroyuki Morimoto

Trifluoromethylation with (phen)CuCF₃



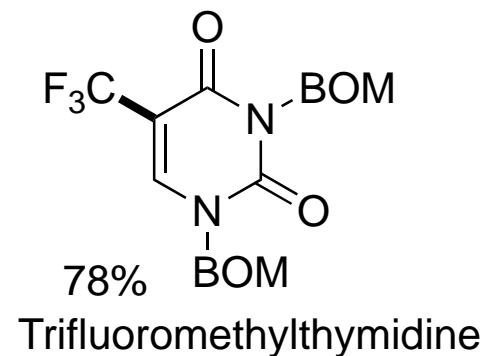
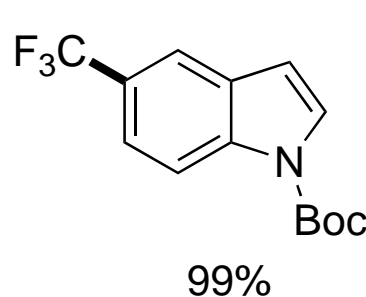
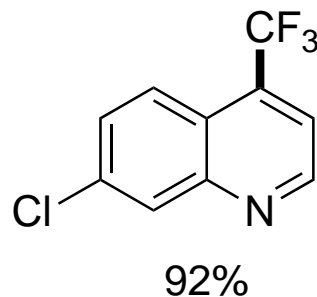
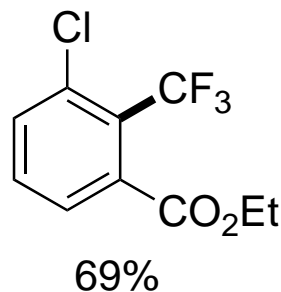
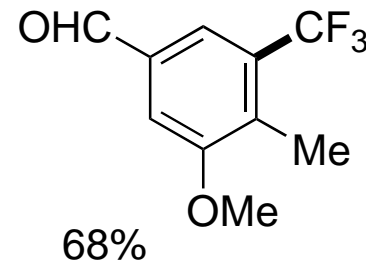
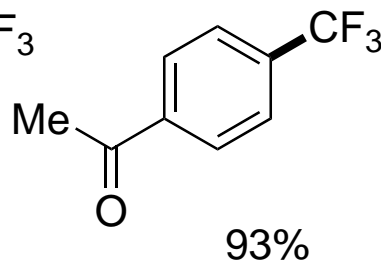
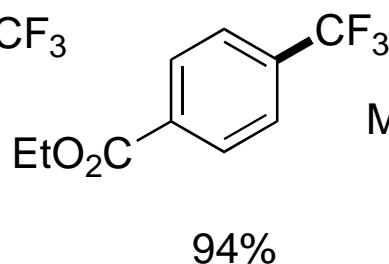
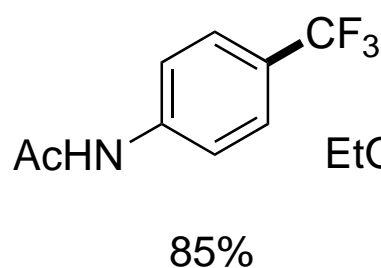
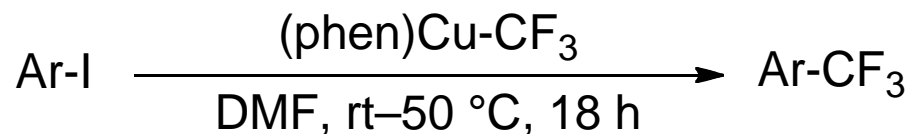
Scope of the Reactions of (phen)CuCF₃

