

Processing aspects of enzymatic rearrangement *(for hardstock production)*

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Contents



- Introduction
- Kinetics & deactivation
- Processing & reactor set-up



History ER process



- '33 earliest publication on (trans)esterification¹
- '79 first patent (interesterification)²
- '83 first application in oils and fats modification³
- '85 publication on biocatalysis in organic media⁴
- '86 solvent free processing in PBR

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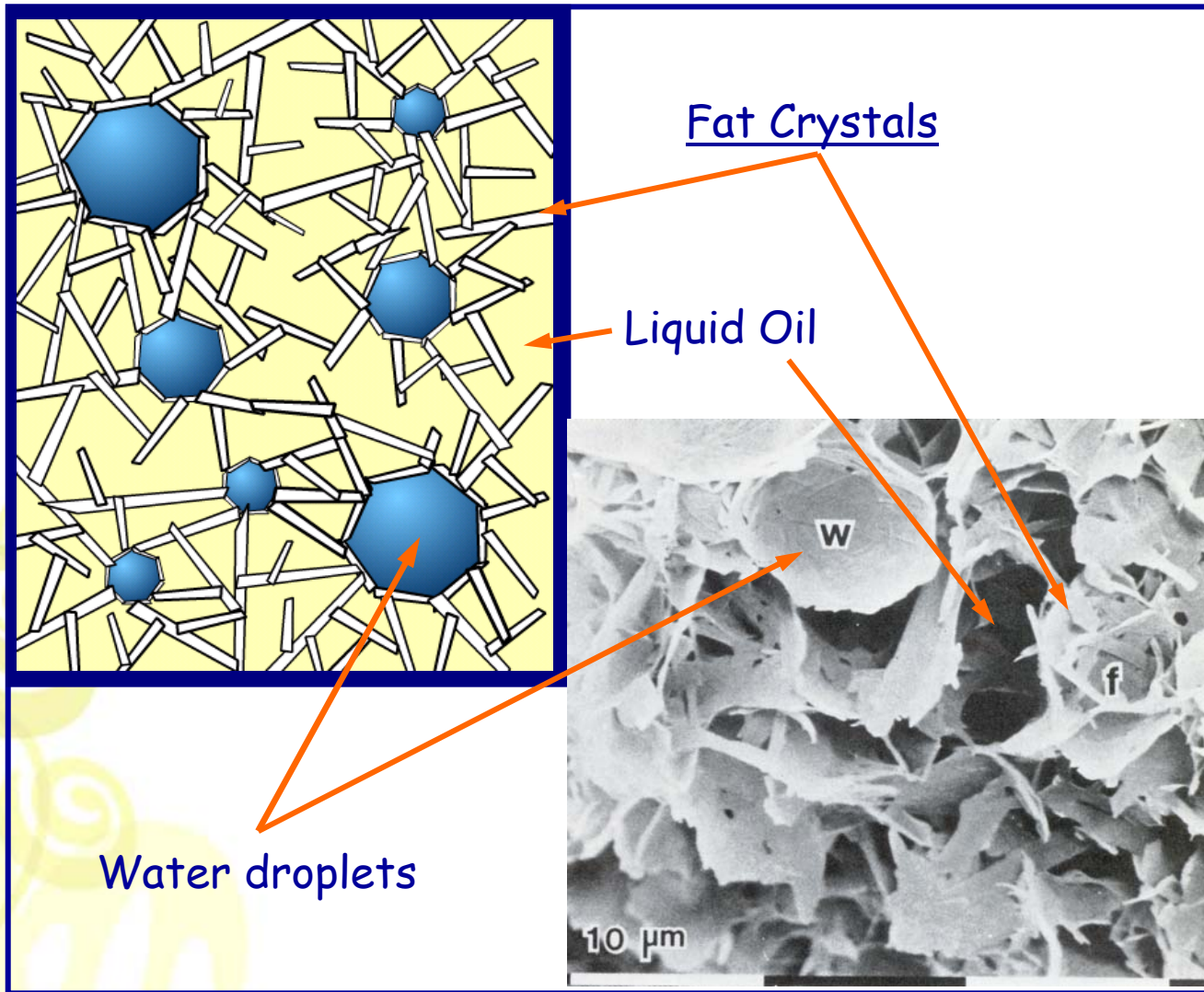
2004 First large scale plant hardstock production US

2007 First large scale plant hardstock production EU

2011 ~ 300 kT capacity (EU)
~ 500 kT capacity (Global)

1. Sym E.A. (1933), Biochem. Z., 258, 304;
2. Coleman M.H., Macrae A.R. (1979), US Patent 4,275,081
3. Macrae A.R. (1983), JAACS, 60, 2, 243.
4. Zaks A., Klivanov A.M. (1985), Proc. Natl. Acad. Sci, 82, 3192.
5. Hansen T.T., Eigtved P. (1986), World Conf. Oils Fats Ind.', S. 365.