Reactions with 1:1 ArI to copper



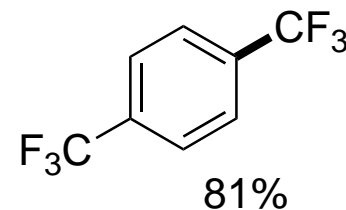
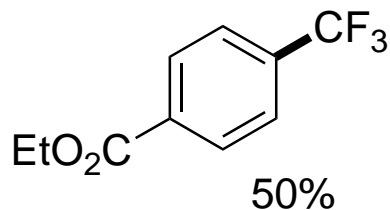
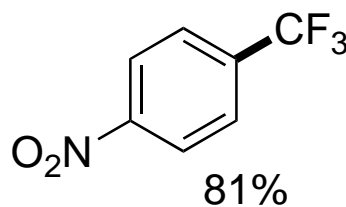
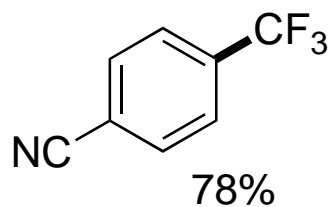
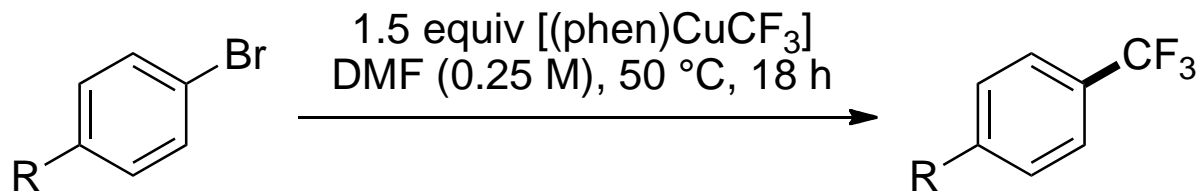
Scope of the Reactions of (phen)CuCF₃

Reactions with 1:1 ArI to copper

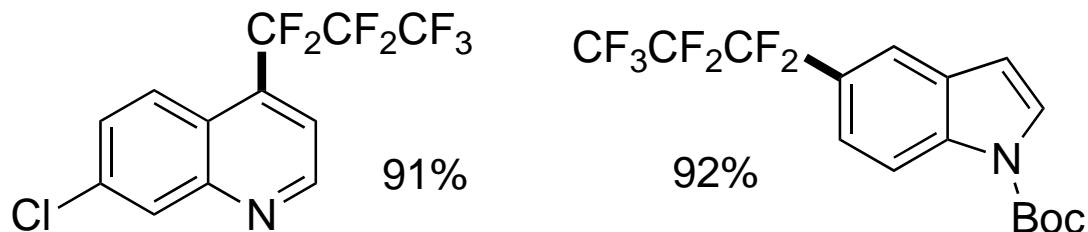
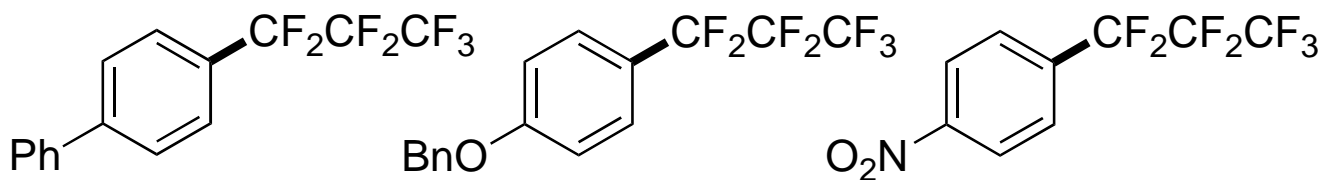
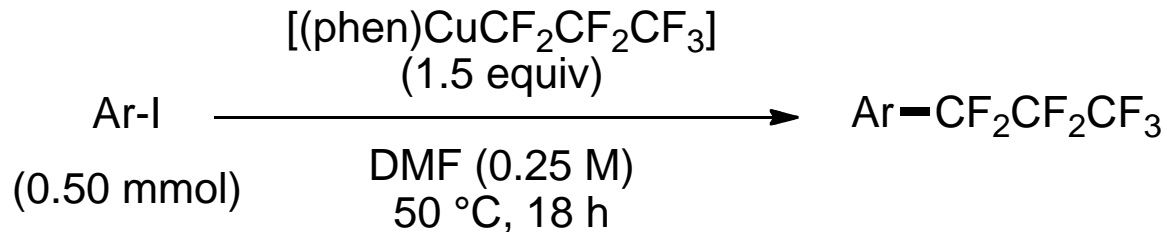