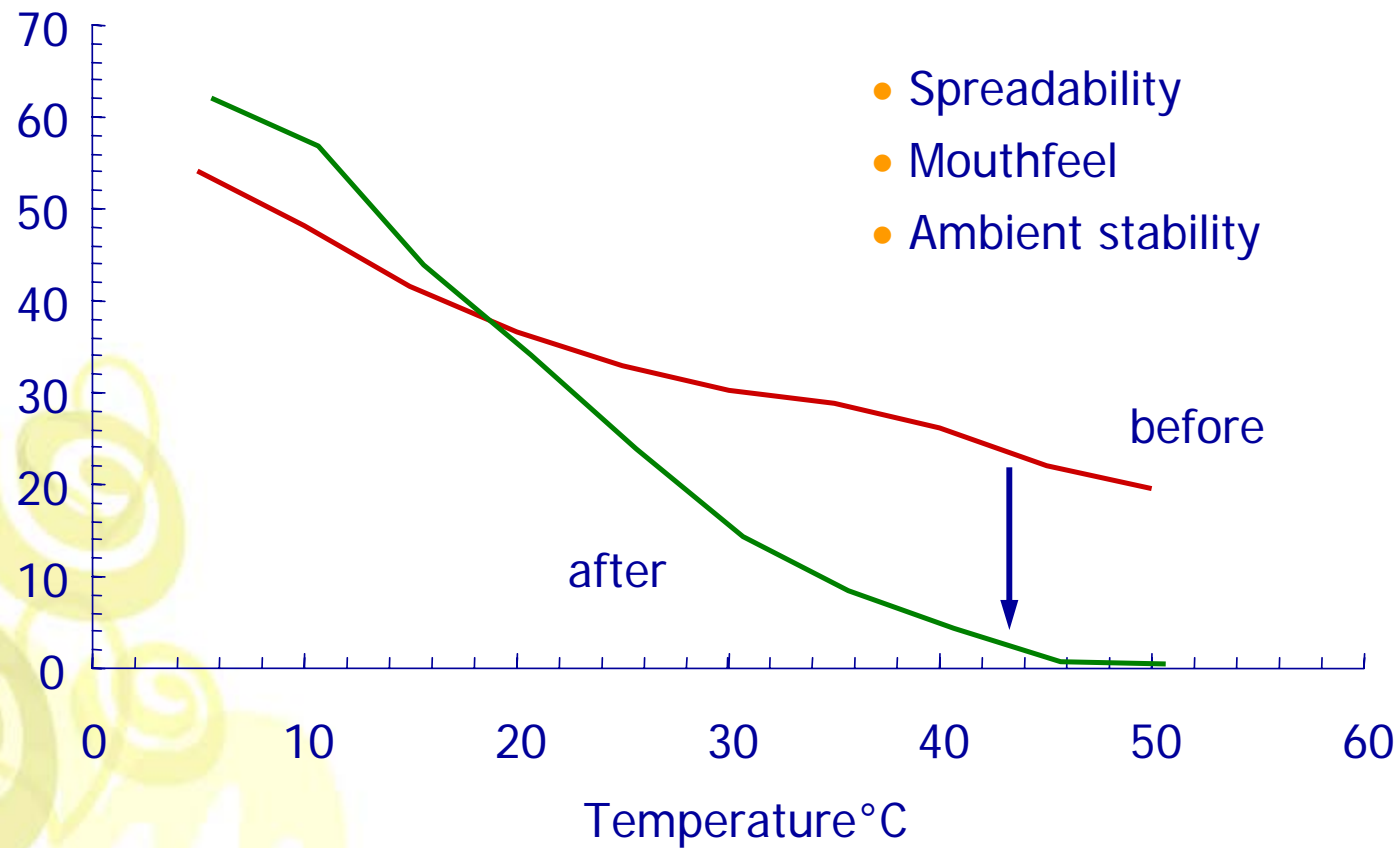
Structure of a fat continuous spread



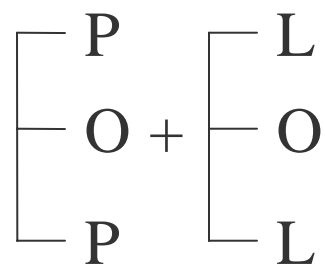
Fat melting properties



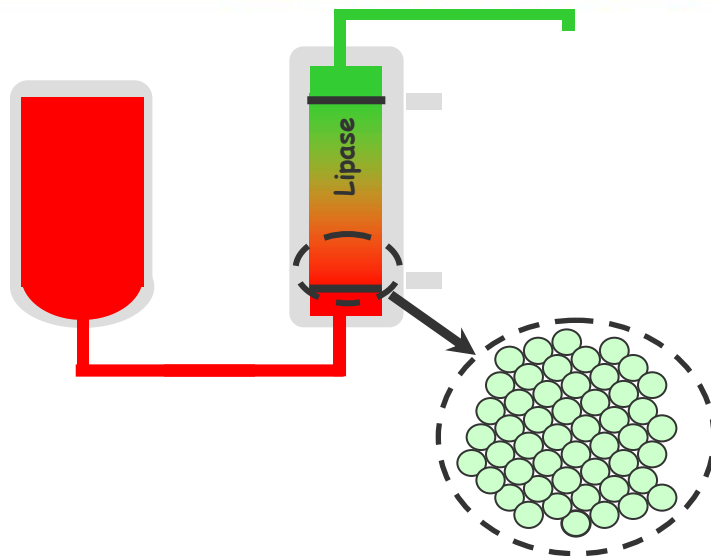
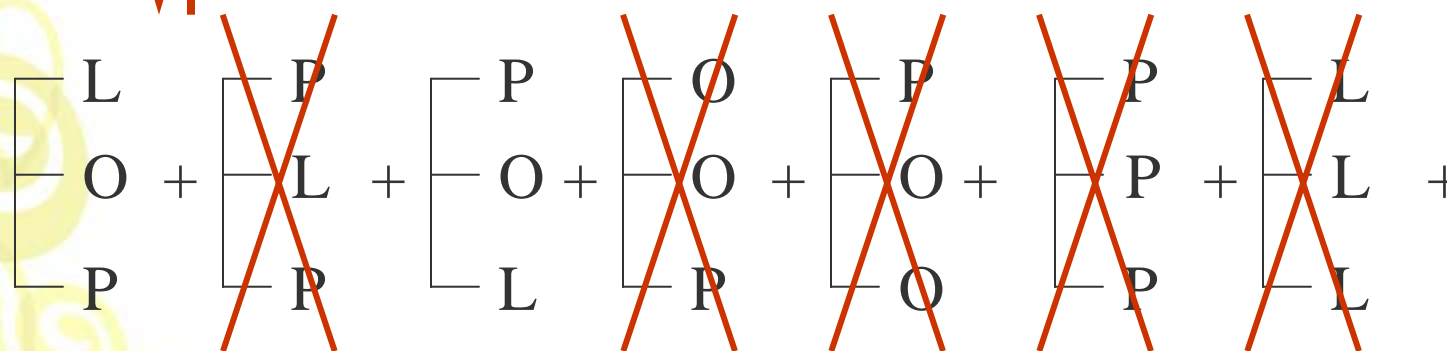
Solid Fat Content [SFC %-w/w]



Enzymatic Rearrangement

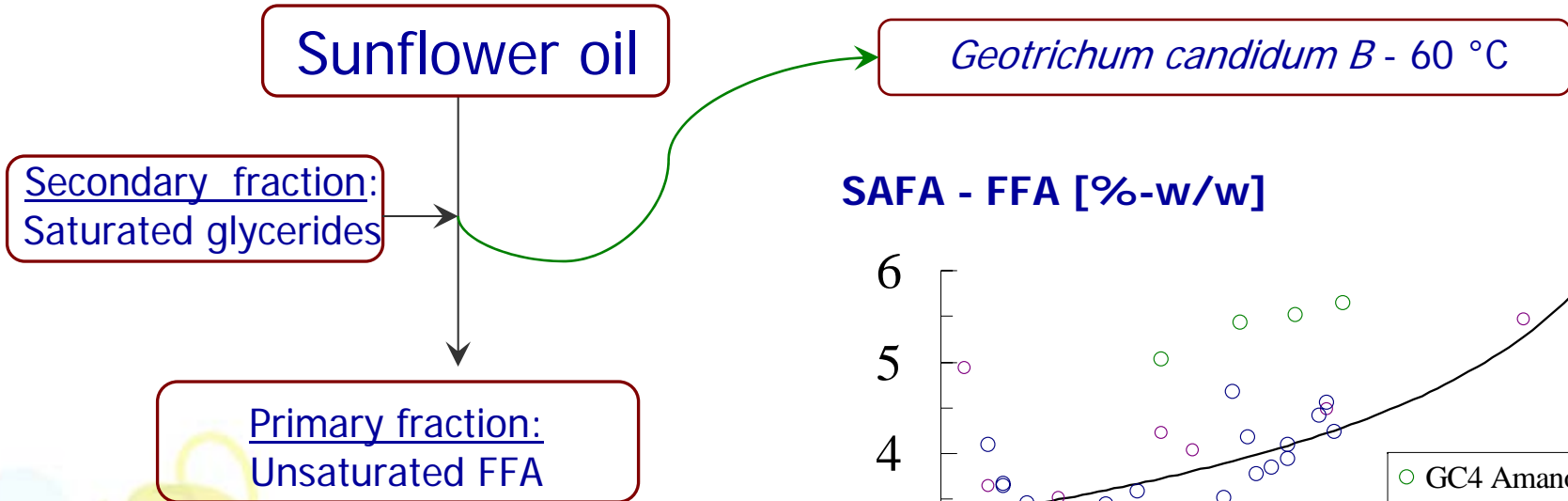


Lipase 60 °C

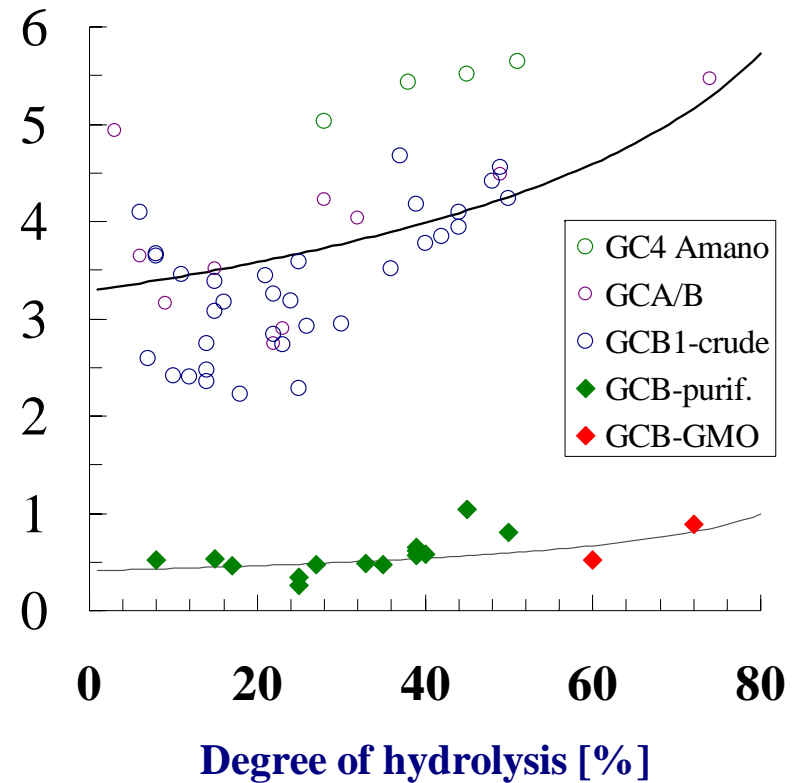


Lipases generally *sn*-1,3 specific $\xleftrightarrow{\text{acyl-migration}}$ catalyst random