Scope of the Reactions of (phen)CuCF₃

Reactions of aryl bromides



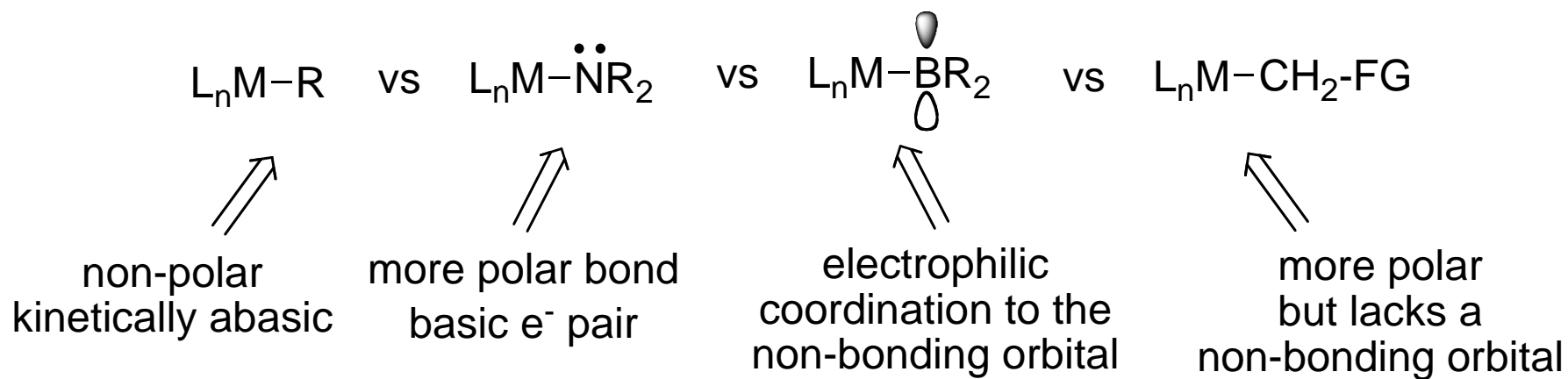
Trifluoroalkylation by (phen)CuR_f



Organometallic Chemistry of Metal-Heteroatom vs Metal-Carbon Bonds

Underlying Principles

comparison of the properties of "anionic" ligands



Pauling Electronegativities: Pd: 2.20

C: 2.55

N: 3.04

B: 2.04

Acknowledgments

C-H Bond Functionalization

Karen Waltz
Huiyuan Chen
Sabine Schlecht
Kazumori Kawamura
Kevin Cook
Natia Anastasi
Makato Takahashi
Domingo Garcia
Yuichiro Kondo
Chulsung Bae
Doris Kunz
Joshua Lawrence
Timothy Boller
Naofumi Tsukada (Si)
Jaclyn Murphy
Christoph Tzschucke
Carolyn Wei
Tim Boebel
Dan Robbins
Carl Liskey
Eric Simmons

Shell:

Tom Semple (Shell)

Minnesota:

Marc Hillmyer
Nicole Wagner, Nicole Boaen

Hokkaido:

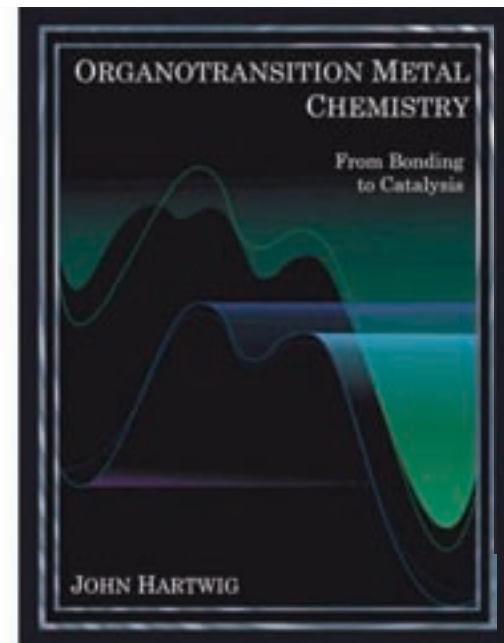
Tatsuo Ishiyama
Norio Miyaura
Jun Takagi, Yusuke Nobuta

Texas A&M:

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