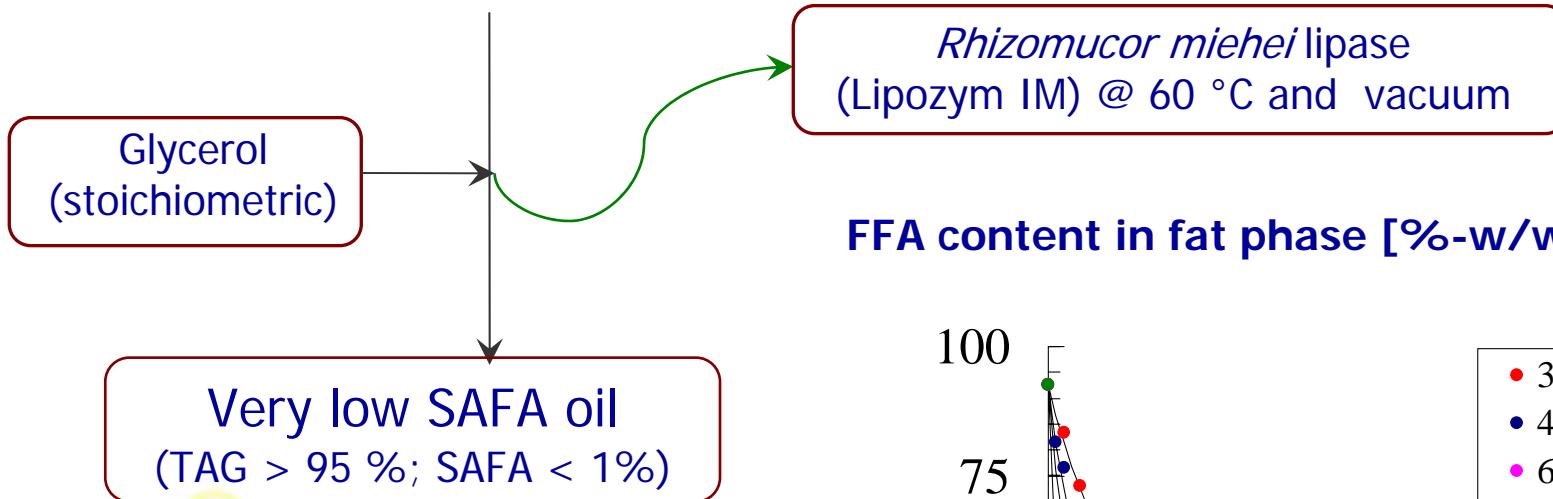
Cis- $\Delta 9$ fatty acid hydrolysis



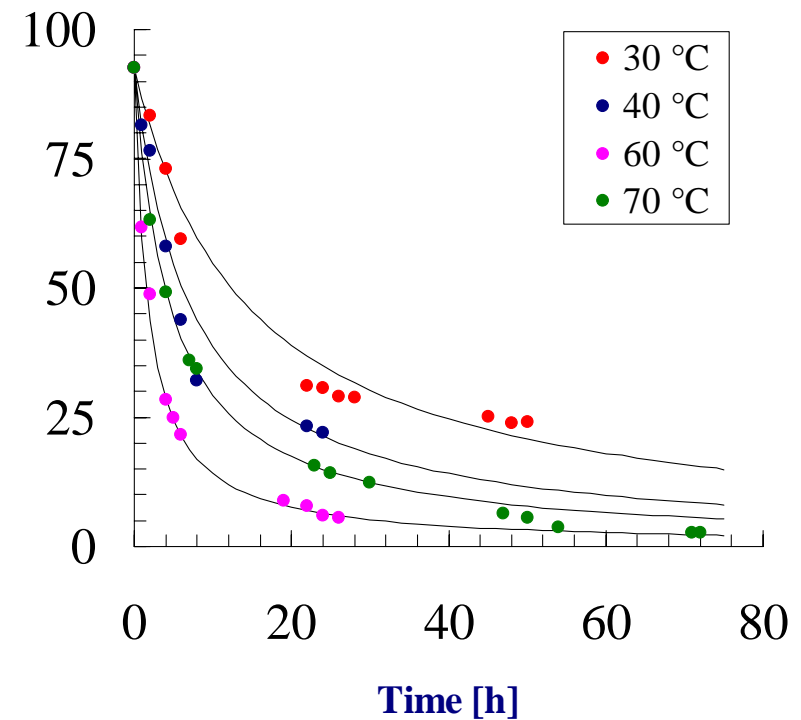
SAFA - FFA [%-w/w]



Impact acylmigration



FFA content in fat phase [%-w/w]



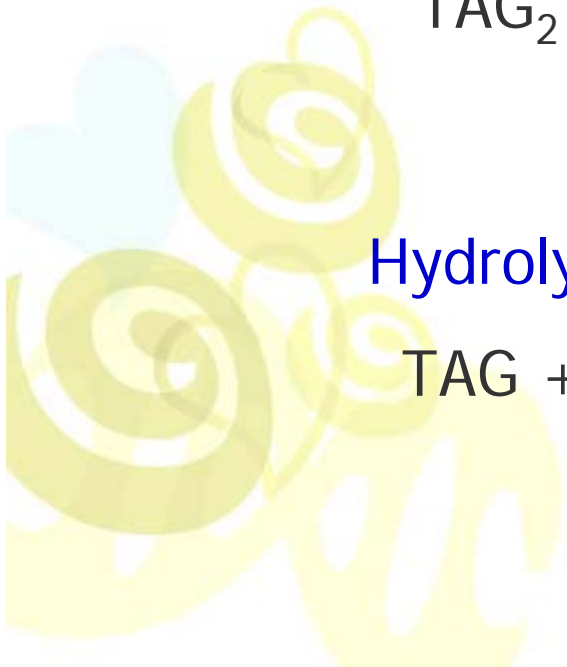
ER kinetics



TAG Rearrangement (primary - slow)



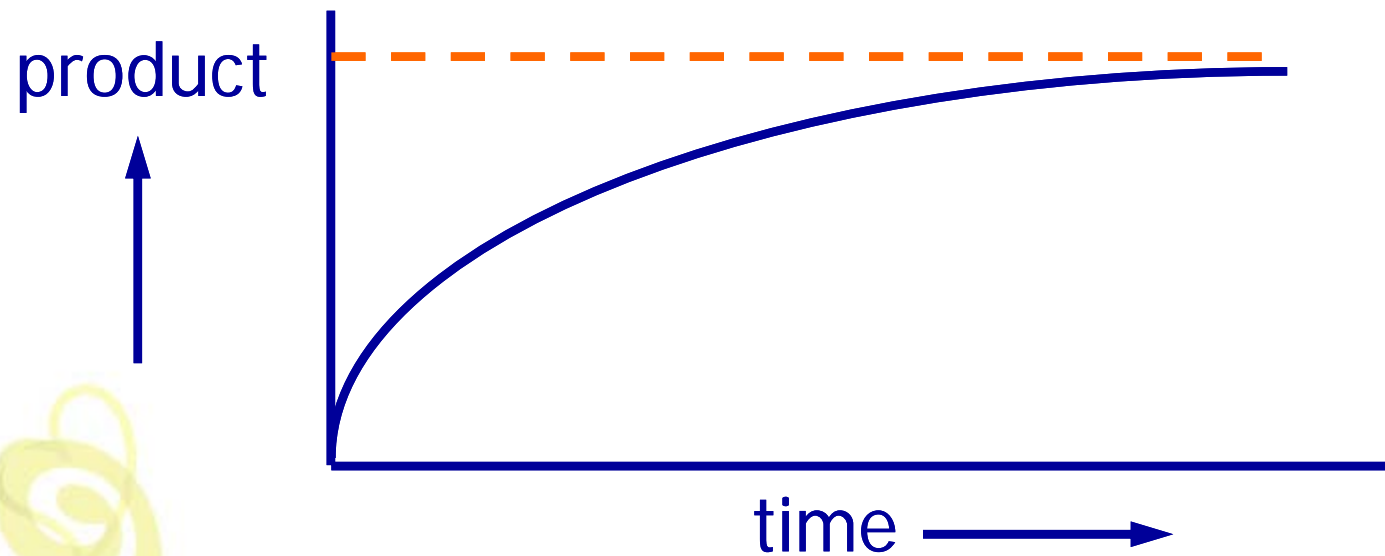
Hydrolysis (secondary - quick)



Modeling - Pseudo 1st-order



“Just swapping the fatty acids...”



$$k_{\text{cat}} \cdot C_{\text{cat}} \cdot t = -\ln(1-X)$$

X = relative degree of conversion

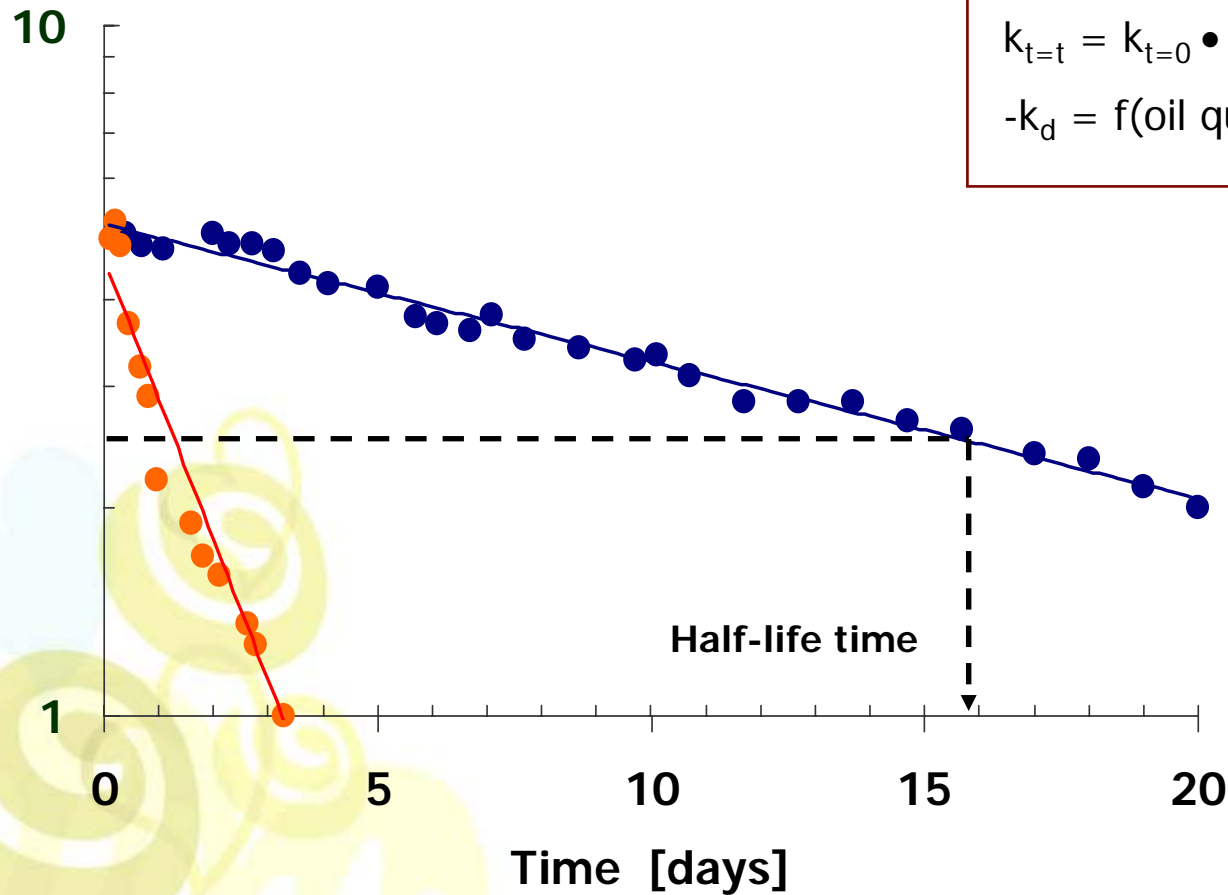
(representative change between starting blend and product)



Catalyst deactivation (1st order)



Activity (ton/m³. h)



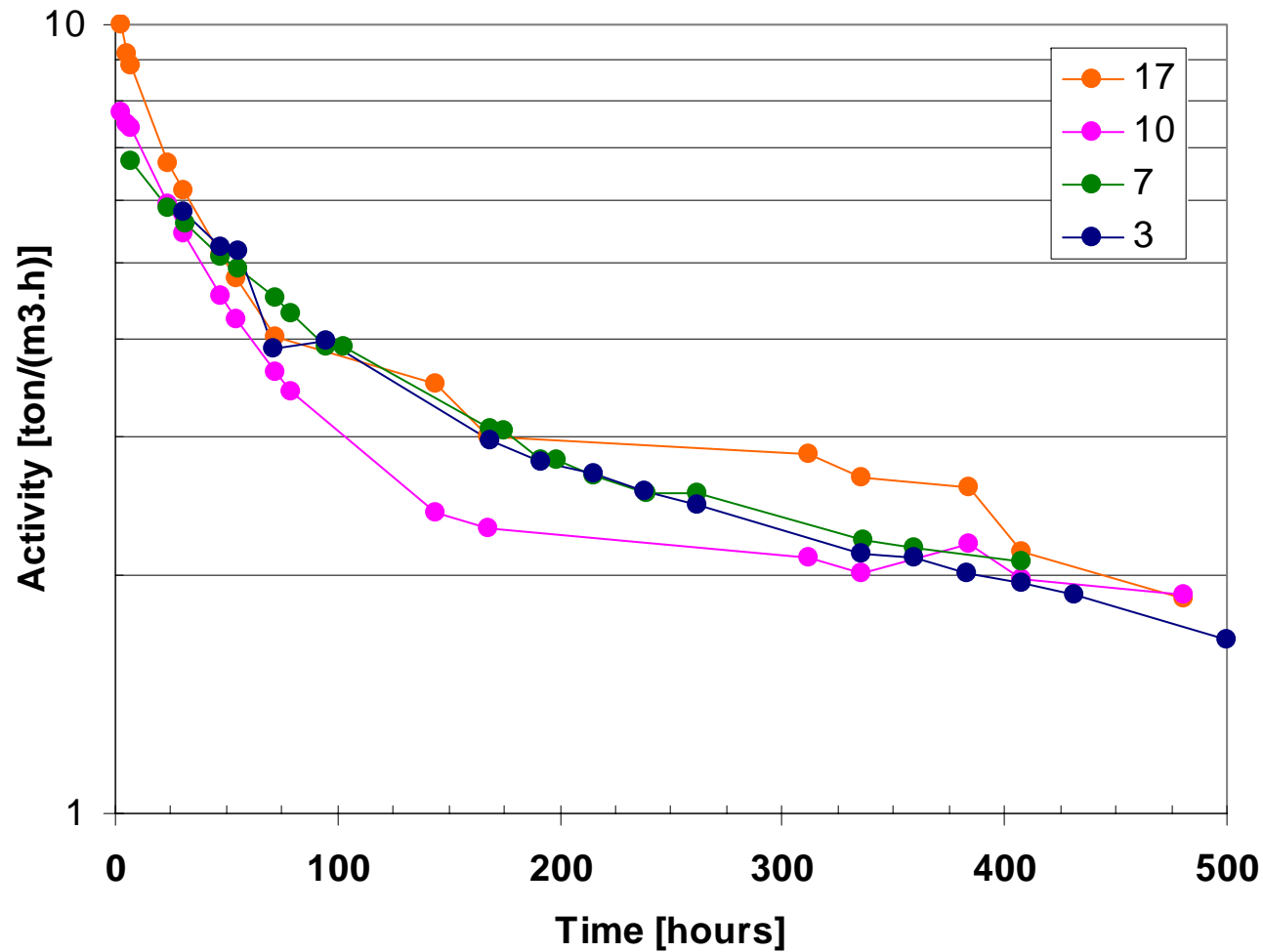
First-order deactivation kinetics:

$$k_{t=t} = k_{t=0} \cdot e^{-k_d \cdot t}$$

$$-k_d = f(\text{oil quality}, T, a_w)$$

R. oryzae lipase / Accurel EP100

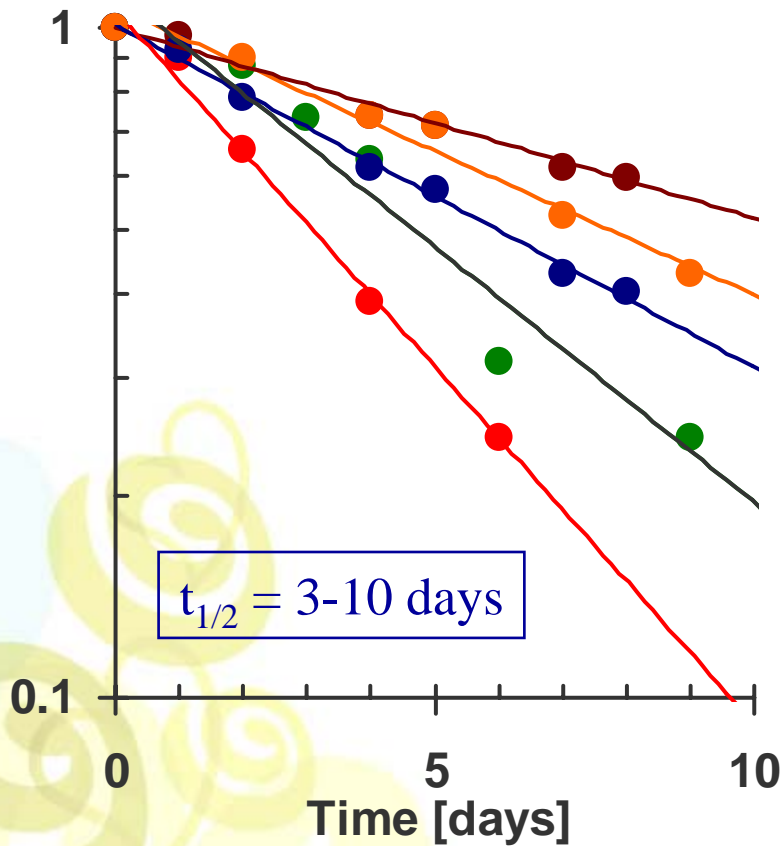
Catalyst deactivation (2nd order)



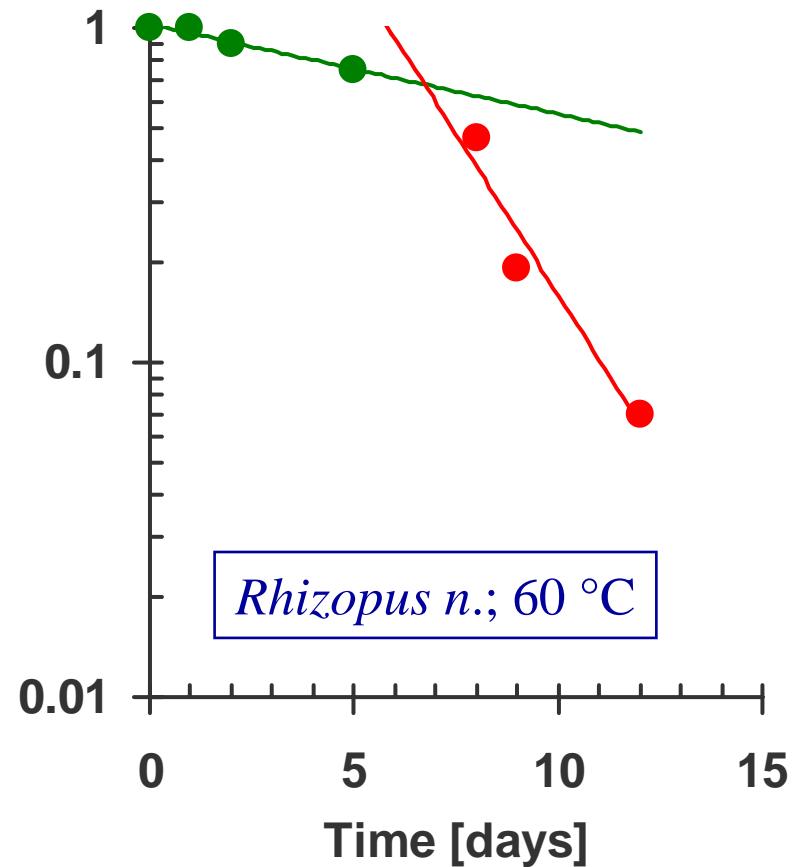
Effect of oil quality



Relative catalyst activity



Relative catalyst activity

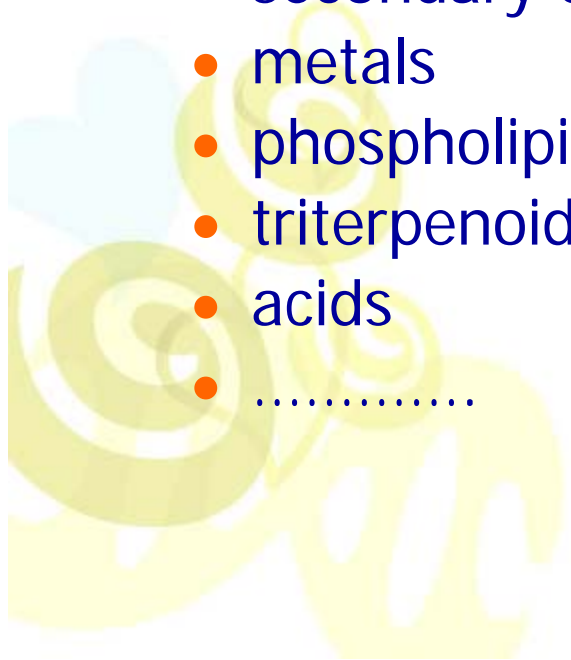


R. oryzae lipase / Accurel EP100

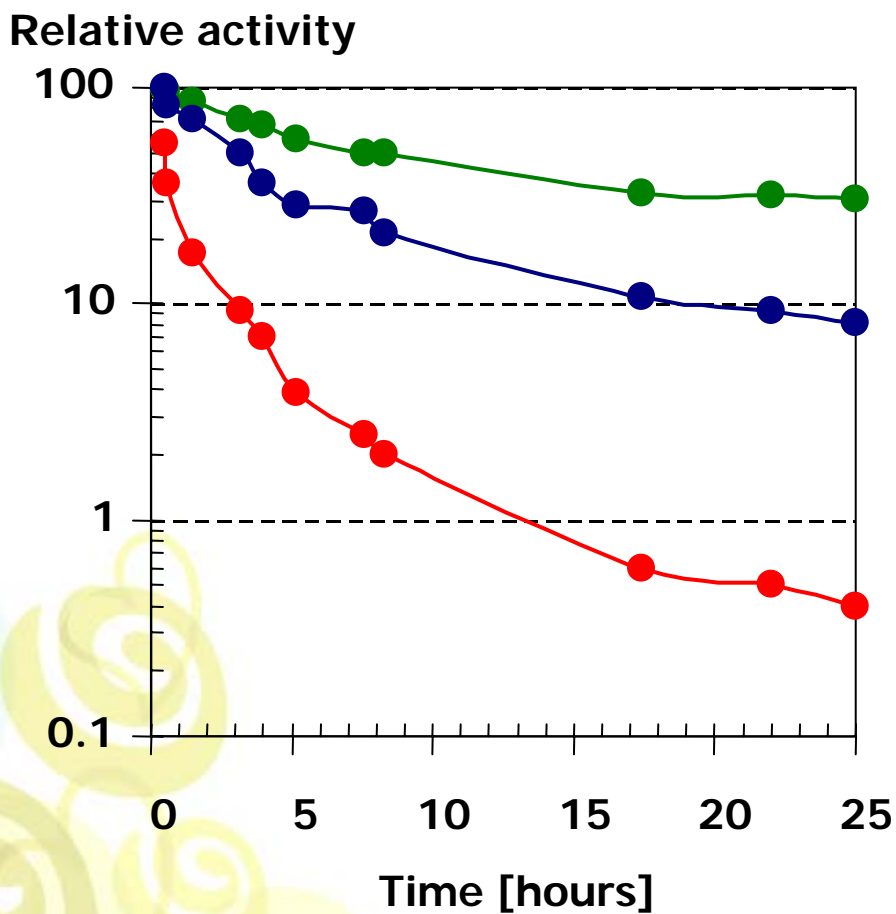


Deactivation mechanisms

- Thermal (denaturation)
- Fouling (apparent)
- Poisoning
 - hydroperoxides (POV)
 - secondary oxidation products (PAV)
 - metals
 - phospholipids
 - triterpenoids
 - acids
 -



Oxidation products



- Fresh
- Oxidised (POV=5)
- Oxidised (POV=110)

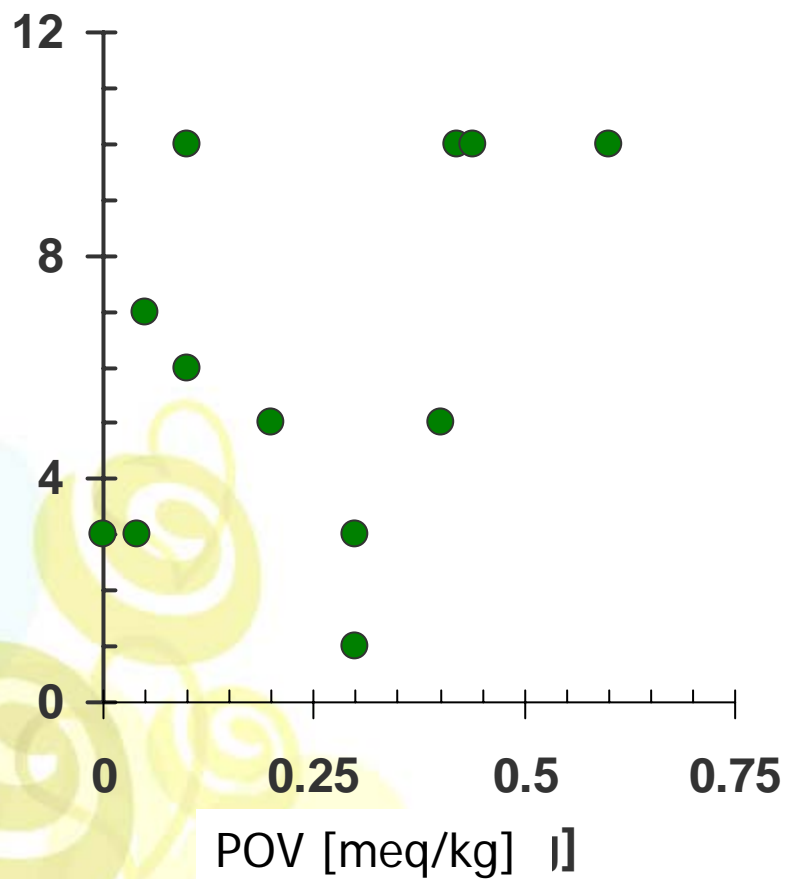
Lipozyme IM20
Butylether; 80 °C



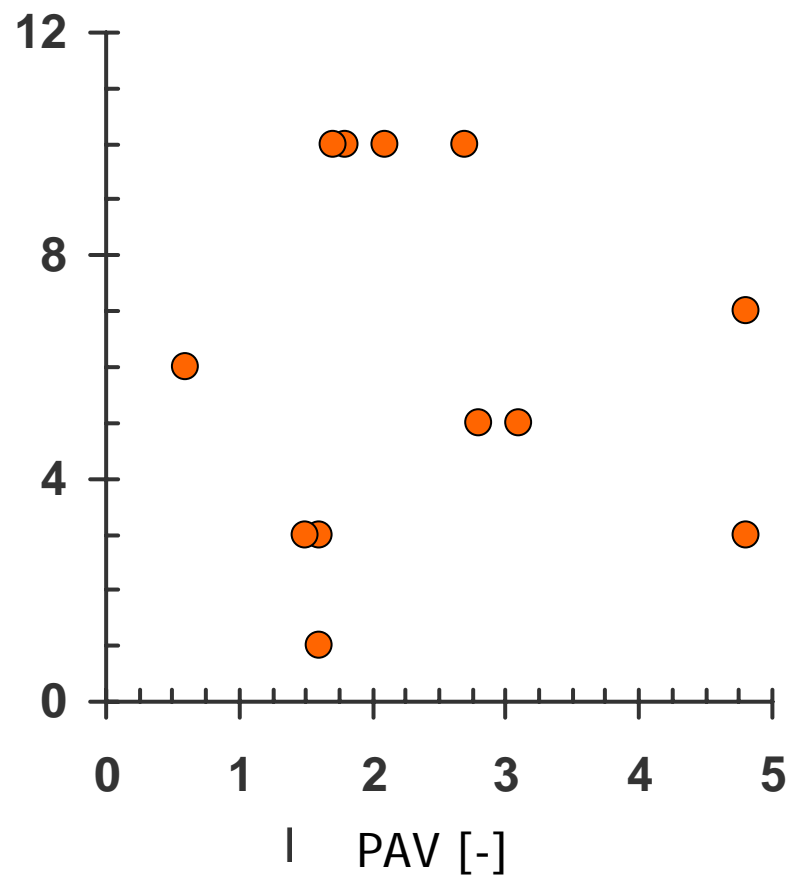
Oil quality parameters (2001)



Half-life time [days]

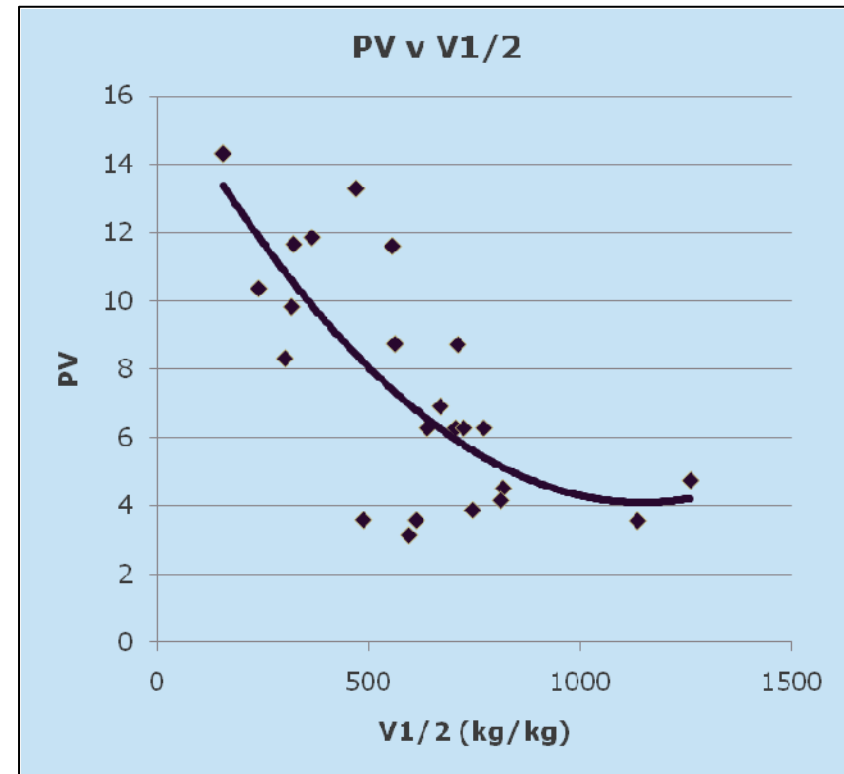
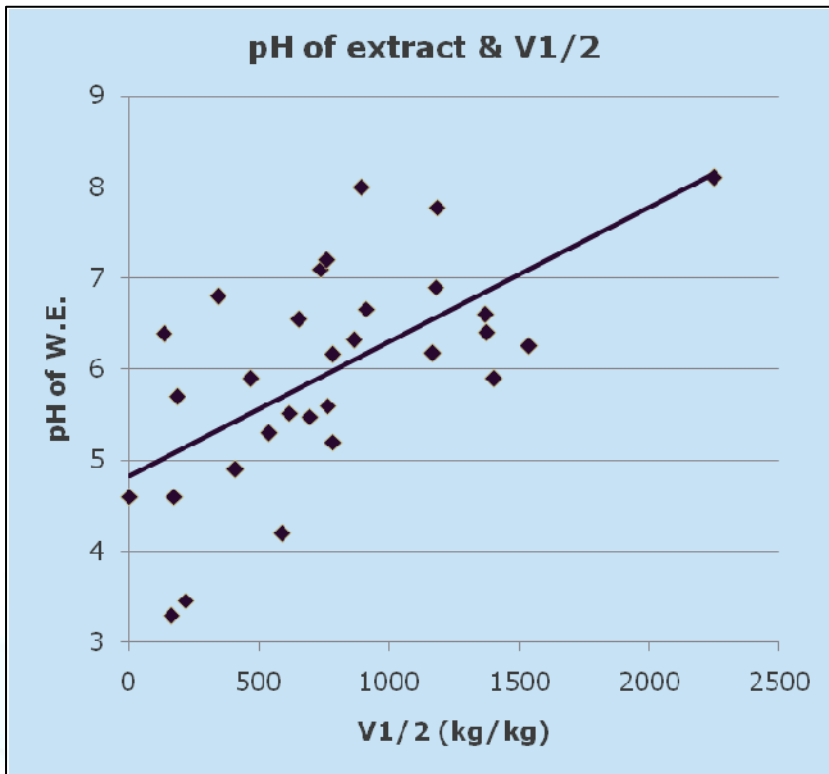


Half-life time [days]



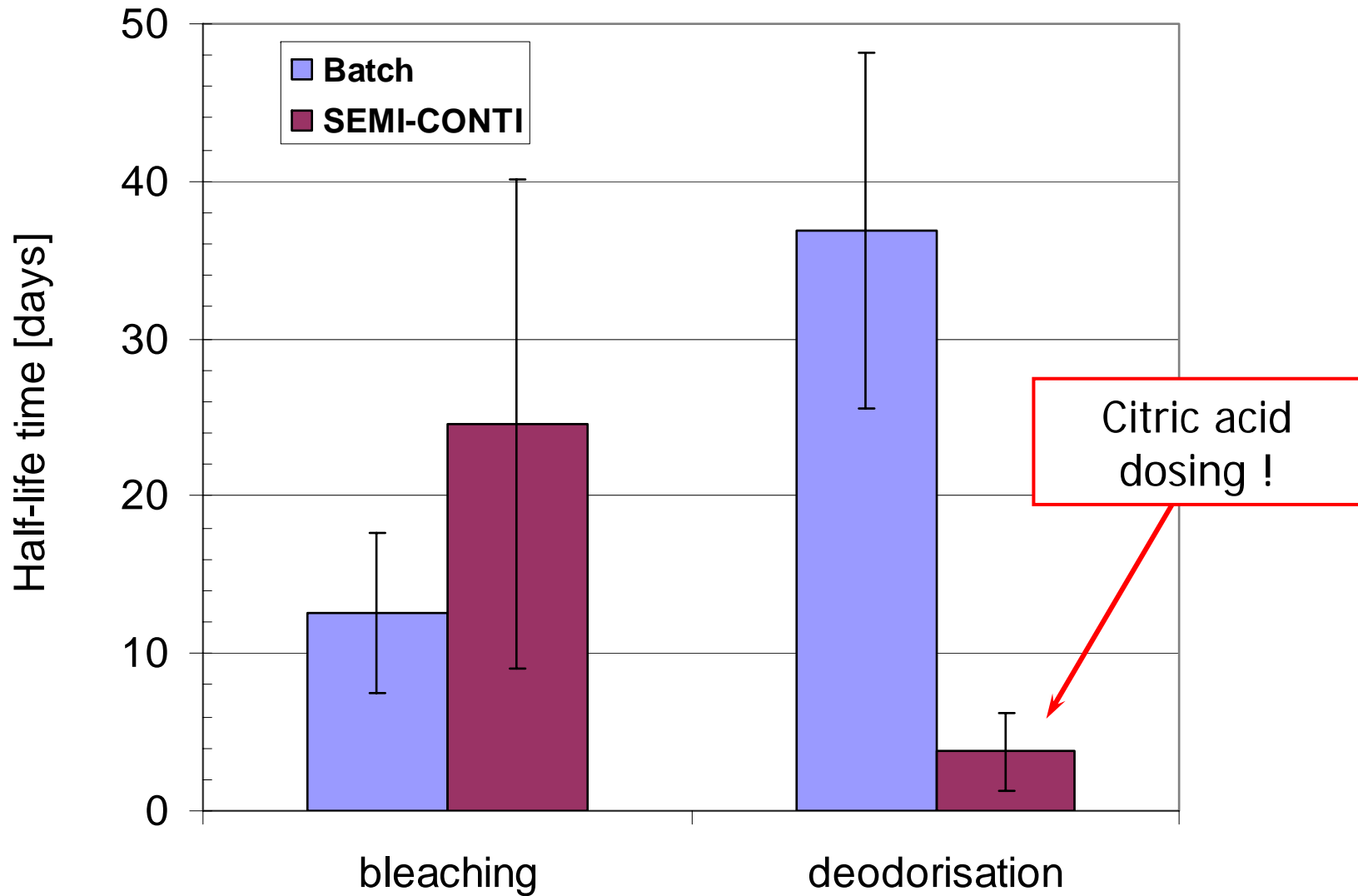
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Oil quality parameters (2005)

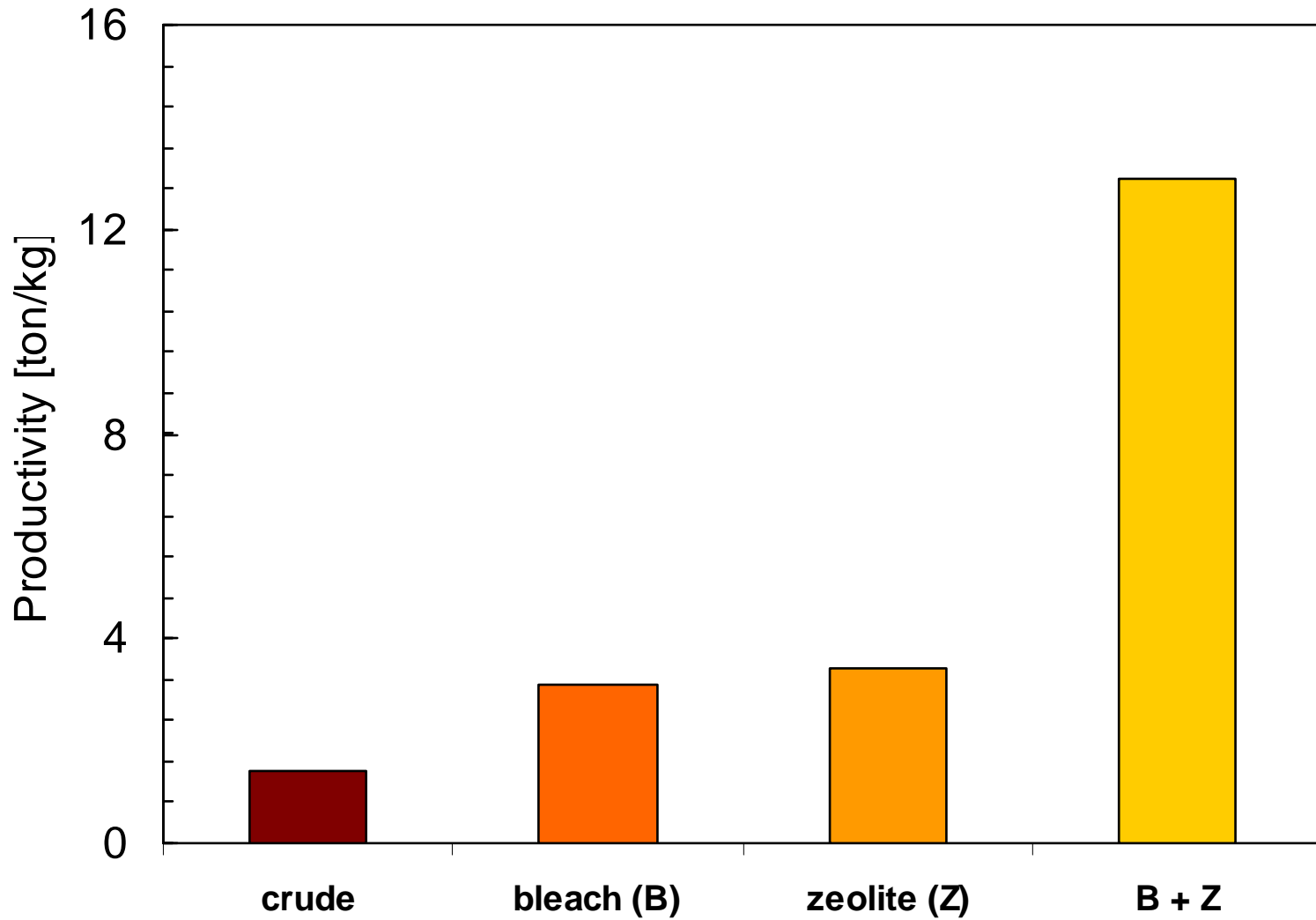


V1/2= the amount of oil processing when reaching half the initial activity

Pretreatment – processing



Pretreatment – silica /zeolites

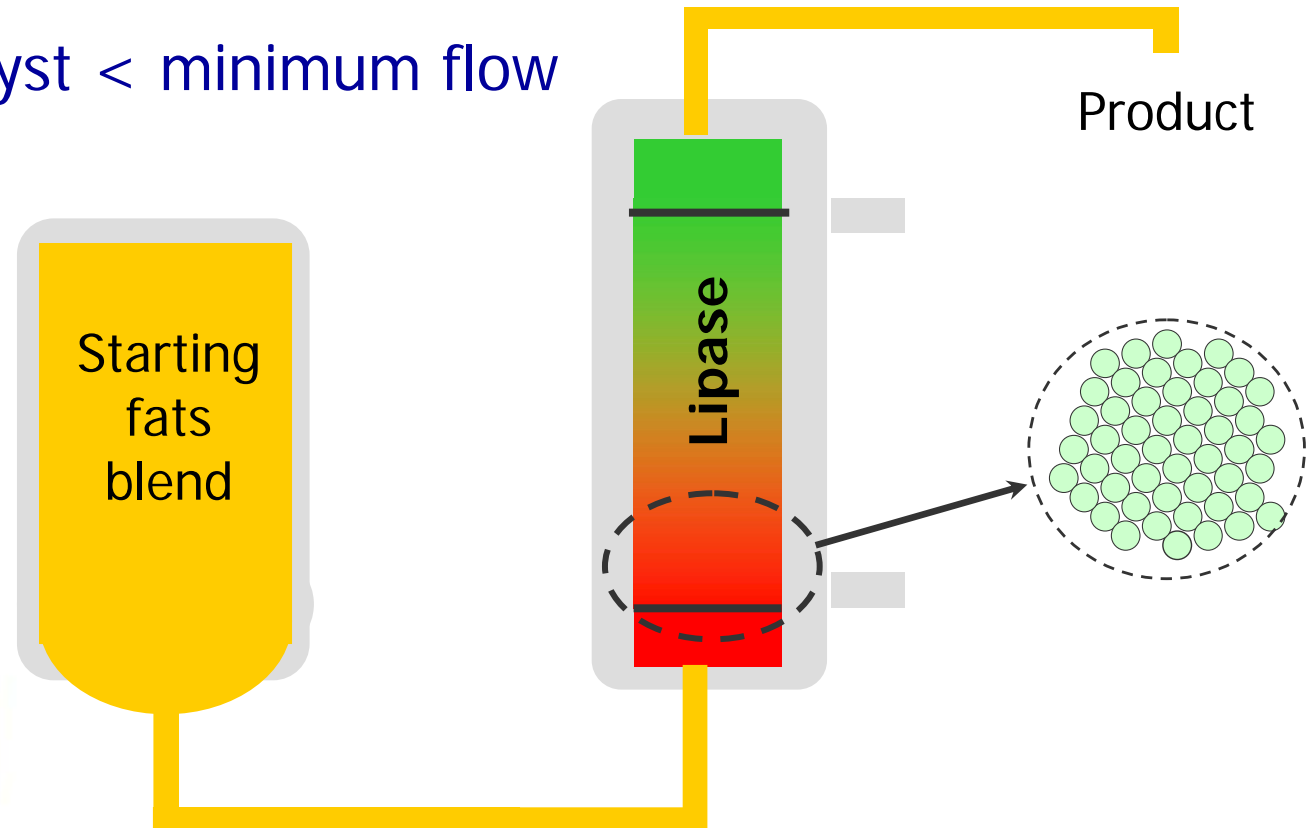


process conditions

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Single Packed Bed Reactor (PBR)

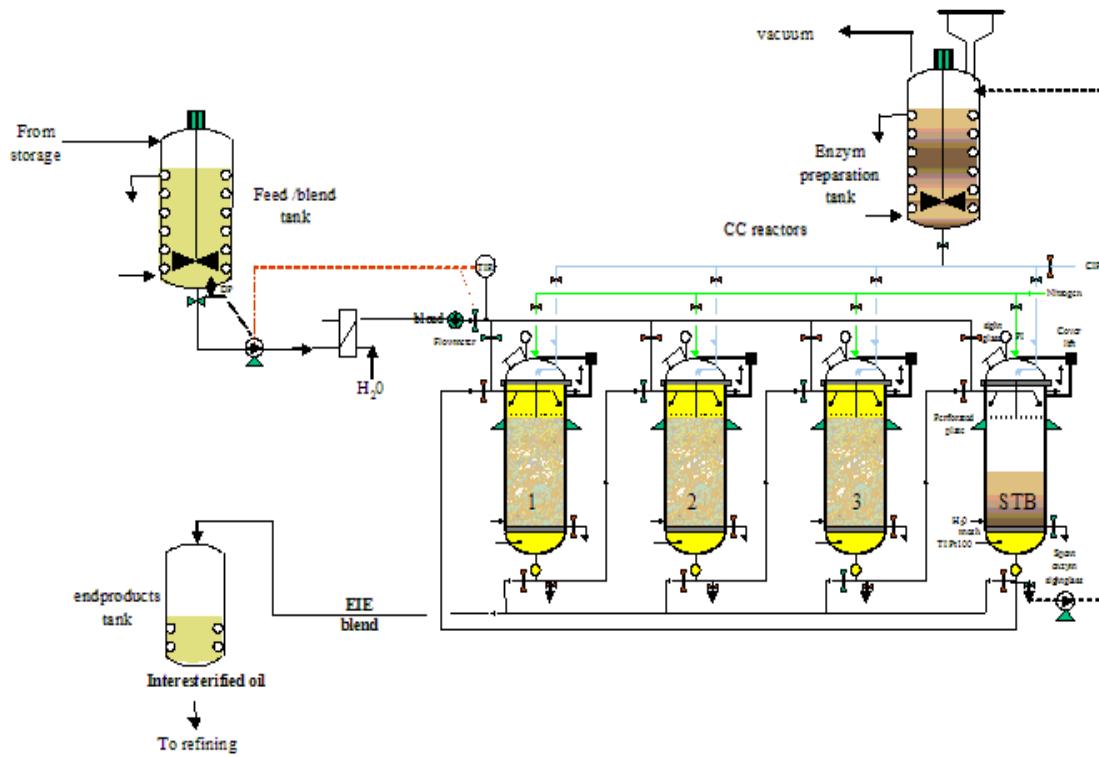
- How to deal with deactivation ?
 - Constant degree of conversion
 - Decreasing flow
 - Change catalyst < minimum flow



Series of PBRs



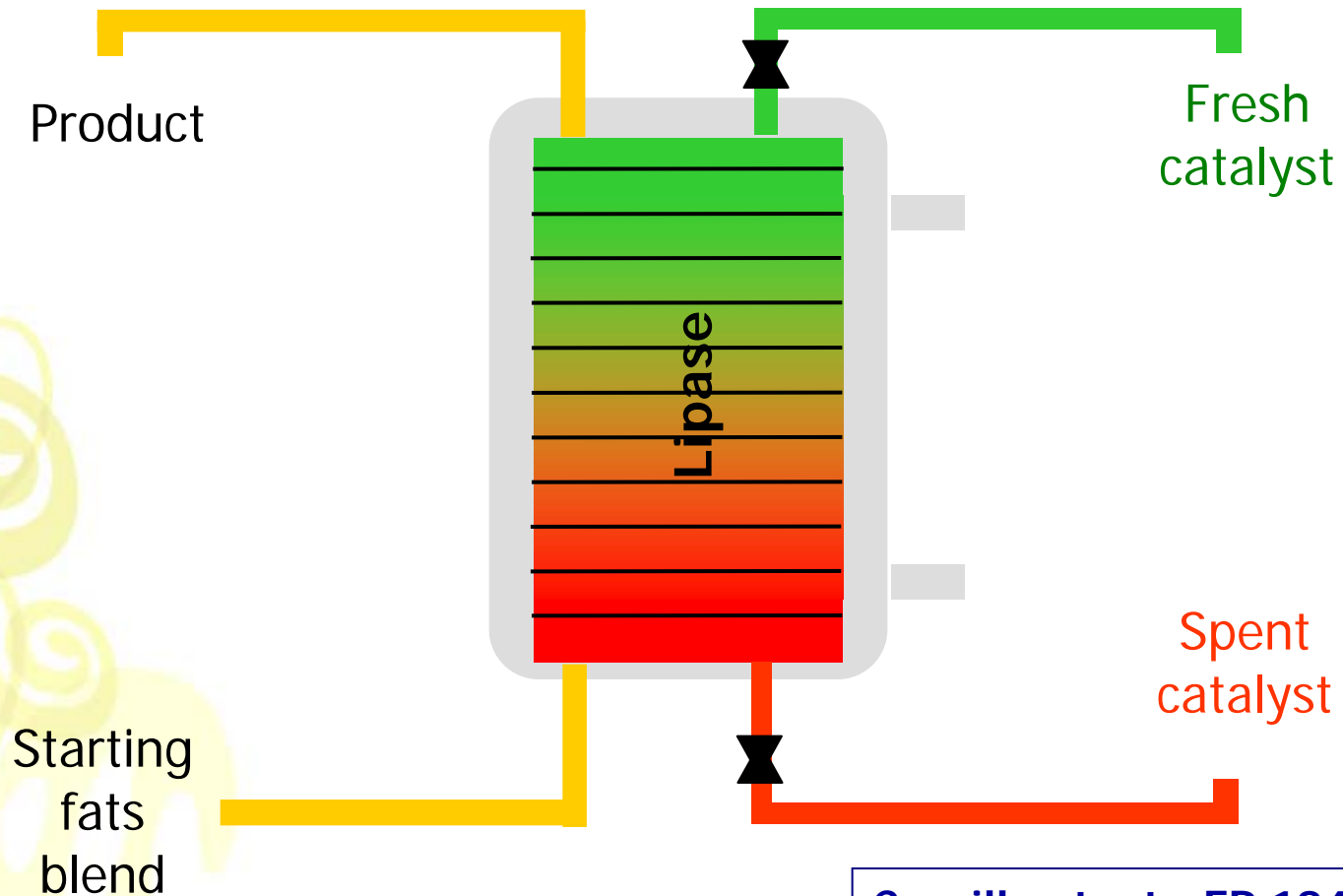
- Catalyst refreshment in the stand-by reactor



Semi-continuous PBR



- Semi-continuous catalyst refreshment
- Minimum catalyst exposure



Cargill patent - EP 1840 204 B1

Summary



- Enzymatic rearrangement has developed into a commercially viable alternative to chemical interesterification.
- The process kinetics can adequately be described using simple first-order modelling.
- Lipase deactivation:
 - seems mainly due to poisoning (oil quality & acids);
 - requires adequate pre-refining;
 - can be described by simple kinetics (enzyme specific).
- Several reactor systems have been developed to deal with catalyst deactivation.



Thank You !

